Environmental Due Diligence for Purchase of Silicon Metaltech, Inc. at Rock Island, Washington

November 1992

Prepared for:

(b) (6)

Prepared by:
Environmental Engineering & Consulting, Inc.
19125 Northcreek Parkway, Suite 111
Bothell, Washington 98011-8002
(206) 485-3437



(b) (6)

Camas, Washington 98607

Dear Mr. (b) (6)

As you and (b) (6) requested, Environmental Engineering & Consulting, Inc. (EEC) has undertaken an evaluation to address environmental issues, including those included in your November 9 letter as well as others I thought prudent for purposes of due diligence related to your planned purchase of the operating and tangible assets of Silicon Metaltech, Inc. (SMI) at Rock Island, Washington.

I have been familiar with this facility since 1988, but not for 15 years as your letter suggested.

Responses to the issues you raised and additional observations are as follow:

## 1.Underground Storage Tanks

There is presently one active underground storage tank (UST) at SMI. This 5,000 gallon tank is used for boiler fuel. This tank was "tightness" tested in 1988. Subsurface soil samples were also collected near the tank in 1988. Results of the "tightness" and soil sample tests indicated no apparent leakage from this tank. This tank is exempt from UST regulations since it is used for onsite heating fuel.

A 1,000 gallon gasoline UST was closed and removed in 1989 based on EEC recommendation that it was a potential liability. A closure report by EEC (dated January 24, 1990, copy enclosed) documents this closure and the absence of any apparent leakage from this tank. The 1988 "tightness" testing of this tank and soil sample tests also had indicated no apparent leakage from this tank, although a low level of petroleum in the soil was detected.

Another UST was closed prior to the 1988 audit. Subsurface soil samples were collected near this former UST in 1988 and no contaminants were detected.



November 16, 1992 (b) (6)

Page 2

#### 2.Laboratory Dry Well

The cleanup of mercury contamination found at the laboratory dry well has not been completed. It is our understanding that prior owner(s) are currently preparing to correct this, but we have not been advised of the details of this planned remediation, so it would not be possible to offer an opinion as to its suitability.

SMI has been listed as a contaminated site in the Washington Department of Ecology (WDOE) Affected Media and Contaminants (AMC) report. This listing is believed to be due primarily to the mercury contamination at the former laboratory. SMI has been scored using the Washington ranking model. The site remains a category C1 (confirmed hazardous substance site, a relatively high ranking) according to the most recent AMC report, dated October 9, 1992. It had been understood earlier that the site's ranking had been lowered from a 4 to a 5, although this is not reflected in the AMC reports.

Onsite disposal of fume waste in unlined ponds and in piles may have contributed to the C1 ranking.

It is recommended the State's ranking be reviewed. An addendum to this report will be prepared on results of this review.

# 3. Waste Oil Disposal

The contaminated area resulting from past waste oil disposal to the north and east of the maintenance shops has not been corrected nor has it been investigated further to define the extent and probable corrective action costs. Corrective action for this contaminated area was estimated in 1988 at about \$50,000, but these costs were not reviewed again. It is our understanding that corrective action for this matter was not undertaken by SMI due to the Hanna indemnity.

This waste oil has been sent to offsite waste oil processors since 1988 as recommended by EEC. This should have eliminated the major source of waste oil contamination.

Steam cleaning of oily parts could also be a source of waste oil contamination in soils at SMI. This can be corrected with a proper collection at minimal cost.



November 16, 1992 (b) (6)

Page 3

#### 4. PCBs

No remedial actions have been taken related to the PCBs to our knowledge. It is possible no cleanup would be needed. Corrective action for this contaminated area was estimated in 1988 at about \$18,000, but these costs were not reviewed again.

#### 5. Fume Hazardous Waste Determination

Fume waste was tested further in 1991 as provided in our letter report to SMI of April 17, 1991, a copy of which is enclosed. New fume samples were tested via the TCLP test method that had replaced the EP Toxicity method for hazardous wastes. This testing by the new method confirmed our prior assessment that the fume was not a hazardous waste. The possibility that lead contamination in the fume would cause the fume to be a classified as a hazardous waste was reduced since the results for lead in the recent TCLP tests were substantially less than the 1988 EP Toxicity tests. The fume continues not to be a "listed waste," according to our review of federal regulations (40 CFR 261.31–.32) and telephone conversations this week with EPA regarding any recently listed waste.

Onsite disposal of fume waste in unlined ponds and in piles may have contributed to the C1 ranking in the WDOE AMC list, as noted above. The TCLP and EP Toxicity analytical data discussed above and other data on fume would not suggest that onsite disposal of fume waste as currently practiced would be a threat to ground water.

# 6. Effluent Discharge

No adverse issues have been raised about SMI's effluent discharge by WDOE or others to our knowledge. An NPDES discharge permit renewal application was filed with WDOE in 1991. A storm water permit application was also filed at that time. A copy of these permit applications are enclosed.

Recent data on effluent contaminants were provided with these applications. There has been no comment or response from WDOE to our knowledge. The NPDES discharge permit has not yet been renewed. It is common for permit renewals to lag applications by two years or more. Until renewal, the former NPDES permit remains in effect. If WDOE has concerns about the effluent discharge, it is considered most likely that more monitoring would be required at minimal additional expense.



November 16 1992 (b) (6)

Page 4

An annual permit fee of \$7,666 was charged for effluent discharges for the period July 1992 through June 1993.

#### 7. Ground Water Quality

Subsequent to the December 5, 1988 SMI post—audit site characterization report, the City of Rock Island had several wells tested in 1989. This testing was reportedly due to concern over the nearby Douglas County landfill and a proposal to dispose of sewage sludge near the airport. The uncertified laboratory which performed the testing reported elevated levels of mercury, lead, arsenic, chromium and selenium, with most being above drinking water standards. The Washington Department of Social and Health Services (DSHS) followed up by testing these wells again (as were many other wells) using a state lab and certified private labs. These later results were found not to be elevated and less than drinking water standards. A copy of the DSHS report on this matter is enclosed.

Ground water quality standards were adopted by WDOE about two years ago. Shortly thereafter, in early 1991 WDOE met with SMI and EEC. WDOE initially expressed concern that onsite fume disposal could be a problem and prior Hanna or WDOE ground water monitoring at the site may have indicated ground water contamination. To help resolve this issue, all relevant data (1988 to present) were provided to WDOE shortly after the meeting. No subsequent actions are known to have been taken by WDOE that would indicate a continuing concern on their part. It is possible that some of these data were used in the WDOE AMC report ranking noted in item 10. The WDOE hydrogeologist involved in the early 1991 meeting was queried during this work about the status of this issue. During this conversation he indicated he had been moved to other, higher priority projects. His recollection was unclear, but he may not have been convinced in early 1991 that ground water monitoring at SMI was not necessary to address fume disposal, the mercury contamination and compliance with ground water standards.

If ground water monitoring were required to address fume disposal and compliance with ground water standards, this could be accomplished for about \$50,000 or less.



November 16 1992 (b) (6)

Page 5

#### 8. SARA 313

We have been advised by EPA Region 10 that the non-fibrous forms of aluminum oxide were delisted from the SARA 313 hazardous chemical list (40 CFR 372) on February 14, 1990. Accordingly, it can probably be assumed that aluminum oxide at SMI is not reportable under 313.

As far as we know, SMI did not make a study fugitive emission study such as was suggested in the December 1988 report. However the 1990 and 1991 annual emission inventories prepared by EEC might be considered to be such studies but these inventories did not specifically address 313.

#### 9. Furnace Waste Hazardous Waste Determination

Tests in late 1988 determined that the "hard pan" and "carbon block" were not classified as hazardous wastes according to the EP Toxicity method. These tests (also for most priority pollutants, including cyanide) found no other contaminants at levels of concern.

A copy of the EEC May 2, 1991 letter report discussing this information is enclosed.

# 10.Solid Waste Disposal

State regulations (WAC 173-304) require SMI to apply for and obtain a solid waste disposal permit for onsite disposal of fume waste, hard pan, carbon block and other wastes from the local health district. As noted in items 5 and 9 above, these wastes have been tested extensively and there was no evidence of contaminants at levels of concern. These wastes would be considered "inert waste" under the regulations, which minimizes solid waste disposal operational requirements. If and when the health district demand such a permit be obtained, additional cost would be incurred to prepare the application and possibly to modify waste handling and disposal procedures. Ground water monitoring may also be necessary as noted in item 7 above.



November 16, 1992 (b) (6)

Page 6

#### 11.Air Pollution Control

Observations about air pollution issues are as follows.

a.1990 Clean Air Act Amendments/NESHAP (National Emission Standards for Hazardous Air Pollutants)

EEC recently prepared a response for SMI to the EPA Ferroalloy NESHAP Information Collection Request, a copy of which is attached. As indicated in this document, emissions of antimony (Sb) compounds, arsenic (As) compounds, cadmium (Cd) compounds, chromium (Cr) compounds, cobalt (Co) compounds, cyanide (CN) compounds, lead (Pb) compounds, mercury (Hg) compounds, nickel (Ni) compounds, selenium (Se) compounds and polycyclic organic matter (POM) were reported to EPA. It is understood EPA will use this information along with similar information from other sources to develop maximum achievable control technology (MACT) standards. It is unknown whether the eventual MACT standards will affect SMI. It can be expected to require one to several years for emission standards to be adopted for the ferroalloy industry as a result of this information collection effort by EPA.

#### b. Permit fees

Permit fees may be expected to increase in the future. Presently SMI is charged \$10/ton annually. Estimated emissions for 1991 were 1,745.9 tons per year, resulting in a fee of about \$17,459. A copy of the 1991 emission inventory prepared by EEC is attached for reference.

It is understood from discussions with a person on a WDOE permit fee advisory committee that a proposal will be sent to the Legislature that could increase SMI air permit fees to the range of \$40,000 to \$53,000 per year. Apparently WDOE intends to fund its air program through such fees as it has other programs.

The permit fees, even at current levels, may provide an incentive to fine tune the annual emission inventory and perhaps implement better controls.



November 16, 1992 (b) (6)

Page 7

#### c. WDOE Air Toxics

WDOE has adopted air toxics regulations to control emissions of some 500 pollutants from new and modified sources. These regulations do not appear to apply to SMI, provided SMI does not make a modification under definition of "new toxic air pollution source", see attached regulations, and pamphlet. As long as SMI does not make an "alteration..which may increase emissions or increase ambient air concentrations of any regulated air pollutant.." these regulations would not apply.

#### d.Related Air issues

WDOE may require some emissions at SMI to be better controlled, e.g., uncontrolled tapping emissions and fugitive (i.e., uncaptured emissions) furnace emissions.

Particulate matter smaller than 10 microns (PM10) may require better controls in the future; the principal means of control for PM10 would be the two large baghouses.

These considerations probably necessitate better air pollution controls at SMI in the future.

Please do not hesitate to call me if there are any questions.

Singerely,

Patrick H. Wicks, PE, CHMM

President

**Enclosures** 

UST closure report, January 24, 1990

# UNDERGROUND TANK CLOSURE SILICON METALTECH, INC. ROCK ISLAND, WASHINGTON

January 24, 1990

## UNDERGROUND TANK CLOSURE SILICON METALTECH, INC. ROCK ISLAND, WASHINGTON

January 24, 1990

Patrick H. Wicks, PE, CHMM

att Huil

President

Roger Wilson

Geologist

Prepared for:

Silicon Metaltech, Inc. Seattle, Washington

Prepared by: Environmental Engineering & Consulting, Inc. 19125 Northcreek Parkway, Suite 111 Bothell, Washington 98011-8002 (206) 485-3437

# TABLE OF CONTENTS

1	PROJECT DESCRIPTION AND SCOPE OF WORK	1
2	2.1 Tank Removal Documentation	2 2 2 3
3	3.1 UST Inspection 3.2 Subsurface Conditions 3.3 Ground Water 3.4 Analytical Results 3.5 Tank Cleaning and Disposal	4 4 4 4 4 5 5
4	CONCLUSIONS	6
Та	TABLES & FIGURES able 1. Sample Results for Petroleum Hydrocarbons	
Fiç	gure 1. Site Location Map - SMI Rock Island, WA	
Fig	gure 2. Sample Locations - SMI Tank Removal	
	APPENDIX	
	Laboratory Results	
	Chain-of-Custody	
	Photographs	
	Tank Cleaning Certificate	

#### 1 PROJECT DESCRIPTION AND SCOPE OF WORK

Environmental Engineering and Consulting, Inc. (EEC) was retained by Silicon Metaltech, Inc. (SMI), to document the closure by removal of an underground storage tank (UST) located at the facility in Rock Island, Washington. The scope of the project included:

- (a) monitoring and documenting the removal of an approximately 1,000 gallon unleaded gasoline UST,
- (b) collecting soil samples from the tank excavation, and from soil stockpiled around the perimeter of the excavation,
- (c) laboratory analysis of soil samples for total petroleum hydrocarbons (TPH), and volatile aromatic hydrocarbon compounds principally benzene, toluene, ethyl benzene and xylene (BTEX), and
- (d) preparation of this report.

The location of the SMI plant at Rock Island is shown on Figure 1. The location of the UST within the plant is also shown on the figure.

#### 2 METHODS OF INVESTIGATION

#### 2.1 Tank Removal Documentation

The 1,000 gallon unleaded gasoline UST to be removed was located south of the change building at the SMI facility. The removal was completed on November 27, 1989, by SMI personnel using a backhoe and rubber tired front end loader under the direction of a representative from B&C Equipment Company (B&C) Seattle, Washington. Figure 2 shows the approximate location of the UST excavation. SMI had removed as much gasoline as possible before removal.

Prior to removal approximately 50 pounds of dry ice was placed into the UST by a representative of B&C to displaced potentially explosive vapors. This minimized explosion and fire hazards during removal.

The UST removal was observed by Chief Coombs of Douglas County Fire Protection District 2, in accordance with his request (EEC had notified the District in the event a permit for the removal was needed, but no permit was required). The tank and tank excavation was inspected and photographed by an EEC field representative.

The removed tank was loaded onto a B&C company truck and reportedly driven to Northwest EnviroService, Inc. in Seattle, Washington for cleaning and disposal.

#### 2.2 Sample Collection

Soil samples were collected from the bottom and walls of the tank excavation, and stockpiled excavated soil. Sample S-1 and S-3 were collected from the west and east walls of the excavation, respectively. These samples consisted of soil collected at four locations from each wall at depths of 4.5 and 8 feet below grade. Samples S-2 and S-4 were collected from the north and south walls of the excavation, respectively. These samples consisted of soil collected at six locations from each wall at depths of 4.5 and 8 feet below grade. Sample S-5 was collected at five locations from the bottom of the excavation at a depth of 8 feet below grade. Samples S-6 and S-7 were collected from excavated soil stockpiled to the south and west of the excavation. Soil samples were screened with an HNu photoionization detection instrument upon removal from the excavation and stock piled soil.

The composite soil samples were collected with a stainless steel spoon and bowl and consisted of scraping equal amounts of soil from each location in to the stainless steel bowl. The soil in the bowl was mixed thoroughly and transferred into sterilized

glassware with Teflon sealed lids provided by the project laboratory. The composite samples were stored on site and transported to the laboratory in an ice chest chilled to 4 degrees Celsius to minimize dissipation of volatile hydrocarbons. EPA recommended protocols for sample management, including maintenance of chain-of-custody documents, were followed at each stage of the project.

#### 2.3 Analytical Methods

AmTest, Inc. was chosen to perform the laboratory analysis on the soil samples. Soil samples were analyzed for TPH (nC-1 to nC-22) by EPA Method 8015 (Modified). Soil samples were also analyzed for BTEX by EPA Method 8020.

Sample S-1 and S-2, S-3 and S-4, and S-6 and S-7 were composited by the laboratory prior to analysis.

#### 3 RESULTS OF INVESTIGATION

#### 3.1 UST Inspection

Inspection of the UST upon removal revealed moderate corrosion on the exterior surfaces. No obvious holes or evidence of leakage was observed. After removal, the UST was reportedly transported to Northwest EnviroService, Inc. for cleaning and disposal.

#### 3.2 Subsurface Conditions

Soils logged from the walls of the tank excavation were comprised of fill material and consisted of medium to very coarse grained sand with an occasional cobble and boulder up to one foot in diameter. Asphalt chunks were observed in the fill down to a depth of four and one-half feet below grade. Thin discontinuous layers of black sooty material was noted throughout the walls of the excavation. No hydrocarbon odors were noticed in the excavated material or soil samples. The results of the HNu screening of soil samples are presented below in parts per million (ppm):

<u>Sample</u>	Concentration (ppm)
S-1	<5
S-2	<5
S-3	<5
S-4	<5
S-5	<5
S-6	<5
S-7	<5

#### 3.3 Ground Water

No ground water was encountered during the tank removal. According to a well log for SMI well number three, the depth to ground is roughly 30 feet below grade.

#### 3.4 Analytical Results

The results of laboratory tests performed on selected soil samples are presented in Table 1. A review of the results in Table 1 reveal that all the composite soil samples had no detectable concentrations of BTEX, gasoline, diesel #1 and diesel #2.

#### 3.5 Tank Cleaning and Disposal

The tank was cleaned and disposed of in accordance with all federal, state, and local rules and regulations as reflected in the December 8, 1989 certificate by Northwest EnviroService, Inc.

#### 3.6 Backfilling of Excavation

On December 14, 1989, EEC advised SMI that the laboratory results indicated no hydrocarbon contamination. SMI advised that it backfilled the excavation on December 15, 1989 with soil/fill from the tank removal and clean fill from elsewhere on the plant site.

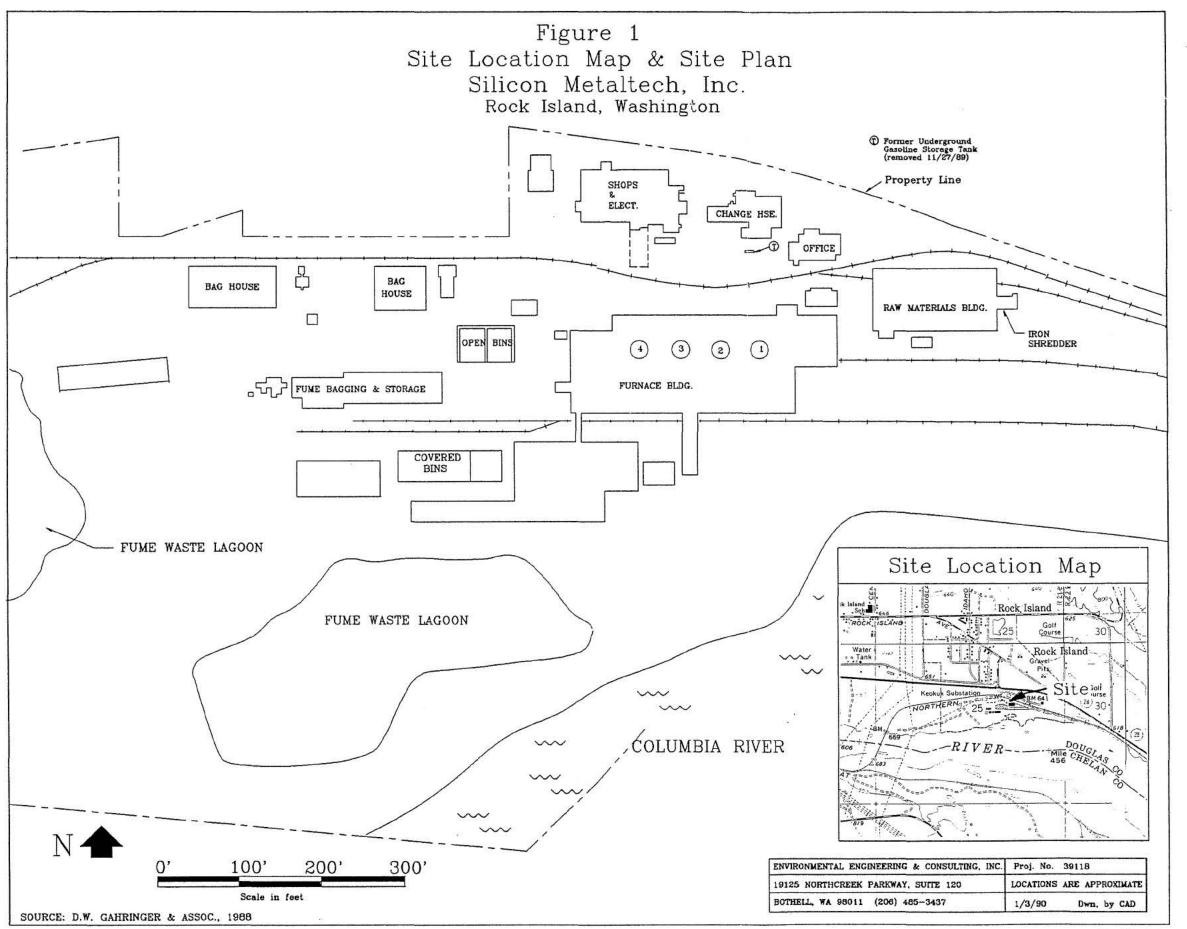
#### 4 CONCLUSIONS

Our interpretation of appropriate guidelines for allowable concentration levels are based on the Washington Department of Ecology (WDOE) 8/1/88 draft document "Policies and Procedures for Underground Storage Tank Removal." In that document the WDOE has established recommended cleanup level goals for petroleum hydrocarbons. These cleanup level goals are listed in Table 1.

An inspection of the tank revealed moderate corrosion over the exterior surfaces of the tank, but no obvious evidence of leaks or holes were visible.

Based on information developed through observation of the UST removal, field screening with a photoionization instrument, soil sampling, laboratory analysis, and WDOE cleanup guidelines, it appears that soil, in the vicinity of the tank excavation is not contaminated by residual petroleum hydrocarbons, or BTEX that would warrant cleanup or remediation.

This report should satisfy EPA regulations (40 CFR, Part 280) for closure of petroleum USTs.



# Figure 2. Sample Locations SMI Tank Removal

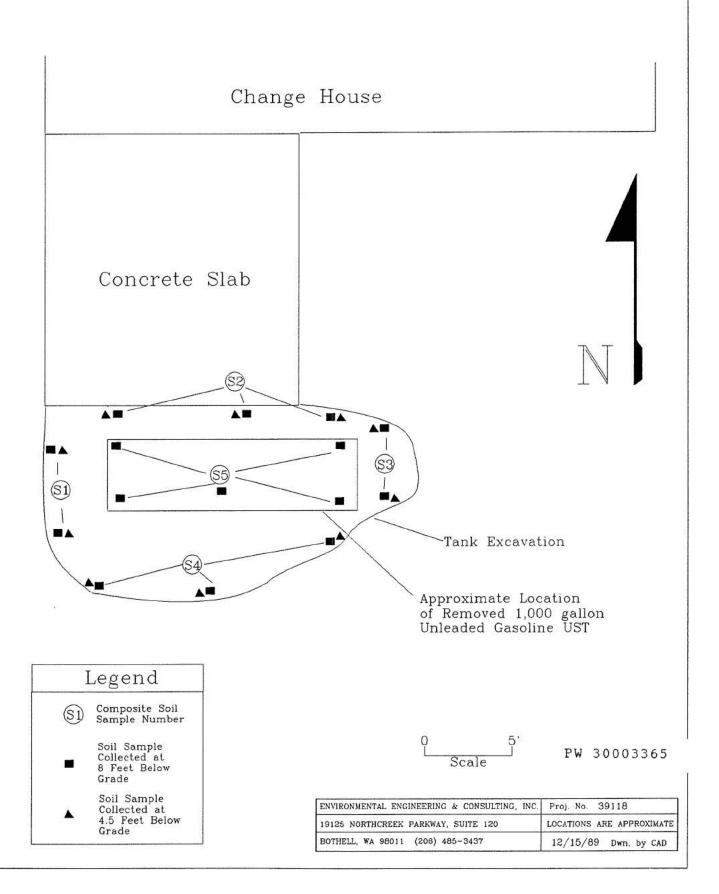


TABLE 1

# Sample Testing Results for Volatile Aromatic Hydrocarbons, Petroleum Hydrocarbons, and Field Screening mg/kg (ppm)

Sample					WDOE Soil Cleanup Guidelines
Number	S-1,2	S-3,4	S-5	S-6,7	(1)
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Method 8020 Re	<u>esults</u>				
Benzene	<0.005	<0.005	<0.005	<0.005	0.66
Toluene	<0.005	<0.005	<0.005	<0.005	143
Ethylbenzene	<0.005	<0.005	<0.005	<0.005	14
Total Xylenes	<0.005	<0.005	<0.005	<0.005	NA
Method 8015M	Results				
TPH	<u>rtocarto</u>				200
Gasoline	<20	<20	<20	<20	
Diesel #1	<25	<25	<25	<25	
Diesel #2	<40	<40	<40	<40	
HNu screening levels ppm	NR ,	NR	NR	NR	NA

TPH - Total Petroleum Hydrocarbons

NR - No Response = <5ppm

NA - Not Applicable

(1) WDOE, 1988, Policies and Procedures for Underground Tank Removal (Draft)

# **APPENDIX**

AMIEST

AmTest Inc.

Professional Analytical Services

14603 N.E. 87th St. Redmond, WA 98052

Fax: 206 883 3495

ANALYSIS REPORT

Tel: 206 885 1664

CLIENT: Environmental Engineers

DATE RECEIVED: 11/29/89

DATE ANALYZED: DATE REPORTED:

12/4/89 12/5/89

REPORT TO: Roger Wilson

19125 North Creek Parkway, Suite 111

Bothell, WA 98011-8002

#### ANALYSIS OF FUELS IN SOIL BY EPA 8015 (MODIFIED)

Laboratory Sample Nos.	Client Identification	Gasoline (mg/kg)	Diesel #1 (mg/kg)	Diesel #2 (mg/kg)
921945	S-1 + S-2	ND	ND	ND
921946	S-3 + S-4	ND	ИD	ND
921947	S-6 + S-7	ND	ND	ND
921948	S-5	ND	ND	ND
921948	S-5 Spike F* (%)	**************************************	106.	7 <u>—</u>
921948	S-5 Spike G* (%)	-	102.	-
BLANK		ND	ND	ND
DETECTION LI	MIT	20.	25.	40.

<sup>\*</sup>Spike Recovery Data - reported in percent (%).

ND = Not Detected.

-2-

CLIENT: Environmental Engineers

DATE RECEIVED:

11/29/89

REPORT TO: Roger Wilson

DATE ANALYZED: 12/4/89

DATE REPORTED: 12/5/89

#### GC ANALYSIS OF PURGEABLE AROMATIC COMPOUNDS IN SOIL BY EPA 8020

Laboratory Sample Nos. Client Identification	921945 S-1 + S-2	921946 S-3 + S-4	921947 S-6 + S-7	DETECTION LIMIT (ug/kg)
Benzene	ND	ND	ND	5.
Toluene	ND	ND	ND	5.
Chlorobenzene	ND	ND	ND	5.
Ethylbenzene	ND	ND	ND	5.
m+p-Xylene	ND	ND	ND	5.
o-Xylene	ND	ND	ND	5.
1,3-Dichlorobenzene	ND	ND	ND	5.
1,4-Dichlorobenzene	ND	ND	ND	5.
1,2-Dichlorobenzene	ND	ND	ND	5.

m-Xylene & p-Xylene coelute.

ND = Not Detected.

All results are reported in ug/kg.

# Internal Surrogate (%)

Bromofluorobenzene 97. 85. 66.

CLIENT: Environmental Engineers

DATE RECEIVED: 11/29/89

REPORT TO: Roger Wilson

DATE ANALYZED: 12/4/89

DATE REPORTED: 12/5/89

GC ANALYSIS OF PURGEABLE AROMATIC COMPOUNDS IN SOIL BY EPA 8020

Laboratory Sample Nos.	921948	DETECTION LIMIT
Client Identification	S-5	(ug/kg)
Benzene	ND	5.
Toluene	ND	5.
Chlorobenzene	ND	5.
Ethylbenzene	ND	5.
m+p-Xylene	ND	5.
o-Xylene	ND	5.
1,3-Dichlorobenzene	ND	5.
1,4-Dichlorobenzene	ND	5.
1,2-Dichlorobenzene	ND	5.
<pre>m-Xylene &amp; p-Xylene coelute. ND = Not Detected. All results are reported in ug/kg</pre>	•&	
<pre>Internal Surrogate (%)</pre>		

70.

JJMcA/pb

Bromofluorobenzene

REPORTED BY

James J. McAteer,

MILSI

PW

30003371

Professional
Analytical
Services

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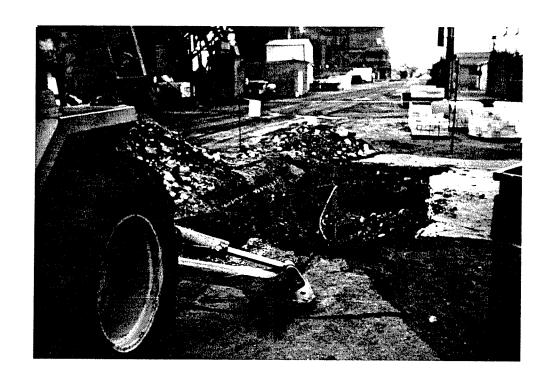
Redmond, WA 98052

Tel: 206 885 1664

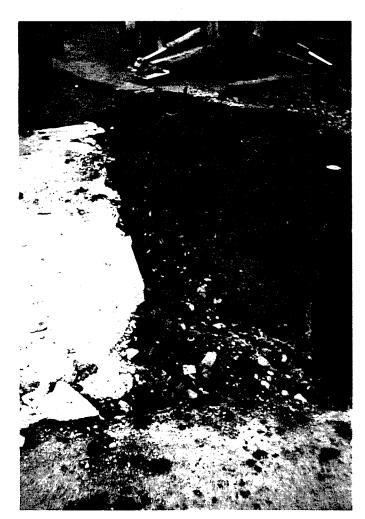
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**CHAIN OF CUSTODY RECORD** 

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Photograph 1 - Looking west at tank excavation prior to tank 11/27/89 removal.



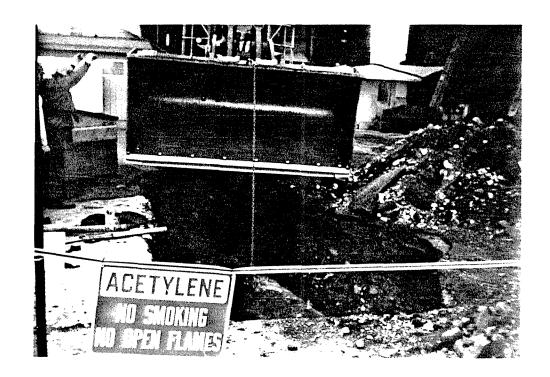
Photograph 2- View looking at 11/27/89 tank excavation prior to tank removal



Photograph 3 - View showing B & C Equipmment Co. personnel 11/27/89 preparing tank for removal.



Photograph 4 - View showing removed soil stckpiled south and 11/27/89 west of excavation.



Photograph 5 - Rubber tire loader removing tank. 11/27/89



Photograph 6 - View showing removed tank. 11/27/89



#### DISPOSAL CERTIFICATION

DATE:

December 8, 1989

TO:

B & C Equipment

19009 - 16 Avenue South

Seattle, WA 98188

REFERENCE P.O. #n/a

Dear Sir,

This letter is to certify that Northwest EnviroService, Inc. has received the following tank(s) for cleaning and disposal in accordance with all federal, state and local rules and regulations:

1.) One (1) 1,000 gallon gas

NWES JOB #:

32-11925

DATE RECEIVED:

11-27-89

DATE CLEANED:

12-07-89

DATE OF DISPOSAL:

12-07-89

LOCATION OF TANK ORIGIN: Eastern Washington

If you have any questions or requests for service, feel free to contact this office at (206)-762-1190.

Thank you for your business and we look forward to being of service in the future.

Sincerely,

Northwest EnviroService, Inc.

 $\mathcal{L}$  A orm  $\infty$   $\mathcal{L}$  . X Thomas R. Gremel

Underground Tank Division Manager

TRG:cal

PW 30003375

April 17, 1991 letter report to SMI on TCLP testing results C:\reports\229r.wr1

April 17, 1991

Mr. Jim Trunzo Executive Vice President Silicon Metaltech, Inc. 100 4th Street Rock Island, Washington 98850

Dear Jim:

Results of lab testing of fume waste via the TCLP method were recently received. I have attached a copy of revised Table C from the recent excerpts report to which these data have been added. The data has been graphed also for comparison with the prior EP Toxicity data and with the regulatory limits, see attached.

The results show no samples exceed the TCLP regulatory limits. We did not test for eleven volatile organics nor eight pesticides because it would be so improbable that these could be present, considering the source of the fume and the high temperatures to which it is exposed.

The 1991 TCLP lead results were substantially lower than the previous EP Tox results, which allayed what could have been a major concern. Generally other elements/metals in the 1991 sample results were similar to or lower than the 1988 results. None of the semivolatile organics were detected above the method reporting limit. Enclosed also is a copy of the lab report.

It is recommended these results be forwarded to the WDOE along with other data not provided in the excerpts report. Please do not hesitate to call me if there are any questions.

Sincerely,

Patrick H. Wicks, PE, CHMM President

Enclosures

cc: Robert L. Miller

PW 30001168

# PW 30001169

# TABLE C (updated) FUME WASTE ANALYTIC DATA SUMMARY

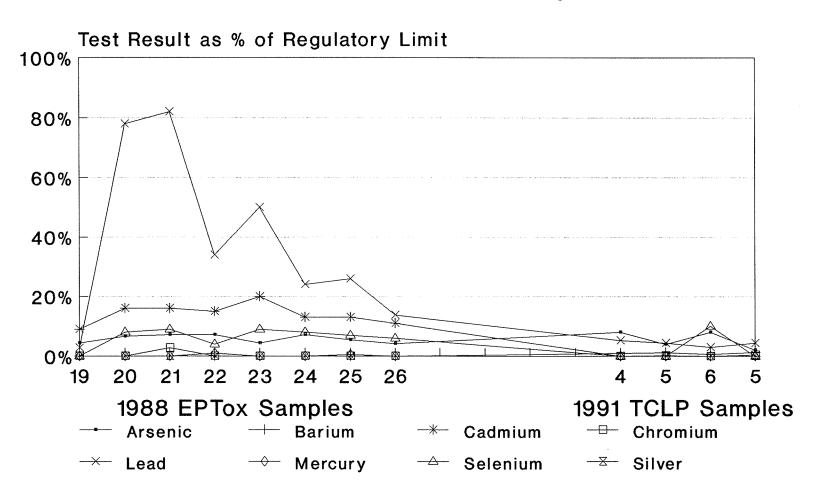
## Constituent Concentration, parts per million (ppm, mg/l, mg/kg)

Sample Infernation   Sample	Sample Location	NWFL (1)	FWL (2)	FWL	FWDA (3)	FWDA	FWDA	FWL	FWL	Fume (4)	Fume (4)			Pond 1,2,8	Pond 4	Pond 6	Pond 4	
Sumple Date   14-ce   14-ce	Sample Identification									_	•				rume		rume	
Part			3M1F0-0010	M110-11012	SMIFU-00/	3M1FU-203	3m11-0-483	m1F0-13014	M1FU-15016	SMI SLU-I				Composite		Sidily	dualianta	
Sample Type	Sample Date	04_Oct	na_Oct	O4_Oct	M-Oct	04_0ct	04_Oct	M_Oct	M_Oct	05.Oct	•			13_Mar	13_Mar	12 Mar		
Sumple Depth   Profession   P	Sample Date																	
Sample Deph. Fi.   Sample Deph. Fi.   Sample Deph. Fi.   Sample Deph. Fi.   Sample Sample Deph. Fi.   Sample Sample Deph. Fi.   Sample Sample Minister of Date   Sa	Samnie Tyne																	
An Test   An T	,									·	Tunio masto	FP Toxicity	TCLP			•		
Repert Number or Date   March Number or Dat										Am Test	Am Test	-						
Machanisty Sample Number   102019   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   102012   1	· ·										_							
CIC   O E PTOXICITY   E PTOX   E PTOX	•																	
Elements/Metals																		
Elements/Metals	TCLP or E P TOXICITY	E P TOX	EPTOX	EPTOX	EPTOX	EPTOX	EPTOX	EPTOX	E P TOX					TCLP	TCLP	TCLP	TCLP	
Salum																		
Calmium	Arsenic	0.22	0.33	0.36	0.36	0.22	0.36	0.27	0.21			5	5	0.4	0.2	0.4	0.1	ALL SAMPLES < TCLP/EPT
Chomism	Barium	<1	<1	<1	<1	<1	<1	<1	<1			100	100	1	1.2	0.6	1.3	ALL SAMPLES < TCLP/EPT
Copper	Cadmium	0.09	0.16	0.16	0.15	0.2	0.13	0.13	0.11			1	1	<0.01	<0.01	<0.01	<0.01	ALL SAMPLES < TCLP/EPT
Lead	Chromium	<0.05	<0.05	0.14	<0.05	<0.05	<0.05	<0.05	<0.05			5	5	<0.01	0.01	<0.01	0.02	ALL SAMPLES < TCLP/EPT
Mercury	Copper	<0.05	1.4	0.82	0.39	0.42	0.35	0.34	0.25			-	_					
Nicker	Lead	0.14	3.9	4.1	1.7	2.5	1.2	1.3	0.69	1		5	5	0.26	0.22	0.15	0.22	ALL SAMPLES < TCLP/EPT
Selenium	Mercury	<0.001	<0.001	<0.001	0.002	<0.001	<0.001	0.001	<0.001			0.2	0.2	<0.001	<0.001	<0.001	<0.001	ALL SAMPLES < TCLP/EPT
Silver	Nickel	0.11	0.16	0.07	0.07	0.07	0.16	<0.05	<0.05		•	_	-	·				
Semivolatile Organics	Selenium	<0.05	0.08	0.09	0.04			0.07	0.06			1		<0.1	<0.1	0.1	<0.1	ALL SAMPLES < TCLP/EPT
Semivolatile Organics         400         <0.05         <0.05         <0.05         ALL SAMPLES < TCLP/EPT           2.4,6-Trichlorophenol         2         <0.05	Silver	<0.05						<0.05				5	5	<0.01	<0.01	<0.01	<0.01	ALL SAMPLES < TCLP/EPT
2.4,5-Trichlorophenol       400       <0.05	Zinc	0.89	16	11	6	9	3	4	2	<u> </u>		_						
2,4,6-Trichlorophenol       2       <0.05																······································		,
2,4-Dinitrotolene       0.13       <0.05																		1
Cresol, total         200         <0.05         <0.05         <0.05         ALL SAMPLES < TCLP/EPT           Cresol, m,p-         <0.05											-							3
Cresol, mp−																		1
Cresol, m-         200           ALL SAMPLES < TCLP/EPT           Cresol, p-         200         <0.05										-			200				····	ALL SAMPLES < TCLP/EPT
Cresol, o-         200         <0.05         <0.05         <0.05         <0.05         ALL SAMPLES < TCLP/EPT           Cresol, p-         200         0.05         <0.05														<0.05	<0.05	<0.05		
Cresol, p−         200         C         C           Hexachrolobenzene         0.13         <0.05																		
Hexachrolobenzene         0.13         <0.05         <0.05         <0.05         <0.05         <0.05         ALL SAMPLES < TCLP/EPT           Hexachrolobutadiene         0.5         <0.05														<0.05	<0.05	<0.05		ALL SAMPLES < TCLP/EPT
Hexachrolobutadiene         0.5         <0.05										***************************************							····	· · · · · · · · · · · · · · · · · · ·
Hexachroloethane         3         <0.05         <0.05         <0.05         ALL SAMPLES < TCLP/EPT           Nitrobenzene         2         <0.05				***************************************														4
Nitrobenzene         2         <0.05         <0.05         <0.05         ALL SAMPLES < TCLP/EPT           Pentachlorophenol         100         <0.2		···														~		4
Pentachlorophenol         100         <0.2         <0.2         <0.2         <0.2         ALL SAMPLES < TCLP/EPT           Pyridine         5         <0.2															<b></b>			4
Pyridine         5         <0.2         <0.2         <0.2         ALL SAMPLES < TCLP/EPT																		
				***************************************														· · · · · · · · · · · · · · · ·
													5	<0.2	<0.2	<0.2		ALL SAMPLES < ICLP/EPT

#### FOOTNOTES

- (1) New Waste Fume Lagoon NWFL
- (2) Fume Waste Lagoon FWL
- (3) Fume Waste Disposal Area-FWDA
- (4) Sample of slurry as discharged into lagoon. Results are reported as liquid in mg/liter. All other fume samples reported as mg/kg, dry weight basis.

# Fume Waste TCLP/EPTox SMI 1988 & 1991 Samples



C:\hgdata\39200tcl.cht



April 9, 1991

Pat Wicks
Environmental Engineering & Consulting, Inc.
19125 Northcreek Parkway
Suite 120
Bothell, WA 98011-8002

Re: Rock Island Wa./Project #39200

Dear Pat:

Enclosed are the results of the soil and water samples submitted to our lab on March 14, 1991. Preliminary results were transmitted via facsimile on April 4, 1991. For your reference, our service request number for this work is K911367.

All analyses were performed in accordance with the laboratory's quality assurance program.

Please call if you have any questions.

Respectfully submitted,

Columbia Analytical Services, Inc.

Abbie Spielman
Project Chemist

AS/das

PW 30001171

#### COLUMBIA ANALYTICAL SERVICES, INC.

#### **Analytical Report**

Environmental Engineering & Consulting, Inc. Client:

Date Received: 03/14/91

Submitted By: Pat Wicks Date TCLP Performed: 03/25/91 Date Analyzed: 03/26/91

Rock Island Wa./#39200 Project:

Work Order #: K911367

Sample Matrix: Soil

Toxicity Characteristic Leaching Procedure (TCLP) EPA Method 1311 Metals mg/L (ppm) in TCLP Extract

		Sample Name	Pond #1,2,8 Comp	Pond #4	
		Lab Code	<b>:</b> :	K1367-4	K1367-5
Analytes	Methods	MRL	Regulatory Limit*		
Arsenic	3010/6010	0.1	5.0	0.4	0.2
Barium	3010/6010	0.1	100	1.0 ′	1.2
Cadmium	3010/6010	0.01	1.0	ND ~	ND -
Chromium	3010/6010	0.01	5.0	ND <	0.01
Lead	3010/6010	0.05	5.0	0.26	0.22
Mercury	7470	0.001	0.2	ND	ND
Selenium	3010/6010	0.1	1.0	ND-	ND ·
Silver	3010/6010	0.01	5.0	ND	ND -

MRL Method Reporting Limit

From 40 CFR Part 261, et. al. and Federal Register, March 29, 1990 and June 29, 1990

ND None Detected at or above the method reporting limit

PW 30001172

00001

#### **Analytical Report**

Client:

Environmental Engineering & Consulting, Inc.

03/14/91 Date Received:

Submitted By: Pat Wicks

Date TCLP Performed: 03/25/91

Project:

Rock Island Wa./#39200

Date Analyzed:

03/26/91

Sample Matrix: Water

Work Order #:

K911367

#### Toxicity Characteristic Leaching Procedure (TCLP) EPA Method 1311 Metals mg/L (ppm) in TCLP Extract

		Sample Name Lab Code		Pond #6 K1367-6	Method Blank K1367-MB
Analytes	Methods	MRL	Regulatory Limit*		
Arsenic	3010/6010	0.1	5.0	0.4-	ND
Barium	3010/6010	0.1	100	0.6	ND
Cadmium	3010/6010	0.01	1.0	ND -	ND
Chromium	3010/6010	0.01	5.0	ND	ND
Lead	3010/6010	0.05	5.0	0.15	ND
Mercury	7470	0.001	0.2	ND -	ND
Selenium	3010/6010	0.1	1.0	0.1	ND
Silver	3010/6010	0.01	5.0	ND	ND

MRL Method Reporting Limit

From 40 CFR Part 261, et. al. and Federal Register, March 29, 1990 and June 29, 1990

ND None Detected at or above the method reporting limit

#### Analytical Report

Client: Environmental Engineering & Consulting, Inc.

Submitted By: Pat Wicks

Project: Rock Island Wa./#39200

Sample Matrix: Soil

 Date Received:
 03/14/91

 Date TCLP Performed:
 03/25/91

 Date Extracted:
 03/28/91

 Date Analyzed:
 03/29/91

 Work Order #:
 K911367

Toxicity Characteristic Leaching Procedure (TCLP)
EPA Method 1311
Semivolatile Organic Compounds
mg/L (ppm) in TCLP Extract

Sample Name: Pond #1,2,8 Pond Comp #4
Lab Code: K1367-4 K1367-5

	Analytes	Methods	MRL	Regulatory Limit*		
	Allarytes	Metrious	WINTE	Little		
	Hexachloroethane	3510/8270	0.05	3	ND	ND
	Nitrobenzene	3510/8270	0.05	2	ND	ND
	Hexachlorobutadiene	3510/8270	0.05	0.5	ND	ND
	2,4-Dinitrotoluene	3510/8270	0.05	0.13	ND	ND
	Hexachlorobenzene	3510/8270	0.05	0.13	ND	ND
	2,4,6-Trichlorophenol	3510/8270	0.05	2	ND	ND
-	2,4,5-Trichlorophenol	3510/8270	0.05	400	ND	ND
	Pentachlorophenol	3510/8270	0.2	100	ND	ND.
-	Pyridine	3510/8270	0.2	5	ND	ND
65	o-Cresol	3510/8270	0.05	200	ND	ND
	m,p-Cresols	3510/8270	0.05	200	ND	ND
	Total Cresols	3510/8270	0.05	200	ND	ND

MRL Method Reporting Limit

From 40 CFR Part 261, et. al. and Federal Register, March 29, 1990 and June 29, 1990

ND None Detected at or above the method reporting limit

peroved by Whi Allelin

Date 4/11/91

PW 30001174

#### Analytical Report

Client:

Environmental Engineering & Consulting, Inc.

Submitted By:

Pat Wicks

Project:

Rock Island Wa./#39200

Sample Matrix: Water Date Received:

03/14/91

Date TCLP Performed: 03/25/91

Date Extracted: Date Analyzed:

03/28/91 03/29/91

Work Order #:

K911367

Toxicity Characteristic Leaching Procedure (TCLP) EPA Method 1311 Semivolatile Organic Compounds mg/L (ppm) in TCLP Extract

Sample Name: Lab Code:			Pond #6 K1367-6	Method Blank K1367-MB	
Analytes	Methods	MRL	Regulatory Limit*		
Hexachloroethane	3510/8270	0.05	3	ND	ND
Nitrobenzene	3510/8270	0.05	2	ND	ND
Hexachlorobutadiene	3510/8270	0.05	0.5	ND	ND
2,4-Dinitrotoluene	3510/8270	0.05	0.13	ND	ND
Hexachlorobenzene	3510/8270	0.05	0.13	ND	ND
2,4,6-Trichlorophenol	3510/8270	0.05	2	ND	ND
2,4,5-Trichlorophenol	3510/8270	0.05	400	ND	ND
Pentachlorophenol	3510/8270	0.2	100	ND	ND
Pyridine	3510/8270	0.2	5	ND	ND
o-Cresol	3510/8270	0.05	200	ND	ND
m,p-Cresols	3510/8270	0.05	200	ND	ND
Total Cresols	3510/8270	0.05	200	ND	ND

MRL Method Reporting Limit

From 40 CFR Part 261, et. al. and Federal Register, March 29, 1990 and June 29, 1990

ND None Detected at or above the method reporting limit

Approved by athi Aprelmen

PW 30001175

# APPENDIX A LABORATORY QC RESULTS

PW 30001176

Client:

Environmental Engineering & Consulting, Inc.

Submitted By:

Pat Wicks

Project: Sample Matrix: Rock Island Wa./#39200

Soil

Date Received:

03/14/91

Date TCLP Performed: 03/25/91 Date Analyzed: Work Order #:

03/26/92 K911367

QA/QC Report **Duplicate Summary** Toxicity Characteristic Leaching Procedure (TCLP) EPA Method 1311 Metals mg/L (ppm) in TCLP Extract

Sample Name:

Pond #4 K1367-5

Lab Code:

Analytes	Methods	MRL	Sample Result	Duplicate Sample Result	Average	Relative Percent Difference
Arsenic	3010/6010	0.1	0.2	0.1	0.2	50
Barium	3010/6010	0.1	1.2	1.3	1.2	8
Cadmium	3010/6010	0.01	ND	ND	ND	
Chromium	3010/6010	0.01	0.01	0.02	0.02	50
Lead	3010/6010	0.05	0.22	0.22	0.22	<1
Mercury	7470	0.001	ND	ND	ND	
Selenium	3010/6010	0.1	ND	ND	ND	
Silver	3010/6010	0.01	ND	ND	ND	

**MRL** 

Method Reporting Limit

ND

None Detected at or above the method reporting limit

alhi Mielman

PW 30001177

Client:

Environmental Engineering & Consulting, Inc.

Submitted By:

Pat Wicks

Project:

Rock Island Wa./#39200

Sample Matrix: Soil

Date Received:

03/14/91

Date TCLP Performed: 03/25/91

Date Analyzed:

03/26/91

Work Order #:

K911367

QA/QC Report Matrix Spike Summary Toxicity Characteristic Leaching Procedure (TCLP) EPA Method 1311 Metals mg/L (ppm) in TCLP Extract

Sample Name: Pond #4 Lab Code: K1367-5

					Spiked	
		Spike		Sample	Sample	Percent
Analytes	Methods	Level	MRL	Result	Result	Recovery
Arsenic	3010/6010	5.0	0.1	0.2	5.6	108
Barium	3010/6010	5.0	0.1	1.2	6.8	112
Cadmium	3010/6010	1.0	0.01	ND	1.03	103
Chromium	3010/6010	5.0	0.01	0.01	5.31	106
Lead	3010/6010	5.0	0.05	0.22	5.48	106
Mercury	7470	0.01	0.001	ND	0.010	100
Selenium	3010/6010	1.0	0.1	ND	1.1	110
Silver	3010/6010	1.0	0.01	ND	1.07	107

MRL

Method Reporting Limit

ND

None Detected at or above the method reporting limit

PW 30001178

Client: Environmental Engineering & Consulting, Inc.

Submitted By: Pat Wicks

Project: Rock Island Wa./#39200

Sample Matrix: Soil/Water

 Date Received:
 03/14/91

 Date TCLP Performed:
 03/25/91

 Date Extracted:
 03/28/91

 Date Analyzed:
 03/29/91

Work Order #: K911367

QA/QC Report
Surrogate Recovery Summary
Toxicity Characteristic Leaching Procedure (TCLP)
EPA Method 1311
Semivolatile Organic Compounds
(EPA Methods 3510/8270)
in TCLP Extract

#### Percent Recovery

Sample Name:	Pond #1,2,8 Comp	Pond #4	Pond #6	Percent Recovery Acceptance
Lab Code:	K1367-4	K1367-5	K1367-6	Criteria
Analytes				
2-Fluorophenol	63.6	64.3	58.4	21-100
Phenol-D <sub>6</sub>	60.8	64.4	56.1	10-94
2,4,6-Tribromophenol	85.4	90.7	87.1	10-123
Nitrobenzene-D <sub>6</sub>	66.7	71.7	61.1	35-114
2-Fluorobiphenyl	66.5	74.2	63.9	43-116
Terphenyl-D <sub>14</sub>	111	101	110	33-141

Approved by affic Affelha

Date 4/11/41

W 30001179

Client:

Environmental Engineering & Consulting, Inc.

Submitted By:

Pat Wicks

Project:

Rock Island Wa./#39200

Sample Matrix:

Soil

Date Received: 03/14/91

Date TCLP Performed: 03/25/91

Date Extracted:

03/28/91

Date Analyzed:

03/29/91

Work Order #:

K911367

QA/QC Report
Surrogate Recovery Summary
Toxicity Characteristic Leaching Procedure (TCLP)
EPA Method 1311
Semivolatile Organic Compounds
(EPA Methods 3510/8270)
in TCLP Extract

#### Percent Recovery

Sample Name: Lab Code:	Method Blank K1367-MB	EPA Percent Recovery Acceptance Criteria
Analytes		
2-Fluorophenol	72.0	21-100
Phenol-D <sub>6</sub>	66.2	10-94
2,4,6-Tribromophenol	87.4	10-123
Nitrobenzene-D <sub>5</sub>	76.8	35-114
2-Fluorobiphenyl	72.0	43-116
Terphenyl-D <sub>14</sub>	99.5	33-141

Approved by

No

PW 30001180

# APPENDIX B CHAIN OF CUSTODY INFORMATION

PW 30001181

#### **ENVIRONMENTAL ENGINEERING & CONSULTING, INC.**

19125 Northcreek Parkway, Suite 120 Bothell, Washington 98011–8002 (206) 485–3437 FAX (206) 483–1058

#### **CHAIN OF CUSTODY RECORD**

Sampled by: Tom Clymer	Project #:_39200	
Date: 3-/3-9/	Location: Rock Island Wa	<u>.</u>

	T		ı	Y		
Sample	Date/Time	Type of	# of	Lab	Analyses Required	
Number	Sampled	Sample Fes./s.	Containers	Sample #	and Comments	
Pond#1	3-13-91	Solid Fume	1			
Pond #2	3-13-91	FeSi/Si Solid Fume				
Pond#4	3-13-91	Fesilsi Soldfune	1			
Pond #6	3-13-9/	Liquid 9. Fume	2			
Pond #8	3-13-91	Solid Fesi fune	1			
-						
Relinquished	by: Tom C	lymer		Relinguished	d by:	
Organization:	Signatur	é	Inc.	Signature Organization:		
Date: <u>3-/3-9</u>		100 FM		Date:Time:		
Received by:	france	Addu	:	Received by	s:Signature	
Organization:	Asignature 3	114/91 8	30	Signature Organization:		
Relinquished t	•			Additional C	omments:	
Organization:_	Signature					
Date:	Time:					
Received by:_						
Organization:_	Signature	9				
				<u> </u>		

Laboratory Please Note: Results are to be reported as detected values or as less than detected values with detection limit; do not use U, J, B, K, M or similar data reporting qualifiers.



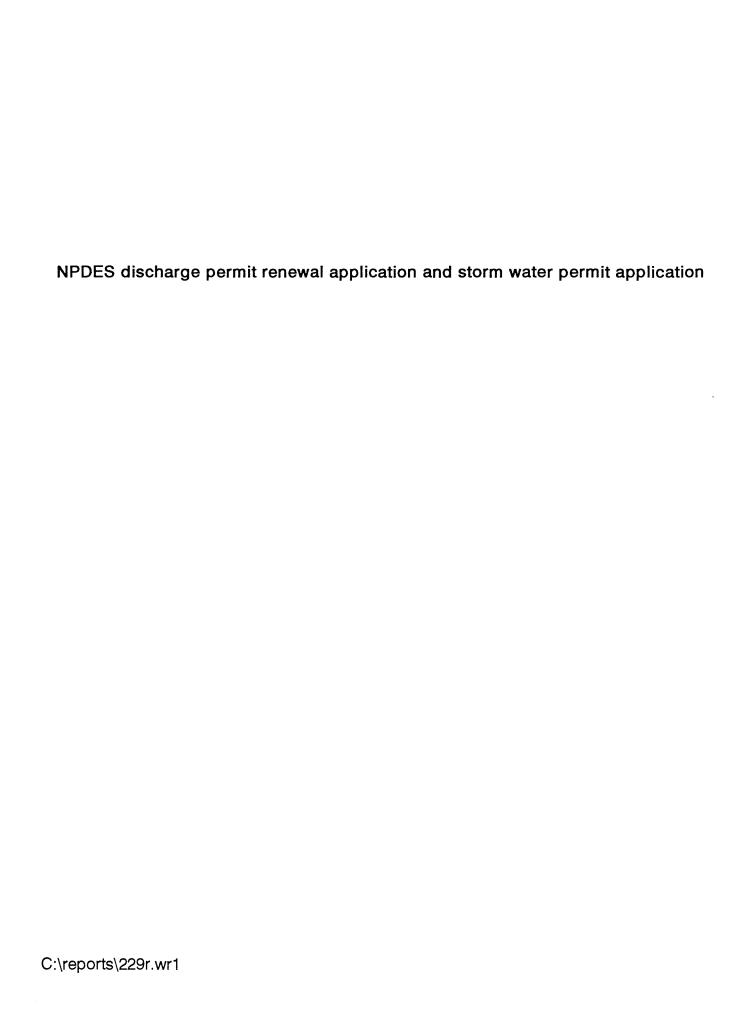
### **INSTRUCTIONS FOR LABORATORY ANALYSIS**

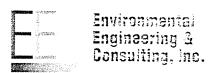
<b>Date:</b> 3/11/91	Project Number 39300 Please show this Project no. on invoices, charge slips and statements
Test 2 so Turnaround Needed: Normal	emples: one from pond 4 and one composite  of ponds 1, 2 and 8
One week	
Other	
Compositing:	Yes No # Composite samples from ponds 1, 2 and 8 in to 0
Chromatograms required:	Yes No
Client to pay invoice:	No, EEC to pay
Fax results as soon as available:	Yes No
Analyses required: Prices:	TCLPSV TCLP Chamber Compositing  595 \$105 \$380 \$38 \$10
Analyses required: Prices:	

T	t de la company	Environmental
<b> -</b>	Walter	Engineering &
	A CALOREST STATE OF THE STATE O	Consulting, Inc.

INSTRUCTIONS FOR L	ABORATORY ANALYSIS
Date: 3/11/91	Project Number 39260 Please show this Project no. on invoices, charge slips and statements
	Fume Slurry Somples only in one l. battles; only one X somple to be tested
One week	
Other	
Compositing:	Yes No, unless needed
Chromatograms required:	Yes
Client to pay invoice:	Yes, lent send towork to EEC, to be forwarded:  No, EEC to pay
Fax results as soon as available:	Yes No TUP TUP discountingous
Analyses required: Prices:	TCLP TCLP TCLP chromatogous  SV organis for  extruction netals 8270 &  \$ 95 \$ 105 \$ 380 \$ 38
Analyses required: Prices:	

PW 30001184





Mr. Jim Trunzo Executive Vice President Silicon Metaltech, Inc. 100 4th Street Rock Island, Washington 98850

#### Dear Jim:

Enclosed are completed NPDES permit application forms (except for entries to be made by you or Robert) for renewal of the effluent discharge permit and for the storm water discharge permit, including related information:

- a. NPDES permit application 2C forms for renewal of the effluent discharge permit
- b. NPDES permit application 2F forms for storm water discharge permit
- c. Complete set of instructions and forms for storm water discharge permit application
- d. Schematic water flow diagram to be attached to permit application Forms 2C and 2F
- e. Site Plan (two pages) to be attached to the storm water discharge permit application
- f. Topographic map to be attached to permit application Form 1
- g. Monitoring data summary 1986-1990 (FYI only, not to be attached to applications)

Please review all information in the application (including the schematic water flow diagram site plan), since you are responsible that it is correct. In particular, review the storm water permit application, Table 2F-2. Any items from that table that I have not included in VII.E. as being used at the plant should be added to VII.E. If you would like to know the basis for any information, calculation or data on the forms, please call.

I have assumed that with the exception the two areas shown on the site plan and possibly the raw materials storage area, there is no surface runoff to the Columbia River from any area of the plant. If this is not correct, please advise.



Mr. Jim Trunzo March 19, 1991 Page 2

Section III of the storm water permit application requests hazardous waste accumulation areas to be marked on a site map. I assumed the only hazardous waste you generate is spent Safety-Kleen solvent from parts washers and that this material remains in use and does not become a waste until it is picked up by Safety-Kleen. Consequently you would not "accumulate" hazardous waste and nothing has been so marked on the site plan. If this is not correct or other wastes are generated or handled differently, please advise.

When you submit the storm water permit application, please attach a site drawing showing topographic information. I would suggest the survey/topographic map prepared in 1988 by Gahringer & Associates of Wenatchee.

Please send a copy of the storm water permit application to the U S Environmental Protection Agency, Region 10, 1200 Sixth Avenue, Seattle, Washington 98101.

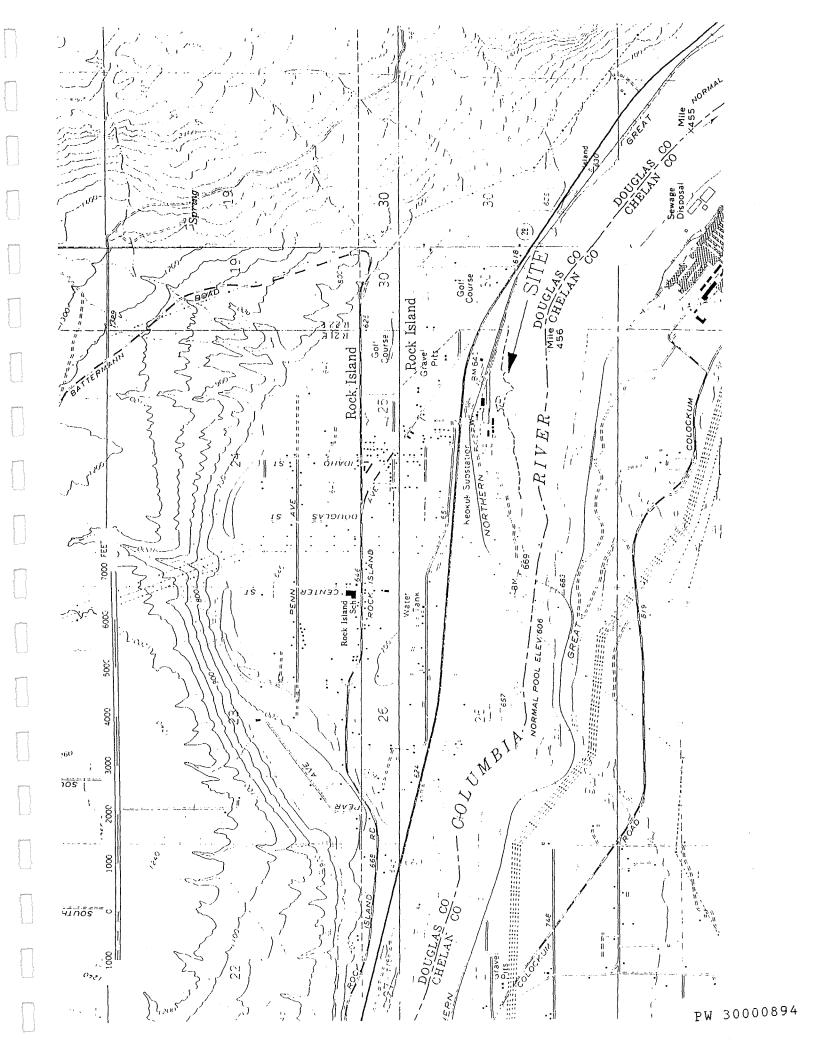
I would appreciate your sending me a copy of the applications submitted also. Please do not hesitate to call me if there are any questions.

Sincerely,

Patrick H. Wicks, PE, CHMM President

Enclosures

cc: Robert L. Miller



20 SEPA

U.S. ENVIRONMENTAL PROTECTION AGENCY APPLICATION FOR PERMIT TO DISCHARGE WASTEWATER

EXISTING MANUFACTURING, COMMERCIAL, MINING AND SILVICULTURAL OPERATIONS

1110,	COMMENCIAL, MINING AND SI	
	Consolidated Permits Program	

				on Charles and the
For each outfall, list the la	atitude and longitude of	of its location to the nearest	15 seconds and the name of the receiving	water.

NUMBER	8.	LATITUDE	Ε	. c. r	ONGITUD	E	D. RECEIVING WATER (name)
(list)	1. 0 EG.	2. MIH.	3, SEC.	1. DEG.	2. MIN.	3. SEC.	D. RECEIVING WATER MAINEY
001	47	22	11	120	08	16	Columbia River
		-					
		ļ			ļ		

I. OUTFALL LOCATION

#### II. FLOWS, SOURCES OF POLLUTION, AND TREATMENT TECHNOLOGIES

- A. Attach a line drawing showing the water flow through the facility. Indicate sources of intake water, operations contributing wastewater to the effluent, and treatment units labeled to correspond to the more detailed descriptions in Item B. Construct a water balance on the line drawing by showing average flows between intakes, operations, treatment units, and outfalls. If a water balance cannot be determined (e.g., for certain mining activities), provide a pictorial description of the nature and amount of any sources of water and any collection or treatment measures.
- B. For each outfall, provide a description of: (1) All operations contributing wastewater to the effluent, including process wastewater, sanitary wastewater cooling water, and storm water runoff; (2) The average flow contributed by each operation; and (3) The treatment received by the wastewater. Continue on additional sheets if necessary.

Furnaces	b. AVERAGE FLOW (include units) 2,187,000gpd	a. DESCRIPTION	b. LIST CODES FRO
	1 2 187 000 and		
Storm Wat-	1 - 1 - 0 1 1 0 0 0 9 0 0	Discharge	A-4
Storm Water		Discharge	A-4
Fume_Slurry_Ponds	120 qpd	Evaporation	1-F
		Drying Beds	5-H
		Sedimentation	1-U
		1	
			30000895
		PW	300007
	Fume Slurry Ponds	Fume Slurry Ponds 120 qpd	Drying Beds Sedimentation

				3. FRE	QUENCY			4. FLOW		
OUTFALL	Z. OPER			a. DAYS	b. MONTHS	a. FLOV		b. TOTAL (specify u	VOLUME vith units)	c DUF
(list)	CONTRIBU (li	TING	FLOW	PER WEEK (specify average)	(specify average)	1. LONG TERM LVERAGE	2. MAXIMUM DAILY	I. LONG TERM AVERAGE	2, MAXIMUM DAILY	(in days
	t guideline limitation		ulgated by EPA	under Section 30	4 of the Clean		oly to your factor (IV)			<del></del>
3. Are the limitati	ons in the applicable	efflue	nt guideline ex	pressed in terms of	production (or	other measu		n/?		
. If you answere	d "yes" to Item III-8 olicable effluent gui	list the	quantity whi	ch represents an a	ctual measure ls.				sed in the term	s and uni
			1. AVERA	GE DAILY PRODUC	TION			)	2. AFFI	ECTED
a. QUANTITY PER C	b. UNITS OF	MEASL	RE	C. OPI		OUTFALLS (list outfall number				
. IMPROVEMEN	TS									
water treatment	equired by any Fede t equipment or praced to, permit conditions.	tices or	any other en iministrative o	vironmental progra	ms which may ers, enforcemen	affect the di	scharges desc	ribed in this a	pplication? Th	is include
DENTIFICATIO	N OF CONDITION, INT, ETC.	2. A	FFECTED OL		3. BR	IEF DESCRI	PTION OF PR	OJECT	4. FIN	AL SOM
								***	PW 3000	
									FM 2000	0070

PW 30000897

II. BIOLOGICAL TOXICITY TESTING DA		(日本) 日本 (日本) 日本 (日本) 日本 (日本) 日本 (日本) 日本 (日本) 日本 (日本) (日本)	
Do you have any knowledge or reason to bell	eve that any biological test for acute or chronic tox	icity has been made on any of	your discharges or on a
receiving water in relation to your discharge v	within the last 3 years?		
YES (identify the to	est(s) and describe their purposes below)	⊠NO (go to Sect	ion VIII)
HILCONTRACT ANALYSIS INFORMATION			
Were any of the analyses reported in Item V	performed by a contract laboratory or consulting fi	rm?	
YES (list the name, analyzed by, e	address, and telephone number of, and pollutants ach such laboratory or firm below)	NO (go to Sect	ion IX)
A. NAME			
	8. ADDRESS	(area code & no.)	B. POLLUTANTS ANALYZED
		(area code & no.)	(list)
Lauchs Testing	940 South Harney St.	(area code & no.)	Priority Polluta
Lauchs Testing Laboratories, Inc.		(area code & no.)	Priority Polluta and other hazard
The state of the s	940 South Harney St.	(area code & no.)	Priority Polluta
The state of the s	940 South Harney St. Seattle, WA 93108	(206)767-506	Priority Polluta and other hazard substances.  Priority Pollutant
Laboratories, Inc.	940 South HarneySt. Seattle, WA 93108	(206)767-506	Priority Polluta and other hazard substances.  Priority Pollutant metals/elements
Laboratories, Inc.	940 South Harney St. Seattle, WA 93108	(206)767-506	Priority Polluta and other hazard substances.  Priority Pollutant metals/elements
Laboratories, Inc.	940 South Harney St. Seattle, WA 93108	(206)767-506	Priority Polluta and other hazard substances.  Priority Pollutant metals/elements
Laboratories, Inc.	940 South Harney St. Seattle, WA 93108	(206)767-506	Priority Polluta and other hazard substances.  Priority Pollutant metals/elements
Laboratories, Inc.	940 South Harney St. Seattle, WA 93108	(206)767-506	Priority Polluta and other hazard substances.  Priority Pollutant metals/elements
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Laboratories, Inc.	940 South Harney St. Seattle, WA 93108	(206)767-506	Priority Polluta and other hazard substances.  Priority Pollutant metals/elements
Laboratories, Inc.	940 South Harney St. Seattle, WA 93108	(206)767-506	Priority Polluta and other hazard substances.  Priority Pollutant metals/elements
Laboratories, Inc.	940 South Harney St. Seattle, WA 93108	(206)767-506	Priority Polluta and other hazard substances.  Priority Pollutant metals/elements
Laboratories, Inc.  AM Test, Inc.  X. CERTIFICATION  // Certify under penalty of law that this docume	940 South Harney St. Seattle, WA 93108  14603 NE 87th St. Redmond, WA 98052	(206)767-506 (206)885-166	Priority Polluta and other hazard substances.  Priority Pollutant metals/elements cyanide, phenol
Laboratories, Inc.  AM Test, Inc.  X. CERTIFICATION  // certify under penalty of law that this docume assure that qualified personnel properly gather	940 South Harney St. Seattle, WA 93108  14603 NE 87th St. Redmond, WA 98052	(206) 767-506 (206) 885-166 (206) 885-166	Priority Polluta and other hazard substances.  Priority Pollutant metals/elements cyanide, phenol
AM Test, Inc.  AM Test, Inc.  X. CERTIFICATION  // Certify under penalty of law that this docume assure that qualified personnel properly gathe those persons directly responsible for gathering	940 South Harney St. Seattle, WA 93108  14603 NE 87th St. Redmond, WA 98052	(206)767-506 (206)885-166 (206)885-166 irrection or supervision in accomplying uirry of the person or pute best of my knowledge and be	Priority Polluta and other hazard substances.  Priority Pollutant metals/elements cyanide, phenol
AM Test, Inc.  AM Test, Inc.  X. CERTIFICATION  // Certify under penalty of law that this docume assure that qualified personnel properly gathe those persons directly responsible for gathering	940 South Harney St. Seattle, WA 93108  14603 NE 87th St. Redmond, WA 98052  ant and all attachments were prepared under my divided information submitted. Based on the information, the information, including the ses for submitting false information, including the	(206) 767-506  (206) 885-166  (206) 885-166  irrection or supervision in accompy inquiry of the person or property in the person of the person o	Priority Polluta and other hazard substances.  Priority Pollutant metals/elements cyanide, phenol
AM Test, Inc.  X. CERTIFICATION  [certify under penalty of law that this docume assure that qualified personnel properly gather those persons directly responsible for gathering am aware that there are significant penaltic	940 South Harney St. Seattle, WA 93108  14603 NE 87th St. Redmond, WA 98052  ant and all attachments were prepared under my divided information submitted. Based on the information, the information, including the ses for submitting false information, including the	(206) 767-506  (206) 885-166  (206) 885-166  irrection or supervision in accompy inquiry of the person or property in the person of the person o	Priority Polluta and other hazard substances.  Priority Pollutant metals/elements cyanide, phenol cross who manage the system or elief, true, accurate, and complete. Imment for knowing violations.
AM Test, Inc.  X. CERTIFICATION  Acceptify under penalty of law that this docume assure that qualified personnel properly gather those persons directly responsible for gathering am aware that there are significant penaltic A. NAME & OFFICIAL TITLE (type or principle)	940 South Harney St. Seattle, WA 93108  14603 NE 87th St. Redmond, WA 98052  ant and all attachments were prepared under my divided information submitted. Based on the information, the information, including the ses for submitting false information, including the	rection or supervision in accordingly inquiry of the person or person of the person of the possibility of fine and imprison in according to the person of th	Priority Polluta and other hazard substances.  Priority Pollutant metals/elements cyanide, phenol cross who manage the system or elief, true, accurate, and complete. Inment for knowing violations.
AM Test, Inc.  X. CERTIFICATION  [certify under penalty of law that this docume assure that qualified personnel properly gather those persons directly responsible for gathering am aware that there are significant penaltic	940 South Harney St. Seattle, WA 93108  14603 NE 87th St. Redmond, WA 98052  ant and all attachments were prepared under my divided information submitted. Based on the information, the information, including the ses for submitting false information, including the	(206) 767-506  (206) 885-166  (206) 885-166  irrection or supervision in accompy inquiry of the person or property in the person of the person o	Priority Polluta and other hazard substances.  Priority Pollutant metals/elements cyanide, phenol cyanide, phenol
AM Test, Inc.  AM Test, Inc.  C. CERTIFICATION  Correctly under penalty of law that this docume assure that qualified personnel properly gather those persons directly responsible for gathering am aware that there are significant penaltic.  A. NAME & OFFICIAL TITLE (type or prince)	940 South Harney St. Seattle, WA 93108  14603 NE 87th St. Redmond, WA 98052  ant and all attachments were prepared under my divided information submitted. Based on the information, the information, including the ses for submitting false information, including the	rection or supervision in accordingly inquiry of the person or person of the person of the possibility of fine and imprison in according to the person of th	Priority Polluta and other hazard substances.  Priority Pollutant metals/elements cyanide, phenol cyanide, phenol ersons who manage the system of elief, true, accurate, and complete inment for knowing violations.

PA Form 3510-2C (Rev. 2-85)

PLEASE PRINT OR TYPE IN THE UNSHADED AREAS ONLY. You may report some or all of this information on separate sheets (use the same format) instead of completing these pages. SEE INSTRUCTIONS.

EPA I.D. NUMBER (copy from Item 1 of Form 1)

WADD00756940

Form Approved.
OMB No. 2000-0059
Approval expires 12-31-85

oval expires 12-31-85
OUTFALL NO.

V. INTAKE AND EFFLUENT	CHARACTERISTICS (continued	d from page 3 of Form 2-C)
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001

PART A - You must provide the results of at least one analysis for every pollutant in this table. Complete one table for each outfall. See instructions for additional details.

				EFFLUENT				3. UN		4, IN	TAKE (option	al)
1. POLLUTANT	a. MAXIMUM	DAILY VALUE	b. MAXIMUM 3	ODAY VALUE	CLONG TERM	allable)		(specify i	1	a. LONG	TERM E VALUE	b, NO. OF
	(i)	(2) MASS	CONGENTRATION	(2) MASS	CONCENTHATION	(2) MASS	ANALYSES	B. CONCEN- TRATION	b, MASS	CONCENTRATION	(2) MASS	ANALYSES
a. Biochemical Oxygen Demand (IIOD)							The second secon					
b. Chemical Oxygen Demand, (COD)		Ŋ	OT PRESE	NTITA WON:	TORED, E	BUT WOULD						
c. Total Organic Carbon (TOC)		E	E WILLIN	G TO SAMI	LE AND J	EST IF R	EQUIRED	•1				
d. Total Suspended Solids (TSS)												
e. Ammonia (as N)												
f. Flow	2,987,	432	VALUE	<b>-</b>	2,187,	000			pdp	VALUE		
g. Temperature <sup>(winter)</sup> all yea	VALUE	4	VALUE		VALUÉ 22			°C		VALUE		
h. Temperature (summer)	VALUE		VALUE		VALUE			°C		VALUE		
l, pH	7.4 avg	7.8	MUMINIM	MAXIMUM				STANDAR	DUNITS		><<	

PART B - Mark "X" in column 2-a for each pollutant you know or have reason to believe is present. Mark "X" in column 2-b for each pollutant you believe to be absent. If you mark column 2a for any pollutant which is limited either directly, or indirectly but expressly, in an effluent limitations guideline, you must provide the results of at least one analysis for that pollutant. For other pollutants for which you mark column 2a, you must provide quantitative data or an explanation of their presence in your discharge. Complete one table for each outfall. See the instructions for additional details and requirements.

1. POLLUT-	2. MAF	K 'X'			3. E	FFLUENT				4. UI	STIP	5. INT	AKE (option	al)
ANT AND CAS NO.	A. BE- LIEVEL PRE- SENT	b. ue-	a. MAXIMUM I	DAILY VALUE	b. MAXIMUM 30	b. MAXIMUM 30 DAY VALUE		C.LONG TERM AVRG. VALUE		a. CONCEN-		a, LONG AVERAGE	TERM VALUE	b. NO. OF
(if available)	SENT	SENT	CONCENTRATION	(2) MASS	CONCENTRATION	(2) MASE	CONCENTHATION	(2) MASS	YSES	RATION	b, MASS	CONCENTRATION	(Z) MASS	YSES
a. Bromide (24959-67-9)	4	Х												0
b. Chlorine, Total Residual	30	Х				3 1 1 1 1 2 3		Tall						0
c. Color	8000	Х												0
d, Fecal Collform	999	х												0
e. Fluoride (16984-48-8)	x		No prod	ess sour	ce expect	ed; mayb	e present	in inta	ake				** 15, y = 31 (\$100, y = 41) to	0
f. Nitrate- Nitrite (as N)	х		339 600 10		ce expect									0

1. POLLUT-	2. MA	RK 'X'	N.4865.110		NIIVE SA					UENŤ								4. UI	NITS	5. INT	AKE (optional	)
ANT AND	8.06-	b.ez-	a. MA	MUMIX	DAILY	VALUE	b.	MAXIMUM 3	9.RAY	VALUE	C.L	ONG TERM	A.Y.	g. V	ALUE	d. NO. OF		MCEN		* & FHONE		b. NO. OF
CAS NO.	PRE-	VO-		(1)	T	MASS	+	HEENTHATION	100000000000000000000000000000000000000	MASS	-	(i)	-	(z) M	7	ANAL-	TR	ATION	b, MASS	CONCENTRATION	(2) MASS	YSES
g. Nitrogen, Total Organic (as N)		Х	CUNCE	NINATION			150	NCENTRATION				CENTRATION	POCTA							CONCENTRATION		
h. Oll and Grease		Х																			<u> </u>	
. Phosphorus (as P), Total (7723-14-0)	Х		no	proc	ess	sour	се	expect	ed;	may	be	prese	h t	in	int	ake .	or	sto	cm wate	r runoff		
. Radioactivity						VI. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1.											-					
(1) Alpha, Total		x	1																			
(2) Beta, Total	•	х																				
(3) Radium, Total		x																				
(4) Radium 226, Total		х	)																			
k. Sulfate (as SO <sub>4</sub> ) (14808-79-8)	Х		no	proc	ess	sour	¢е	expect	ed;	may	be	prese	nt	in	int	ake o	r	stor	water	runoff		
l. Sulfide (de S)		X.	•																	Idiloii		
m. Sulfite (as SO3) (14265-45-3)		Х													CC 484-14-00-05							
n. Surfactants		Х													*****						Total Company	<b> </b>
o. Aluminum, Total (7429-90-5)	х		no	proc	ess	sour	ce	expect	ed.	may	he	nrese	4-	in	int	also o				runoff		
p. Barlum,					- 22		<del></del>			y	-	preser	-	111	TILL	ake o	L S	COLI	water	runoff		-
Total (7440-39-3)	х		no	proc	ess	sour	de	expect	ed.	mav	he	nreser	+	in	int	ako o		. + 0 ~~		runoff	Ŷ	
q. Boron, Total		7,										preser	-	1.11	TILL	are o	μ 5	COLI	water	runori	<u> </u>	
(7440-42-8) r. Cobalt, Total		Х			-		-				<u> </u>					1						
(7440-48-4) I. Iron, Total		Х					-				-											-
(7439-89-6) t. Magneslum,	Х		no	proc	ess	sour	е	expect	ed;	may	be	preser	t	in	int	ake o	r s	torm	water	runoff		
Total (7439-95-4) . Molybdenum,		Х				-4	-				_			127500000								
Total (7439-98-7) v. Manganese,		х																				
Total (7439-96-5)	χ		no	proc	ess	sour	е	expect	ed;	may	be	presen	t_	in	int	ake o	c s	torm	water	runoff		
w. Tin, Total (7440-31-5)	Χ		no	proc	ess	sour	е	expect	ed;	may	þе	presen	t	in	int	ake o	r s	torm	water	runoff		
x. Titenium, Totel (7440-32-6)		х					1															

EPA I.D. NUMBER (copy from Hem 1 of Form 1) OUTFALL NUMBER

WAD000756940

001

Form Approved. OMB No. 2000-0059 Approval expires 12-31-85

CONTINUED FROM PAGE 3 OF FORM 2-C

PART C - If you are a primary industry and this outfall contains process wastewater, refer to Table 2c-2 in the instructions to determine which of the GC/MS fractions you must test for. Mark "X" in column 2-a for all such GC/MS fractions that apply to your industry and for ALL toxic metals, cyanides, and total phenols. If you are not required to mark column 2-a (secondary industries, nonprocess wastewater outfalls, and nonrequired GC/MS fractions), mark "X" in column 2-b for each pollutant you know or have reason to believe is present. Mark "X" in column 2-c for each pollutant you believe is absent. If you mark column 2a for any pollutant, you must provide the results of at least one analysis for that pollutant. If you know or have reason to believe it will be discharged in concentrations of 10 ppb or greater. If you mark column 2, you must provide the results of at least one analysis for each of these pollutants which you know or have reason to believe that you discharge in concentrations of 100 ppb or greater. Otherwise, for pollutants for which you mark column 2b, you must either submit at least one analysis or briefly describe the reasons the pollutant is expected to be discharged. Note that there are 7 pages to this part; please review each carefully. Complete one table (all 7 pages) for each outfall. See instructions for additional details and requirements.

1. POLLUTANT	2.	MARK	·x·			3.	EFFLUENT				4, U	INITS	5. IN'	TAKE (optio	onal)
	1			A MAXIMUM	DAILY VALUE			c.LONG TERM	AYRG. VALUE	d NO.OF	_	1		IG TERM SE VALUE	b, NO, C
(if available)	ING RE- QUIR- ED	T D. BE- LIEVED PHE- SENT	SENT	CONCENTRATION		(I) ava		(If avai		ANAL- YSES	a, CONCEN- TRATION	b, MASS	(I) CONCENTRATION	(2) MASS	YSES
METALS, CYANID								l l							
1M. Antimony, Total (7440-36-0)			Х							3					0
2M. Arsenic, Total (7440-38-2)		Х		0.004	0.07					<sub>-</sub> 3	ppm	lb/d	0,003	0.05	3
3M. Beryllium, Total, 7440-41-7)			Х							3					0
4M. Cadmium, Total (7440-43-9)		,	х							3			ND		3
5M. Chromium, Total (7440-47-3)		х		0.017	0.31					3	ppm	lb/d			0
6M. Copper, Total (7440-50-8)		х		0.089	1.6					3	ppm	lb/d			0
7M. Lead, Total (7439-92-1)		х		ND						3	ppm	lb/d	0.001	0.02	3
8M. Mercury, Total (7439-97-6)	.1		х	,						3			ND		3
9M. Nickel, Total (7440-02-0)		Х		0.008	0.14					3	ppm	lb/d			0
10M. Selenium, Total (7782-49-2)			х							3			ND		3
11M. Silver, Total (7440-22-4)		,	X							3					0
12M. Thallium, Total (7440-28-0)	PW 3		Х	1						3					0
13M. Zinc, Total (7440-66-6)	0000	Х		0.011	0.2					3	ppm	lb/d			0
14M. Cyanide, Total (57-12-5)	0901		x			- Committee and	1			3					0
16M. Phenols, Total	1		х							3					0
DIOXIN	-	***************************************	<u></u>	*****	<u> </u>			<del>-1</del>			4	<u> </u>			
2,3,7,8-Tetra-	T	T	Т	DESCRIBE RES	CIN TC										

2,3,7,8-Tetrachlorodibenzo-PDioxin (1764-01-6)

CONTINUED FROM													<b></b>		
1. POLLUTANT AND CAS		MARK					EFFLUENT	LONG TERM	AVDC VALUE		4, UI	VITS		TAKE (optio	
NUMBER	ING RE-	D, WE- LIEVED PHE- SENT	LIEVEU C. BE-	a. MAXIMUM (				C.LONG TERM		I ANAL.	а. СОИСЕН- ТВАТІОЙ	b, MASS	A LONG		b, NO.OF
(if available) GC/MS FRACTION					(2) MASS	CONCENTRATION	(2) MASS	CONCENTHATION	(z) MASS	YSES			(I) CONCEN- THATION	(2) MABS	YSES
	- 70	LATIL	E COM	POUNDS								• .			
1V. Acrolein (107-02-8)			Х		**************************************					0		400			0
2V. Acrylonitrile (107-13-1)			Х							0					0
3V. Benzene (71-43-2)			Х							2					0
4V. Bis (Chloro- methyl) Ether (542-88-1)			Х		- variable and a second					0					0
5V. Bromoform . (75-25-2)			Х							2	•				0
6V. Carbon Tetrachloride (56-23-5)			Х							2					0
7V. Chlorobenzene (108-90-7)			Х							2					0
8V, Chlorodi- bromomethane (124-48-1)			Х							2					0
9V. Chloroethane (75-00-3)	**********		X							1					0
10V. 2-Chloro- ethylvlnyl Ether (110-75-8)			Х							1					0
11V. Chloroform (67-66-3)			Х							2					0
12V. Dichloro- bromomethane (75-27-4)			Х							2					0
13V. Dichloro- difluoromethana (75-71-8)			Х							0					0
14V. 1,1-Dichloro- ethane (75-34-3)			Х							2					0
15V. 1,2-Dichloro- ethane (107-06-2)			Х							2					0
16V. 1,1-Dichloro- ethylene (75-35-4)			Х							2					0
17V. 1,2-Dichloro- propane (78-87-5)	P €		Х							2					0
18V. 1,3-Dichloro- propylene (542-75-6)	3000		Х							2					0
19V. Ethylbenzene (100-41-4)	0600	)	Х							2					0
20V. Methyl Bromide (74-83-9)	,	<b>.</b>	Х							2					0
21V. Methyl Chloride (74-87-3)			Х			·				2		,			0
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I. POLLUTANT AND CAS		MARK					EFFLUENT				4. UI	VITS		AKE (optio	onal)
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23V, 1,1,2,2·Tetra- .hloroethane 79-34-5)			Х							2		***************************************			0
24V, Tetrachloro- othylene (127-18-4)			X	- · · · · <del></del> · · ·						2					0
25V, Toluene 108-88-3)			Х							1					0
26V. 1,2-Trans Dichloroethylene (156-60-5)			X							2					0
27V. 1,1,1-Tri- ;hloroethane (71-55-6)			X				CHINESE SERVICE STATE OF THE S			2					0
28V. 1,1,2-Trl- chloroethane (79-00-5)			Х				Very appropriate age plants are assessed in the contraction of the con			2					0
29V. Trichloro- ethylene (79-01-6)			Х				***************************************			2					0
30V, Trichloro- fluoromethane (75-69-4)			X						***************************************	0					0
31V. Vinyl Chloride (75-01-4)		10.001	X	100						2					0
GC/MS FRACTION		ID CON	IPOUN	102											
1A. 2-Chlorophenol (95-57-8) 2A. 2,4-Dichloro-			. X						and the second s	2					0
phonol (120-83-2)			X				-			2					0
3A. 2,4-Dimethyl- phenol (105-67-9) 4A. 4,6-Dinitro-O-			X				~			2					0
Cresol (534-52-1)			Х							2	Santakora via tak salatana valata garang anaka.			and a finite layer of another hands and another finite for	0
phenol (51-28-5)	-		X							2					0
(88-75-5) 7A. 4-Nitrophenol	_	-U	X						-1	2					0
(100-02-7) 		<b>Σ</b> ω	X							2					0
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phenol (87-86-5)		903	X							2					0
(108-95-2) 11A. 2,4,6-Tri-		ı ——	X			<b>t</b> ,	~			2.					0
chlorophenol			X							2			,		0

CONTINUED FROM THE FRONT 1. POLLUTANT 3. EFFLUENT 4. UNITS 2. MAHH 'X' 5. INTAKE (optional) AND CAS NUMBER b. MAXIMUM 30 DAY VALUE | C.LONG TERM AVRG. VALUE (if quallable) A LONG TERM AVERAGE VALUE STEST PRE- CRE-8. MAXIMUM DAILY VALUE b. NO. OF d NO. OF a. CONCEN ANALb. MASS ANAL. TRATION (I) CONCEN-(if available) (I) MANS (1) MASS (c) MAGE (1) MASS GC/MS FRACTION - BASE/NEUTRAL COMPOUNDS 1B. Acenaphthene X 2 0 (83-32-9) 2B. Acenaphtylene (208-96-8) Х 2 0 3B. Anthrucene 2 X 0 (120 - 12 - 7)48. Benzidine 2 X 0 (92-87-5) 5B. Benzo (a) Anthracena X 2 0 (56-55-3)6B. Benzo (a) X 2 0 Pyrene (50-32-8) 7B. 3.4-Bonzofluoranthene 0 X 2 (205 - 99 - 2)88. Benzo (ghi) Perylene X 2 0 (191-24-2) 9B. Benzo (k) Fluoranthene 0 X 1 (207-08-9) 10B. Bis (2-Chloroethoxy) Methane X 2 0 (111-91-1) 11B. Bis (2-Chloroethyl) Ether X 2 0 (111-44-4) 12H. Bis /2-Chloroiso-0 propyl) Ether (102-60-1) X 2 13B. Bls (2-Ethylhexyl) Phthalate X 2 0 (117-81-7) 14B. 4-Bromophenyl Phenyl X 2 0 Ether (101-55-3) 15B. Butyl Benzyl Phthalate (85-68-7) 2 0 X 16B. 2-Chloronaphthalene 2 0 X (91-58-7) 4 E 17B. 4-Chlorophenyl Phenyl 2 0 3000 X Ether (7005-72-3) 18B. Chrysene 2 0 (218-01-9) X 09 19B. Dibenzo (a,h) 0 Anthracene 2 0 X (53.70.3)20B. 1,2-Dichlorobenzene (95-50-1) X 2 0 21B. 1,3-Dichloro-0

benzene (541-73-1)

X

I.D. BER from of F 1) OU LNU **CONTINUED FROM PAGE V-6** WAD000756940 0.01 Approval expires 12-31-85 1. POLLUTANT 2, MARK 'X' 3. EFFLUENT 4. UNITS 5. INTAKE (optional) AND CAS ATEST D. ME C. MEING LIEVEL LIEVEL
QUIM SENT SENT b. MAXIMUM 30 DAY VALUE | C.LONG TERM AVRG. VALUE (if available) A. LONG TERM AVERAGE VALUE a. MAXIMUM DAILY VALUE d NO.OF NUMBER b. NO. OF a. CONCEN-ANAL b. MASS ANAL CONCINTUATION TRATION (If available) (1) (I) CONCEN-YSES YSES GC/MS FRACTION - BASE/NEUTRAL COMPOUNDS (continued) 22B. 1.4-Dichloro-2 benzene (106-46-7 Х 0 23B. 3,3'-Dichloro benzidine 2 X (91-94-1) 0 24B. Diethyl Phthalate 2 Χ (84-66-2)0 26B. Dimethyl Phthalate 2 Χ 0 (131-11-3)26B, DI-N-Butyl Phthalate 2 Х (84-74-2)0 27B. 2,4-Dinitro-2 toluene (121-14-2) Х 0 28B. 2.6-Dinitrotoluene (606-20-2) 2 Χ 0 29B. Di-N-Octyl Phthalate 2 Х 0 (117-84-0)30B. 1,2-Diphenylhydrazine (as Azo-2 Χ 0 benzene) (122-66-7 31B. Fluoranthene 2 (206-44-0) Χ 0 32B. Fluorene (86-73-7) 2 X 0 33B. Hexachlorobenzene (118-74-1) Χ 0 34B. Hexachlorobutadiene 2 X 0 (87-68-3)35B. Hexachlorocyclopentadiene Х 0 (77-47-4)36B. Hexachloroethane (67-72-1) 2 Х 0 37B. Indeno (1,2,3-cd) Pyrene 2 Х 0 (193-39-5) 38B. Isophorone (78.59.1)щ 2 Χ 0 Σ 39B. Naphthalene  $\omega$ 2 (91-20-3)0 000 40B. Nitrobenzene X (98.95.3)0 0 9 41B. N-Nitro-0 sodimethylamine 0 0 (62-75-9) 42B, N-Nitrosodi-2 N-Propylamine 0 Χ (621-64-7)

		MAHR 'R'				1	1 1 1 1 1 1 1 1 1 1 1				4. UNITS		5. INT	5. INTAKE (optional)	
AND CAS NUMBER	8 11 81	D	(. w	B. MAXIMUM L	AILY VALUE	b. MAXIMIM a	ODAY VALUE	6.1.0NG 13.JIM	Tables. VALUE	d NO.01	a, CONCEN-			. II HM	b NO.01
(if available)	HU-	FMA	AM-	6. MAXIMUM L	(1) MALS	1 7.1	(2) MASS.	(1)	(/) MAUL	ANAL. YSES	TRATION	b, MASS	(1) CONCEN-	(/) MANN	YSE5
GC/MS FRACTION	L RA	SE/NEI	ITRAI	COMPOUNDS	continued)	Leones librations	The second secon	T GREEN HALLON					1001000		I
43B. N-Nitro- sodiphenylamine										2					0
(86-30-6)	ļ		X							. 2					
44B. Phenanthrene (85-01-8)			х							2					0
45B. Pyrene (129-00-0)			Х							2					0
46B. 1,2,4 - Tri- chlorobenzene (120-82-1)			Х							2					0
GC/MS FRACTION	- PES	TICID	ES												
1P. Aldrin (309-00-2)			Х						-	2					0
2P, α-BHC (319-84-6)			Х		•					2					0
3P. β-BHC (319-85-7)			Х							2					0
4P. γ-BHC (58-89-9)			Х							2					0
5P. δ·BHC (319-86-8)			Х							2					0
6P. Chlordane (57-74-9)			Х							2					0
7P. 4,4'-DDT (50 29-3)			Х							2					0
8P. 4,4'-DDE (72-55-9)			Х		, and the second annual annual and the second and the second and the second annual ann		W W W I			2					0
9P. 4,4'-DDD (72-54-8)			Х							2					0
10P. Dieldrin (60-57-1)			Х							2					0
11P. <i>a</i> -Endosulfan (115-29-7)			Х							2					0
12P. β·Endosulfan (115-29-7)			Х							2	er ye inga sa panganda manananananananan da panganda da	er first studiet ritte der tilstellerstamming vivour er remande	en demokratiskop i removember versi i kingeren i		0
13P. Endosulfan Sulfate (1031-07-8)	₹ ₹	:	Х		A 40		<del>.</del>	THE REST OF STREET, AND RESIDENCE	> >	2					0
14P. Endrin (72-20-8)	3000		Х							2					0
15P. Endrin Aldehyde (7421-93-4)	0090		Х			THE CASE OF A CONTRACTOR AND A CONTRACTO				2					0
16P. Heptachlor (76-44-8)	σ		Х						resonancialistic custinistic are the resonancialistic relationship	2					0

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AND CAS NUMBER	ATEST	D. BE.	C BE.	8. MAXIMUM E	DAILY VALUE	b. MAXIMUM 3	ODAY VALUE	c.LONG TERM	AVRG. VALUE	d NO.OF	a. CONCEN-	b. MASS	8. LONG	TERM E VALUE	b. NO.C
(if available)	auia.	SENT	SENT	(1)	(2) MASS	CUNCL NTHATION	(2) MASS	CONCENTHATION	(2) MASS	YSES	TRATION	U. MA33	(I) CONCEN-	(2) MASS	YSES
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17P, Heptachlor Epoxide (1024-57-3)			Х							2					
18P. PCB-1242 (53469-21-9)			Х							2					
19P. PCB-1254 (11097 69 1)			Х							2				. , -	
20P, PCB-1221 (11104-28-2)			х							2					
21P, PCB-1232 (11141-16-5)			Х							2					
22P. PCB-1248 (12672-29-6)			х							2					
23P, PCB-1260 (11096-82-5)			х							2					
24P, PCB-1016 (12674-11-2)			Х							2					
25P, Toxaphene (8001-35-2)			Х	-						2					

EPA Form 3510-2C (Rev. 4-84)

PAGE V-9

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Please print or type in the unshaded areas only

Form

2F

NPDES

WAD000756940

United States Environmental Protection Agency Washington, DC 20460

### Application for Permit To Discharge Stormwater Discharges Associated with Industrial Activity

Paperwork Reduction Act Notice

Public reporting burden for this application is estimated to average 28.6 hours per application, including time for reviewing instructions, searching existing data sources, gathering and maintaining the data needed, and completing and reviewing the collection of information. Send comments regarding the burden estimate, any chairmaning the data needed, and completing and reviewing the collection of information. Send comments regarding the burden estimate, any other aspect of this collection of information, or suggestions for improving this form, including suggestions which may increase or reduce this burden to: Chief, Information Policy Branch, PM-223, U.S. Environmental Protection Agency, 401 M St., SW, Washington, DC 20460, or Director, Office of Information and Regulatory Affairs, Office of Management and Budget, Washington, DC 20503.

1. Outfall Location   For seen outfall, list the latitude and longitude of its location to the nearest 15 seconds and the name of the receiving water.	Washington: BC 20000.									
A. Outfall Number  (list)  B. Latitude  C. Longitude  C. Longitude  (name)  O01  47  22  11  120  08  16  Columbia River  II. Improvements  A. Are you now required by any Federal. State, or local authority to meet any implementation schedule for the construction, upgrading or operation of wastewater treatment equipment or practices or any other environmental programs which may affect the discharges described in this application? This includes, but is not limited to, permit conditions, administrative or enforcement compliance schedule letters, stipulations, court orders, and grant or loan conditions.  1. Identification of Conditions.  2. Affected Outfalls  D. Receiving Water  (name)  Columbia River  D. Receiving Water  (name)  (name)	I. Outfall Location		•			2	( Jr			
(list)  8. Latitude  C. Longitude  (name)  001  47  22  11  120  08  16  Columbia River  II. Improvements  A. Are you now required by any Federal, State, or local authority to meet any implementation schedule for the construction, upgrading or operation of wastewater treatment equipment or practices or any other environmental programs which may affect the discharges described in this application? This includes, but is not limited to, permit conditions, administrative or enforcement orders, enforcement compliance schedule letters, stipulations, court orders, and grant or loan conditions.  1. Identification of Conditions.  2. Affected Outfalls  Compliance Cate	For each outfall, list the la	atitude and	longitude o	f its locat	ion to the	nearest 1	5 seconds a	and the name of the receiving water.		
II. Improvements  A. Are you now required by any Federal. State, or local authority to meet any implementation schedule for the construction, upgrading or operation of wastewater treatment equipment or practices or any other environmental programs which may affect the discharges described in this application? This includes, but is not limited to, permit conditions, administrative or enforcement orders, enforcement compliance schedule letters, stipulations, court orders, and grant or loan conditions.  1. Identification of Conditions.  2. Affected Outfalls  Compliance Cate	A. Outfall Number							D. Receiving Wa	ter	
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	1 Identification of Conc	atatu u u	_	\ \#aata4	- O Halla				100	(0.000000)
Agreements, Etc.   number   Source of discharge   J. Brief Description of Project   a. reg.   b. proj.							1	C Dated Description of Brainst		
	Ağlesinenis, c.c.	·	number	Sourc	ze or discharge			3. Biter Describtion of Project	a. req.	, b. proj.
			++				<del> </del>	SC - 0000 - The substitution of the substituti		<del> </del>
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You may attach additional sheets describing any additional water pollution (or other environmental projects which may affect your discharges) you now have under way or which you plan. Indicate whether each program is now under way or planned, and indicate your actual or planned schedules for construction.

#### III. Site Drainage Map

Attach a site map showing topography (or indicating the outline of drainage areas served by the outfall(s) covered in the application if a topographic map is unavailable) depicting the facility including; each of its intake and discharge structures; the drainage area of each storm water outfall; paved areas and buildings within the drainage area of each storm water outfall, each known past or present areas used for outdoor storage or disposal of significant materials, each existing structural control measure to reduce pollutants in storm water runoff, materials loading and access areas, areas where pesticides, herbicides, soil conditioners and fertilizers are applied; each of its hazardous waste treatment, storage or disposal units (including each area not required to have a RCRA permit which is used for accumulating hazardous waste under 40 CFR 262.34); each well where fluids from the facility are injected underground; springs, and other surface water bodies which receive storm water discharges from the facility. receive storm water discharges from the facility.

V. Narrative Desc	ription of Pollu		s) of impervious	us surfaces (including paved ar	
Outfall Area of Impe	rvious Surface	Total Area Drained	Outfall Number	Area of Impervious Surface (provide units)	Total Area Drained
	Sq. Ft.	21,700 Sq.			
a manner to allow employed, in the	v exposure to stor last three years, t	m water; method of treatn o minimize contact by the	nent, storage, o se materials wi	n the past three years have bee or disposal; past and present m th storm water runoff; materials ditioners, and fertilizers are app	naterials management practice: loading and access areas; and
or no surface runoff	uartz rock, coal	River from this area.		rials storage area as noted or the area noted on the Site Pla	
There is no surface Fume (baghouse du	runoff to the Columbia sty slurry ponds	umbia River from this are and piles of dried fume ( from this area.	ea. uncovered) a	re in the areas shown on the	Site Plan. There is no
Putrescible garbage	is disposed at le	ocal landfill, not in this a tioners or fertilizers have	rea. There is e been applied	lid waste disposal area show no surface runoff to the Colu d to these area. One catch be or no surface runoff discharg	asin in Area PR may be
storm water runo	ff; and a description		rm water receiv	al and nonstructural control mees, including the schedule and so other than by discharge.	
Outfall umber		¥ 00	tment		List Codes fro
		ollutants in storm water rater runoff is collected a		rolled mainly by limiting area d into outfall.	s of
nonstormwater di	enalty of law th	at the outfall(s) covered at all nonstormwater discha		cation have been tested or e se outfail(s) are identified in eith	
ame and Official Title (I		Signature			Date Signed
B. Provide a descrip	tion of the method	used, the date of any testi	ng, and the on	site drainage points that were d	irectly observed during a test.
effluent was collect total phenol. Comp (0.017 ppm versus	ed November 21 ared to other effi <0.003 ppm) and	, 1988 during a storm. T luent samples in August I more zinc (0.011 ppm v	his sample w 1988, the No	ount of storm water runoff. C as tested for priority pollutan vember 21 effluent sample ha opm) than the August dry wea	t metals, total cyanide and ad more chromium
VI. Significant Lea	ks or Spills mation regarding		eaks or spills	of toxic or hazardous pollutants	
years, including the a	pproximate date a	and location of the spin of it	ak, and the ty	pe and amount of material relea	seu.
None					
				12	ē

## EPA ID Number (copy from Item I of Form 1) WAD000756940 Continued from Page 2 VII. Discharge Information A.B.C. & D: See instructions before proceeding. Complete one set of tables for each outfall. Annotate the outfall number in the space provided. Tables VII-A, VII-B, and VII-C are included on separate sheets numbered VII-1 and VII-2. E: Potential discharges not covered by analysis - Is any pollutant listed in Table 2F-2 a substance or a component of a substance which you currently use or manufacture as an intermediate or final product or byproduct? No (go to Section IX) Yes (list all such pollutants below) Chlorine -Oil and grease Sulfate Aluminum Iron Tin titanium 11 7 12 34 14 15 15 VIII. Biological Toxicity Testing Data Do you have any knowledge or reason to believe that any biological test for acute or chronic toxicity has been made on any of your discharges or on a receiving water in relation to your discharge within the last 3 years? Yes (list all such pollutants below) X No (go to Section IX) IX. Contract Analysis Information Were any of the analyses reported in Item V performed by a contract laboratory or consulting firm? V Vac fliet the name address and telephone number of and collutants No fon to Section Y

A, Name	B. Address	C. Area Code & Phone No.	D. Pollutants Analyzed
Lauchs Testing Laboratories, Inc.	940 South Harney St. Seattle, WA 98108	(206)767-5060	Priority Polutant and other hazardou substances.
AM Test, Inc.	14603 NE 87th St Redmond, WA 98052	(206)885-1664	Priority Pollutar metals/elements, cyanide, phenol

I certify under penalty of law that this document and all attachments were prepared under my direction or supervision in accordance with a system designed to assure that qualified personnel properly gather and evaluate the information submitted. Based on my inquiry of the person or persons who manage the system or those persons directly responsible for gathering the information, the information submitted is, to the best of my knowledge and

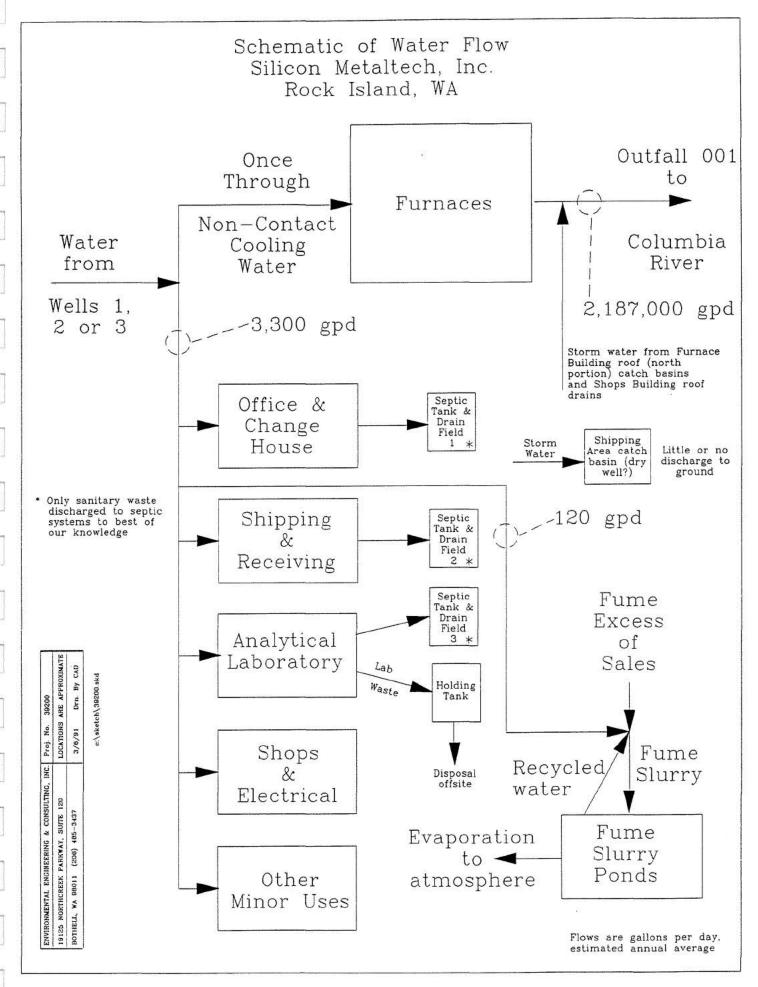
belief, true, accurate, and complete. I am aware that there are significant penalties for submitting false information, including the possibility of fine and imprisonment for knowing violations.

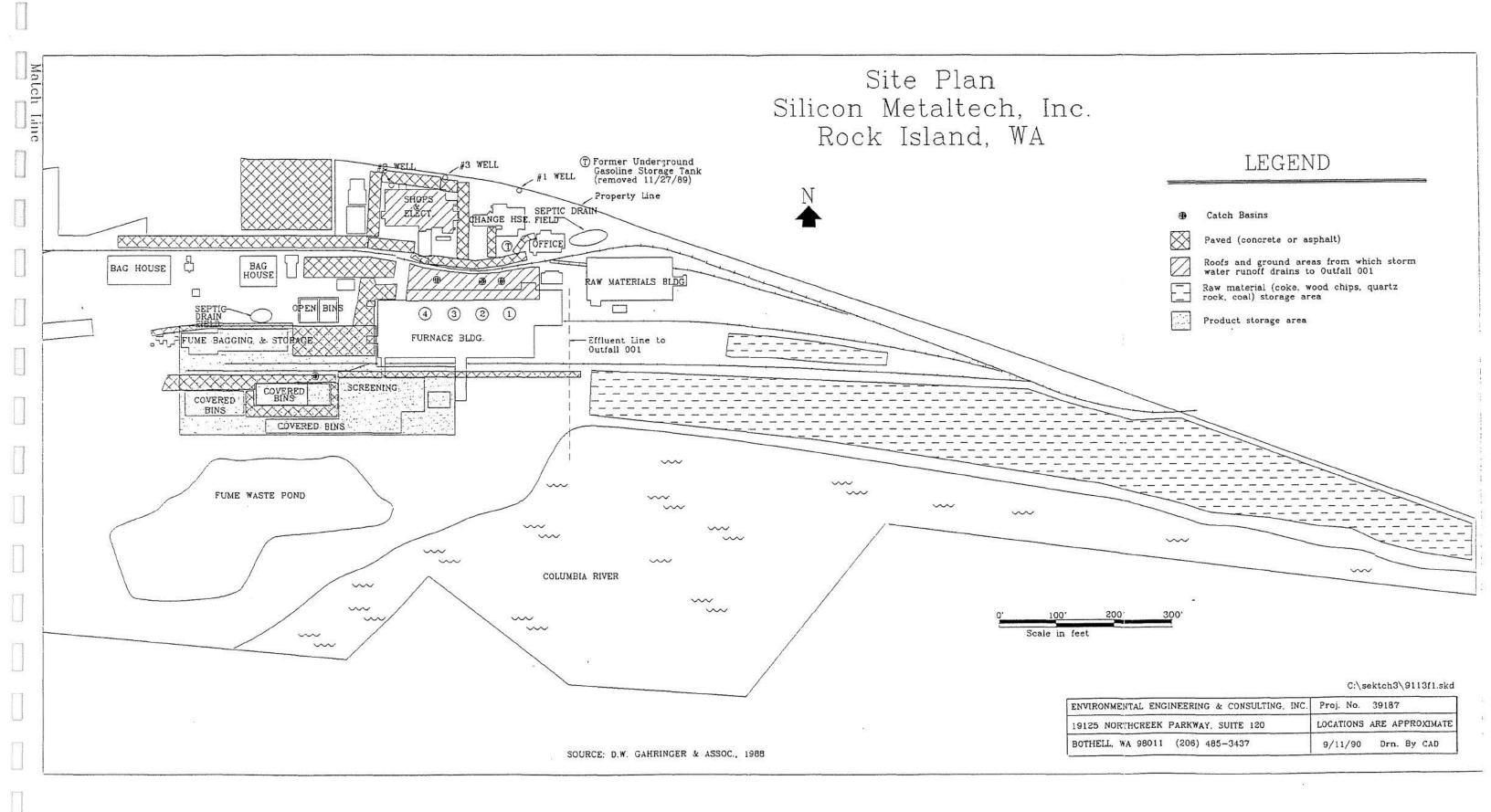
A. Name & Official Title (type or print)	B. Area Code and Phone No.
	€
C. Signature	D. Date Signed

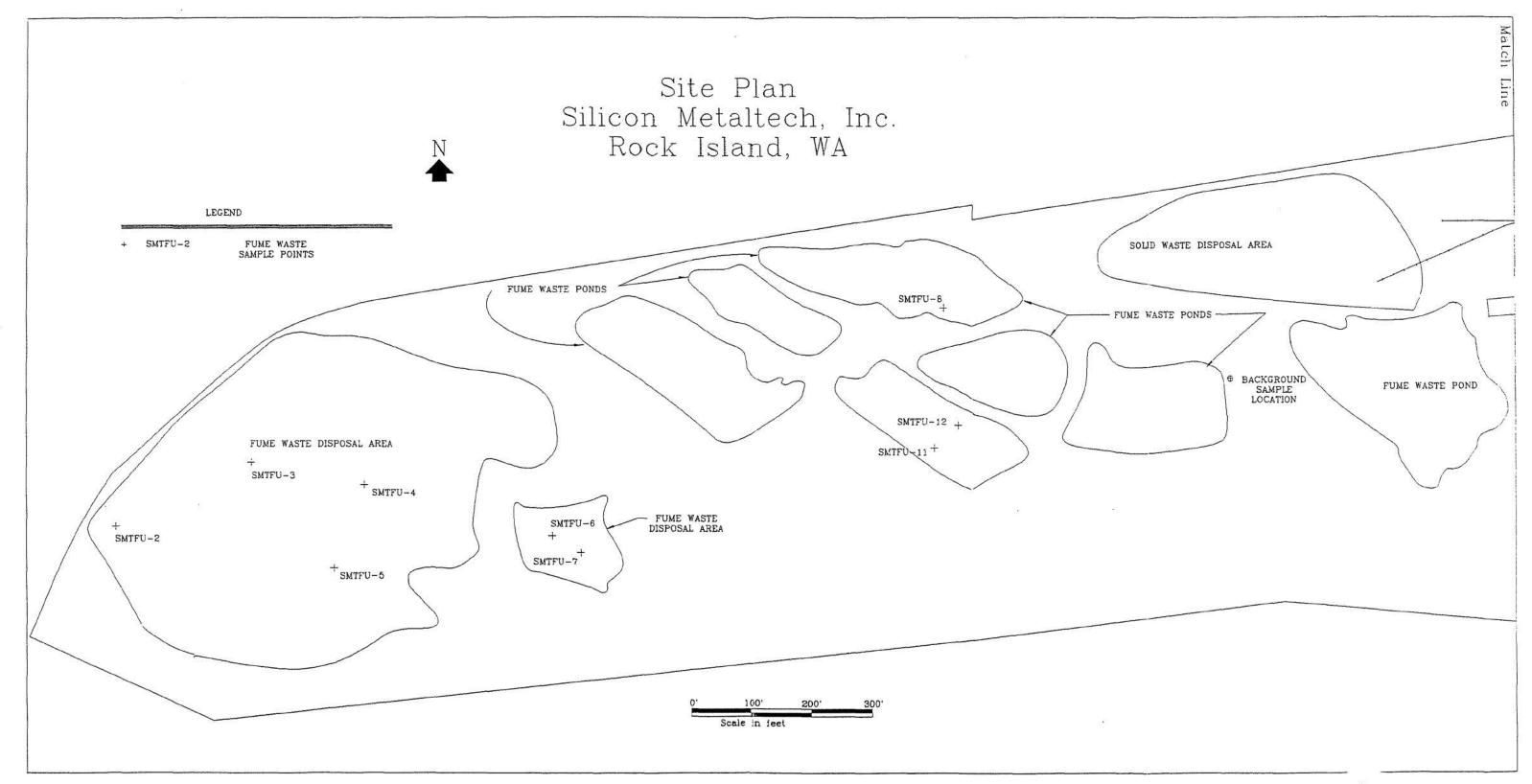
WAD 000756940 VII. Discharge Information (Continued from page 3 of Form 2F) Part A - You must provide the results of at least one analysis for every pollutant in this table. Complete one table for each outfail. See instructions for additional details. Average Values Number Maximum Values (include units) Pollutant (include units) of Grab Sample Grab Sample Storm and Taken During First 30 Taken During CAS Number Flow-weighted Flow-weighted Events First 30 Minutes Composite (if available) Composite Sampled Minutes Sources of Pollutants Oil and Grease Biological Oxygen These parameters not Demand (BCD5) Chemical Oxygen presently monitored, but Demand (CCD) Total Suspended would be willing to sample Solids (TSS) Total Kjeldani and test if required Nitrogen Nitrate plus Nitrite Nitrogen Total Phosphorus Minimum Minimum Maximum Maximum Part 8 -List each pollutant that is limited in an effluent guideline which the facility is subject to or any pollutant listed in the facility's NPDES permit for its process wastewater (if the facility is operating under an existing NPDES permit). Complete one table for each outrail. See the instructions for additional details and requirements. Average Values Maximum Values Number Pollutant (include units) (include units) of Outfall 001 Storm and Grap Sample Grab Sample Taken During Taken During CAS Number Flow-weighted Flow-weighted Events First 30 First 30 (if available) Minutes Composite Minutes Composite Sampledi Sources of Pollutants 7.4 7.8 Intake piping and PH 72.2 F 76 F storm water runoff Temperature The values above are from effluent monitoring and are not first during last five (5)years 30 minute nor flow weighted composite values PW 30000911

	Maxim	um Values	Ave	rage Va	lues	Number			
Pollutant		ıde units)	1	clude un		of			
and CAS Number	Grab Sample Taken During	Flow-weighted	Grab Sample Taken Durin	e	low-weighted	Storm	Ou	tfall	001
(if available)	First 30 Minutes	Composite	First 30 Minutes		Composite	Sampled		Sources o	f Pollutants
- Result:	s of testing Out	fall 001 (non-cor			and small a	amount of	storm	water ru	noff)
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-		Maximur	n Sto	rm					
<del></del>		value							
Polluta	ints CAS Num		•		ource of Po	llutants			
- Arsenio			4	1 In	take, piping	g or storm	water	runoff	
- Chromi	ium 7440–47–	3 0.01	7	1 Int	take, piping	or storm	water	runoff	
- Copper	7440–50–	8 0.08	9	1 Int	take, piping	or storm	water	runoff	
Lead	7439-92-	1 n	d		take, piping	•			
Nickel	7440-02-	0.00	8		ake, piping				
Zinc	7440-66-				ake, piping	•			
•		2.37				,	401	2011011	
Note: T	hese values ar	e for grab sample	es but not fire	st 30 m	ninute sami	oles nor fi	7\#/ \#/ <i>E</i>	inhted co	mnoeitae
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Part D - Provid	de data for the storr	m event(s) which resul	ted in the maxim	num vait	s.	v weighted c	ompos	ite sampie. 7.	8.
ite of Durati		infall Number of	hours between	Mavim	o. num floe rate	Total flow	from	Season	Form
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vent (in minu		ured and e	nd of previous ble rain event		ns/minute or ecify units	(gallons specify ui		taken	(rainfa snowm
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<u>ollected</u>	at 16:30	, 1.5 hours	before (	end o	of above	e recor	cded	storm	s event
9. Provide a de	escription of the me	ethod of flow measure	ment or estimate	9.		<del></del>			
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									30000912

Continued from the Front



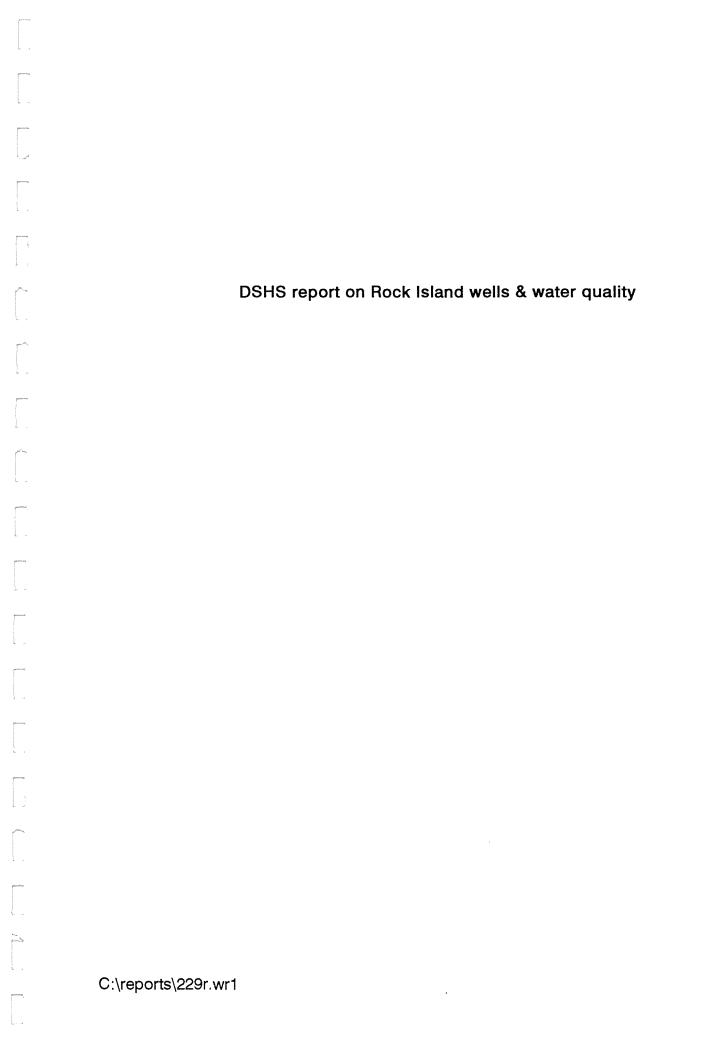




### monitoring period reported

		flow, g	gpd	p	)H	tem	p, F 
Start	End	avg	max	avg	max	avg	max
01-Jun-90	01-Jan-91	1,663,412	2,771,814	7.4	7.7	73	74
01-Jan-90	01-Jul-90	2,207,097	2,987,432	7.5	7.6	73	75
01-Jul-89	01-Jan-90	1,884,368	2,881,056	7.5	7.8	73	75
01-Jan-89	30-Jun-89	1,449,461	1,898,838	7.5	7.7	74	75
30-Jun-88	01-Jan-89	2,511,969	2,685,393	7.45	7.5	75.2	76
01-Jan-88	30-Jun-88	2,701,855	2,875,354	7.5	7.6	74	75
03-Aug-87	01-Jan-88	2,125,688	2,852,709	7.3	7.4	72	73
06-Jan-87	02-Jul-87	2,247,357	2,247,357	7.2	7.4	71	72
03-Aug-86	06-Jan-87	2,422,670	2,498,361	7.2	7.3	69	73
01-Jan-86	02-Jul-86	2,660,761	2,971,500	7.2	7.2	68	72
		_		_			
<u>Maximum</u>			2,987,432		7.8		76.0
	<u>Average</u>	2,187,464		. 7.4		72.2	

period report 5 years





#### STATE OF WASHINGTON

### DEPARTMENT OF SOCIAL AND HEALTH SERVICES

West 924 Sinto Avenue, L32-4 • Spokane, Washington 99201-2595 • (509) 456-3115

April 6, 1989

The Honorable Clyde Ballard Washington State Representative 410 Legislative Building, AS-33 Olympia WA 98504

Re: Water Quality Report and Results for Drinking Water at Rock Island

Dear Representative Ballard:

As the result of reported high levels of metals from the Rock Island wells, a great deal of follow-up sampling has been done by our department and Rock Island with samples being analyzed in the state lab as well as some in certified private labs. Water from the Rock Island system was also checked for pesticides and volatile organic chemicals

All of the re-samples and results to date have not confirmed the first results that were reported from the uncertified lab that did the initial analysis. Recent results given the town by phone indicate no metals were found in the fish samples that were checked.

To stay on top of this issue and hopefully ease some citizen concern, we are recommending Rock Island continue to sample for Arsenic, Lead, Mercury, and Selenium every three months to assure there are no seasonal trends and to take another complete inorganic chemical analysis in February, 1990.

As requested, we have prepared the attached brief report covering our findings and included the sampling results for your information. We appreciate the assistance you provided as well as your patience and understanding as we worked through this difficult situation.

Sincerely,

Tom Justus, P.E. Regional Engineer (509) 456-2453

TJ:vw

cc: Chelan-Douglas Health District
Town of Rock Island



#### STATE OF WASHINGTON

### DEPARTMENT OF SOCIAL AND HEALTH SERVICES

West 924 Sinto Avenue, L32-4 • Spokane, Washington 99201-2595 • (509) 456-3115

April 6, 1989

The Honorable Alex W. McLean Washington State Representative 421 House Office Building, AS-33 Olympia WA 98504

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Town of Rock Island



#### STATE OF WASHINGTON

### DEPARTMENT OF SOCIAL AND HEALTH SERVICES

West 924 Sinto Avenue, L32-4 • Spokane, Washington 99201-2595 • (509) 456-3115

April 6, 1989

The Honorable George L. Sellar Washington State Senator 312 Legislative Building, AS-32 Olympia WA 98504

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Tom Justus, P.E. Regional Engineer (509) 456-2453

TJ:vw

cc: Chelan-Douglas Health District
Town of Rock Island

### ROCK ISLAND, WASHINGTON SUMMARY REPORT MARCH, 1989

#### SUMMARY

The town of Rock Island submitted water samples to National Chem Lab in February for inorganic analyses due to concern over the location and operation of the Douglas County landfill as well as a proposal to dispose of sewage treatment sludge from Wenatchee at airport property near Pangborn Field. Preliminary results from the lab showed high levels of mercury (1.5 parts per million), lead, arsenic, and selenium. Final results reported from the lab showed mercury levels just below the Maximum Contaminant Level (MCL) of 0.002 ppm, but levels of arsenic, cadmium, chromium, iron, and selenium above the MCL. National Chem Lab currently is not certified. The Chelan-Douglas Health Department collected another set of samples and sent them to the DSHS State laboratory. The DSHS lab did not confirm the previous results. Additional water samples were collected from 13 wells and a pond on February 23, 1989 by DSHS personnel and analyzed at the DSHS lab. All results were under state and federal drinking water standards. town of Rock Island sent an additional set of samples to ABC Laboratories and the DSHS Pesticide lab. All these results were also under levels set by state and federal drinking water standards.

### INTRODUCTION

Rock Island is located in central Washington, approximately 20 miles southeast of Wenatchee (Figure 1). The geology of the area mainly consists of glacial, fluvial and volcanic rocks. The most recent materials are glacio-fluvial deposits of the two prominent terraces in the area, and stream deposits. The upper terrace is composed of coarse sand to gravel and boulders. The lower terrace is composed of sand and gravel, and cobbles and boulders of predominantly granitic and metamorphic composition. Basalt of the Columbia River Basalt Group crops out along the north side of the river and in the river. Basalt has been penetrated by wells in the eastern end of the area. The general direction of groundwater flow is southeast.

Washington State University (WSU) collected samples in Rock Island in September 1978 and April 1979 as part of a three year study to investigate water quality in the Rock Island area at the request of the Chelan County PUD. As a result of this study, the Chelan-Douglas Health Department informed

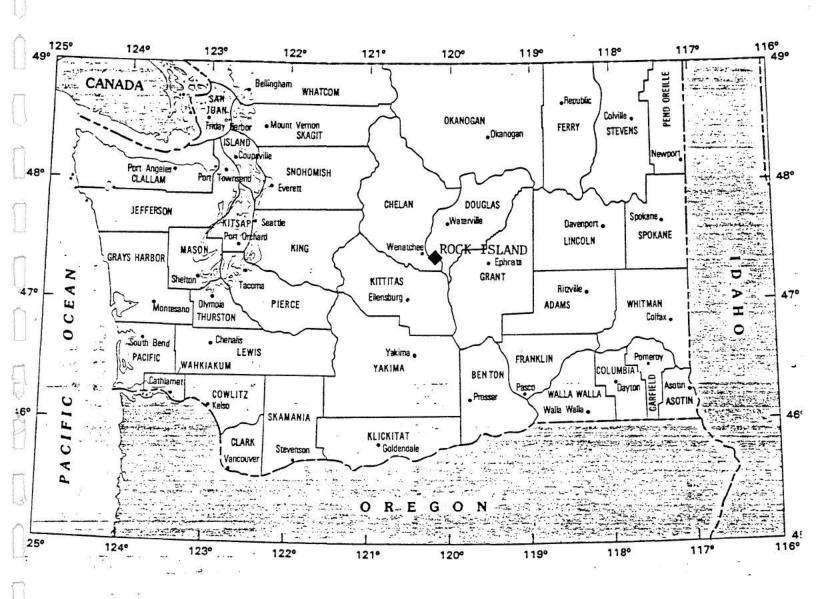


Figure 1. General location map.

DOE in July, 1980, that concentrations of arsenic and selenium had been detected in groundwater and surface waters and that two well water samples exceeded "Washington State Standards". Because the levels of arsenic and selenium posed possible environmental problems and health hazards, both DOE and DSHS became involved. Samples were collected on August 8, 1980 by DOE and WSU personnel. One surface water and eleven well samples were collected in and around Rock Island. In addition, two samples were collected from Hanna Mining Company (now Silicon Metaltech) baghouse dust disposal system, one from a lagoon, and one from the recycled slurry. The samples were split three ways for analysis by DOE, WSU, and DSHS. Soil samples were also collected and analyzed by the DOE laboratory. Comparative analysis between DOE, WSU, and DSHS of water samples showed fairly close agreement. The results showed no concentrations of arsenic above the recommended drinking water levels. Two wells contained concentrations of selenium just above the drinking water standards (by DOE analysis only). Since these results were not confirmed by the other two labs, no resamples were taken. Arsenic concentrations detected in the soil samples were not above levels that have been observed in soils from orchards where arsenic was used as a pesticide. Concentrations of arsenic, selenium, lead, cadmium and mercury in the Hanna Lagoon and Hanna Lagoon Influent samples were all high.

As a result of the high levels detected in the lagoons, two observation wells were established around the lagoons and samples were collected periodically in 1982. Hanna also monitored a control well for the same period and agreed to do optional sampling of the Columbia River. Nothing was detected in any of these samples.

The town of Rock Island has been conducting routine monitoring of the city wells as required under the Safe Drinking Water Act. All samples have been within set drinking water standards. However, concern over the proposal to dispose of sludge near Pangborn Field prompted the town to submit samples to an uncertified lab that subsequently reported high levels of arsenic, cadmium, chromium, iron, and selenium in city wells. DSHS and the Chelan-Douglas Health Department became involved after these results were reported.

### SAMPLING METHODOLOGY

Samples were collected February 23, 1989 in the Rock Island area by DSHS personnel (Figure 2). Sampling procedures were conducted as described in the Hazardous Waste Drinking Water Program (HWDWP) Field Protocol. All samples were analyzed for arsenic, lead, mercury, and selenium at the DSHS laboratory.

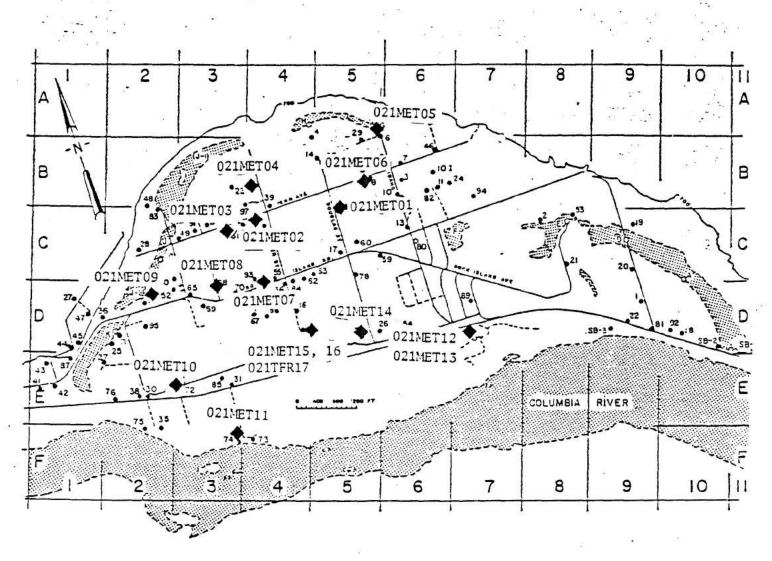


Figure 2. DSHS sample locations.

→ - well site

#### RESULTS AND DISCUSSION

Final results from National Chem Lab for samples collected in early February are as follows (in ppm):

	<u>Well #2</u>	<u>Well #3</u>	Town Hall	MCL
Arsenic		0.187		0.05
Cadmium	0.075	0.038	0.051	0.01
Chromium	0.981	<0.001	0.086	0.05
Iron	0.066	7.02	0.080	0.30
Lead	0.271	0.268	0.266	0.05
Mercury	0.0019	0.0019	0.0016	0.002
Selenium	0.602	1.015	0.425	0.01

Results from the DSHS State Lab for samples collected by the Chelan-Douglas Health Department February 10 are as follows (in ppm):

	<u>Well #2</u>	<u>Well #3</u>	MCL
Arsenic	<0.010	0.021	0.05
Cadmium	<0.002	<0.002	0.01
Chromium	<0.010	<0.010	0.05
Lead	<0.002	<0.010	0.05
Mercury	0.0028*	0.0005	0.002
Selenium	<0.005	<0.005	0.01

\*Note: These samples were collected in non acid-rinsed cubitainers. Mercury is a fairly common contaminant. Whenever mercury is positive, a resample is collected in an acid rinsed container.

Results from the DSHS State Lab for samples collected February 23 indicated all samples were less than the detection limit for mercury (0.0005 ppm), lead (0.002 ppm), and selenium (0.005 ppm). Arsenic was detected in five wells and in the surface sample, but all were below the MCL of 0.05 ppm. All samples were collected in acid rinsed containers.

Arsenic (ppm)
0.017
0.014
0.036
0.024
0.014
0.017

The town of Rock Island collected samples from wells #2 and 3 on February 17. These samples were sent to ABC Laboratories and complete inorganic chemical analyses were performed. All results were below the MCL, except iron and turbidity in well #3. Some arsenic was detected in well #3 (0.016 ppm - below the MCL). Wells #2 and 3 were sampled again on February 21 and sent to ABC Laboratories for a Volatile Organic Scan and to the DSHS Pesticide lab for pesticide scans. Nothing was detected in any of these samples.

Copies of all laboratory results are in Appendix 1.

#### CONCLUSIONS AND RECOMMENDATIONS

Samples collected by Health Department and DSHS personnel and analyzed at the DSHS State lab indicted no health concerns with the drinking water at this time. Although one well showed mercury levels just above the MCL, this was most likely due to the type of container the sample was collected in. A subsequent sample indicted mercury levels below the detection limit.

It is recommended that the town of Rock Island monitor for arsenic, lead, mercury, and selenium every three months for the next year and resample for complete inorganics in February, 1990 to assure residents that the drinking water meets drinking water standards.

APPENDIX 1

### \*\*\* NATIONAL CHEM LAR TOST REPORT \*\*\*

(City of Rock Island ) February 10, 1989 P.O. Ser 99

+ ... 3

5 M. Garden

Roc. Island, WA 99850 [Attn: Robert Knight Jr.

[Invoice #1053

Purchase Order #8686

NCL #031.001

Sample I.D.: WATER

	· · · · · · · · · · · · · · · · · · ·				
	Analysis Sequested	•••	Band's	R MCL	
•					
	Coliform	2.2	MF(0:100 ml	:212	
	Nitrale Nitrogen	1.5	m <u>ç</u> ./⊑	10.0	
	Total Phosphats	.545	mg/L	•	
	Arsenic	0.187		0.05	
	Cadmium	0.038	mg/L	0.01	
	Calcium	56.1	mg/ <b>'</b> _		
	Chromium	<0.001	mg/L	o.⊙₹	
	Inda	7.02		O.3	
	Lead	0.242		0.05	
	Manganese	10.001		©_0 <del>5</del>	
•	Mercury	0.0019	•	0.002	
	Selesius	1.015		0.01	
	211 ven	0.001	•	0.05	
•	Epaium	24.4			
	Catter	0.199		1 5	
•	Zina	.6.601		5.4	

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Analyst Signature 1000 S. M. M. Co. E

### MAILUMAL LIBERT I SE Water Test Report

Date: 02/14/89

Time: 11:55

rage: 1 of 1

NCL#: WE9021302

INVOICE #W9BD02

CITY OF ROCK ISLAND SAMPLE #: YOWN HALL

LOT #:

PO #: 858a COMMENT:

DATE RECVD: 02/09/89 TIME RECVD: 10:00 RECVD BY: MICKEY HUNACEK

TRANSPORTATION MODE: COSTUMER DECIVER

SAMPLE SOURCE: TOWN HALL

'SAMPLE SIZE: 1 DUAR!

COMPAINER TYPE: bease

DELIVERED BY: MR. WHITIC

TEST	MOL	REBULTS	DATE TIME ANALYZED ANALYZE	D CHEMIST
COLIFORM BACTERIA			92/13/83	HICKEY HUNACE
NITRATE (AD)			02/13/87	
			02/13/89	
		(.001 eg/L	02/13/89	MICKEY HUNACE
CADHIUM		0.051 ag/L	02/13/89	MICKEY HUNACE
SALOTIM		52.9 mg/L	02/13/89	MICKEY HUNACE
MERCURI	0.001	0.0018 ag/L	02/13/89	MICKE: HUNACE
SILVEF		K.001 aq/L	92, 13, 89	HICKEY HUNACE
CHROMIUM		0.086 ag/L	02/13/89	MICKE'S HUNACE
LEAC		0.25a sa/L	02/13/89	MICKET HUANCE
IRON		0.08v ag/L	02/13/69	MICKEY HUNACE
MANGHNESE		(.001 mg/L	02/13/89	MICKEY HUNACE
SELEHIUM		0.425 eg/L	02/13/89	MICKEY HUNACE
SODIUM		21.3 mo/L	JZ/13/89	MICKEY HUNGCE
COPPER		C. Oul more	02/13/89	MICKE: HUNACE
NICKEL		1.24 80/1	92/15/89	MICKE, HUNACE

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PW 30002682

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USE HEAVY PENCIL

DO NOT WRITE IN SHADED AREAS

#828 P01

Étate of Weshington of at Social and Health Services Dinaion of Health

PUBLIC HEALTH LABORATORIES 1910 N.E. 160th St. . Seettle WA 98155

SEE BACK FOR INSTRUCTIONS

### WATER SAMPLE INFORMATION FOR INCREANIC CHEMICAL ANALYSES USE THIS FORM FOR THE COMPLETE CHEMICAL ANALYSIS ONLY

SYSTEM I.O. NO.	SYSTEM NAME	3.89 Patt	Carter 5-5475 STEM CLASS CIO ONE) 1 2 3 4 Douglas
SOURCE TYPE	SOURCE NO! IF SOURCE IS LAKE OR STREAM ENTER NAME (WHI NO!)  IF SAMPLE WAS CRAWN FROM DISTRIBUTION SYSTEM	FEES ARE CHARGED FOR A fee achedule is available	SCHOOL SAND CONTRACT OF CONTRACT SHAPE
Before Treatment After Treatment  TAKEN AFTER TREATMENT	WAS ITFILTEREDFLUCRIDATED	PARTY TO PAY FOR F	FEE FOR SERVICE TESTING  (Print Pall Name & Address)
	roblems, address for additional copies, etc.)	Haz. Wa Office of Envir West 924 Sinto	ronmental Health Programs
		Spokane	Street WA. 99201-2595 Zip Code
	LABORATORY	Telephone: ( Scan	545-2475

					TORY REPOR	
TESTS	THEL	HAAN	RESULTS	UNITS	Compliance YES / NO	CHEMIS
rsenic 4	0.06	40	0.010	) mg/l		PO
Parium =	1.0 7	<u>                                    </u>		mg/l		
admium •	0.01*			mg/l		
Chromium >	0.05	<u>   _</u>	_!	mg/l		
on •	0.3	<u>  _</u>		l mg/l		<u> </u>
Lead •	0.05	20	000	2   mg/l	1	10
langanese -	0.05	<u> </u>	_•	mg/l		
Mercury *	0.002	4.	000	5 190/1	4	ne
alenium =	0.01	$\leq$	<u> 2.00.</u>	5 mg/i	/	1-24
ulver 4	0.05*			mg/1		
codium -		<u>   _</u>		mg/l		
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reidity	1.0*			! NTU		
Color	16.0	! _		Cotor Units	a Sun a soorman	
uoride •	2.0 "	!_		mg/l		
intrate **	10.0	<u>   _</u>		mg/l		
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LABORATORY SUPERVISOR

\$ 79,00 TO HAZ. WASTE ACCOUNT

REMARKS:

PW 30002684

ara, i randa "Leidur taad water lest medont

- Date: 02/14/89

Time: 11:55

rade: 1 of (

MCLH: WEROZIACI -

INVOICE #W9BE01

CIT: OF ROCK ISLAND SAMPLE #: #2 WELL

LOT #: PO #: 8686 COMMENT:

DATE RECVD: 02/09/89 TIME RECVD: 10:00 RECVD BY: MICKEY HUNACEK

\* TRANSFORTATION MODE: CUSTOMER DELIVER

SAMPLE SOURCE: #2 WELL

SAMPLE SIZE: 1 DUART CONTAINER TYPE: GLASS

DELIVERED BY: MR. WHITIC

TEST	HÜL	RESULTS	DATE TIME ANALYZED AMALYZED	CHEMIST	
COLIFORM BACTERIA		2.2 HPM/100mi	02/14/89	Micaey HUNACEX	١
MITPATE (AD)		2.9 mg/L	02/14/89	MICKEY HUNACEK	:
PHOSPHATE TOTAL (AB)		0.145 ag/L	92/14/89	MICKEY HUNACEK	
EINC		0.056 <b>2</b> 9/L	v2/14/5°	MICKEY HUNACER	
Cabalea	9,910	0.075 ag/L	02/14/ <b>3</b> 9	MICKE! HUMACEK	
CALSIUM		57.7 mg/L	92/14/89	MICKEY HUNACEX	
MERCUPY	0.003	0.0017 49/1	02/14/89	MICKEY HUMACEK	
SILVEE	0.050	7.001 *g/L	02/14/85	MICKEY HUNACEK	
CH994164	0.050	0.99: <b>10</b> /L	02/14/89	MICKE: HUMACEX	
LESS	9.950	9.271 20/L	92/14/89	MICKEY HUNGER	
IRON	0.300	მ.შგა ∰შ∕ი	92/14/69	MICKEY HUNAGEK	
MANGANESE	0.050	(,001 ag/L	02/14/89	MICKE: HUNACEK	
SELENIUM	0.010	0.602 mg/L	02/14/69	MICKEY HUNACEL	
SODIUM		25.9 ag/L	02/14/89	MICKE'S HUNACEK	
COPPER		.0la mg/L	92/14/89	HICKEY HUNACEK	
NICKEL		1.3a mg/L	02/14/89	MICKEY MUNACER	

Wieling Hunard

#### FOR INCREAMIC CHEMICAL EEEYJAKA

telt "T	MX	124	U	SE THIS	FORM FOR	THE CO	MPLE	TE C	HEMICAL	ANALYSIS ONLY	
J.		*1,5				W. 7 12 75	3, <u>2</u> 3	920	9 COLLECTED	Patt: Carter Scan 545-2475	
$\bigcap$	373	TEM 1.0. M	0.		SYSTEM NAME	MET	01	,		8YSTEM ILASS (curcio see) 1 2 3 4	Douglas
		• =		(Wet No.)	IF SOURCE IS LAKE OR	STREAM ENTER N.	AME	300000		SED FOR CHEMICAL TEST	
-	Before	WAS TAKES Tresiment Tresiment		I IF SAMPLE	was drawn from Diath Lected from System A	BUTION SYSTEM T: (Address)		] '	PARTY TO PA	Y FOR FEE FOR SERVIC	E TESTING
	CHLO	RINATED		WATER SCI	FILTERED FTENER: TYPE USED	)		] :		ste Program	Print Full Name & Address)
REMAI	RKS:	(Water q	quality p	problems, ac	dress for additions	d copies, etc.)	01		Office of	Environmental H	ealth Programs
						(i)			Spokane,	Şireel	9201-2595
					V				City Telephone: (	Scan 545	-2475
						- LABORAT	TORY	REPO	RT	Ares Code	
			LESS			(DO NOT WA	TE BELOW		NE) CHEMIST	7 001 TO 12 SHILLS	AND THE RESERVE OF THE
TEST	8	MCL	HAN	100	RESULTS	UNITS	YES	Y 40	IMITIALS		
rsenic		0.05	-	0.	0 10	j mg/1			PO		
านกา	-	1.0 '	-			mg/l	1			LARCRATOR	Y SUPERVISOR
admium	<b>a</b>	0.01	-	•-		mg/l				Otame o	e (netiate)
romium		0.05	-	<del> •</del>		mg/1		1	-	1.7-6	cens
-011 -010	-	0.05	2	0.	002	mg/l			PO	CHARGE: \$ 79	20125 200 To
nganes	. *	0.05		1		mg/l				مداه	E. WAST ACCT
Yelchia	5	0.002	1	. 0	005	mg/1	V		750	REMARKS: MTZ	
enium		0.01	Z	0.	005	mg/l			1		
Cilver	-	0.06				mg/l					
dium	•					mg/l					
ardness						Mg/I AS CaCoS					
nductiv	ity	700				Micromnos/cm 25° C					
urbidity		1.0*			•	NTU					
olor		15.0	<u> </u>	<u> </u>		Color Units					
oride	•	2.0 '	<u> </u>		•	mg/l					
htrats	n#	10.0				mg/l					
oride		250				mg/l					
Oulfate	ж.	250				mg/l					
<u>s</u>		500				mq/l		!		PW	30002686

CL s the Maximum Contaminant Level Allowed \* Primery Standard

25HS 4-92F (Rev. 7/85) -873- DSHS Laboratory & Data Processing Copy DSHS Silling Copy Water Supplier Copy District Engineer Copy Local Health Dept. Copy

1.0

5.0

Copper

mg/i mg/1

### Division of Health DO NOT WRITE IN SHADED AREAS | PUBLIC HEALTH LABORATORIES | 1610 N.E. IDON St., Seattle WA 38166

SEE BACK FOR INSTRUCTIONS

## WATER SAMPLE INFORMATION FOR INCREAMIC CHEMICAL ANALYSES

			US	E THI	SFO	RM FOR		~		1 =	CHEMIC							
								23.	25	3.8		• \$C	Patt an 545					
	25 10 E	( 1.D. m).				021	ME			2			SYSTEM (circle on	•)	4	Dond	glas	
SOURCE			10	Well Ho.)	D. # 30U	RCE IS LAKE OR	STREAM EN	TER NAM	E	3	ES ARE CH							51
THIS SAM	APLE WAS	4. PW	1 1	F SAMPLE	NAR ZRAN	WN FRCM DISTR	BUTION BYST	EM	-	┤^	fee schedul							
	Betore Tre		. [	T WAS CO	LLECTEC A	ACM SYSTEM A	T: (Address)				PARTY TO	PAY F	OR FEE A	OR SE	RVICE	TESTING	3	
	After Tree		1-							1			500					
						TYPE USED		TED				enure (Re			(P	ring Full House	4 Address)	
			_			or additions		etc.)		1	Hazard.				1 11-	-116-7		
									3		Office				пне	a itn i	rogra	ms
	1000 m		Ĵ-	4.55							West 92	4 51	nto Ave	anue siree	,			
-								-			Spokane	·,		WA.	99	201-2	595	
						<del></del>					City		-				Zip Code	
											Telephone:	(	Scan	ا د	545-	2475		
							- LABO	RATO	RY F	REPO	RT -		Area Code					MAN.
			ERR .				i i	1	Compli		CHEMIST	<del>-</del> 1	P#37	# 10 m		47.75 -57		-5000
TESTS	<del>-</del>		RAN		RESUL	r8	UNITS		YES	WO.	INITIALS							
rsenic	-)	0.06	<del>-</del>	•		<u> </u>	mg/I		V		PC	4		F.			Carrier.	
um	•	1.0	_¦-		<u>•</u> -		mg/l				-	_	\$1150 MARCH   15					= <del>,==,</del> ;
admium	×	0.01		•			mg/l			_		_	LAB	CRAT	CRY	SUPE	RVISOF	7
minuc	۵	0.05		•			mg/I		[			_		1		rein		
5/1	•	0.3	_				mg/l		_/			_1	CHARG	مرايك			4	
ecd	•	0.05	$\leq 1$	<u></u>	<u> </u>	<u>ಲ                                    </u>	mg/l		/		PC	2	CHANG	ب	17	$\omega_{\mu}$	$\omega$	Λ.
ganese	-	0.05	_	•			mg/l						REMARKS	:	4	a.u	ts Jesti	icot
ercury	=	0.002	4.	<u> </u>	0	05	mg/l		ノ		1-201		V CE	4)	(	0		
nium	-1	0.01	<u>S</u> .	<u>o.</u>	<u>O C</u>	25	mg/l	1			100	<del>L'</del>	-, ~	- •				
lver	4	0.05	_	•			mg/1											
um							mg/l											
ardness				N EIII			mg/I AS CaCo	23										No.
uctivity	70	00 _	_				Micromhos 26° C			) 								
roidity		1.0	] -			_•	NTU					j						
بمار		16.0	_			_•	Color Unite											
ide -		2.0					mg/l											
trate •	-	10.0*				_•	mg/1											
ide	11 21	50					mg/l		Ť									
-	so, 21	50					mg/l	$\neg$										
		00					mg/l					7						
Oper	<b>C4</b>	1.0					mg/I		Ť			7	923		PW	300	02687	7
parties.	2n	8.0		0/10			mg/l	T			1	7	n					
											1	acced?						

6 the Maximum Contaminant Lavel Allowed Primary Standard
3H3 4-92F (Rev. 7/86) -873- DSMS Laboratory & Data Processing Copy DSMS Billing Copy Water Supplier Copy District Engineer Copy Local Health Dapt. Copy

1.0 \*

0.01

0.06\*

0.3 0.05

0.05

0.002

0.01

0.05

700

1.0

15.0 2.0

10.0

260

500 1.0

5.0

14

-a : 160

nty

### Division of Health PUBLIC HEALTH LABORATORIES

FOR INSTRUCTIONS

MIND IN STREET	ALL ANCAG				1610	N.E. 160th 3t., , 3	AW ettine	98155			
RETAY	SAMP	LI	INFOF	LAK	TICH	FCR	INC	DIKADA	CHEMIC	L	<b>ESSYLLKA</b>
	USE	THIS	FORM	FOR	THE	COMPL	STE	CHEMICAL	ANALYSIS	ONL	Y
	3,405	13/3	List of London		/ V	DATE COLLECT	ED	CCLLECTED	avPatti Car	tar	

Surface3. Well  Surface3.	
Barore Treatment After Treatment  N AFTER TREATMENT WAS ITFILTEREDFLUORIDATED  CHLORINATEDWATER 90FTENER: TYPE USED  CS: (Water quality problems, address for additional oopies, etc.)  West 924 Sinto Avenue  Spokane,WA. 99201-2595  City	1
HAZARDWATER SOFTENER: TYPE USED	PARTY TO PAY FOR FEE FOR SERVICE TESTING
Office of Environmentathelalth Programs  West 924 Sinto Avenue  Spokane, WA. 99201-2595  City Ze Code  Telephone: (Scan , 545-2475	Print Full reason & Address
Spokane, WA. 99201-2595 City Ze Code  Telephone: ( Scan , 545-2475	The state of the s
Telephone: Scan , 545-2475	
Telephone: ( Scan , 545-2475	Spokane, wa 99201-2595
Area Code	Scan 545-2475

mq/

mg/l

mg/1 mg/1

mg/I

mg/l

mg/!

mg/1

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mg/I mg/l S CaCo3 Micromhas/cm 25° C

> NTU Color

mg/1

mg/1

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mail

CONTRACTOR OF THE PROPERTY OF THE PARTY OF T

LABORATORY SUPERVISOR

CHARGE \$79.00 to
REMARKS: Hay. Waste Rect

PW 30002688

s the Maximum Conteminant Level Allowed Primary Standard

DEHS Laboratory & Data Processing Copy

DSHS Billing Copy Water Supplier Copy District Engineer Copy Local Health Dect. Copy

#### WATER SAMPLE INFORMATION FOR IMORGANIC CHEMICAL **EESYLANA**

SYSTEM LO. NO.		O21 MET	04	SYSTEM GLASS (circle one)	Douglas
DURCE TYPE	(Mem Her)	IF SOURCE :S LAKE OR STREAM ENTER NAME		ED FOR CHEMICAL TO available from this de	
Setore Treatment  After Treatment		AB DRAWN FROM DISTRIBUTION SYSTEM EGTED FROM SYSTEM AT: (Address)	PARTY TO PAY	FOR FEE FOR SERV	VICE TESTING
CHLORINATED	water sof	FILTEREDFLUORIDATED TENER: TYPE USED  Zrees for additional copies, etc.)	- Hazardous	Required Waste Program	(Print Pull Name & Address)
nonco. (Water quality )	or continue, and	:		Environmental	Health Programs
			Spokane,	51.5	99201 - 2595 Zio Code
			Telephone: (	Scan , 54	5-2475

(DO NOT WRITE BELOW THIS LINE) Compliance LESS THAN CHEMIST TESTS RESULTS MCL UNITS mg/1 ALZEUIC 0.05 1.0 \* mg/1 rum LABORATORY SUPERVISOR 0.01\* mg/I admium . J. Davies CHARGE: \$ 79,00 65 0.05 mg/1 mumo œ

mg/l 0.3 70 0.05 820 mg/1 Iganese

0 5 0.002 mg/1 **Hercury** 0.01 . mg/I BRIUM 0.05\* mg/1 Silver mg/1 fium tardness Micromhos/cm 700 ductivity

1.0" .. bidity NTU Color Units 15.0 color 2.0 mg/l iride 10.0 mg/1 Vitrate 250 mg/I \*1 oride

Sulfate mg/1 250 mg/l 500 mg/1 1.0 Copper 8.0 mg/1

REMARKS:

HAZE WASTE ACCT

PW 30002689

and She !

£.W	S RETA	ELCKAS HT BEU	INFORMA IS FORM FO					CHEM ANALYS			323
					OLLECTED	<u> </u>	COLLECTE	<sup>0 sy:</sup> Patti Scan 545-2	475		2022841345
	Y81EM LD. MO.		SYSTEM NAME  OR 1	MET	05	<del></del> 8		SYSTEM (ourose one	3 4	Douglas	\$
SOURCE TYP	PE 8001 We	(Well No.)	O .F SOURCE IS LAKE O	N RETHE MASRIE N	AME	And Harriston		GED FOR CHE		120/14/76	
2. Spring		IF SAMPLE	WAS DRAWN FROM DIST			A 100	schedule is	available from	n this depar	tment.	
	re Treatment r Treatment	IT WAS C	OLLECTED FROM SYSTEM	AT: (Accress)		P	ARTY TO PA	Y FOR PEE P	OR SERVICE	TESTING	
IF TAKEN A	FTER TREATM	MENT WAS IT_	FILTERED	FLUORIDATED		-	Signature	(Asqueed)	0	Print Full Name & A	407044)
			OFTENER: TYPE USE			На	zardous	Waste Pro	gram		
ACMARINO.	(maior qua	in, prociona,	action to accinion	er copies, sid.,	1.0	Of	fice of	Environme	ntel He	alth Pro	grams
			`			We	st 924 S	into Ave			
						_	196		Street	004 0505	
-2			17	-		25	okane,		WA. 99	201-2595	
						Те	dephone: (_	Scan Area Coos	545-		
			~~~~	- LABORAT	ORY R	EPOR	T	A 54 COO			
: <del></del>	7	255 HAN		(DO NOT WRI	Compil		CHEMIST	10 X 10		E-LOWE-L	San Francis
TESTS	MCL 1	HAN	RESULTS	UNITS	725	HO	INITIALS			- /-	
(Arsenic M	0.05		036	_   mg/l	V	🚽	PO				
Barium -	1.0			mg/l							
Cadmium =	0.01'			neg/I				LABO	RATORY	SUPERVI	SOR
Chromium =	0.05	•						ال ا	100	رعام	
Iron 🏲	0.3			mg/I				SUA TO	- 400	· —	
Lead P	0.05	40.	002	mg/l	/		PO	CHARGE	\$19.	00 70	HAZ
Manganese =	0.05			mg/l				REMARKS:	MAS	TE ACO	JUNT.
Mercury	0.002	4.0	200	mg/I	~		7	-			
Seienium *	0.01'	10.	005	mg/l	/		1	-			
Silver 4	0.05*		<del></del>	mg/I							
Sodium	† T			mg/l							
Hardness	t i			mg/l A8 CaCo3			i		E.		
7	700			Micromnos/cm							
Conductivity	1.0'			- 25°C		$\dashv$					
Turbidity				Color		-					
Color	15.0	-		Units							
Fluoride '	2.0	=		_ mg/l				÷.			
Nitrate **	10.0	=	<del></del>	_ mg/1							
Chloride m	260			mg/l							

PW 30002690

Sulfate

TDS

Zinc

Copper

250

500

1.0

5.0

mg/1

mq/1

mg/l

mg/1

## WATER SAMPLE INFORMATION FOR INORGANIC CHEMICAL ANALYSES

(Barriot)		U	se inis form fo	N (1987)	MPLZ		MICAL		
				A Same of the	1,23	<u>8,2</u>		<sup>o av:</sup> Patti Carter Scan 545-2475	
57	/STEMJW		SYSTEM VAME  CQ			6		1 2 3 4	Douglas
SOURCE TYP	•••	Weil Purchase	JOLACE NO. F BOURCE S ARE (Well No.)	E REINE MABRIC RC	AME			GED FOR CHEMICAL TE	
Bero	E WAS TAKE TO Treatment Treatment		IF SAMPLE WAS DRAWN PROM DI IT WAS COLLECTED FROM SYSTE	STRIBUTION SYSTEM M AT: (Address)		] :	PARTY TO PA	Y FOR FEE FOR SERVI	CE TESTING
Man and a second			WAS ITFILTERED	FLUORIDATED		-		Waste Program	(Print Futi Name & Ascress)
REMARKS:	(Water o	quality p	problems, address for additili		VA. 2			Environmentant h	los 1+h Dynamams
1)				î, î	7	1			lea ich Frograms
						W	est 924 S	Sinto Avenue	
( <b></b> .						-	2 21 2 2 2		0001 0005
				ý.		غ ا	pokane,	WA. S	9201-2595 Zap Come
				******		,	elephone: (		5-2475
				LABORAT	ORY F	EPO	RT	Area Code	w-w
~ <del></del>		1.700		(DO NOT WHI	Compli				Y
TE818	'MCL	HAH	RESULTS	UNITS	YES	× 40	CHEMIST		
Arsenic =	0.05		O a 4	mg/1	V		PO		
larium &	1.0			mg/I	Ì				
Cadmium □	0.01*			mg/i			l	LABORATOR	RY SUPERVISOR
Chromium =	0.05			mg/1				15	De 10'es
fron re	0.3			mg/t		· ·		- / LT	
ead -	0.05	1	0.002	mg/l	1		PO	CHARGE: 5	19,00 to 179.
.Manganese =	0.05			mg/l	Ī	Mic Timber		REMARKS:	West accid
Mercury	0.002	4	.0005	img/I	V		سيد	veker	001 0000
Selenium .	0.01*	<	0.005	mg/l			A		
Silver	0.06			mg/l					
Sodium =				mg/l					
Hardness				mg/l AS CACOS		33 1348			
Conductivity	700			Micromhos/cm 25° C					
Turbidity	1.0*			NTU					
Color	15.0	<u> </u>		Cólor Units					
Fluoride •	2.0 '	<u> </u>		mg/1					
Nitrate **	10.0			mg/l					
Chloride es	250	<u> </u>		mg/t					
Sulfate 🐍	250	<u> </u>		mg/1					
2DT	500		**************************************	mg/l					PW 30002691
Copper ≈	1.0	.		mg/t					
7:00		1	All	mad			1	14 .	

Primary Standard 'MCL a the Maximum Contaminant Level Allowed Primary Standard
DSHS 4-92F (Rev. 7795) -873- DSHS Laboratory & Data Processing Copy DSHS Billing Copy Water Supplier Copy District Engineer Copy Local Health Debt. Copy

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### WATER SAMPLE INFORMATION FOR INORGANIC CHEMICAL ANALYSES

	22,23,89   Scan 545-2475
SYSTEM I.D. NO. SYSTEM NAME	avarem ilasa county Douglas
OURCE TYPE   SQURCE NO. F SOURCE IS LAKE OR STREE    Surface	A fee schedule is available from this department.
M8 SAMPLE WAS TAKEN  IF SAMPLE WAS COLLECTED FROM BYSTEM AT: (App. After Treatment)	PARTY TO PAY FOR FEE FOR SERVICE TESTING
TAKEN AFTER TREATMENT WAS ITFILTEREDFLUC CHLORINATEDWATER SOFTENER: TYPE USED	Warandous Waste Program
MARKS: (Water quality problems, address for additional cop	Office of Environmental Health Programs West 924 Sinto Avenue
	Spokane, WA 99201-2595
	Telephone: (Scan 545-2475
	PRATORY REPORT

	400			(DO NOT WA	TE BELOW THIS LINE)		
TESTS	MCL	HARA	RESULTS	UNITS	Compliance YES / NO	CHEMIST	
Arsenic	0.05		.014	mg/l		PO	
rium •	1.0 *	<u> </u>		. ma/l			
admwm ⇔	0.01*			mg/l			
romium »	0.05			mg/I		1	
ron 🏊	0.3	<u>l</u> .		mg/l		<u> </u>	
gd	0.05	$\leq$	0.002	mg/l		PO	
nganese =	0.05	<u></u>		mg/l			
dercury =	0.002	4	0005	. mg/l	V _	1-2-	
enium =	0.01	4	0.005	mg/l		1	
Silver 4	0.06		•	mq/l			
dium 🛰		<u></u>  .		mg/l			
fardness				mg/l AS CaCoS			
nductivity	700			Micromhos/cm 25° C			
urbidity	1.0'			. UTU			
or	15.0	<u> </u>		Cotor Units			
oride •	.2.0	<u></u> :		mg/l			
litrate 5.c.	ים.םנ	<u>                                      </u>		mg/l			
oride .9,	.280	<u></u>		mg/l			
ulfate .m.	250	<u>.</u>		mg/I			
1	500			mg/l		!	
ooper 🛥 .	1.0	<u> </u>  .		mg/l	1		
	5.0	<u> </u>		mg/I			



LABORATORY SUPERVISOR

J. Davies CHARGE \$79.00 TO HAS. REMARKS: WASTE ALCOUNT

PW 30002692

### Division of meanth PUBLIC HEALTH LABORATORIES

FOR INSTRUCTIONS

## WATER SAMPLE INFORMATION FOR INORCAMIC CHEMICAL AMALYSES

SYSTEM LO. NO.		OQ 1	MET	3. <u>89</u> scar	3Y37EM TLA88 (curcie one) 1 2 3 4	Douglas
	(West No.)	F SOURCE IS LAKE OR		FEES ARE CHARGED A fee schedule is av		
IS BAMPLE WAS TAKEN  Before Treatment  After Treatment		AS DRAWN FROM DISTRI ECTED FROM SYSTEM A		PARTY TO PAY F	OR FEE FOR SERVI	CE TESTING
AKEN AFTER TREATMENT	WATER SOF	TENER: TYPE USED		Hazardous Wa		(Print Full Name & Address)
ARKS: (Water quality p	problems, add	iress for additiona	copies, etc.)			ealth Programs
				West 924 Sin	to Avenue	<del></del>
		#3 W		Spokane.	wa. 9	9201-2595 zp coor
				<del></del> †		-2475

					TORY REPOR	
TESTS	MCL	THAN	REBULTS	UNITS	YES NO	CHEMIST
Arsenic **	0.05	4	0.010	mg/l		PO
rium -	1.0 '			mg/I		
o Cadmium ⇔	0.01*			mg/l		
iromium =	0.05	<u> </u>		mg/I		
ron *	0.3	<u> </u>		mg/l		
ad n	0.05	12	0.002	mg/l		PO
.anganese -	0.05			mg/l		1 7 7 7 7 7 1
Mercury M	0.002	4	.0005	mg/l	v !	m
lenium »	2017	$\leq$	0.005	mg/l		A
Silver ~	0.06			mg/l		3
dium 🖦				mg/l		
Hardness				Mg/I AS CaCo3		
nductivity	700			Micromnos/cm		
urbidity	1.0*			NTU		
kor	15.0			Color Units		
Joride F	2.0 *			mg/I		
litrate	10.0			mg/I		
oride «	260			mg/l		
Sulfate	250			mg/l		
\$	500			mg/l		
соррет съ	1.0			mg/I		
	6.0	!		! mg/l		



LABORATORY SUPERVISOR

PW 30002693



DEE DAVA FOR INSTRUCTIONS

## WATER SAMPLE INFORMATION FOR INCREANCE CHEMICAL ANALYSES

SYSTEMD. NO.	OQI MET	3.89   Scan 545-24	
SOURCE TYPE   SOURCE NO. If (West No.)	F SOURCE IS LAKE OR STREAM ENTER NAME	FEES ARE CHARGED FOR CHEMIC A fee schedule is available from t	
	CTED MORN SYSTEM AT: (Address)	PARTY TO PAY FOR FEE FOR	SERVICE TESTING
TAKEN AFTER TREATMENT WAS IT	ENER: TYPE USED	Hazardous Waste Prog	(Prom Full Name & Address)
MAJE (Water quality problems son			
MARKS: (Water quality problems, addi	ess for additional copies, etc.)	Office of Environmen West 924 Sinto Avenu	1770
MARKS: (Water quality problems, addr	ess for additional copies, etc.)	West 924 Sinto Avenu	1770

•				(DO NOT WRI	RITE SELOW THIS LINE)			
TESTS	.WCT	LESS THAN	RESULTS	UNITS	Comp	ance NO	CHEMST	
rsenic *	0.05		7	mg/I			PO	
rum •	1.0 *			_   mg/l		2011-08WH		
admium 👊	0.01			mg/l				
omium •	0.05	<u>  </u>		mg/l				
on 4	0.0			mg/l				
id *	0.05	<	0.002	mg/!	/		100	
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MCL is the Maximum Contaminant Level Allowed Primary Standard

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3943 4-92F (Rev. 7/86) 473 DSHG Laboratory & Data Processing Copy DSHS Billing Copy Water Supplier Copy District Engineer Copy Local Health Dept. Copy

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### WATER SAMPLE INFORMATION FOR INCRGANIC CHEMICAL ANALYSES

· _	A San Sa San San San Ta	U	SE THIS FORM FOR			TE C			ONLY	5000
				251	CLECTED	2	COLLECTE	Patti Ca		
3) <u></u>		X			- 184 >		<u> </u>	Scan 545-24		ale Tools
	STEM LO. NO		SYSTEM HAME					GYSTEM CLAS	IS COUNTY	
<u> </u>			021	Me	T 1	1		1 2	3 4 Douglas	
SCURCE TYP	3		SOURCE NO! IF SOURCE IS LAKE OF	STREAM ENTER N	AME	PEE	S ARE CHAR	GED FOR CHEMIC	CAL TESTING	
1. Surre		Well Puronsee				A fe	e achedule i	s evaluation from t	his department.	
THIS SAMPLE			IF SAMPLE WAS CRAWN FROM DISTRICT WAS COLLECTED FROM SYSTEM A			7 .				
	• Treatment		II WAS SOCIED FROM STOTEM A	ii. Vida essy		P	PARTY TO PA	AY FOR FEE FOR	SERVICE TESTING	
Affer	Treatment					1				
IF TAKEN AF	TER TREA	THENT	WAS ITFILTERED	_FLUORIDATED		-	Signatur	e (Retwred)	(Frint Pell Harre & Address	•
	RINATED		WATER SOFTENER: TYPE USE	생물에서 있다면 하는 그런데 나는	- A	- Н	azardous	Waste Progr	ram	
HEMARINS:	(Magies di	mailty p	problems, address for additions	ii cobies, etc.)	771.	F1 12.5			tal Health Progra	ms
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							7 951E	Cana	EAE DATE	
<u> </u>						1	elephone: (_	Scan )	545-2475	
1				- LABORAT						
TESTS	'MCL	LESS	RESULTS	UNITS	Compil		CHEMIST		45 A 12	Contract of the
12310			0010		YES	NO	DETALS	SOME THE		
Arsenic -	0.05	=	0.010	mg/(			10			
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\*MCL is the Maximum Conteminant Level Allowed Primary Standard

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DEHS 4-82F Rev 7/56) -875 DSH8 Laboratory & Date Processing Copy D8HS Billing Copy Water Supplier Copy District Engineer Copy Local Health Dept. Copy

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### WATER SAMPLE INFORMATION FOR INCREANIC CHEMICAL ANALYSES

<u></u>			U	SE THIS	FORM F			TE C		ANALYSIS ONLY	
	B		*	2 (1)		2 - tun 1 1 1 1	COLLECTED	2 50	COLLECT	Patti Cartar	
		40.15	-3		15 12		<u> </u>	2 = -	L   1	Scan 545-2475	
(***)	era	TEM LO. M	o		SMAN MOTEVE	227				SYSTEM CLASS COUNTY	
					0.	21 ME	ア	12		1 2 3 4 Douglas	
SOURCE	TYPE			SOURCE NO.	IF BOURCE IS LA	KE OR STREAM ENTER A	WE	PEES	ARE CHAP	RGED FOR CHEMICAL TESTING	
	Sertac Lorreg	٠	Wed Purchases							le available from this department.	
_		WAS TAKE		F SAMPLE .	AS CRAWN FROM	DISTRIBUTION SYSTEM		1			
		Treatment		11 1110 502	20,20 1,000 210			P	ARTY TO P	AY FOR FEE FOR SERVICE TESTING	
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general,		0.0000000000000000000000000000000000000			TENER: TYPE	USED		- Ha	zardous	Waste Program	39
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								Te	lephone: (_	Scan 545-2475	
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Γ٦		100000000000000000000000000000000000000		v=====================================		(DO HOT WR		CONTRACTOR CO.			
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um	_	0.01*	†			mg/1				LABORATORY SUPERVISO	R
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nc	~	0.3	-			mg/l			~~	CHARGE: & Da a	
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\*MCL is the Maximum Contaminant Lover Allowed Primary Standard D3H8 4-92F (Rev. 7/85) -873- D3H3 Laboratory & Data Processing Copy DSHS Billing Copy Water Supplier Copy District Engineer Copy Local Health Dept. Copy

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\*MCL a the Maximum Contaminant Laver Allowed

W.A	אבר		SE THIS FORM FO				HEMICAL		(YIA23		
				Carried Rose	CLECTED 2,23	3,8	2 9	Patti Carter Scan 545-2475			
U_	*STEM 1.0. 14	o. 	SYSTEM HAME  O2			/3		(cycle one)	Douglas		
THE SAMPLE Befor	WAS TAKEN Treatment Treatment	Purchase	IF SAMPLE WAS DRAWN FROM DIS IT WAS COLLECTED FROM SYSTEM	TRIBUTION SYSTEM AT: (Accress)	AME	A 10	e schedule la	GED FOR CHEMICAL TESTING available from this department of the properties of the service terms of the service term			
			WAS ITPILTERED WATER SOFTENER: TYPE US		Sec. 22.22		Signature	(Required) (Print F	ul Neme & Address)		
AARKS:	(Water q	quality ;	problems, address for addition	ō	Hazardous Waste Program Office of Environmental Health Programs West 924 Sinto Avenue						
1		<del></del>					pokane.	WA. 9920			
							elephone: (	Scan 545-24	ZID COOM 75		
				LABORAT					<del></del>		
TESTS	MCL	LESS THAN	REGULTS	UMITS	Compi		CHEMIST	**************************************	· · · · // is		
lic ?	0.05	4	0.010	mg/l	/		PO				
rium 🙀	1.0 *	<u></u>		mg/l				TONO SERVICE RUA SECURIO DE SE			
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n	0.3	-	0.002	mg/l			PO	CHARGE & TADO	75 Hm		
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anganase ···	0.002	1	.0 005	<del></del> .				REMARKS:	· 1001.		
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### WATER SAMPLE INFORMATION FOR INCRCANIC CHEMICAL ANALYSES

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a comment	7						A1	OLLECTED		G COLLEG	Patti C.		
5 am	3			<i>tio</i> ( )				7.5		T Thursday	Scan 545-24		5052
	-	STEM 1.3. N	····		_ AVSTEM	02 I	MET	- 14	1		SYSTEM CL (circle one)	Douglas	
3 <del>0</del> 707.730	CE TYP			SOURCE (Wall No.		E IS LAKE OF	STREAM ENTER N			ES ARE CHA	RGED FOR CHEM	IICAL TESTING	
100	1. Surre Soring	STORE RESERVED	Puronae		_				A	fee schedule	is available from	this department.	
TH-8 2		WAS TAKE			E WAS DRAWN COLLECTED FRO		BUTION SYSTEM			PARTY TO P	AY FOR FEE FO	R SERVICE TESTING	
<u>, Ē</u>		Tresment							_				
TAK					FR.TEF		_FLUORIDATED			Signat	ure (Required)	(Print Full Name & Address)	
EMAI			10.12		TE-SX		i copies, etc.)		→       F	lazardous	Waste Prog	ram	
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П									1 ,	lest 924	Sinto Avenu	Street	11-17-0
				The state of the s		•			-  :	pokane,		WA. 99201-2595	
									-	Telephone: (	Scan )	545-2475	
-		-					- LABORAT						
TEST		-MCL	LESS		RESULTS		UNITS		DIIAnce	CHEMIST	( = 1/2 ×		THE STATE OF
rsenic	<u>.</u>	0.05	+5	10	. O	10	i mg/i	YES	1 40	PO		eren in the state of the state	
um		1.0			<u></u>		mg/i		$\dagger$	1	Sec. Marie		
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omium		0.06		]	•		mg/l				] .1.	Dades	
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Silver		0.05*		.	•		mg/l						
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nuoride	,	2.0 '				•	mg/l						
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2m :

Copper

mq/I

mg/l

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### WATER SAMPLE INFORMATION FOR INORGANIC CHEMICAL ANALYSES

			U	SE THIS	FORM FOR	THE CO	MPLET	E CH		ANALYSIS ONLY
	V.					100	10 LECTED	3.27	COLLECTE	Patti Carter Scan 545-2475
	EYST	EM 1.0. A	ó. 		OZ/	ME	7 /	5		SYSTEM CLASS COUNTY Douglas
2. 84 THIS SAL	Surrece Orwig MPLE W		Purchase	(Wed No.)	IF SOURCE IS LAKE OF	RIBUTION SYSTEM	IAMÉ	A fae	achedule i	GED FOR CHEMICAL TESTING a evaluable from this department.  AY FOR FEE FOR SERVICE TESTING
TAKEN	N AFTI				FILTERED			-	-0.000000000000000000000000000000000000	e (Required) (Print Full Name & Aedrese)
EMARK	(S: (	Water o	quality p	robiems, ac	odress for addition	ai copies, etc.)	7	0f	fice of	Waste Program Environmental Health Programs Sinto Avenue
					e. 1				okane.	WA. 99201-2595 Zip Code
						- LABORA	TORY R	1	ephone: (_	Scan ) 545-2475
<i>p</i>						DO NOT WA				
TESTS		MCL	HESS FESS		RESULTS	UNITS	Compil	ence NO	CHEMIST INTIALS	
rsenic	-	0.05	1	0.	010	mg/l	V		PO	之 可能性 (1975年) 1970 V
rium	•	1.0 '				mg/l				
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ed	3	0.05 *	1	0.	003	mg/l			PO	CHARGE Charge \$79,00 to
manganese		0.05		•_		mg/l		İ		REMARKS:
Freury	7	0.002	1	0_	002	mg/l	ー		<u> </u>	remarks: Haz. Wiste
lenium	3	0.01	<	0.	005	mg/l -	1			Vacit
Silver	*	0.05*		•_		mg/l			196 (1977)	
nuibr	•					mq/l				
Hardness						Mg/I AS CaCoS				
anductivity		700				Micromnes/cm 25° C				
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TOIC		15.0	<u> </u>			Golor Units				
Fluoride	,	2.0 *				mg/l				
itrate		10.0 *			•	mg/l				
hloride	e1	250				mg/l				
Cultura	10,	250				mg/1				PW 30002699
DS .		500				mg/l				£# 3000
Copper	co	1.0				mg/l				

MCL s the Maximum Contaminant Level Allowed Primary Standard

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mg/l

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SEE BACK FOR INSTRUCTIONS

### WATER SAMPLE INFORMATION FOR INORGANIC CHEMICAL ANALYSES

<u> </u>		U	SE THIS FORM FOR					NLY			
				200	2,2		Patti Cart Scan 545-2475				
			v~	1 ///=		76	1 2 3	4   0009103			
SOURCE TY	no	Well Perchase	SOURCE NO! IF BOURCE IS LAKE O	20	FEES ARE CHARGED FOR CHEMICAL TESTING  A fee schedule is available from this department.						
which the same of	E WAS TAKE		IF SAMPLE WAS DRAWN FROM DIST.	RIBUTION SYSTEM		7		WAS TEATING			
h	ore Treatment r Treatment			PARIT TO P	AY FOR FEE FOR SE	AVICE TESTING					
In the secondary	11 Hand 2000		WAS ITFILTERED			Signatur	re (Required)	(Print Full Name & Address)			
AMERICAN STREET			propiems, address for addition				Waste Program				
						Who was seemed the total		Health Programs			
\				0-		West 924	Sinto Avenue				
	-				5 10 <b>76</b> 5	. Spokane.	WA.	99201-2595			
						City		Zip Code			
<b></b>				*		Telephone: (_		45-2475			
	**********			- LABORAT	TORY	REPORT -	Area Code				
The control of the co		11866		(DO NOT WAI	Comp						
TESTS	THICL	THAN	RESULTS	UNITS	YES	NO BHITIALS					
Arsenic *	0.05	$\leq$	0.010	mg/l	V	PO					
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admium 👊	0.01*	<u> </u>		mg/l			LABORATO	ORY SUPERVISOR			
Chromium .	0.06			mg/l			<i>با</i> ر ا	Paries			
on ~	0.3			mg/l			CHARGE &	P.DO TO HAZ WASTR ACCT,			
Lead	0.05	_	0.002	mg/l	/	PO	CHARGE. P.	M.W WHE			
anganese =	0.05	<u> </u>		mg/l			REMARKS:	WASIR ACCI			
Mercury -	0.002		.0005	mq/l	س	-					
elemum +	0.01	$\leq$	0.005	mg/1			<del></del>				
Silver "	0.08*			_   mg/l		<u> </u>					
^odium ►				mg/I							
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arbidity	1.0*			_ NTU							
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ntorade ex	250			mg/l							
Sulfate ».	250			mg/l							
75	500			mg/1				<b>7</b> 4000			
opper co	1.0			mg/1				PW 30002700			
Zinc a	5.0			mg/I			20				

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#### PUBLIC HEALTH LABORATORIES FOR INSTRUCTIONS DO NOT WRITE IN SHADED AREAS water sample information for increanic chemical analyses USE THIS FORM FOR THE COMPLETE CHEMICAL ANALYSIS ONLY COLLECTED BY: Patti Carter SYSTEM TLASS (DECIS ORS) SYSTEM NAME SYSTEM I.D. NO. Douglas OBI TER 1 2 3 SOURCE NO. F SOURCE IS LAKE OR STREAM ENTER NAME (West No.) SOURCE TYPE FEES ARE CHARGED FOR CHEMICAL TESTING A fee schedule is available from this department. 2. Soring 4. Puranzas THIS SAMPLE WAS TAKEN IF SAMPLE WAS DRAWN FROM DISTRIBUTION SYSTEM IT WAS COLLECTED FROM SYSTEM AT: (Address) PARTY TO PAY FOR FEE FOR SERVICE TESTING Before Treetment After Treatment TAKEN AFTER TREATMENT WAS IT\_\_\_\_FILTERED \_\_\_\_FLUORIDATED Signature (Required) CHLORINATED \_\_\_\_\_WATER SOFTENER: TYPE USED \_ Hazardous Waste Program REMARKS: (Water quality propiems, address for additional copies, etc.) Office of Environmenta? Health Programs West 924 Sinto Avenue WA 99201-2595 Spokane 545-2475 Scan Telephone: ( Ares Coos LABORATORY REPORT (DO NOT WRITE BELOW THIS LINE) Compliance LESS CHEMIST TEST8 'MCL RESULTS UNITS YES mg/I 0.05 שותש 1.0 mg/I areum LABORATCRY SUPERVISOR 0.01 mg/l nium 0.06 mg/1 muimore 0.3 mg/l PO 0.06 mg/I REMARKS: HAZ, WASTE AZEOUNT isusze. 0.05 mg/l mg/I 0.002 atenta 0.01 mg/1 ทษก 0.05" mg/1 VEF mg/1 . mr. ness AS CaCos Micromhos/cm 700 nductivity 25° C 1.0" NTU dity Cotor 15.0 or 2.0 mg/l ide 10.0 ma/I rate 250 mg/i de -

PW 30002701

rate

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24

280 500

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mg/I

mg/1

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mg/l

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O NOT W	RITE	IN SHADE						-51	<b>S</b> PCKANE. (509) 5	34-016	1	.~		FOR INSTRUCTIONS	
20.78				-	:047	T 35		-	DATE COLL	ECTED	-110	COLLECTED BY	r: Rand	CAL ANALYSES dy Blackburn	
s this a f					-	-	olianc	e samo	0_2/1 ie? Yes		<u> </u>	**************************************	-1251		
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MPLE LOC	CATI	ON -	28 - 220	IS SAM		TAKEN	BEFOR	T.	AFTER 📮					FILTEREDFLUORIDATED TENER: TYPE USED	
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PE: .		SURFACE		PURCHA	SE	0	2							1 #2 Center	
		OF FINAL	7.0		1.0		e 7			SEN	D REPO	RT TO: (PRINT F	TULL NAME &	ADDRESS)	
8 <sub>1</sub> ,1	E C	RT				89	2.4					City of	Rock	Island	
	£.				<u> </u>		====	8				P.O. Bo	× 99		
REMARKS	16								ŧs	-	-		Street		
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						X o Con Jones			BORATOR'						
TESTS		*MCL	Loss Than		F	RESUL	.TS			Comp		Chemist Initials		Laboratory Number (If different than above)	-000
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Сристи	Cr	0.05	Y		•	0_	<u></u>	0	mg/I	X		JGH			
Iron		0.3	y	1			0	5	mg/i	X		JGH	İ		
Lead		0.05 P	y	1		0	1	0	mg/l	X	i	JGH			
Manganese	-		X	i	•	0	<u> </u>	0	mg/l	x		JGH			=
Mercury	Нд		y		0	0_		0	mg/l	x		JGH			_
Seienium	Se	0.01	y	Ī	•	0_		5 -3	mg/l	x		JGH			_
Silver	Ag	0.05	X	Ī	•	0	1	0	mg/l	x		JGH			200000
Sodium	Na						1	4	mg/I	İ		JGH			_
Hardness				ı	-	2	2	2 1	mg/I As CaCO3	i	i	JKC			
Conductivity	y	700				4	2	0	Micromnos/cm 25° C	X		JGH	III		_
Turbidity	T	1.0						4	NTU	х		JGH	I	**AC 7 *****	<u>= 1000</u> 1300
Color	T	15.0					5 .	0	Color	X.		JKC			_
Fluoride	-	2.0						2	mg/l	x		JGH	1	USHS - UIV. OF HEALTH	
Nitrate	38 N	10.0 P					4	9 1	mg/I	x		JMS			_

CI 250

so 250

Chloride

Suifate

v.

JMS

JGH

mg/I

mg/l

PW 30002702

## ABC LABORATORIES, INC. EAST 4922 UNION AVENUE SPOKANE, WA 99212 509-534-0161

REPORT TO: Town of Rock Island

P.O. Box 99

Rock Island, WA 98850

LAB NO: 32298-89

DATE: 3-08-89

DATE REC'D: 2-24-89

P.O.#:

ATTN:

Gwen

DESCRIPTION: Perform Volatile Organic Scans with quality control on two submitted samples.

## YOLATILE ORGANIC SCAN

DETECTION LIMIT: 1 ppb Parameter	ND: Not Detected	IS: Internal Standard #2	#3
aaa Trifluorotoluene		1\$	15
Chlorobenzene		ND	ND
1,3,Dichlorobenzene		ND	ND
1, 4, Dichlorobenzene		ND	ND
1,2,Dichlorobenzene		ND	ND
Bromodichloromethane		ND	ND
Bromoform		ND	ND
Carbon Tetrachloride		ND ·	ND
Chloroform		ND	ND
Dibromochloromethane		ND	ND
1, 1, Dichloroethane		ND	ND
1,2,Dichloroethane		ND	ND
1, 1, Dichloroethylene		ND	ND
Trans 1, 2, Dichloroethyle	ene	ND	ND
1,2,Dichloropropane		ND	ND
Cis 1, 3, Dichloropropane		ND	ND
Trans 1, 3, Dichioropropy	/lene	ND	ND
Methylene Chloride		ND	ND
1, 1, 2, 2, Tetrachloroeth	nane	ND	ND
Trichloroethylene		ND	ND
1, 1, 1, Trichloroethane	•	ND	ND
1, 1, 2, Trichloroethane	•	ND	ND
Tetrachloroethylene		ND	ND
2, Chioroethylvinyl Ether		ND	ND

Respectfully submitted, ABC LABORATORIES, INC.

W.E. Burkhardt

Manager

### ABC LABORATORIES, INC. EAST 4922 UNION AVENUE SPOKANE, WA 99212 509-534-0161

REPORT TO: Town of Rock Island

P.O. Box 99

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Parameter ND: No	it Detected IS: Internal Standard #2	#3
aaa Trifluorotoluene	IS	15
Chlorobenzene	ND	ND
1,3,Dichlorobenzene	ND	ND
1, 4, Dichloropenzene	ND	ND
1,2,Dichlorobenzene	ND	ND
Bromodichloromethane	ND	ND
Bromoform	ND	ND
Carpon Tetrachloride	ND	ND
Chloroform	ND	ND
Dibromochloromethane	ND	ND
1, 1, Dichlorcethane	ND	ND
1, 2, Dichlorcethane	ND	ND
1, 1, Dichloroethylene	ND	ND
Trans 1, 2, Dichloroethylene	ND	ND
1,2,Dichloropropane	ND	ND
Cis 1, 3, Dichloropropane	ND	ND
Trans 1, 3, Dichloropropylene	ND	ND
Methylene Chloride	ND	ND
1, 1, 2, 2, Tetrachloroethane	ND .	ND
Trichloroethylene	ND	ND
1, 1, 1, Trichloroethane .	ND	ND
1, 1, 2, Trichloroethane	ND	ND
Tetrachloroethylene	ND	ND
2, Chioroethylvinyl Ether	ND	ND

Respectfully submitted, ABC LABORATORIES, INC.

W.E. Burkhardt

Manager

	RE	SULTS OF ANALYSIS	Date	e: 2/2	4/89
Sample Identification .	Lab Number	Test or Residue	Results	Units	MRL
GWEN HOUCK, CITY OF ROCK ISCAND, WA.		e e		e Stern	
WELL =2, WATER	89W0005	CHC PESTICIDE SCAN	N. D.	PPB	0.010
2/21/89		ORGAND PHOS. SCAN	N.D.	,,	0.050
	a				
	8.	€9 S2			
WELL = WATER  INCK ISLAND CITY  WELL 2/21/89	89W0006	OHC PESTICIDE SCAN	N.D. N.D.	P.P.B.	0.010
		ja .			
REPORTED RESULTS	TO GWE	N HONCIC, 10:00 AM	, 2/24/	189. HK.	
			•		
				***	
		· ·//.			
		- Jan	tal c	- Kua PW 30	002705

# NATIONAL CHEM LAB Water Test Report

Date: 03/08/89

Time: 12:14

Page: 1 of 1

Thank you for using National Chem Lab. If you have any questions regarding these results please contact the lab at 509-754-5725.

NCL#: WE9030802

Chemist: MICKEY HUNACEK Michael Shiper cele

CITY OF ROCK ISLAND

SAMPLE #:

LOT #:

PO #:

COMMENT:

DATE RECVD: 02/09/89 TIME RECVD: 12:00 RECVD BY: MICKEY

TRANSPORTATION MODE: CUSTOMER DELIVER

SAMPLE SOURCE: WELL #2

SAMPLE SIZE:

CONTAINER TYPE:

DELIVERED BY: CUSTOMER

TEST	MCL	RESULTS		DATE ANALYZED	TIME ANALYZED	CHEMIST
ARSENIC	0.050	<.001 ppm		03/06/89	12:00	MICKEY HUNACEK
SELENIUM	0.010	<.001 mg/L	,	03/06/89	12:00	MICKEY HUNACEK
CHROMIUM	0.050	0.02 mg/L		03/06/89	12:00	MICKEY HUNACEK

#### NATIONAL CHEM LAB Water Test Report

Date: 03/08/89

Time: 12:12

Page: 1 of 1

Thank you for using National Chem Lab. If you have any questions regarding these results please contact the lab at 509-754-5725.

NCL#: WE9030801

Chemist: MICKEY HUNACEK Thurach

CITY OF ROCK ISLAND

SAMPLE #:

LOT #:

PO #:

COMMENT:

DATE RECVD: 01/31/89 TIME RECVD: 12:00 RECVD BY: MICKEY

TRANSPORTATION MODE: CUSTOMER DELIVER

SAMPLE SOURCE: WELL #3

SAMPLE SIZE:

CONTAINER TYPE:

DELIVERED BY: CUSTOMER

TEST	MCL	RESULTS	DATE ANALYZED	TIME ANALYZED	CHEMIST
ARSENIC	0.050	<.001 ppm	03/06/89	09:30	MICKEY HUNACEK
SELENIUM	0.010	<.001 mg/L	03/06/89	09:30	MICKEY HUNACEK
CHROMIUM	0.050	0.03 mg/L	03/06/89		MICKEY HUNACEK

#### NATIONAL CHEM LAB Water Test Report

Date: 03/08/89

Time: 12:13

Page: 1 of 1

Thank you for using National Chem Lab. If you have any questions regarding these results please contact the lab at 509-754-5725.

NCL#: WE9030803

Chemist: MICKEY HUNACEK Muckey Afunacel

CITY OF ROCK ISLAND

SAMPLE #:

LOT #:

PO #:

COMMENT:

DATE RECVD: 02/09/89 TIME RECVD: 12:00 RECVD BY: MICKEY

TRANSPORTATION MODE: CUSTOMER DELIVER

SAMPLE SOURCE: TOWN HALL

SAMPLE SIZE:

CONTAINER TYPE:

DELIVERED BY: CUSTOMER

	TEST	MCL	RESULTS	DATE ANALYZED	TIME ANALYZED	CHEMIST
	ARSENIC	0.050	<.001 ppm	03/06/89	12:00	MICKEY HUNACEK
*	SELENIUM	0.010	<.001 mg/L	03/06/89	12:00	MICKEY HUNACEK
•	CHROMIUM	0.050	0.02 mg/L	03/06/89	12:00	MICKEY HUNACEK

PW 30002709

0-

### ABC LABORATORIES, INC. EAST 4922 UNION AVENUE SPOKANE, WA 99212 509-534-0161

REPORT TO: Town of Rock Island

P.O. Box 99

Rock Island, WA 98850

LAB NO: 32298-89

DATE: 3-08-89

DATE REC'D: 2-24-89

P.O.#:

ATTN:

Gwen

DESCRIPTION: Perform Volatile Organic Scans with quality control on two submitted samples.

### YOLATILE ORGANIC SCAN

DETECTION LIMIT: 1 ppb ND: Not Detected Parameter	is: Internal Standard #2	#3
aaa Trifluorotoluene	1\$	15
Chlorobenzene	ND	ND
1,3,Dichlorobenzene	ND	ND
1, 4, Dichlorobenzene	ND	ND
1, 2, Dichlorobenzene	ND	ND
Bromodichloromethane	ND	ND
Bromoform	ND	ND
Carbon Tetrachloride	ND ·	ND
Chloroform	ND	ND
Dibromochloromethane	ND	ND
1, 1, Dichloroethane	ND	ND
1, 2, Dichloroethane	ND	ND
1, 1, Dichloroethylene	ND	ND
Trans 1, 2, Dichloroethylene	ND	ND
1, 2, Dichloropropane	ND	ND
Cis 1, 3, Dichloropropane	ND	ND
Trans 1, 3, Dichioropropylene	ND	ND
Methylene Chloride	ND	ND
1, 1, 2, 2, Tetrachloroethane	ND	ND
Trichloroethylene	ND	ND
1, 1, 1, Trichloroethane	ND	ND
1, 1, 2, Trichloroethane	ND	ND
Tetrachloroethylene	ND	ND
2, Chloroethylvinyl Ether	ND	ND

Respectfully submitted, ABC LABORATORIES, INC.

W.E. Burkhardt

Manager

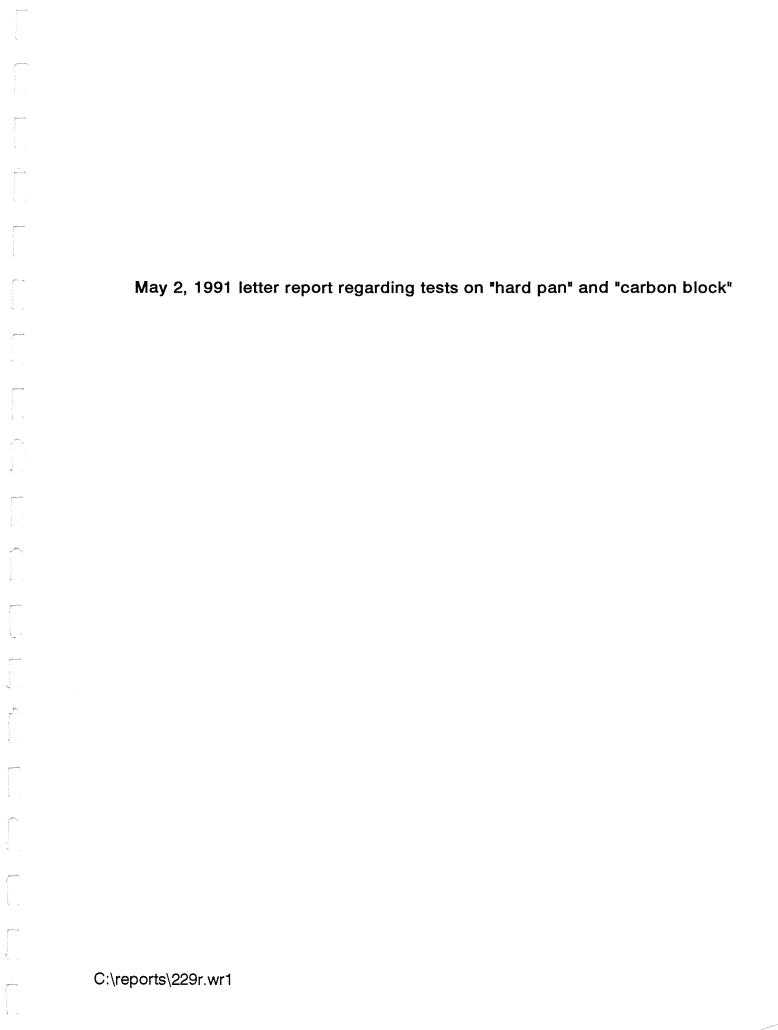
ase Print Plainly USE HEAVY PENCIL DO NOT WHITE IN SHADED AREAS

## LABORATORY NAME 4922 E UNION AVE SPOKANE, WA 99212



# WATER SAMPLE INFORMATION FOR THORGANIC CHEMICAL ANALYSES

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yes, wh	at w	as the la	borato	ory numb	er of t	the pre	vious s	ample? —				* .
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7 3 4				ock Is							1 O 2	
MPLE LO	CATI	on 	0.0000000000000000000000000000000000000	IS SAMPLE REATMENT		<b>↓</b> BEFOR		AFTER T	IF TAP			T WAS ITFILTEREDFLUORIDATED WATER SOFTENER: TYPE USED
JURCE			***		SOUR	CE NO.	IF SOU	RCE IS LAKE O	R STRE	AM, ENT	ER NAME	IF SAMPLE WAS DRAWN FROM DISTRIBUTION SYSTE IT WAS COLLECTED FROM SYSTEM AT: (ADDRESS)
		SURFACE		PURCHASE	0	3						Well #3 Penn
- E	DATE	OF FINAL			12.50	===			SEN	D REPO	RT TO: (F) INT F	FULL NAME & ADDRESS)
	REPO		1		1		38				8778	tock Island
24	**					=3			No.		O. Box	Name 99
REMARKS	:					35.	53	*		<u> </u>	U. BOX	Street
4									-		ck Isla	
Cop	per	, Cu,	mg/	<u>′L</u>	0.0	046			Teres	CITY	09) 88	ZIP CODE 44-1261
Zinc	Ξ,	Zn, m	g/L		.01	10			1000	Are Co	••	4-1201
								BORATORY		ORT	a.	ş
							(DO N	ISE STIRW TO	OW THI	S LINE)		
	i		Less Than	1				L.	Comp	NO	Chemist	Laboratory Number (If different than above)
TESTS	41	*MCL 0.05	<	1	RESU	LTS 1	5	mg/i		140	Initials	(in directif than above)
Barium		1.0	X	•	<u> </u>	<del></del>		mg/I	X		JGH JGH	
admium		0.01 P	X		0	0	2	/ng/l	X	ĺ	JGH	
Chromium			X		0	1	0	mg/I	X		JGH	
-4					1.	5	2	ang/l		х	JGH	
ron	Fe I	0.3				1				137.0	0 0 11	
-	Pb	Р	X		0		_0	mg/l	X		JGH	
Lead	РЬ	0.05 P	X		<u> </u>	<u>-</u>		mg/l	X		JGH JGH	
Lead  Manganese	РЬ	0.05 P	X		0 0	<u>-1</u> -1 -1			х		JGH	
Lead	Pb Mn	0.05 P	X		0	1 1 1 0	5	mg/l	x		JGH JGH	
Lead Manganese Mercury Selenium	Pb Mn Hg	0.05 P 0.05 0.002 P 0.01 P	x		0	1 1 0 1	50_	mg/l mg/l	х		JGH JGH JGH	
Aanganese Mercury Selenium	Pb Mn Hg	0.05 P 0.05 0.002 P 0.01 P	X	0	0 0		5 0 5	mg/l mg/l mg/i	x x		JGH JGH JGH	
Lead Manganese Mercury Selenium Silver Sodium	Pb Man Hg Se Ag	0.05 P 0.05 0.002 P 0.01 P	X	0	0 0	1 1 0 1 1 2	5 0 5 0	mg/l mg/l mg/l mg/l mg/l mg/l	x x		JGH JGH JGH JGH	
Lead Manganese Mercury Selenium Silver Sodium Hardness	Pb Man Man Man Man Man Man Man Man Man Man	0.05 P 0.05 0.002 P 0.01 P	X	0	0 0 0	1 0 1 1 2 2	_5 _0 _5 _0 _5	mg/l mg/l mg/l mg/l mg/l mg/l As CaCO3	x x		JGH JGH JGH JGH JGH	
Aller Sodium Hardness Conductivity	Pb Man Man Man Man Man Man Man Man Man Man	0.05 P 0.05 0.002 P 0.01 P 0.05 P	X	0	0 0 0		5 0 5 0 5 5 6	mg/l mg/l mg/l mg/l mg/l mg/l mg/l As CaCO3	x x	X	JGH JGH JGH JGH JGH JGH JGH	
Manganese Mercury Selenium Silver Sodium Haraness Conductivit	Pb Man Man Man Man Man Man Man Man Man Man	0.05 P 0.05 0.002 P 0.01 P 0.05 P	X		0 0 0	2		mg/l mg/l mg/l mg/l mg/l mg/l McGCG3 Micromnos/cm 25° C NTU Color	X	X	JGH JGH JGH JGH JGH JGH JGH	10000000000000000000000000000000000000
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All All All All All All All All All All	Pb   Mn   Hg   Se   Ag   Na	0.05 P 0.05 0.002 P 0.01 P 0.05 P 700 1.0 P 15.0 P	X		0 0 0	2 2 5 •		mg/l mg/l mg/l mg/l mg/l mg/l As CaCO3 Micromnos/cm 25° C NTU Color Units	X	X	JGH JGH JGH JGH JGH JGH JKC JGH JKC JGH JKC	10000000000000000000000000000000000000
All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese All Anganese Anganese Anganese All Anganese All Anganese All Anganese All Anganese Anganese Anganese Anganese Anganese Anganese Anganese Anganese Anganese Anganese Anganese Anganese Anganese Anganese Anganese Anganese Anganese Anganese Anganese Anganese Anganese Anganese Anganese Anganese Anganese Anganese Anganese Anganese Anganese Anganese Anganese Anganese Anganese Anganese Anganese Anganese Anganese Anganese Anganese Anganese Anganese Anganese Anganese An	Pb Mn Hg Se Ag Na Y	0.05 P 0.05 0.002 P 0.01 P 0.05 P 700 1.0 P 15.0 P 10.0 P	X	0	0 0 0	2 •		mg/l mg/l mg/l mg/l mg/l mg/l Mg/l As CaCO3 Micromnos/cm 25° C NTU Color Units mg/l	X	X	JGH JGH JGH JGH JGH JGH JGH JGH JKC	10 = 2 = 1 型   1





Mr. Robert L. Miller Assistant Plant Manager Silicon Metaltech, Inc. 100 4th Street Rock Island, Washington 98850

#### Dear Robert:

In accordance with our telephone conversation last week, enclosed are data on furnace wastes (carbon block and hard pan) which were not available when the December 5, 1988 report was These are data that were mentioned in my March 14, issued. 1991 letter to Jim Trunzo which were not included in the data package recently provided to WDOE. It was recommended these additional data also be submitted to WDOE.

Testing of carbon block and hard pan samples was recommended in 1988 to assess whether any wastes disposed in your onsite landfill/piles were hazardous wastes for the same reason that characterization of the fume waste had been carried out earlier.

The carbon block and hard pan samples were collected in November 1988 when Furnace 1 was undergoing a maintenance overhaul. Three samples were collected of each of the two types of material. Your chemist assisted with the sampling such that the samples would be representative of these wastes. The three samples of carbon block were composited in the laboratory and tested for most priority pollutants (metals, cyanide, semi-volatile organics [base-neutral extractables and acid extractables], nonpriority pollutant semi-volatile organics. EP Toxicity tests were also performed. Likewise a hard pan composite prepared by the laboratory was tested for the same parameters. The samples were not tested for volatile organics because it was so unlikely that any volatile materials could be present at the high operating temperatures of the furnace.

The laboratory results have been summarized in the attached table. Copies of the laboratory reports for this testing are also attached.

These data indicated that these furnace liner wastes have levels of metals/elements similar to background soil, although chromium was higher in the carbon block sample. Carbon block and hard pan samples had no detectable cyanide and no detectable semi-volatile organics. None of the EP Toxicity



Mr. Robert L. Miller May 2, 1991 Page 2

metal/element test parameters were above detectable levels. Accordingly the carbon block and hard pan would not be classified as hazardous or dangerous under the former EP Toxicity test. Generally the results indicate these wastes do not have levels of contaminants that should be of concern.

These data are potentially relevant to ground water quality at your facility because these wastes are piled/disposed onsite.

Please do not hesitate to call me if there are any questions.

Sincerely,

Patrick H. Wicks, PE, CHMM President

Enclosures

# CARBON BLOCK & HARD PAN FURNACE WASTES ANALYTIC DATA SUMMARY

# Constituent Concentration, parts per million (ppm, mg/l, mg/kg)

Sample Location	Furnace 1	Furnace 1
Sample Identification	Carbon	Hard
	Block	Pan
	Composite	Composite
Sample Date	15-Nov	15-Nov
	1988	1988
Sample Type	Carbon Block	Hard Pan
Sample Depth, Ft		
Laboratory	Am Test	Am Test
Report Number or Date	36-88	36-88
Laboratory Sample Number	823986	823987

#### **PRIORITY POLLUTANTS**

ME	ГΑ	LS

Antimony	<2.0	<2.0
Arsenic	<3.0	<3.0
Beryllium	<0.70	<0.70
Cadmium	<0.20	<0.20
Chromium	187	3.8
Copper	10	40
Lead	<2.0	<2.0
Mercury	< 0.0096	<0.0092
Nickel	14	21
Selenium	<3.0	<3.0
Silver	<1.0	<1.0
Thallium	<2.0	<2.0
Zinc	13	61

#### **MISCELLANEOUS**

Total Cyanide	<0.06	<0.06

#### **BASE-NEUTRAL EXTRACTABLES**

Acenaphthene	<0.0333	< 0.0333
Acenaphthylene	< 0.0333	< 0.0333
Anthracene	< 0.0333	<0.0333
Benzidine	<0.1000	<0.1000
Benzo(a)anthracene	< 0.0333	< 0.0333
Benzo(a)pyrene	< 0.0333	< 0.0333
Benzo(b)fluoranthene	<0.0333	< 0.0333
Benzo(ghi)perylene	< 0.0333	< 0.0333
Bis(2-chloroethoxyl) methane	< 0.0333	< 0.0333
Bis(2-chloroethyl) ether	< 0.0333	< 0.0333
Bis(2-chloroisopropyl) ether	< 0.0333	< 0.0333
Bis(2-ethylhexyl) phthalate	0.06	0.04

# CARBON BLOCK & HARD PAN FURNACE WASTES ANALYTIC DATA SUMMARY

## Constituent Concentration, parts per million (ppm, mg/l, mg/kg)

Sample Location	Furnace 1	Furnace 1
Sample Identification	Carbon	Hard
•	Block	Pan
	Composite	Composite
Sample Date	15-Nov	15-Nov
·	1988	1988
Sample Type	Carbon Block	Hard Pan
Sample Depth, Ft		
Laboratory	Am Test	Am Test
Report Number or Date	36-88	36-88
Laboratory Sample Number	823986	823987
4-Bromophenyl phenyl ether	< 0.0667	< 0.0667
Butyl benzyl phthalate	< 0.0333	< 0.0333
2-Chloronaphthalene	< 0.0333	< 0.0333
Chrysene	< 0.0333	< 0.0333
4-Chlorophenyl phenyl ether	< 0.0333	< 0.0333
Dibenzo(a,h)anthracene	< 0.0333	< 0.0333
1,2-Dichlorobenzene	< 0.0333	< 0.0333
1,3-Dichlorobenzene	< 0.0333	< 0.0333
1,4-Dichlorobenzene	< 0.0333	< 0.0333
3,3-Dichlorobenzidine	<0.1000	<0.1000
Diethyl phthalate	< 0.0333	< 0.0333
Dimethyl phthalate	<0.0333	< 0.0333
Di-n-Butyl phthalate	0.0433	< 0.0333
2,4-Dinitrotoluene	< 0.0667	< 0.0667
2,6-Dinitrotoluene	< 0.0667	< 0.0667
Di-n-octyl phthalate	< 0.0333	< 0.0333
1,2-Diphenylhydrazine		
Fluoranthene	< 0.0333	< 0.0333
Fluorene	< 0.0333	< 0.0333
Hexachlorobenzene	< 0.0667	<0.0667
Hexachlorobutadiene	< 0.0333	< 0.0333
Hexachlorocyclopentadiene	< 0.0667	< 0.0667
Hexachloroethane	< 0.0333	< 0.0333
Indeno(1,2,3-cd)pyrene	< 0.0333	< 0.0333
Isophorone	<0.0333	< 0.0333
Naphthalene	<0.1667	<0.1667
Nitrobenzene	< 0.0333	< 0.0333
N-nitrosodi-n-propylamine	< 0.0333	< 0.0333
N-Nitrosodiphenylamine	< 0.0333	< 0.0333
Phenanthrene	< 0.0333	< 0.0333
Pyrene	< 0.0333	< 0.0333
1,2,4-Trichlorobenzene	< 0.0333	< 0.0333

# CARBON BLOCK & HARD PAN FURNACE WASTES ANALYTIC DATA SUMMARY

## Constituent Concentration, parts per million (ppm, mg/l, mg/kg)

Sample Location	Furnace 1	Furnace 1
Sample Identification	Carbon	Hard
	Block	Pan
	Composite	Composite
Sample Date	15-Nov	15-Nov
	1988	1988
Sample Type	Carbon Block	Hard Pan
Sample Depth, Ft		
Laboratory	Am Test	Am Test
Report Number or Date	36-88	36-88
Laboratory Sample Number	823986	823987

#### **ACID EXTRACTABLES**

2-Chlorophenol	<0.0333	<0.0333
2,4-Dichlorophenol	<0.0333	< 0.0333
2,4-Dimethylphenol	<0.0333	< 0.0333
4,6-Dinitro-o-cresol	<0.1667	<0.1667
2,4-Dinitrophenol	<0.1667	<0.1667
2-Nitrophenol	<0.0333	<0.0333
4-Nitrophenol	<0.1333	<0.1333
p-Chloro-m-cresol	<0.0333	< 0.0333
Pentachlorophenol	<0.1667	<0.1667
Phenol	<0.0333	<0.0333
2,4,6-Trichlorophenol	<0.0667	<0.0667

#### **NON-PRIORITY POLLUTANTS**

#### OTHER ORGANICS

	·
<0.0333	<0.0333
<0.0333	<0.0333
<0.0333	<0.0333
<0.0333	< 0.0333
<0.0333	< 0.0333
<0.0333	< 0.0333
<0.0333	< 0.0333
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<0.0333	< 0.0333
	<0.0333 <0.0333 <0.0333 <0.0333 <0.0333 <0.0333

# CARBON BLOCK & HARD PAN FURNACE WASTES ANALYTIC DATA SUMMARY

## Constituent Concentration, parts per million (ppm, mg/l, mg/kg)

Sample Location	Furnace 1	Furnace 1
Sample Identification	Carbon	Hard
·	Block	Pan
	Composite	Composite
Sample Date	15-Nov	15-Nov
	1988	1988
Sample Type	Carbon Block	Hard Pan
Sample Depth, Ft		
Laboratory	Am Test	Am Test
Report Number or Date	36-88	36-88
Laboratory Sample Number	823986	823987
2-Methyl naphthalene	<0.0333	<0.0333
2-Nitroaniline	<0.0333	<0.0333
3-Nitroaniline	<0.0667	<0.0667
4-Nitroaniline	<0.1000	<0.1000
Styrene		
2,4,5-Trichlorophenol	<0.0667	<0.0667
Vinyl acetate		
total 1,2-Dichloroethylene		

EP Toxicity Maximum

Allowable

TCLP or E P TOXICITY	E P TOX	E P TOX	Concentr.
Elements/Metals			·
Arsenic	<0.02	<0.02	5
Barium	<1.0	<1.0	100
Cadmium	<0.05	<0.05	1
Chromium	<0.05	<0.05	5
Copper			
Lead	<0.10	<0.10	5
Mercury	<0.001	<0.001	0.2
Nickel			_
Selenium	<0.05	<0.05	1
Silver	<0.05	<0.05	5
Zinc			



Sample Chain of Custody

/.O.No.:	36-8	8	Project	Name	SILICON	METALTECH			/ / /		ERM-NW, Inc
mpler:	P. WI	K-S / T CLIMER Number / / /		T. CLIMER Number / / / 2535/52 nd					2535 152 md AVE, NE, 50 Redmond, WA 98052		
ERM ample umber	Date	Time	C O M P	G R A B	Samp	le Location	Containers				Remarks P85-1787
	11-15-58	12:30		1	F#1 CAR30	NBLOCK #1	1 1			7	ANALIZE ONE COMPOSITE
	11-15-8	12:30		レ	F#1 CARB	W BLUCK #3	1			()	FROM THESE THREE
25 0 22	11-15-88	12:30			22	N BLOCK #2	1			J	SAMPLES FOR PRIORITY
	11-15-88	12:30		ı	F#1 HARD	PAN #1					POLLUTANT METALS
	11-15-88	12:30		v	F#1 HARD	PAN #2	r				TOTAL CYANIDE (P.P.
	11-15-88			200000000000000000000000000000000000000		PAN #3					MEMOD) AND JEMI-
											VOLATILE ORGANICS
SPE	CIAL	INSTR	Puci	rion	15: RETA	IN UNUSED	SAMPLE	SFOR			VIA EPA METHOD
Po	SSIBL	E AL	201	TION	IAL AN	YLYSES. P.	FOUNDE	FLOI	ppy		8270 OR 8250.
		400*1111 #15T4 (#1000)		100		F REPORT!					Action and the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Control of the Contro
	Relinqu		T	ate	Time	Sample Receive		Date	Time	7731 AVEL	Reason for Transfer
NICK	(18H	wil	1/-	17-8	8 8:35	In Benn	10.00	17-50	8.35	tras	roport & lab
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### 14603 N.E. 87th St. • REDMOND, WASHINGTON 98052 • 206/885-1664

#### ANALYSIS REPORT

CLIENT: ERM Northwest, Inc.

DATE RECEIVED: 11/29/88

REPORT TO: Pat Wicks

DATE REPORTED: 12/7/88

2535 - 152nd Avenue NE Suite B2

PROJECT NO.: 36-88

Redmond, WA 98052

SILICON METAL TECH

Laboratory Sample Nos. 823986 823987

Client Identification F#1 Carbon F#1 Hard

Block Comp Pan Comp

Cyanide (ug/g) <0.06 <0.06



-2-

CLIENT: ERM Northwest, Inc.

DATE RECEIVED: 11/29/88 DATE REPORTED: 12/7/88

REPORT TO: Pat Wicks

PROJECT NO.: 36-88

#### E.P. TOXICITY

Laboratory Sample Nos. Client Identification	823986 F#1 Carbon Block Comp	823987 F#1 Hard Pan Comp	MAXIMUM ALLOWABLE CONCENTRATION (mg/1)
Arsenic	<0.02	<0.02	5.0
Selenium	<0.05	<0.05	1.0
Barium	<1.0	<1.0	100.
Cadmium	<0.05	<0.05	1.0
Lead	<0.10	<0.10	5.0
Mercury	<0.001	<0.001	0.20
Silver	<0.05	<0.05	5.0
Chromium	<0.05	<0.05	5.0

All values are reported in mg/l.



-3-

CLIENT: ERM Northwest, Inc. REPORT TO: Pat Wicks		DATE RECEIVED: 11/29/88 DATE REPORTED: 12/7/88 PROJECT NO.: 36-88
		149
Laboratory Sample Nos.	823986	823987
Client Identification	F#1 Carbon Block Comp	F#1 Hard Pan Comp
Total Solids (%)	100.	100.
Arsenic	<3.0 <3.0]	<3.0
Silver	<1.0 <1.0]	<1.0
Antimony	<2.0 <2.0]	<2.0
Beryllium	<0.70 <0.70]	<0.70
Cadmium	<0.20 <0.20]	<0.20
Chromium	187. 171.]	3.8
Copper	10. 10.]	40.
Mercury	<0.0096	<0.0092

All results are reported in ug/g on a dry weight basis.



-4-

CLIENT: ERM Northwest, Inc. REPORT TO: Pat Wicks		DATE RECEIVED: 11/29/88 DATE REPORTED: 12/7/88 PROJECT NO.: 36-88
Laboratory Sample Nos.	823986	823987
Client Identification	F#1 Carbon Block Comp	F#1 Hard Pan Comp
Nickel	14.	21.
Lead	<2.0 <2.0]	<2.0
Selenium	<3.0 <3.0]	<3.0
Thallium	<2.0 <2.0]	<2.0
Zinc	13. 11.]	61.

All results are reported in ug/g on a dry weight basis.



-5-

CLIENT: ERM Northwest, Inc.

REPORT TO: Pat Wicks

DATE RECEIVED: 11/29/88
DATE REPORTED: 12/7/88

PROJECT NO.: 36-88

#### NBS 1645 REFERENCE MATERIAL

Element	Determination (ug/g)	True Value (ug/g)	Recovery (%)
Arsenic	45.	46.	68.
Silver	NC	NC	NC
Antimony	NC	NC	NC
Beryllium	NC	NC	NC
Cadmium	9.1	10.2	89.
Chromium	24,970.	29,600.	84.
Copper	103.	109.	95.
Mercury	THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE STATE OF THE S	-	_
Nickel	41.	45.8	89.
Lead	598.	714.	84.
Selenium	NC	NC	NC
Thallium	<5.54	1.44	-
Zinc	1,518.	1,720.	88.



DATE RECEIVED:

DATE REPORTED:

11/29/88

12/7/88



CLIENT:

ERM Northwest, Inc.

REPORT TO: Pat Wicks PROJECT NO.: 36-88 Laboratory Sample Nos. BNA 823987 DETECTION Method F#1 Hard LIMIT Client Identification Blank Pan Comp. (ug/kg) Aniline ND ND 33.3 2-Chlorophenol ND ND 33.3 Bis (2-Chloroethyl) Ether ND ND 33.3 Phenol 33.3 ND ND 1,3-Dichlorobenzene ND 33.3 ND ND 33.3 1,4-Dichlorobenzene ND 1,2-Dichlorobenzene ND ND 33.3 Benzyl Alcohol 33.3 ND ND 33.3 Bis (2-Chloroisopropyl) Ether ND ND 2-Methyl Phenol 33.3 ND ND Hexachloroethane ND ND 33.3 N-Nitrosodipropylamine ND ND 33.3 Nitrobenzene ND ND 33.3 4-Methylphenol ND ND 33.3 Isopherone ND ND 33.3 2-Nitrophenol 33.3 ND ND 33.3 2,4-Dimethylphenol ND ND Bis (2-Chloroethoxy) Methane 33.3 ND ND 2,4-Dichlorophenol ND ND 33.3 1,2,4-Trichlorobenzene ND ND 33.3 Naphthalene ND ND 166.7 Benzoic Acid ND ND 33.3 4-Chloroaniline ND 33.3 ND Hexachlorobutadiene ND ND 33.3

ND

ND

ND

ND

ND

ND

ND

ND

ND

ND

ND = Not Detected.
All values are reported in ug/kg.

2-Methylnaphthalene

4-Chloro-3-Methylphenol

2,4,6-Trichlorophenol

2,4,5-Trichlorophenol

Hexachlorocyclopentadiene

33.3

33.3

66.7

66.7

66.7



A

CLIENT: ERM Northwest, Inc.

DATE RECEIVED: 11/29/88
DATE REPORTED: 12/7/88
REPORT TO: Pat Wicks

PROJECT NO.: 36-88

Laboratory Sample Nos.	BNA Method	823987 F#1 Hard	DETECTION LIMIT
Client Identification	Blank	Pan Comp.	(ug/kg)
2-Chloronaphthalene	ND	ND	33.3
2-Nitroaniline	ND	ND	33.3
Acenaphthylene	ND	ND	33.3
Dimethyl Phthalate	ND	ND	33.3
2,6-Dinitrotoluene	ND	ND	66.7
Acenaphthene	ND	ND	33.3
3-Nitroaniline	ND	ND	66.7
2,4-Dinitrophenol	ND	ND	166.7
Dibenzofuran	ND	ND	33.3
2,4-Dinitrotoluene	ND	ND	66.7
4-Nitrophenol	ND	ND	133.3
Fluorene	ND	ND	33.3
4-Chlorophenyl Phenyl Ether	ND	ND	33.3
Diethyl Phthalate	ND	ND	33.3
4-Nitroaniline	ND	ND	100.0
2-Methyl-4,6-Dinitrophenol	ND	ND	166.7
N-Nitrosodiphenylamine	ND	ND	33.3
Azobenzene	ND	ND	33.3
4-Bromophenyl Phenyl Ether	ND	ND	66.7
Hexachlorobenzene	ND	ND	66.7
Pentachlorophenol	ND	ND	166.7
Phenanthrene	ND	ND	33.3
Anthracene	ND	ND	33.3
Di-N-Butyl Phthalate	ND	ND	33.3
Fluoranthene	ND	ND	33.3
Pyrene	ND	ND	33.3
Benzidine	ND	ND	100.0
Benzyl Butyl Phthalate	ND	ND	33.3
Benzo (a) Anthracene	ND	ND	33.3

ND = Not Detected.
All values are reported in ug/kg.



	-8-		
CLIENT: ERM Northwest, Inc.		DATE RECEI	
REPORT TO: Pat Wicks		PROJECT NO	
Laboratory Sample Nos.	BNA	823987	
Client Identification	Method Blank	F#1 Hard Pan Comp.	
Compound	Concentra	ation (ug/kg)	DETECTION LIMIT (ug/kg)
Chrysene	ND	ND	33.3
3,3-Dichlorobenzidine	ND	ND	100.0
Bis (2-Ethylhexyl) Phthalate	ND	40.0	33.3
Di-N-Octyl Phthalate	ND	ND	33.3
Benzo (b) Fluoranthene	ND	ND	33.3
Benzo (k) Fluoranthene	ND	ND	33.3
Benzo (a) Pyrene	ND	ND	33.3
Indeno (1,2,3-cd) Pyrene	ND	ND	33.3
Dibenzo (a,h) Anthracene	ND	ND	33.3
Benzo (g,h,i) Perylene	ND	ND	33.3
Surrogate Compounds	Reco	very (%)	AMOUNT
			SPIKED (ug/kg)
2-Fluorophenol	81.	79.	50.
D6-Phenol	69.	71.	50.
D5-Nitrobenzene	69.	51.	25.
O Tilmanah Jahansai			II.

73.

65.

90.

77.

69.

65.

ND = Not Detected.

2-Fluorobiphenyl 2,4,6-Tribromophenol D14-Terphenyl

25.

50.

25.

-9-

CLIENT: ERM Northwest, Inc. REPORT TO: Pat Wicks	DATE RECEIVED: 11/29/8 DATE REPORTED: 12/7/88 PROJECT NO.: 36-88		
Laboratory Sample Nos.	823986 Duplicate	823986	DETECTION LIMIT
Client Identification	F#1 Carbon Block Comp	F#1 Carbon Block Comp	(ug/kg)
Aniline	ND	ND	33.3
2-Chlorophenol	ND	ND	33.3
Bis (2-Chloroethyl) Ether	ND	ND	33.3
Phenol	ND	ND	33.3
1,3-Dichlorobenzene	ND	ND	33.3
1,4-Dichlorobenzene	ND	ND	33.3
1,2-Dichlorobenzene	ND	ND	33.3
Benzyl Alcohol	ND	ND	33.3
Bis (2-Chloroisopropyl) Ether	ND	ND	33.3
2-Methyl Phenol Hexachloroethane	ND	ND	33.3
N-Nitrosodipropylamine	ND ND	ND	33.3
Nitrobenzene	ND	ND ND	33.3 33.3
4-Methylphenol	ND	ND	33.3
Isopherone	ND	ND	33.3
2-Nitrophenol	ND	ND	33.3
2,4-Dimethylphenol	ND	ND	33.3
Bis (2-Chloroethoxy) Methane	ND	ND	33.3
2,4-Dichlorophenol	ND	ND	33.3
1,2,4-Trichlorobenzene	ND	ND	33.3
Naphthalene	ND	ND	166.7
Benzoic Acid	ND	ND	33.3
4-Chloroaniline	ND	ND	33.3
Hexachlorobutadiene	ND	ND	33.3
2-Methylnaphthalene	ND	ND	33.3
4-Chloro-3-Methylphenol	ND	ND	33.3
Hexachlorocyclopentadiene	ND	ND	66.7
2,4,6-Trichlorophenol	ND	ND	66.7
2,4,5-Trichlorophenol	ND	ND	66.7

ND = Not Detected.
All values are reported in ug/kg.



-10-

CLIENT: ERM Northwest, Inc.	DATE RECEIVED: 11/29/88 DATE REPORTED: 12/7/88		
REPORT TO: Pat Wicks		PROJECT NO	
Laboratory Sample Nos.	823986 Duplicate	823986	DETECTION LIMIT
Client Identification	F#1 Carbon Block Comp	F#1 Carbon Block Comp	(ug/kg)
2-Chloronaphthalene	ND	ND	33.3
2-Nitroaniline	ND	ND	33.3
Acenaphthylene	ND	ND	33.3
Dimethyl Phthalate	ND	ND	33.3
2,6-Dinitrotoluene	ND	ND	66.7
Acenaphthene	ND	ND	33.3
3-Nitroaniline	ND	ND	66.7
2,4-Dinitrophenol	ND	ND	166.7
Dibenzofuran	ND	ND	33.3
2,4-Dinitrotoluene	ND	ND	66.7
4-Nitrophenol	ND	ND	133.3
Fluorene	ND	ND	33.3
4-Chlorophenyl Phenyl Ether	ND	ND	33.3
Diethyl Phthalate	ND	ND	33.3
4-Nitroaniline	ND	ND	100.0
2-Methyl-4,6-Dinitrophenol	ND	ND	166.7
N-Nitrosodiphenylamine	ND	ND	33.3
Azobenzene	ND	ND	33.3
4-Bromophenyl Phenyl Ether	ND	ND	66.7
Hexachlorobenzene	ND	ND	66.7
Pentachlorophenol	ND	ND	166.7
Phenanthrene	ND	ND	33.3
Anthracene	ND	ND	33.3
Di-N-Butyl Phthalate	36.7	43.3	33.3
Fluoranthene	ND	ND	33.3
Pyrene	ND	ND	33.3
Benzidine	ND	ND	100.0
Benzyl Butyl Phthalate	ND	ND	33.3
Benzo (a) Anthracene	ND	ND	33.3

ND = Not Detected. All values are reported in ug/kg.



-11-

CLIENT:	ERM Northwest,	Inc.	DATE RECEIVED:	11/29/88
	×.		DATE REPORTED:	12/7/88

REPORT TO: Pat Wicks PROJECT NO.: 36-88

Laboratory Sample Nos. 823986 823986

Duplicate

Client Identification F#1 Carbon F#1 Carbon Block Comp Block Comp

Compound	Concentra	DETECTION LIMIT (ug/kg)	
Chrysene	ND	ND	33.3
3,3-Dichlorobenzidine	ND	ND	100.0
Bis (2-Ethylhexyl) Phthalate	43.3	60.0	33.3
Di-N-Octyl Phthalate	ND	ND	33.3
Benzo (b) Fluoranthene	ND	ND	33.3
Benzo (k) Fluoranthene	ND	ND	33.3
Benzo (a) Pyrene	ND	ND	33.3
Indeno (1,2,3-cd) Pyrene	ND	ND	33.3
Dibenzo (a,h) Anthracene	ND	ND	33.3
Benzo (g,h,i) Perylene	ND	ND	33.3
Surrogate Compounds	Recovery (%)		AMOUNT SPIKED
			(ug/kg)
2-Fluorophenol	82.	72.	50.
D6-Phenol	75.	65.	50.
D5-Nitrobenzene	66.	56.	25.
2-Fluorobiphenyl	81.	74.	25.
2,4,6-Tribromophenol	56.	60.	50.
D14-Terphenyl	86.	71.	25.

ND = Not Detected.





CLIENT: ERM Northwest, Inc.

DATE RECEIVED: 11/29/88

REPORT TO: Pat Wicks

DATE REPORTED: 12/7/88

PROJECT NO.: 36-88

Laboratory Sample No.

823987 Spike

Client Identification

F#1 Hard Pan Comp

Matrix Spike Compounds	Recovery (%)	Amount Residue (ug/g)	Amount Expected (ug/g)
2-Clorophenol	74.	37.2	50.
Phenol	74.	36.9	50.
1,4-Dichlorobenzene	66.	16.4	25.
N-Nitrosodipropylamine	60.	14.9	25.
1,2,4-Trichlorobenzene	58.	14.5	25.
4-Chloro-3-Methylphenol	70.	34.9	50.
Acenaphthene	84.	20.9	25.
2,4-Dinitrotoluene	31.	7.7	25.
4-Nitrophenol	30.	14.9	50.
Pentachlorophenol	46.	23.1	50.
Di-N-Butylphthalate	72.	18.	25.
Pyrene	94.	23.6	25.
Surrogate Compounds	Recovery (%)	Amount Spiked (ug/kg)	Amount Found (ug/kg)
2-Fluorophenol	88.	50.	43.9
D6-Phenol	80.	50.	39.9
D5-Nitrobenzene	60.	25.	15.
2-Fluorobiphenyl	82.	25.	20.4
2,4,6-Tribromophenol	84.	50.	41.8
D14-Terphenyl	96.	25.	24.
DATE REPORTED	Jan O	John	



CLIENT: ERM Northwest, Inc.		DATE RECEIVED: 11/29/88 DATE REPORTED: 12/7/88		
REPORT TO: Pat Wicks		PROJECT NO	.: 36-88	
Laboratory Sample Nos.	BNA Method	823987 F#1 Hard	DETECTION LIMIT	
Client Identification	Blank	Pan Comp.	(ug/kg)	
Aniline 2-Chlorophenol	ND ND	ND ND	33.3	
Bis (2-Chloroethyl) Ether	ND	ND	33.3	
Phenol	ND	ND	33.3	
1,3-Dichlorobenzene	ND	ND	33.3	
/1,4-Dichlorobenzene	ND	ND	33.3	
/1,2-Dichlorobenzene	ND	ND	33.3	
Benzyl Alcohol	ND	ND	33.3	
/Bis (2-Chloroisopropyl) Ether	. ND	ND	33.3	
√2-Methyl Phenol	ND	ND	33.3	
/Hexachloroethane	ND	ND	33.3	
M-Nitrosodipropylamine \\	ŭ¦ ∪ <b>₹ h</b> D	ND	33.3	
Nitrobenzene \ \\\ \\\\\\\\\\\\\\\\\\\\\\\\\\\	ND	ND	33.3	
√4-Methylphenol  √1/√√2  √1/√2  √2  √3  √4-Methylphenol  √1/√2  √3  √4  √4  √4  √4  √4  √4  √4  √4  √4	Y ND	ND	33.3	
Asopherone \(\tau\)	× OX.ND	ND	33.3	
2-Nitrophenol	/ O. S. ND	ND	33.3	
2,4-Dimethylphenol  Bis (2-Chloroethoxy) Methane	TX F ND	ND	33.3	
· (- · · · · · · · · · · · · · · · ·	v -:-	ND	33.3	
2,4-Dichlorophenol	ND	ND	33.3	
√1,2,4-Trichlorobenzene	ND	ND	33.3	
Waphthalene	ND	ND	166.7	
Æenzoic Acid	ND	ND	33.3	
4-Chloroaniline	not ND	ND	33.3	
A-Chloroaniline  Whexachlorobutadiene  A-Methylnaphthalene	ND	ND	33.3	
	70770	ND	33.3	
A-Chloro-3-Methylphenol ✓	ND	ND	33.3	
√Hexachlorocyclopentadiene	ND	ND	66.7	
2,4,6-Trichlorophenol	ND	ND	66.7	
√2,4,5-Trichlorophenol	ND	ND	66.7	

ND = Not Detected. All values are reported in ug/kg.



CLIENT: ERM Northwest, Inc. REPORT TO: Pat Wicks		DATE RECEIVED: 11/29/88 DATE REPORTED: 12/7/88 PROJECT NO.: 36-88	
Laboratory Sample Nos.	BNA Method	823987 F#1 Hard	DETECTION LIMIT
Client Identification	Blank	Pan Comp.	(ug/kg)
√2-Chloronaphthalene	ND	ND	33.3
✓2-Nitroaniline	ND	ND	33.3
✓Acenaphthylene	ND	ND	33.3
Dimethyl Phthalate	ND	ND	33.3
√2,6-Dinitrotoluene	ND	ND	66.7
Acenaphthene	ND	ND	33.3
✓3-Nitroaniline	ND	ND	66.7
√2,4-Dinitrophenol	ND	ND	166.7
√Dibenzofuran	ND	ND	33.3
→2,4-Dinitrotoluene	ND	ND	66.7
4-Nitrophenol	ND	ND	133.3
✓Fluorene	. \ ND	ND	33.3
← Chlorophenyl Phenyl Ether	ND O	ND	33.3
Diethyl Phthalate	ND	ND	33.3
4-Nitroaniline	ND /	ND	100.0
✓2-Methyl-4,6-Dinitrophenol \	ND	ND	166.7
N-Nitrosodiphenylamine	ND	ND	33.3
Azobenzene	. / ND	ND	33.3
4-Bromophenyl Phenyl Ether	ND	ND	66.7
Hexachlorobenzene	XV X ND	ND	66.7
/Pentachlorophenol	ND ND	ND	166.7
Whenanthrene . ~	" X ND	ND	33.3
Anthracene (N	× 3 ND	ND	33.3
✓Di-N-Butyl Phthalate	ND ND	ND	33.3
Fluoranthene D	OND	ND	33.3
√Pyrene J	ND ND	ND	33.3
✓Benzidine	ND ND	ND	100.0
Benzyl Butyl Phthalate	ND	ND	33.3
Benzo (a) Anthracene	ND	ND	33.3
517		<del>-                                    </del>	5.5 5.4.

ND = Not Detected. All values are reported in ug/kg.



DATE RECEIVED: CLIENT: ERM Northwest, Inc. 11/29/88 DATE REPORTED: 12/7/88 PROJECT NO.: 36-88 REPORT TO: Pat Wicks Laboratory Sample Nos. BNA 823987 Method F#1 Hard Client Identification Blank Pan Comp. DETECTION Concentration (ug/kg) Compound LIMIT (ug/kg) ∨Chrysene ND ND 33.3 √3,3-Dichlorobenzidine ND ND 100.0 Fis (2-Ethylhexyl) Phthalate ND 40.0 33.3 √Di-N-Octyl Phthalate ND ND 33.3 Benzo (b) Fluoranthene Benzo (k) Fluoranthene ND 33.3 ND ND ND 33.3 ∠Benzo (a) Pyrene ND ND 33.3 ✓Indeno (1,2,3-cd) Pyrene ND ND 33.3 ✓Dibenzo (a,h) Anthracene ND ND 33.3 Æenzo (g,h,i) Perylene ND ND 33.3 Surrogate Compounds Recovery (%) THUOMA SPIKED (ug/kg) 2-Fluorophenol 81. 79. 50. D6-Phenol 69. 71. 50. D5-Nitrobenzene 69. 51. 25. 2-Fluorobiphenyl 73. 77. 25. 2,4,6-Tribromophenol 65. 69. 50. D14-Terphenyl 90. 65. 25. Colore prison de de de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la colore de la color ND = Not Detected.

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CLIENT: ERM Northwest, Inc. REPORT TO: Pat Wicks	DATE RECEIVED: 11 DATE REPORTED: 12 PROJECT NO.: 36-8		
Laboratory Sample Nos. Client Identification	823986 Duplicate F#1 Carbon Block Comp	823986 F#1 Carbon Block Comp	DETECTION LIMIT (ug/kg)
		Brock Comp	
✓Aniline	ND	ND	33.3
~2-Chlorophenol	ND	ND	33.3
✓Bis (2-Chloroethyl) Ether	ND	ND	33.3
Phenol	ИD	ND	33.3
1,3-Dichlorobenzene	ND	ND	33.3
√1,4-Dichlorobenzene	ND	ND	33.3
1,2-Dichlorobenzene	ND	ND	33.3
Benzyl Alcohol	ND	ИD	33.3
Bis (2-Chloroisopropyl) Ether	ND	ND	33.3
2-Methyl Phenol	ND	ND	33.3
-Hexachloroethane	ND	ND	33.3
Nitrosodipropylamine	ND	ND	33.3
Nitrobenzene	ND	ИD	33.3
-Methylphenol	ND	ND	33.3
/Isopherone	иD	ND	33.3
√2-Nitrophenol	ND	ND	33.3
√2,4-Dimethylphenol	ND	ND	33.3
✓Bis (2-Chloroethoxy) Methane	ND	ND	33.3
✓2,4-Dichlorophenol	ND	ND	33.3
X,2,4-Trichlorobenzene	ND	ND	33.3
Naphthalene	ND	ND	166.7
Benzoic Acid	ND	ND	33.3
√4-Chloroaniline	ND	ND	33.3
Mexachlorobutadiene	ND	ND	33.3
2-Methylnaphthalene	ND	ND	33.3
√ 4-Chloro-3-Methylphenol	ND	ND	33.3
✓Hexachlorocyclopentadiene	ND	ND	66.7
√2,4,6-Trichlorophenol	ND	ND	66.7
√2,4,5-Trichlorophenol	ND	ND	66.7

ND = Not Detected. All values are reported in ug/kg.



DATE RECEIVED:

11/29/88

REPORT TO: Pat Wicks		DATE REPORTED: 12/7/88 PROJECT NO.: 36-88		
Laboratory Sample Nos.	823986 Duplicate	823986	DETECTION LIMIT	
Client Identification	F#1 Carbon Block Comp	F#1 Carbon Block Comp	(ug/kg)	
✓2-Chloronaphthalene ✓2-Nitroaniline	ND ND	ND ND	33.3 33.3	
'Acenaphthylene	ND	ND	33.3	
✓Dimethyl Phthalate	ND	ND	33.3	
√2,6-Dinitrotoluene	ND	ND	66.7	
Acenaphthene	ND	ND	33.3	
✓3-Nitroaniline	ND	ND	66.7	
√2,4-Dinitrophenol	ND	ND	166.7	
Dibenzofuran	ND	ND	33.3	
₩2,4-Dinitrotoluene	ND	ND	66.7	
✓4-Nitrophenol	ND	ND	133.3	
#luorene	ND	ND	33.3	
✓4-Chlorophenyl Phenyl Ether	ND	ND	33.3	
✓Diethyl Phthalate	ND	ND	33.3	
<pre>/4-Nitroaniline</pre>	ND	ND	100.0	
√2-Methyl-4,6-Dinitrophenol	ND	ND	166.7	
N-Nitrosodiphenylamine	ND	ND	33.3	
Azobenzene	ND	ND	33.3	
✓4-Bromophenyl Phenyl Ether	ND	ND	66.7	
~Hexachlorobenzene	ND	ND	66.7	
▶Pentachlorophenol	ND	ND	166.7	
Phenanthrene	ND	ND	33.3	
-Anthracene	ND	ND	33.3	
✓Di-N-Butyl Phthalate	36.7	43.3	33.3	
√Fluoranthene	ND	ND	33.3	
Pyrene	ND	ND	33.3	
✓Benzidine	ND	ND	100.0	
Benzyl Butyl Phthalate	ND	ND	33.3	
→Benzo (a) Anthracene	ND	ND	33.3	

ND = Not Detected.
All values are reported in ug/kg.

CLIENT: ERM Northwest, Inc.

CLIENT: ERM Northwest, Inc.

DATE RECEIVED: 11/29/88
DATE REPORTED: 12/7/88

PROJECT NO.: 36-88

REPORT TO: Pat Wicks

.....

Laboratory Sample Nos.

823986

823986

Client Identification

Duplicate

F#1 Carbon F#1 Carbon Block Comp

Compound	Concentra	Concentration (ug/kg)		
Chrysene  3,3-Dichlorobenzidine  Bis (2-Ethylhexyl) Phthalate  Di-N-Octyl Phthalate  Benzo (b) Fluoranthene  Benzo (k) Fluoranthene  Benzo (a) Pyrene	ND ND 43.3 ND ND ND ND ND ND	ND ND 60.0 ND ND ND ND ND ND	33.3 100.0 33.3 33.3 33.3 33.3 33.3	
✓Indeno (1,2,3-cd) Pyrene  ✓Dibenzo (a,h) Anthracene  ✓Benzo (g,h,i) Perylene	ND ND	ND ND	33.3	
Surrogate Compounds	Recov	ery (%)	AMOUNT SPIKED	

			SPIKED (ug/kg)	
2-Fluorophenol	82.	72.	50.	
D6-Phenol	75.	65.	50.	
D5-Nitrobenzene	66.	56.	25.	
2-Fluorobiphenyl	81.	74.	25.	
2,4,6-Tribromophenol	56.	60.	50.	
D14-Terphenyl	86.	71.	25.	

ND = Not Detected.

unia cuty extra copy as needed (warehouse)

CLIENT: ERM Northwest, Inc.

DATE RECEIVED: 11/29/88

DATE REPORTED: 12/7/88

PROJECT NO.: 36-88

REPORT TO: Pat Wicks

823987 Spike

Laboratory Sample No. Client Identification

F#1 Hard Pan Comp

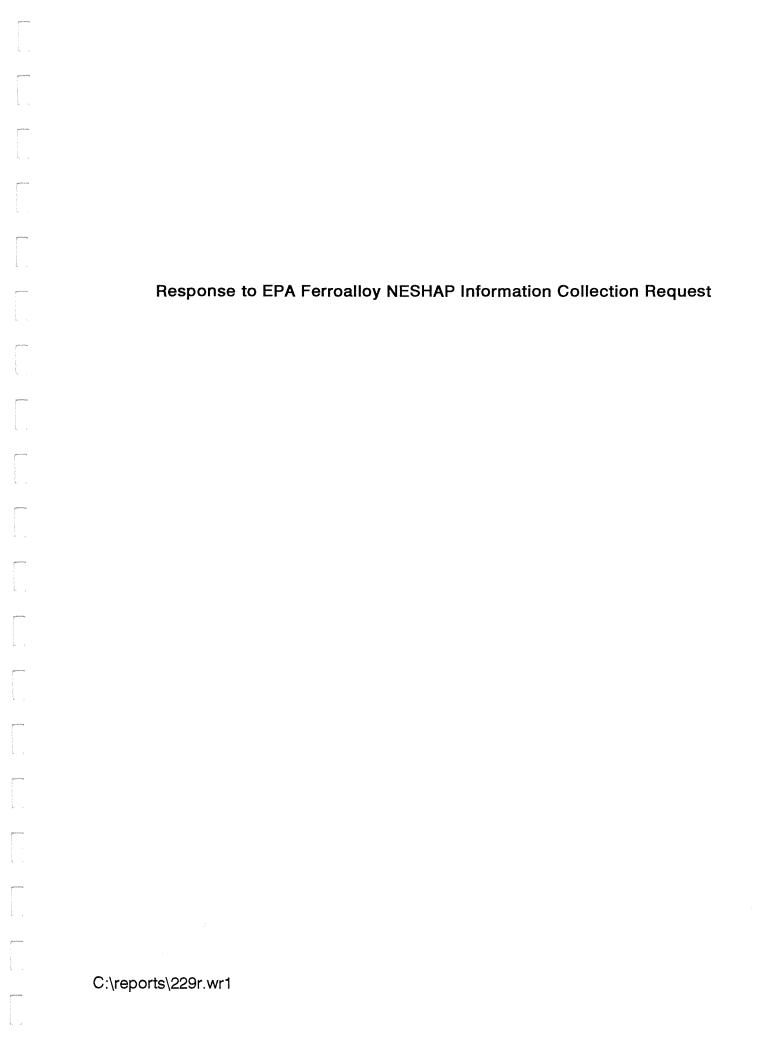
Matrix Spike Compounds	Recovery (%)	Amount Residue (ug/g)	Amount Expected (ug/g)
2-Clorophenol	74.	37.2	50.
Phenol	74.	36.9	50.
1,4-Dichlorobenzene	66.	16.4	25.
N-Nitrosodipropylamine	60.	14.9	25.
1,2,4-Trichlorobenzene	58.	14.5	25.
4-Chloro-3-Methylphenol	70.	34.9	50.
Acenaphthene	84.	20.9	25.
2,4-Dinitrotoluene	31.	7.7	25.
4-Nitrophenol	30.	14.9	50.
Pentachlorophenol	46.	23.1	50.
Di-N-Butylphthalate	72.	18.	25.
Pyrene	94.	23.6	25.
Surrogate Compounds	Recovery (%)	Amount Spiked (ug/kg)	Amount Found (ug/kg)
2-Fluorophenol	88.	50.	43.9
D6-Phenol	80.	50.	39.9
D5-Nitrobenzene	60.	25.	15.
2-Fluorobiphenyl	82.	25.	20.4
2,4,6-Tribromophenol	84.	50.	41.8
D14-Terphenyl	96.	25.	24.
		Wind	

DATE REPORTED

JAH/pb

THE WILL THE SE WITH BUT LAND

John A. Hicks



October 21, 1992



Mr. Conrad K. Chin Environmental Engineer Industrial Studies Branch Office of Air Quality Planning and Standards U S Environmental Protection Agency Research Triangle Park, N C 27711

Dear Mr. Chin:

Enclosed are the completed NESHAP request for information with Tables 1, 2 (3 sets), 3-A, 3-B and 4, Figure 1 (process flow diagram) and seven page process flow calculations summary for Silicon Metaltech, Inc. at Rock Island, Washington.

This information is submitted in response to EPA's letter of July 9, 1992 from Mr. Bruce C. Jordan of your office to Mr. Jim Trunzo of Silicon Metaltech, Inc.

We appreciate extensions of your deadline that were allowed in your August 18 letter and subsequent telephone conversations.

We hope this information will be satisfactory for your purposes. Please do not hesitate to call me or Mr. Trunzo if there are any questions.

Sincerely,

Patrick H. Wicks, PE, CHMM

President

Enclosures

cc: Mr. Jim Trunzo

Silicon Metaltech, Inc.

Enclosures 100 4th Street

Rock Island, Washington 98850

Mr. Bruce C. Jordan

Enclosures Office of Air Quality Planning and Standards

U S Environmental Protection Agency Research Triangle Park, N C 27711

# Enclosure 1

Ferroalloy Industry NESHAP Information Collection Request

U.S. Environmental Protection Agency Office of Air Quality Planning and Standards Emissions Standards Division

June 1992

For questions, contact Jeffrey Telander at (919) 541-5427

Form Approved OMB No. 2060-0239 Approval Expires: 03/31/95

Public reporting burden for this collection of information is estimated to average 85 hours per response, including the time for reviewing instructions, searching existing data sources, gathering and maintaining the data needed, and completing and reviewing the collection of information. This estimate is based on reported response times of pretest surveys, which ranged from 3 to 717 hours per facility. The majority of facilities estimated that it took less than 100 hours to complete the survey. Send comments regarding this burden estimate or any other aspect of this collection of information, including suggestions for reducing the burden, to Chief, Information Policy Branch. PM-223Y, U. S. Environmental Protection Agency, 401 M Street, S.W., Washington, D.C. 20460, and to Office of Information and Regulatory Affairs, Office of Management and Budget, Washington, D.C. 20503. Include the OMB number in any correspondence.

# MAXIMUM ACHIEVABLE CONTROL TECHNOLOGY (MACT) STANDARDS DEVELOPMENT INFORMATION REQUEST

#### I. Instructions

This information request is to be completed for operations that comprise ferroalloys production at your plant. The ferroalloys production source category includes any facility engaged in producing ferroalloys such as ferrosilicon, silicon metal, ferromanganese, and ferrochrome.

A ferroalloy is typically an alloy of iron and one or more other elements, such as silicon, manganese, or chromium; however, the ferroalloy source category also includes metals that don't contain iron but are used in the primary metals industry. Ferroalloys are typically used as additives to impart unique properties to steel and cast iron.

Ferroalloys production includes, but is not limited to, the following operations: ore unloading, ore storage, sintering, crushing, weigh-feeding, smelting, tapping, casting, and screening. The smelting process can be performed in a variety of furnace types including, but not limited to: submerged arc furnaces, induction furnaces, vacuum furnaces, exothermic reaction furnaces, and electrolytic cells.

We are requesting information regarding each compound identified as a hazardous air pollutant (HAP) that is used in or emitted by any operations, including fugitive emission sources, occurring from the furnaces, ore storage and handling, and product storage and handling operations at your facility. Fill out this information request as completely as possible from existing information. At a minimum, provide (1) information on the presence of HAP emissions and (2) HAP emission estimates based on previously obtained test data or on engineering calculations provided there is a basis for such calculations. No additional monitoring or emission testing is required by your company to respond to this request.

For your convenience, we have provided in Attachment 1 additional information on the scope and purpose of this survey. Respondents should read this material before attempting to complete the survey. Attachment 2 is a copy of an example figure and example tables for the survey. Refer to these examples in completing your response. If you have any questions regarding this request, please contact Jeffrey Telander at (919) 541-5427.

Return this information request and any additional information to:

Emission Standards Division (MD-13)
U. S. Environmental Protection Agency
Office of Air Quality Planning and Standards
Research Triangle Park, NC 27711

Attention: Bruce C. Jordan, Director

#### II. General Information

Α.	Name of legal owner of plant: SILICON METALTECH, INC.
в.	Name of legal operator of plant, if different from legal owner:
c.	Address of legal owner/operator (please specify which): 100 - 4th Street  ROCK ISLAND, WA 98850

### D. Size of Company

1. Approximate number of employees of the business enterprise that owns this plant, including where applicable, the parent company and all

subsidiaries, branches, and unrelated establishments owned by the parent company (answer may be given using the following ranges: 0-100; 101-250; 251-500; 501-750; 751-1,000; 1,001-1,500; and >1,500)

0 - 100

2.	Number	of	plant	employees	attached	to	the
	ferroa]	Lloy	opera	ation:			

0 - 100

Ε.	Name	of	plant:	SILICON METALTECH

- F. Street address of plant: Same as legal owner
- G. Latitude and longitude coordinates of plant (see Appendix A of Attachment 1):

47° 21' 55" North, 120° 8' 20" West

H. Name of contact(s) able to answer technical questions about the completed survey:

Jim Trunzo, President, 509-884-4009

- I. Title(s): Pat Wicks, Consultant, 206-485-3437
- J. Telephone Number: ( ) \_\_\_\_\_\_

#### III. Plant Operations

Complete Table 1 (page 8) for the most recent calendar year (unless the respondent can justify selection of an alternate base year) for all processes at your plant that are covered by the ferroalloy source category. For each type of process (i.e., process line), provide a process flow diagram that includes all sources of air emissions (e.g., stack emissions, process fugitive emissions, and area fugitive emissions [including fugitive dust emissions]). Also include all activities that generate HAP emissions, including the storage, transfer, handling, and processing of the materials, and wastewater and solid waste handling. Indicate all feedstocks, products, and emissions that contain compounds that are listed in Table 2 (pages 9-15), below. Use the same terminology/codes in identifying unit operations and emissions points in this figure as you will use in completing Table 3 (pages 16-17), below.

B. List the products, coproducts, and by-products identified in the process flow diagram and indicate for each how much is produced annually.

Si Metal (@98.9% Si)		13,076 tons in CY1991
Dross	_	1,364 tons in CY1991
Fume		3,357 tons in CY1991

#### IV. HAP's--Usage and Emissions

- A. Complete Table 2 (pages 9-15) for each emission point identified in the process flow diagram(s) developed for Part III.A., above. For each HAP listed on the table, indicate the likelihood, using the codes defined in Table 2, that the HAP is emitted from a given emission point within the source category. Identify the appropriate emission points using the same terminology/codes you used in completing the process flow diagram(s) in Part III.A, above.
- B. Using copies of Tables 3A (page 16) and 3B (page 17), complete the table for each process and emission point identified in Part III, with the following exceptions.
  - For those emission points from units with Resource Conservation and Recovery Act (RCRA) Part B permits, it is not necessary to complete Table 3 for wastewater and solid waste handling operations;
  - 2. Sources with no air pollution capture or control systems will only complete columns 1-3, 8, and 10 of Table 3-A; and
  - 3. Provide HAP data only for those HAP's identified with code "A" in Table 2, above.
- C. Complete Table 4 (pages 18-25) for any air pollution capture or control equipment identified in Table 3, above.
- D. For calculations based on emission factors, material balances, or engineering principles, submit a step-by-step description of the calculations, including assumptions used, and a brief rationale for the

validity of the calculation method used. (See guidance documents listed in Attachment 1, Section IV). If test reports are listed as the basis for emissions estimates or capture system and control device efficiencies, provide a brief summary of the relevant tests. Include information such as the purpose of the test, when it was conducted, what test methods were used, and information on the process operation during the test. It is not necessary to submit copies of actual test reports at this time although EPA may request additional documentation on a plant-specific basis in the future.

V. Factors That Affect HAP Emission Reductions

Completion of Section V is optional. If you choose to respond, clearly distinguish between pollution reduction and source reduction measures. Pollution reduction measures alter the physical, chemical, or biological characteristics or the volume of a HAP through a process or activity which itself is not integral to and necessary to produce a product or provide a service. The use of "addon" devices to capture and control (recover or destroy) HAP emissions are considered pollution reduction measures. contrast, source reduction measures reduce the amount of any HAP prior to recycling, treatment, or disposal. reduction measures include equipment or technology modifications, process or procedure modifications, reformulation or redesign of products, substitution of raw materials, and improvements in housekeeping, maintenance, training, or inventory control.

A. For each unit operation for which pollution reduction or source reduction measures have resulted in a decrease in HAP emissions since 1987, provide the following information (use separate sheets if necessary).

1.	Name	of	unit opera	tion:					
2.	Type	of	control or	descr	iption	of	process	change:	

B. If recovery or recycling of feedstocks is used, quantify the effect of the program (e.g., estimated

		recovery/recycling compared to actual annual purchase):
	c.	Are you aware of any alternative processes (feedstock substitutions or eliminations) or control devices that could result in fewer impacts transferred between environmental media (water, air, and land) or reduced total release to all environmental media (e.g., reduced wastewater or solid waste)? Discuss whether these processes could be adapted to the ferroalloy source category and any experience you have with them.  Yes; DC Power Technology utilizing closed furnace design
		commercialization expected to be 5 to 10 years away.
VI.	Mis	cellaneous
	Α.	If any control or process change described in Part V was instituted as a result of new source review requirements pursuant to 40 CFR 51.160, Subpart I, Review of New Sources and Modifications, provide the date at which the lowest achievable emission rate (LAER) came into effect:
	в.	Describe any factors not addressed in the above questions that might serve to distinguish your facility from others in this source category for purposes of developing a separate source category or subcategory and MACT standards.

TABLE 1. SUMMARY OF PROCESS LINES FOR FERROALLOY SOURCE CATEGORY

		Average annual	Operation	ng cycle	Maximum annual	Age of	Remaining economic life of
Process lines using and/or emitting HAP's	No. of lines	production, tons/year <sup>a</sup>	hr/d	d/yr	production capacity, tons/yr <sup>a</sup>	furnace, years	furnace, years
Silicon Mfg.	1	13,076	24	35,8		F1= 50	F1=3-10
						F2= 50	F2=3-10
					8	F3= 50	F3=3-10
				<del> </del>			

<sup>&</sup>lt;sup>a</sup>Provide production in appropriate units, e.g., tons/year.

 $\infty$ 

<sup>\*</sup> Shorter life is if little or no money spent to maintain; longer life if moderate investment to keep up furnaces.

TABLE 2. PRESENCE OF HAZARDOUS AIR POLLUTANTS IN EMISSION POINTSa

CHEMICAL NAME	Emission Points ->	E1	E2	E3	CS1	F1	E4	CS2	F2	E5				
ACETALDEHYDE		<u> </u>		<u> </u>										
ACETAMIDE			A 27							A MARKEY Par			100000000	BARROTTE A
ACETONITRILE														
ACETOPHENONE													- 7	
2-ACETYLAMINOFLUORENE														
ACROLEIN														
ACRYLAMIDE														
ACRYLIC ACID												2 52	il Wessells	
ACRYLONITRILE													1.	
ALLYL CHLORIDE													15%	
4-AMINOBIPHENYL												-		′
ANILINE		4			Ь	15				B				
o-ANISIDINE							1							
ASBESTOS		0200	1 1228 8							THE WAY				
BENZENE(INCLUDING BENZENE FROM	GASOLINE)				1,	9								
BENZIDINE					13	13				13			l karen	
BENZOTRICHLORIDE						1			(6)					
BENZYL CHLORIDE		e estado esta	1							foli Specialists		-0:20		
BIPHENYL													•	
BIS(2-ETHYLHEXYL)PHTHALATE (DEH	P)				13	₽,		11	1		1	T		
BIS(CHLOROMETHYL)ETHER														
BROMOFORM	10 00 00 00 00	A same	. Lawrence		'n	13		News				1		
1,3-BUTADIENE														
CALCIUM CYANAMIDE	•	7												
CAPROLACTAM						-4								
CAPTAN	5 H2 152 1							11910,551195	1	a second	1,523.5			
CARBARYL			A. C. T. C.											
CARBON DISULFIDE	and the second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second second s					_	1-	$\neg$		1	1	1	1	1

<sup>&</sup>lt;sup>a</sup>For each HAP emission point defined in the process flow diagram, write in the applicable letter code defined below:

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C - No reason or data to assume that this HAP is emitted.

D - HAP is believed to be emitted, but no data exists.

TABLE 2. PRESENCE OF HAZARDOUS AIR POLLUTANTS IN EMISSION POINTSª

Process name (as defined on process flow diagram); Raw Materials Handlin	g & S	ilicon	Arc Fu	rnaces	1-3								1
Emission Points ->	E1	E2	E3	CS1	F1	E4	CS2	F2	E5			1	T
CARBON TETRACHLORIDE				15	1/2		1					1	
CARBONYL SULFIDE	la seeste la ne											- 21	77 10
CATECHOL													
CHLORAMBEN							1						8
CHLORDANE				15	(5								
CHLORINE		00222000		600 800									
CHLOROACETIC ACID													
2-CHLOROACETOPHENONE													
CHLOROBENZENE				Ĵρ	12								
CHLOROBENZILATE													700
CHLOROFORM				j-n	i's								.=
CHLOROMETHYL METHYL ETHER													
CHLOROPRENE													
CRESOLS/CRESYLIC ACID (ISOMERS AND MIXTURE)				1.	Ĭ-,								
o-CRESOL													
m-CRESOL													
p-CRESOL													
CUMENE												77	
2,4-D, SALTS AND ESTERS		358,4000	1.WS	971	2 00500	HAMMAR N							
DDE .				1.	i								
DIAZOMETHANE													
DIBENZOFURANS				Ŋ,	10								
1,2-DIBROMO-3-CHLOROPROPANE													
DIBUTYLPHTHALATE				B	17-					199			1
1,4-DICHLOROBENZENE(P)				Ţ.	[4								
3,3-DICHLOROBENZIDENE				1/2	1.1					5. 52853			
DICHLOROETHYL ETHER (BIS(2-CHLOROETHYL)ETHER)				15	:3							T	1
1,3-DICHLOROPROPENE				j.	14	T	1						T
DICHLORVOS	1	_			1	1				1	1	1	1

<sup>&</sup>lt;sup>a</sup>For each HAP emission point defined in the process flow diagram, write in the applicable letter code defined below:

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## TABLE 2. PRESENCE OF HAZARDOUS AIR POLLUTANTS IN EMISSION POINTSa

Emission Points ->	E1	E2	E3	CS1	F1	E4	CS2	F2	E5		T		7.1
DIETHANOLAMINE	Ī	1	I	I	ı	Ī	1	Ī				- 1	
N,N-DIETHYL ANILINB (N,N-DIMETHYLANILINE)													
DIETHYL SULFATE													
3,3-DIMETHOXYBENZIDINE													
DIMETHYL AMINOAZOBENZENE													SW-II-AL
3,3'-DIMETHYL BENZIDINE													
DIMETHYL CARBAMOYL CHLORIDE		20 EU		2000						12			es tr
DIMETHYL FORMAMIDE													
1,1-DIMETHYL HYDRAZINE	24 1707												
DIMETHYL PHTHALATE				13	12				15				
DIMETHYL SULFATE													1
4,6-DINITRO-O-CRESOL, AND SALTS	-13		-72										. B
2,4 DINITROPHENOL				10	B				B				
2,4-DINITROTOLUENE				1 75	17				13				
1,4-DIOXANE(1,4-DIETHYLENEOXIDE)													
1,2-DIPHENYLHYDRAZINE				53400									
EPICHLOROHYDRIN(1-CHLORO-2,1-EPOXYPROPANE)													
1,2-EPOXYBUTANE				12.000							W		
ETHYL ACRYLATE													
ETHYL BENZENE	530			13	i <sup>2</sup>	,				15 M.		- 1	
ETHYL CARBAMATE (URETHANE)													
ETHYL CHLORIDE (CHLOROETHANE)				Y	·   19							•	7. (3.00
ETHYLENE DIBROMIDE (DIBROMOETHANE)													
ETHYLENE DICHLORIDE (1,2-DICHLOROETHANE)				1,	, i	7							
ETHYLENE GLYCOL													1
ETHYLENE IMINE(AZIRIDINE)													
ETHYLENE OXIDE													
ETHYLENE THIOUREA													
ETHYLIDENE DICHLORIDE (1,1-DICHLOROETHANE)				17					- P		1		

<sup>&</sup>lt;sup>a</sup>For each HAP emission point defined in the process flow diagram, write in the applicable letter code defined below:

- A Specific HAP is known to be emitted.
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- C No reason or data to assume that this HAP is emitted.
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TABLE 2. PRESENCE OF HAZARDOUS AIR POLLUTANTS IN EMISSION POINTSa

Process name (as defined on process flow diagram): _ Raw Materials Handlin	ng & S	ilicon	Arc Fu	rnaces	1-3		20: -Wa (22-Va)-	4 C. 18 A. 14 AV		was residual.		38************************************	
Emission Points ->	E1	E2	E3	CS1	F1	E4	CS2	F2	E5		377373		
FORMALDEHYDE				Π									l progression.
HEPTACHLOR .				1	1,	X-12-					4		
HEXACHLOROBENZENE				(4)	P				F				
HEXACHLOROBUTADIENE		201	100000	15	13				13				
HEXACHLOROCYCLOPENTADIENE				3	13				13				
HEXACHLOROETHANE				1/2	17,				8			5285 - 1	
HEXAMETHYLENE-1,6-DIISO-CYANATE													
HEXAMETHYLPHOSPHORAMIDE						473.55		A					
HEXANE					35/								
HYDRAZINE		847						100000					
HYDROCHLORIC ACID													
HYDROGEN FLUORIDE (HYDROFLUORIC ACID)													
HYDROGEN SULFIDE													(5.004lk
HYDROQUINONE									S-420-1.1				F\$4.00991
ISOPHORONE													
LINDANE (ALL ISOMERS)				13	13	All All All All All All All All All All							
MALEIC ANHYDRIDE .													
METHANOL						Tallens T			8 (47 - 24 i 34) 1				
METHOXYCHLOR	1												
METHYL BROMIDE (BROMOMETHANE)		New Control of the		7:	1.1		es elemente el						
METHYL CHLORIDE (CHLOROMETHANE)				15	12,				1				
METHYL CHLOROFORM (1,1,1-TRICHLOROETHANE)					G								
METHYL ETHYL KETONE (2-BUTANONE)				i	1								*
METHYL HYDRAZINE													
METHYL: IODIDE (IODOMETHANE)													
METHYL ISOBUTYL KETONE (HEXONE)				1	is		-V-2		- 500			100 000	
METHYL ISOCYANATE													
METHYL METHACRYLATE											55-15264		
METHYL TERT BUTYL ETHER	1		1	1			10	1	1			1	Г

<sup>&</sup>lt;sup>a</sup>For each HAP emission point defined in the process flow diagram, write in the applicable letter code defined below:

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TABLE 2. PRESENCE OF HAZARDOUS AIR POLLUTANTS IN EMISSION POINTSa

Raw Materials Handling & Silicon Arc Furnaces 1-3

Emission Point	ts -> E1	E2	E3	CS1	F1	E4	CS2	F2	E5				T
4,4-METHYLENE BIS(2-CHLOROANILINE)													
METHYLENE CHLORIDE (DICHLOROMETHANE)			7.58.80	B	7-								
METHYLENE DIPHENYL DIISOCYANATE (MDI)													
4,4'-METHYLENEDIANILINE						J-176325-135			3778				
NAPHTHALENE				1/2	15				13				
NITROBENZENE				10	17				13				nem-car
4-NITROBIPHENYL													
4-NITROPHENOL				1 1/2	1.3			67 IC.	13				
2-NITROPROPANE											ENVIOLENCE		
N-NITROSO-N-METHYLUREA				į.				ij					
N-NITROSODIMETHYLAMINE				(5)	12.				100 TV 100		Comment		
N-NITROSOMORPHOLINE								S. C.					
PARATHION													
PENTACHLORONITROBENZENE (QUINTOBENZENE)			1000				10.7		525.00	2000 (0000)			į
PENTACHLOROPHENOL				į.	12				B				
PHENOL			1 420	13	i's	877	0,00		3				
p-PHENYLENEDIAMINE						Vac us							
PHOSGENE						E7/ES		1000		S.1. (Ave. 1 m.)			Parko R
PHOSPHINE													l.
PHOSPHOROUS										(3)			
PHTHALIC ANHYDRIDE													Larconner
POLYCHLORINATED BIPHENYLS (AROCHLORS)			) (iii									10000	260
1,3-PROPANE SULTONE				1									
BETA-PROPIOLACTONE													
PROPIONALDEHYDE													
PROPOXUR (BAYGON)													
PROPYLENE DICHLORIDE (1,2-DICHLOROPROPANE)				[2]	(°,								
PROPYLENE OXIDE				AF ENGLISHES									
1,2-PROPYLENIMINE (2-METHYL AZIRIDINE)				1		T							

<sup>&</sup>lt;sup>a</sup>For each HAP emission point defined in the process flow diagram, write in the applicable letter code defined below:

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TABLE 2. PRESENCE OF HAZARDOUS AIR POLLUTANTS IN EMISSION POINTSa

Process name (as defined on process flow diagram): Raw Materials Hand			Arc Fu	ırnaces	1-3						
Emission Points -	> E1	E2	E3	C31	F1	E4	CS2	F2	E5	$\top$	T
QUINOLINE						1				7	T
QUINONE											
STYRENE				15	17-	100					
STYRENE OXIDE											
2,3,7,8-TETRACHLORODIBENZO-p-DIOXIN						2200				WE   10	
1,1,2,2-TETRACHLOROETHANE				(')	1 2						
TETRACHLOROETHYLENE (PERCHLOROETHYLENE)					1.						l legat
TITANIUM TETRACHLORIDE											
TOLUENE				(5)	13						
2,4-TOLUENE DIAMINE									41		
2,4-TOLUENE DIISOCYANATE											
o-TOLUIDINE					eracent A						
TOXAPHENE (CHLORINATED CAMPHENE)				13	17						
1,2,4-TRICHLOROBENZENE				1 (3	16						
1,1,2-TRICHLOROETHANE				1.2	1,2						
TRICHLOROETHYLENE											
2,4,5-TRICHLOROPHENOL											
2,4,6-TRICHLOROPHENOL	33/22			17.	17				13		
TRIETHYLAMINE											
TRIFLURALIN				1							
2,2,4-TRIMETHYLPENTANE											
VINYL ACETATE				j.	13						
VINYL BROMIDE							T -	1			
VINYL CHLORIDE		1		13	17			1			
VINYLIDENE CHLORIDE (1,1-DICHLOROETHYLENE)				1 73	i.			1			
XYLENES (ISOMERS AND MIXTURE)						1		1			
o-XYLENES											1
m-XYLENES			T	1		1		1			
p-XYLENES	_	7	1	1	1	1		1			

<sup>&</sup>lt;sup>a</sup>For each HAP emission point defined in the process flow diagram, write in the applicable letter code defined below:

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TABLE 2. PRESENCE OF HAZARDOUS AIR POLLUTANTS IN EMISSION POINTSa

Process name (as defined on process flow diagram)	Raw Materials Handlin	g & S	ilicon	Arc Fu	rnaces	1-3						
	Emission Points ->	E1	E2	E3	CS1	F1	E4	CS2	F2	E5	Π	
ANTIMONY COMPOUNDS					A	A	B	B	B	B		
ARSENIC COMPOUNDS (INORGANIC INCLUE	ING ARSINE)				A	A	13	B	B	B		
BERYLLIUM COMPOUNDS					12	B	B	3	13	13		
CADMIUM COMPOUNDS					A	A	B	13	13	13		
CHROMIUM COMPOUNDS					A	A	A	A	A	A		
COBALT COMPOUNDS					A	A	A	14	A	7		
COKE OVEN EMISSIONS												
CYANIDE COMPOUNDS					A	A				B		
GLYCOL ETHERS												
LEAD COMPOUNDS			A Y		A	A	B	13	13	B	E GOLDANES	
MANGANESE COMPOUNDS							A	A	A			
MERCURY COMPOUNDS		- 2			A	A	B	B	B	B		
FINE MINERAL FIBERS												
NICKEL COMPOUNDS					A	A	A	A	A	A		
POLYCYCLIC ORGANIC MATTER		D	D	り	B	B						
RADIONUCLIDES (INCLUDING RADON)												
SELENIUM COMPOUNDS					A	A	B	B	13	13		

<sup>&</sup>lt;sup>a</sup>For each HAP emission point defined in the process flow diagram, write in the applicable letter code defined below:

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TABLE 2. PRESENCE OF HAZARDOUS AIR POLLUTANTS IN EMISSION POINTSa

Process name (as defined on process flow disgram):	Fume Handling & Coll	ection	Q.											
CHEMICAL NAME	Emission Points ->	E6	BC1	E7	E8	E9	E11	E12	CS3	E10	BC2			
ACETALDEHYDE														
ACETAMIDE								ii garangang					0 000	-ceres
ACETONITRILE									Balling Council					
ACETOPHENONE														
2-ACETYLAMINOFLUORENE										3	y eo ree cry			
ACROLEIN		1												
ACRYLAMIDE			1000	200	0 00			32.1 64	1000 100	(Need of o				
ACRYLIC ACID						(30))								
ACRYLONITRILE .														
ALLYL CHLORIDE														
4-AMINOBIPHENYL								0 ===		1,000				20.
ANILINE		B	B	B	13	B	B	B	B	B	B			
o-ANISIDINE														
ASBESTOS														
BENZENE(INCLUDING BENZENE FROM GASOL	INE)	B	B	13	13	B	13	B	13	13	13	0 9000		
BENZIDINE		B	B	B	B	13	B	13	B	13	13			
BENZOTRICHLORIDE	NATU TOTAL LODGE THE ANGEL CON SOM		135/97.1			1.				1,120,110				
BENZYL CHLORIDE														
BIPHENYL							. 1577/2000							
BIS(2-ETHYLHEXYL)PHTHALATE (DEHP)		B	B	B	13	B	B	B	13	13	13			
BIS(CHLOROMETHYL)ETHER	TANDESCALARIA MAS MARCATANA										A CONTRACTOR			
BROMOFORM		B	B	B	B	B	B	B	B	B	B			••
1,3-BUTADIENE						1,00000								
CALCIUM CYANAMIDE														
CAPROLACTAM														
CAPTAN												72		
CARBARYL		0.17723				(44)								
CARBON DISULFIDE														

<sup>&</sup>lt;sup>8</sup>For each HAP emission point defined in the process flow diagram, write in the applicable letter code defined below:

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TABLE 2. PRESENCE OF HAZARDOUS AIR POLLUTANTS IN EMISSION POINTSa

Process name (as defined on process flow diagram): Fume Handling & College			<b>=</b> - 1	F- 1		F44	Fin	000	E40	DCO	—-r		
Emission Points ->		BC1	E7	E8_	E9	E11	E12	CS3	E10	BC2	_		
CARBON TETRACHLORIDE	B	B	B	B	B	B	B	B	13	B			
CARBONYL SULFIDE													
CATECHOL											_		
CHLORAMBEN													
CHLORDANE	B	B	B	B	B	B	B	B	B	13			
CHLORINE													
CHLOROACETIC ACID													
2-CHLOROACETOPHENONE													
CHLOROBENZENE	B	13	B	B	B	B	13	B	13	B			110
CHLOROBENZILATE													
CHLOROFORM	B	B	B	B	13	B	13	B	B	13			
CHLOROMETHYL METHYL ETHER													
CHLOROPRENE													
CRESOLS/CRESYLIC ACID (ISOMERS AND MIXTURE)	B	B	B	B	13	B	13	B	B	B			
o-CRESOL			531 5			o kanara							
m-CRESOL													700000
p-CRESOL			200 00		1					10 04835100			
CUMENE										-2.48.62-6			
2,4-D, SALTS AND ESTERS													le constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constitution de la constituti
DDE	13	B	13	12	3 13	3 13	13	13	13	13			2000
DIAZOMETHANE .													
DIBENZOFURANS	B	B	B	P	3 12	B	13	, 12	, 12	B			
1,2-DIBROMO-3-CHLOROPROPANE													
DIBUTYLPHTHALATE	B	B	B	B	B	B	P	B	B	B			100
1,4-DICHLOROBENZENE(P)	B	-	B	B			P	The second name of	B	13	1		_
3,3-DICHLOROBENZIDENE	13		B			3 F			B	13			
DICHLOROETHYL ETHER (BIS(2-CHLOROETHYL)ETHER)	B		B	B			_						Г
1,3-DICHLOROPROPENE	B		13	B	-				_				Г
DICHLORVOS		1-0	- 57	1-2			4	4	1	-1-3-	1-	-	

<sup>&</sup>lt;sup>8</sup>For each HAP emission point defined in the process flow diagram, write in the applicable letter code defined below:

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TABLE 2. PRESENCE OF HAZARDOUS AIR POLLUTANTS IN EMISSION POINTSª

Emission Points ->	E6	BC1	E7	E8	E9	E11	E12	CS3	E10	BC2			-
DIETHANOLAMINE					1			1	l	l l		-	
N,N-DIETHYL ANILINE (N,N-DIMETHYLANILINE)									14,000			<b>-</b> :+	THE STATE OF
DIETHYL SULFATE								1					
3,3-DIMETHOXYBENZIDINE													
DIMETHYL AMINOAZOBENZENE													
3,3'-DIMETHYL BENZIDINE									Q.11=202-				
DIMETHYL CARBAMOYL CHLORIDE												- V	
DIMETHYL FORMAMIDE								000					
1,1-DIMETHYL HYDRAZINE													
DIMETHYL PHTHALATE	B	B	B	13	1/3	B	B	B	B	B			
DIMETHYL SULFATE													
4,6-DINITRO-O-CRESOL, AND SALTS											7.00		
2,4-DINITROPHENOL	B	B	13	13	13	B	B	B	B	B			
2,4-DINITROTOLUENE	13	B	B	13	B				B	13		-	
1,4-DIOXANE(1,4-DIETHYLENEOXIDE)													
1,2-DIPHENYLHYDRAZINE									1.55	S A MEN			
EPICHLOROHYDRIN(1-CHLORO-2,3-EPOXYPROPANE)													
1,2-EPOXYBUTANE				200	1.55								
ETHYL ACRYLATE									1				
ETHYL BENZENE	B	13	13	B	13	13	13	B	B	13			
ETHYL CARBAMATE (URETHANE)				57 E						901 a Per 11990a.			
ETHYL CHLORIDE (CHLOROETHANE)	13	B	B	12	3 13	B	J.	3 1	3 13	B			
ETHYLENE DIBROMIDE (DIBROMOETHANE)											1	7	
ETHYLENE DICHLORIDE (1,2-DICHLOROETHANE)	1	B	P	1	B	3 14	3 12	3 /3	13	B			
ETHYLENE GLYCOL													
ETHYLENE IMINE(AZIRIDINE)													
ETHYLENE OXIDE													
ETHYLENE THIOUREA													
ETHYLIDENE DICHLORIDE (1,1-DICHLOROETHANE)		3 R	B	12	12	3 1	3 7	B	P	> B		i i	$\top$

<sup>\*</sup>For each HAP emission point defined in the process flow diagram, write in the applicable letter code defined below:

A - Specific HAP is known to be emitted.

B - Specific HAP is known not to be emitted.

C - No reason or data to assume that this HAP is emitted.

D - HAP is believed to be emitted, but no data exists.

TABLE 2. PRESENCE OF HAZARDOUS AIR POLLUTANTS IN EMISSION POINTSa

Process name (as defined on process flow diagram): Fume Handling & Coll	ectior	1											
Emission Points ->	E6	BC1	E7	E8	E9	E11	E12	CS3	E10	BC2			
FORMALDEHYDE	200												
HEPTACHLOR	В	B	3	13	13	B	B	B	B	B			
HEXACHLOROBENZENE	13	B	B	B	B	B	B	B	В	B			
HEXACHLOROBUTADIENE	13	B	B	B	B	B	$\mathcal{B}$	B	B	B			
HEXACHLOROCYCLOPENTADIENE	B	B	B	3	B	B	B	B	B	B			
HEXACHLOROETHANE	B	B	D	B	B	B	B	B	B	B	1377	100	8 1500
HEXAMETHYLENE-1,6-DIISO-CYANATE													
HEXAMETHYLPHOSPHORAMIDE			12.00										
HEXANE													
HYDRAZINE											71		
HYDROCHLORIC ACID													
HYDROGEN FLUORIDE (HYDROFLUORIC ACID)													
HYDROGEN SULFIDE													
HYDROQUINONE													
ISOPHORONE													
LINDANE (ALL ISOMERS)	B	B	B	B	B	B	B	B	B	B			
MALEIC ANHYDRIDE													
METHANOL				132									
METHOXYCHLOR													
METHYL BROMIDE (BROMOMETHANE)	B	B	B	B	B	B	B	13	B	B			
METHYL CHLORIDE (CHLOROMETHANE)	B	B	13	B	B	B	B	B	B	B			
METHYL CHLOROFORM (1,1,1-TRICHLOROETHANE)	B	B	13	B	13	13	B	B	B	13			
METHYL ETHYL KETONE (2-BUTANONE)	B		13	B	B	B	B	B	13	13		i i	
METHYL HYDRAZINE		(5.55)											
METHYL IODIDE (IODOMETHANE)													
METHYL ISOBUTYL KETONE (HEXONE)	B	B	13	B	13	B	B	B	B	B		92.00	
METHYL ISOCYANATE			3/1-12-0-										
METHYL METHACRYLATE													
METHYL TERT BUTYL ETHER	$\top$						7 7						T

<sup>&</sup>lt;sup>a</sup>For each HAP emission point defined in the process flow diagram, write in the applicable letter code defined below:

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TABLE 2. PRESENCE OF HAZARDOUS AIR POLLUTANTS IN EMISSION POINTS<sup>a</sup>
Fume Handling & Collection

Emission Points ->	E6	BC1	E7	E8	E9	E11	E12	CS3	E10	BC2		T	
4,4-METHYLENE BIS(2-CHLOROANILINE)											1	1	
METHYLENE CHLORIDE (DICHLOROMETHANE)	13	B	B	B	B	B	13	B	B	B			
METHYLENE DIPHENYL DIISOCYANATE (MDI)													
4,4'-METHYLENEDIANILINE													
NAPHTHALENE	B	13	B	B	B	B	13	B	B	B			
NITROBENZENE	13	B	B	B	B	B	13	B	B	B			
4-NITROBIPHENYL													
4-NITROPHENOL	B	13	B	B	B	B	B	B	B	B			
2-NITROPROPANE													
N-NITROSO-N-METHYLUREA													
N-NITROSODIMETHYLAMINE	り	B	B	13	B	B	B	B	B	B			
N-NITROSOMORPHOLINE											-		
PARATHION													
PENTACHLORONTTROBENZENE (QUINTOBENZENE)													
PENTACHLOROPHENOL	13	B	ら	13	B	B	B	B	13	B			
PHENOL	B	B	B	B	B	B	B	B	B	B			
p-PHENYLENEDIAMINE													
PHOSGENE													
PHOSPHINE													
PHOSPHOROUS													
PHTHALIC ANHYDRIDE									,				
POLYCHLORINATED BIPHENYLS (AROCHLORS)													
1,3-PROPANE SULTONE													٠.
BETA-PROPIOLACTONE													
PROPIONALDEHYDE													
PROPOXUR (BAYGON)													
PROPYLENE DICHLORIDE (1,2-DICHLOROPROPANE)	8P	B	133	B	113	13	B	B	B	B			
PROPYLENE OXIDE													
1,2-PROPYLENIMINE (2-METHYL AZIRIDINE)													

<sup>&</sup>lt;sup>a</sup>For each HAP emission point defined in the process flow diagram, write in the applicable letter code defined below:

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TABLE 2. PRESENCE OF HAZARDOUS AIR POLLUTANTS IN EMISSION POINTSa

Process name (as defined on process flow diagram): Fume Handling & Coll	ection						<del></del>					
Emission Points ->	E6	BC1	E7	E8	E9	E11	E12	CS3	E10	BC2		
QUINOLINE										1		
QUINONE												
STYRENE	B	B	B	13	B	B	13	13	B	B		
STYRENE OXIDE												
2,3,7,8-TETRACHLORODIBENZO-p-DIOXIN												
1,1,2,2-TETRACHLOROETHANE	B	B	B	B	B	13	13	13	B	B		
TETRACHLOROETHYLENE (PERCHLOROETHYLENE)	13	13	13	13	B	13	B	13	13	B		
TITANIUM TETRACHLORIDE												
TOLUENE	B	13	B	B	B	13	B	B	13	B		
2,4-TOLUENE DIAMINE												
2,4-TOLUENE DIISOCYANATE												
o-TOLUIDINE												
TOXAPHENE (CHLORINATED CAMPHENE)	13	13	B	13	13	13	B	B	B	BB		
1,2,4-TRICHLOROBENZENE	13	日日	B	B	B	13	B	13	13	B		
1,1,2-TRICHLOROETHANE	B	13	B	B	13	13	13	B	13	13		
TRICHLOROETHYLENE												
2,4,5-TRICHLOROPHENOL												
2,4,6-TRICHLOROPHENOL	13	B	B	B	B	B	B	B	B	B		
TRIETHYLAMINE												
TRIFLURALIN												
2,2,4-TRIMETHYLPENTANE												
VINYL ACETATE	B	13	B	13	B	13	13	B	13	13		
VINYL BROMIDE												
VINYL CHLORIDE	B	B	B	B	B	13	B	13	13			
VINYLIDENE CHLORIDE (1,1-DICHLOROETHYLENE)	B		B	13	B	B	13	18	13			
XYLENES (ISOMERS AND MIXTURE)	B	13	13	B	B	B	13	13	B	13		
o-XYLENES												
m-XYLENES												
p-XYLENES	1		1		1							

<sup>&</sup>lt;sup>a</sup>For each HAP emission point defined in the process flow diagram, write in the applicable letter code defined below:

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TABLE 2. PRESENCE OF HAZARDOUS AIR POLLUTANTS IN EMISSION POINTSa

Process name (as defined on process flow diagram):	Fume Handling & Colle	ection	l			4								
	Emission Points ->[	E6	BC1	E7	E8	E9	E11	E12	CS3	E10	BC2			
ANTIMONY COMPOUNDS		A	A	A	A	A	A	A	A	À	A		1	
ARSENIC COMPOUNDS (INORGANIC INCLUDING	GARSINE)	A	A	A	A	A	A	A	A	A	A			
BERYLLIUM COMPOUNDS		B	B	3	B	B	B	B	B	B	13			
CADMIUM COMPOUNDS		A	A	A	A	A	A	A	A	A	A			
CHROMIUM COMPOUNDS		A	A	A	A	A	A	A	A	A	A			
COBALT COMPOUNDS		4	A	A	A	A	A	A	A	A	A			
COKE OVEN EMISSIONS														į
CYANIDE COMPOUNDS		A	A	义	A	A	A	A	A	A	A			
GLYCOL ETHERS														
LEAD COMPOUNDS		14	A	A	A	A	A	A	A	A	A			
MANGANESE COMPOUNDS														
MERCURY COMPOUNDS		A	A	A	A	A	A	A	A	A	A	•		
FINE MINERAL FIBERS														
NICKEL COMPOUNDS	•	4	A	A	A	A	A	A	Pr	A	A			
POLYCYCLIC ORGANIC MATTER		B	B	B	$\mathcal{B}$	3	B	乃	B	B	B			
RADIONUCLIDES (INCLUDING RADON)														
SELENIUM COMPOUNDS		A	A	A	A	A	A	A	A	A	A			

<sup>&</sup>lt;sup>a</sup>For each HAP emission point defined in the process flow diagram, write in the applicable letter code defined below:

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TABLE 2. PRESENCE OF HAZARDOUS AIR POLLUTANTS IN EMISSION POINTSa

Process name (as defined on process flow diagram)	Silicon Product Hand	dling & I	Oross	Handli	ng									
CHEMICAL NAME	Emission Points -:	> E13	F3	CS4	E14	ВСЗ	F4	CS5	BC4	E15	E16	F5	CS6	E17
ACETALDEHYDE							TO INCLINE			1	l	1	1 1	
ACETAMIDE														
ACETONITRILE														VW 8200
ACETOPHENONE														
2-ACETYLAMINOFLUORENE														
ACROLEIN				- C-72-+1-14-5-1-										•
ACRYLAMIDE											Section 201			
ACRYLIC ACID														
ACRYLONITRILE .														
ALLYL CHLORIDE								0 100-0						
4-AMINOBIPHENYL													T	,
ANILINE				The state of										
o-ANISIDINE														
ASBESTOS														
BENZENE(INCLUDING BENZENE FROM GAS	SOLINE)						• •							
BENZIDINE					100		0.000							
BENZOTRICHLORIDE					L	1977								
BENZYL CHLORIDE														
BIPHENYL														
BIS(2-ETHYLHEXYL)PHTHALATE (DEHP)														
BIS(CHLOROMETHYL)ETHER														
вкомогокм						The Health						1		
1,3-BUTADIENE														
CALCIUM CYANAMIDE									1				1	
CAPROLACTAM									1					1
CAPTAN													1	
CARBARYL .													1	
CARBON DISULFIDE												1		

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A - Specific HAP is known to be emitted.

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TABLE 2. PRESENCE OF HAZARDOUS AIR POLLUTANTS IN EMISSION POINTSa

Process name (as defined on process flow diagram):	ilicon Product Handli	ng & l	oross	Handli	ng									
	Emission Points ->[	E13	F3	CS4	E14	ВСЗ	F4	CS5	BC4	E15	E16	F5	CS6	E17
CARBON TETRACHLORIDE						1	1					1		
CARBONYL SULFIDE														
CATECHOL														
CHLORAMBEN														
CHLORDANE									38 3843	-72-179				
CHLORINE														
CHLOROACETIC ACID											3 99			
2-CHLOROACETOPHENONE														
CHLOROBENZENE														
CHLOROBENZILATE														
CHLOROFORM														
CHLOROMETHYL METHYL ETHER													-	
CHLOROPRENE					10000									
CRESOLS/CRESYLIC ACID (ISOMERS AND MIXTO	JRE) ,													
o-CRESOL														
m-CRESOL														
p-CRESOL														i i
CUMENE														
2,4-D, SALTS AND ESTERS														
DDE											1			
DIAZOMETHANE												V		
DIBENZOFURANS											-1451 (4527-7			
1,2-DIBROMO-3-CHLOROPROPANE														
DIBUTYLPHTHALATE														
1,4-DICHLOROBENZENE(P)	WARE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE OF THE COMMENCE										-350 -350			
3,3-DICHLOROBENZIDENE														
DICHLOROETHYL ETHER (BIS(2-CHLOROETHY)	L)ETHER)				nga dan									
1,3-DICHLOROPROPENE														
DICHLORVOS		1		_	$\neg \vdash$	_	_	$\neg \vdash$	_	_		7		

<sup>&</sup>lt;sup>a</sup>For each HAP emission point defined in the process flow diagram, write in the applicable letter code defined below:

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TABLE 2. PRESENCE OF HAZARDOUS AIR POLLUTANTS IN EMISSION POINTSª

Process name (as defined on process flow diagram); Silicon Product Handli										21 DI		x1=2000x1=========	
Emission Points>	E13	F3	CS4	E14	ВСЗ	F4	CS5	BC4	E15	E16	F5	CS6	E17
DIETHANOLAMINE							1	1	l		1	1 7	
N,N-DIETHYL ANIUNE (N,N-DIMETHYLANIUNE)												1	
DIETHYL SULFATE			27-23,500										
3,3-DIMETHOXYBENZIDINE													
DIMETHYL AMINOAZOBENZENE						1.191.							
3,3'-DIMETHYL BENZIDINE													
DIMETHYL CARBAMOYL CHLORIDE													
DIMETHYL FORMAMIDE													
1,1-DIMETHYL HYDRAZINE													
DIMETHYL PHTHALATE													100 100
DIMETHYL SULFATE													
4,6-DINITRO-O-CRESOL, AND SALTS												*	
2,4-DINITROPHENOL							. 1					1	
2,4-DINITROTOLUENE				end Person		ani i Hazarana			953				
1,4-DIOXANE(1,4-DIETHYLENEOXIDE)													
1,2-DIPHENYLHYDRAZINE					D-10 6050								
EPICHLOROHYDRIN(1-CHLORO-2,3-EPOXYPROPANE)													1
1,2-EPOXYBUTANE				-0.0						201			
ETHYL ACRYLATE													
ETHYL BENZENE				$\neg \vdash$									
ETHYL CARBAMATE (URETHANE)													
ETHYL CHLORIDE (CHLOROETHANE)										224.0	271		
ETHYLENE DIBROMIDE (DIBROMOETHANE)													1
ETHYLENE DICHLORIDE (1,2-DICHLOROETHANE)									7				_
ETHYLENE GLYCOL													_
ETHYLENE IMINE(AZIRIDINE)				_ _				_	$\neg$		- -	_	_
ETHYLENE OXIDE												$\neg$	$\top$
ETHYLENE THIOUREA													
ETHYLIDENE DICHLORIDE (1,1-DICHLOROETHANE)		$\neg \vdash$		_		_		_		$\neg$		_	_

<sup>&</sup>lt;sup>a</sup>For each HAP emission point defined in the process flow diagram, write in the applicable letter code defined below:

A - Specific HAP is known to be emitted.

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D - HAP is believed to be emitted, but no data exists.

TABLE 2. PRESENCE OF HAZARDOUS AIR POLLUTANTS IN EMISSION POINTSa

Process name (as defined on process flow diagram): Silicon Product Handling & Dross Handling													
Emission Points ->	E13	F3	CS4	E14	всз	F4	CS5	BC4	E15	E16	F5	CS6	E17
FORMALDEHYDE	1	1	1										ľ
HEPTACHLOR											1		
HEXACHLOROBENZENE													
HEXACHLOROBUTADIENE													
HEXACHLOROCYCLOPENTADIENE													
HEXACHLOROETHANE													
HEXAMETHYLENE-1,6-DIISO-CYANATE													
HEXAMETHYLPHOSPHORAMIDE													
HEXANE													
HYDRAZINE													
HYDROCHLORIC ACID													
HYDROGEN FLUORIDE (HYDROFLUORIC ACID)												1	
HYDROGEN SULFIDE													
HYDROQUINONE			•									<u> </u>	
ISOPHORONE													
LINDANE (ALL ISOMERS)													<u> </u>
MALEIC ANHYDRIDE													
METHANOL													
METHOXYCHLOR													
METHYL BROMIDE (BROMOMETHANE)													
METHYL CHLORIDE (CHLOROMETHANE)													
METHYL CHLOROFORM (1,1,1-TRICHLOROETHANE)													
METHYL ETHYL KETONE (2-BUTANONE)													•
METHYL HYDRAZINE													
METHYL IODIDE (IODOMETHANE)													
METHYL ISOBUTYL KETONE (HEXONE)													
METHYL ISOCYANATE													
METHYL METHACRYLATE													
METHYL TERT BUTYL ETHER													

<sup>&</sup>lt;sup>a</sup>For each HAP emission point defined in the process flow diagram, write in the applicable letter code defined below:

A - Specific HAP is known to be emitted.

B - Specific HAP is known not to be emitted.

C - No reason or data to assume that this HAP is emitted.

D - HAP is believed to be emitted, but no data exists.

All blank nells some serve he has tot .

# TABLE 2. PRESENCE OF HAZARDOUS AIR POLLUTANTS IN EMISSION POINTS<sup>a</sup> Silicon Product Handling & Dross Handling

	Emission Points ->	E13	F3	CS4	E14	ВСЗ	F4	CS5	BC4	E15	E16	F5	CS6	E17
4,4-METHYLENE BIS(2-CHLOROANILINE)	Manufacture or												1	1
METHYLENE CHLORIDE (DICHLOROMETHANE)														
METHYLENE DIPHENYL DIISOCYANATE (MDI)														
4,4'-METHYLENEDIANILINE														
NAPHTHALENE														
NITROBENZENE														
4-NITROBIPHENYL														
4-NITROPHENOL														
2-NITROPROPANE														
N-NITROSO-N-METHYLUREA														
N-NITROSODIMETHYLAMINE														
N-NTTROSOMORPHOLINE												-		
PARATHION														
PENTACHLORONITROBENZENE (QUINTOBENZE	. GN													
PENTACHLOROPHENOL														
PHENOL														
p-PHENYLENEDIAMINE														
PHOSGENE														
PHOSPHINE														
PHOSPHOROUS														
PHTHALIC ANHYDRIDE														
POLYCHLORINATED BIPHENYLS (AROCHLORS)														
1,3-PROPANE SULTONE														• •
BETA-PROPIOLACTONE														
PROPIONALDEHYDE														
PROPOXUR (BAYGON)														
PROPYLENE DICHLORIDE (1,2-DICHLOROPROPA	ANE)													
PROPYLENE OXIDE														
1,2-PROPYLENIMINE (2-METHYL AZIRIDINE)														

<sup>&</sup>lt;sup>a</sup> For each HAP emission point defined in the process flow diagram, write in the applicable letter code defined below:

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TABLE 2. PRESENCE OF HAZARDOUS AIR POLLUTANTS IN EMISSION POINTSa

Process name (as defined on process flow diagram):	Silicon Product Handl	ing & I	Oross	Handl	ing									
	Emission Points ->	E13	F3	CS4	E14	ВСЗ	F4	CS5	BC4	E15	E16	F5	CS6	E17
QUINOLINE														
QUINONE														
STYRENE														
STYRENE OXIDE														
2,3,7,8-TETRACHLORODIBENZO-p-DIOXIN														
1,1,2,2-TETRACHLOROETHANE														
TETRACHLOROETHYLENE (PERCHLOROETHYLL	ENE)													
TITANIUM TETRACHLORIDE														
TOLLIENE														
2,4-TOLUENE DIAMINE														
2,4-TOLUENE DIISOCYANATE														
o-TOLUIDINE														
TOXAPHENE (CHLORINATED CAMPHENE)														
1,2,4-TRICHLOROBENZENE														
1,1,2-TRICHLOROETHANE														
TRICHLOROETHYLENE														
2,4,5-TRICHLOROPHENOL														
2,4,6-TRICHLOROPHENOL														
TRIETHYLAMINE														
TRIFLURALIN														
2,2,4-TRIMETHYLPENTANE														
VINYL ACETATE														
VINYL BROMIDE														
VINYL CHLORIDE														
VINYLIDENE CHLORIDE (1,1-DICHLOROETHYL	ENE)													
XYLENES (ISOMERS AND MIXTURE)														
o-XYLENES														
m-XYLENES														
p-XYLENES														

<sup>&</sup>lt;sup>a</sup>For each HAP emission point defined in the process flow diagram, write in the applicable letter code defined below:

- A Specific HAP is known to be emitted.
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TABLE 2. PRESENCE OF HAZARDOUS AIR POLLUTANTS IN EMISSION POINTSa

Process name (as defined on process flow diagram): _ Silicon Product Handli	Process name (as defined on process flow diagram): Silicon Product Handling & Dross Handling												
Emission Points ->[	E13	F3	CS4	E14	ВСЗ	F4	CS5	BC4	E15	E16	F5	CS6	E17
ANTIMONY COMPOUNDS		B	B	2	B	В	B	B	B	B			İ
ARSENIC COMPOUNDS (INORGANIC INCLUDING ARSINE)		В	B	B	B	B	B	B	B	B			
BERYLLIUM COMPOUNDS		$\mathcal{B}$	$\mathcal{B}$	$\mathcal{B}$	B	$\mathcal{B}$	B	B	B	B			
CADMIUM COMPOUNDS		B	B	B	B	E	B	$\mathcal{B}$	B	$\mathcal{B}$			
CHROMIUM COMPOUNDS		À	A	A	A	A	A	A	A	A			
COBALT COMPOUNDS		A	A	Ř	H	H	A	H	A	Ĥ			
COKE OVEN EMISSIONS													
CYANIDE COMPOUNDS													
GLYCOL ETHERS													
LEAD COMPOUNDS		B	B	$\mathcal{B}$	$\mathcal{B}$	B	B	B	B	$\mathcal{B}$			
MANGANESE COMPOUNDS		A	A	A	A	A	A	A	A	A			
MERCURY COMPOUNDS		B	B	B	B	B	$\mathcal{B}$	B	$\mathcal{B}$	B			
FINE MINERAL FIBERS													
NICKEL COMPOUNDS .		A	A	A	4	A	A	A	A	A			
POLYCYCLIC ORGANIC MATTER													
RADIONUCLIDES (INCLUDING RADON)													
SELENTUM COMPOUNDS		B	$\mathcal{B}$	B	B	$\mathcal{B}$	B	B	B	B			

<sup>&</sup>lt;sup>a</sup>For each HAP emission point defined in the process flow diagram, write in the applicable letter code defined below:

A - Specific HAP is known to be emitted.

B - Specific HAP is known not to be emitted.

C - No reason or data to assume that this HAP is emitted.

D - HAP is believed to be emitted, but no data exists.

1 2 3 4 5 6 7 8 9 10

	Type of Equipment/ Emission Points	Name of HAP	Capture system/device	Capture Efficiency, percent	Basis for reported efficiency	HAP Concentrations in particulate matter, wt %	Total particulate loading, tons/yr	Flow rate of captured stream, acfm	Uncontrolled HAP emissions, tons/yr
Raw materials unloading	E1	POM (1)	None			No data exists	1.9		
Raw materials storage	E2	РОМ	None			No data exists	1.9		
Raw materials conveying/weighing	E3	РОМ	None			No data exists	1.9		
Furnace fume fugitive	F1	Sb compounds As compounds Cd compounds Cr compounds Co compounds CN compounds Pb compounds Hg compounds Ni compounds Se compounds	None			0.0002 0.0012 0.0002 0.0000006 0.0000000 0.02 0.00001 0.00000			0.001764 0.010584 0.001764 0.01764 0.00005292 0.000007938 0.1764 0.0000882 0.00882 0.00002646
Furnace fume	CS1	Sb compounds As compounds Cd compounds Cr compounds Co compounds CN compounds Pb compounds Hg compounds Ni compounds Se compounds	None			0.0002 0.0012 0.0002 0.000000 0.0000000 0.000000 0.000000			0.0173244 0.1039464 0.0173244 0.173244 0.0000519732 0.0000779598 1.73244 0.00086622 0.086622 0.000259866
Onsite solid waste disposal –C	E5	Cr compounds Ni compounds	None			0.00			0.00026 0.000052
Tapping (Furnace 1)	F2	Cr compounds Co compounds Mn compounds Ni compounds	Hood			0.002 0.002 0.004 0.004	2 4		0.0001152 0.0001152 0.0002304 0.0002592

Table 3-A. Information on Hazardous Air Pollutants -- Preair Pollution Control Device Streams

Unit Operation	Type of Equipment/ Emission Points	Name of HAP	Capture system/device	Capture Efficiency, percent	Basis for reported efficiency	HAP Concentrations in particulate matter, wt %	Total particulate loading, tons/yr	Flow rate of captured stream, acfm	Uncontrolled HAP emissions, tons/yr
Tapping	E4	Cr compounds	None			0.002	86.8		0.001736
(Furnaces 2,3)		Co compounds				0.002	2		0.001736
,		Mn compounds				0.004	, l		0.003472
		Ni compounds				0.0045	5		0.003906
Tapping	CS2	Cr compounds	None		*****	0.002	52.3	1	0.001046
(Furnace 1)		Co compounds				0.002	2		0.001046
,		Mn compounds				0.004	1		0.002092
		Ni compounds				0.0045	5		0.0023535
Fume storage	E7	Sb compounds	None			0.0002	7.75	5	0.0000155
& classify		As compounds				0.0012	2		0.000093
•		Cd compounds				0.0002	2		0.0000155
		Cr compounds				0.002	2		0.000155
		Co compounds				0.000000	5		0.0000000465
		CN compounds				0.0000009	)		0.0000000698
		Pb compounds				0.02	2		0.00155
		Hg compounds				0.0000			0.000000775
		Ni compounds				0.00			0.0000775
		Se compounds				0.000003			0.0000002325
Fume storage	E8	Sb compounds	None			0.0003	1	5	0.0000143
		As compounds				0.0012	1		0.0000858
		Cd compounds				0.000			0.0000143
		Cr compounds				0.002	1		0.000143
		Co compounds				0.000000	1		0.0000000429
		CN compounds				0.0000009	1		0.000000644
		Pb compounds				0.03	- 1		0.00143
		Hg compounds				0.0000			0.000000715
		Ni compounds				0.00			0.0000715
		Se compounds				0.000003			0.0000002145
Fume bagging	E9	Sb compounds	None			0.000		<b>9</b>	0.00003458
		As compounds				0.001			0.00020748
		Cd compounds				0.000	1		0.00003458
		Cr compounds				0.00			0.0003458
		Co compounds				0.000000	1		0.0000001037
		CN compounds				0.000000			0.0000001556
		Pb compounds				0.0	_		0.003458
		Hg compounds				0.0000	· 1		0.000001729
		Ni compounds				0.00			0.0001729
		Se compounds				0.00000	3		0.000005187

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21-Oct-92

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Table 3-A. Information on Hazardous Air Pollutants -- Preair Pollution Control Device Streams

1 2 3 4 5 6 7 8 9 10

Unit Operation	Type of Equipment/ Emission Points	Name of HAP	Capture system/device	Capture Efficiency, percent	Basis for reported efficiency	HAP Concentrations in particulate matter, wt %	Total particulate loading, tons/yr	Flow rate of captured stream, acfm	Uncontrolled HAP emissions, tons/yr
Classifier	CS3	Sb compounds	Hood			0.0002			0.0003
cyclone		As compounds				0.0012	2		0.0018
excess air		Cd compounds				0.0002	2		0.0003
		Cr compounds				0.002	2		0.003
		Co compounds				0.000000	6		0.0000009
		CN compounds				0.0000009	)		0.00000135
		Pb compounds				0.02	2		0.03
		Hg compounds				0.0000			0.000015
		Ni compounds				0.00			0.0015
		Se compounds				0.000003	3	1	0.0000045
Fume settling	E11	Sb compounds	None			0.0002	2 0		
ponds		As compounds				0.0012	2		
•		Cd compounds				0.0002	2		
		Cr compounds				0.002	2		
		Co compounds				0.0000000	3		
		CN compounds				0.0000009	9		
		Pb compounds				0.02			
		Hg compounds				0.0000			
		Ni compounds				0.00	I I		
		Se compounds				0.00000			
Onsite solid	E12	Sb compounds	None			0.000	1	6	0.0000052
waste disposal -V	V	As compounds				0.001	- 1		0.0000312
		Cd compounds				0.000	- 1		0.0000052
		Cr compounds				0.00			0.000052
		Co compounds				0.000000			0.000000156
		CN compounds				0.0000009	1		0.0000000234
		Pb compounds				0.03	1		0.00052
		Hg compounds				0.0000	1		0.00000026
		Ni compounds				0.00			0.000026
		Se compounds				0.00000	3		0.00000078
Si product cooling	E13								
Si product	F3	Cr compounds	Hood			0.00	2 1.46	3	0.0000292
crushing		Co compounds				0.00			0.0000292
		Mn compounds				0.00			0.0000584
		Ni compounds				0.004			0.0000657

Table 3-A. Information on Hazardous Air Pollutants -- Preair Pollution Control Device Streams

1 2 3 4 5 6 7 8 9 10

Unit Operation	Type of Equipment/ Emission Points	Name of HAP	Capture system/device	Capture Efficiency, percent	Basis for reported efficiency	HAP Concentrations in particulate matter, wt %	Total particulate loading, tons/yr	Flow rate of captured stream, acfm	Uncontrolled HAP emissions, tons/yr
Si product	CS4	Cr compounds	Baghouse	1	1	0.002	13.1		0.000262
crushing		Co compounds				0.002	2		0.000262
Ü		Mn compounds				0.004	1		0.000524
		Ni compounds				0.0045	5		0.0005895
Si product	F4	Cr compounds	Hood			0.002	2.91		0.0000582
screening		Co compounds				0.002	2		0.0000582
J		Mn compounds				0.004	ı İ		0.0001164
		Ni compounds				0.0045	5		0.00013095
Si product	CS5	Cr compounds	Baghouse			0.002		2	0.000524
screening		Co compounds				0.002	2		0.000524
· ·		Mn compounds				0.004	1		0.001048
		Ni compounds				0.0045	5		0.001179
Si product	E16	Cr compounds	None			0.002	0.66	3	0.0000132
storage & loading		Co compounds				0.002	2		0.0000132
_		Mn compounds				0.004	1		0.0000264
		Ni compounds				0.0045	5		0.0000297
Dross	F5	None	Hood			No data	0.16	3	
crushing						exists			
Dross	CS6	None	Baghouse			No data	1.4	1	
crushing			_			exists			
Dross	E17	None	None			No data	0.07	7	
loading						exists			

<sup>(1)</sup> POM = Polycyclic organic matter

Table 3-B. Information on Hazardous Air Pollutants--Controlled Streams

1	2	3	4	5	6	7	8	9	10
Unit Operation	Type of Equipment/ Emission Points	Name of HAP	Capture system/device	Capture Efficiency, percent	Basis for reported efficiency	HAP Concentrations in particulate matter, wt %	Total particulate loading, tons/yr	Flow rate of captured stream, acfm	Uncontrolled HAP emissions, tons/yr
Fume baghouses	E6	Sb compounds	Baghouse	98.5%	estimate	0.0002	125.4		0.0002508
		As compounds				0.0012			0.0015048
		Cd compounds				0.0002			0.0002508
		Cr compounds				0.002			0.002508
		Co compounds				0.0000006			0.0000007524
		CN compounds				0.0000009			0.0000011286
		Pb compounds				0.02			0.02508
		Hg compounds				0.00001			0.00001254
		Ni compounds				0.001			0.001254
		Se compounds				0.000003			0.000003762
Fume baghouses	BC1	Sb compounds	Baghouse	98.5%	estimate	0.0002	8236.8		0.0164736
		As compounds				0.0012			0.0988416
		Cd compounds				0.0002			0.0164736
		Cr compounds				0.002			0.164736
		Co compounds				0.0000006			0.0000494208
		CN compounds				0.0000009			0.0000741312
		Pb compounds				0.02			1.64736
		Hg compounds				0.00001			0.00082368
		Ni compounds				0.001			0.082368
Classifier	F-10	Se compounds				0.000003			0.000247104
	E10	Sb compounds	Baghouse	99.0%	estimate	0.0002			0.000003
cyclone		As compounds				0.0012	1		0.000018
		Cd compounds				0.0002	i .		0.000003
		Cr compounds				0.002			0.00003
		Co compounds				0.000006			0.000000009
		CN compounds				0.0000009			0.000000135
		Pb compounds				0.02			0.0003
		Hg compounds				0.00001			0.0000015
		Ni compounds				0.001			0.000015
	1	Se compounds	1			0.000003			0.00000045

Table 3-B. Information on Hazardous Air Pollutants--Controlled Streams

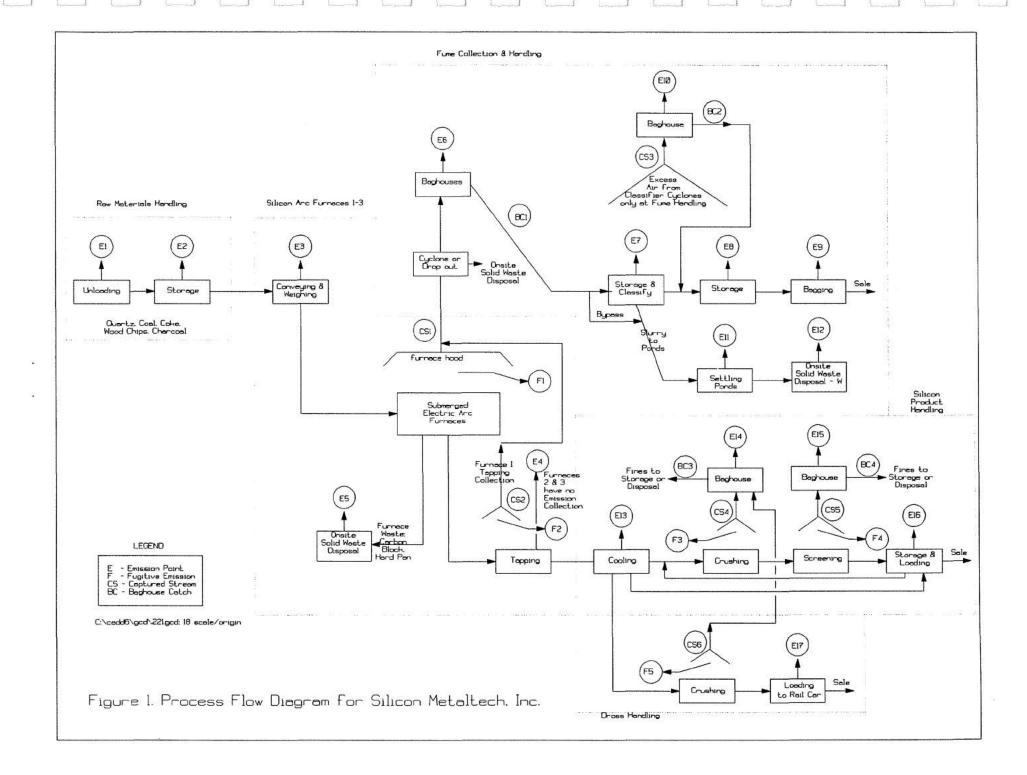
1	2	3	4	5	6	7	8	9	10
Unit Operation	Type of Equipment/ Emission Points	Name of HAP	Capture system/device	Capture Efficiency, percent	Basis for reported efficiency	HAP Concentrations in particulate matter, wt %	Total particulate loading, tons/yr	Flow rate of captured stream, acfm	Uncontrolled HAP emissions, tons/yr
Excess air baghouse	BC2	Sb compounds As compounds Cd compounds Cr compounds Co compounds	Baghouse	99.0%	estimate	0.0002 0.0012 0.0002 0.002 0.0000006	148.5		0.000297 0.001782 0.000297 0.00297 0.00000891
		CN compounds Pb compounds Hg compounds Ni compounds Se compounds				0.000000 0.000000 0.02 0.00001 0.001 0.000003			0.00000031 0.0000013365 0.0297 0.00001485 0.001485
Si/Dross product crushing	E14	Cr compounds Co compounds Mn compounds Ni compounds	Baghouse	99.5%	estimate	0.002 0.002 0.004 0.0045			0.0000014 0.0000014 0.0000028 0.00000315
Si/Dross product crushing	ВСЗ	Cr compounds Co compounds Mn compounds Ni compounds	Baghouse	99.5%	estimate	0.002 0.002 0.004 0.0045			0.0002606 0.0002606 0.0005212 0.00058635
Si product screening	E15	Cr compounds Co compounds Mn compounds Ni compounds	Baghouse	99.5%	estimate	0.002 0.002 0.004 0.0045			0.0000026 0.0000026 0.0000052 0.00000585
Si product screening	BC4	Cr compounds Co compounds Mn compounds Ni compounds	Baghouse	99.5%	estimate	0.002 0.002 0.004 0.0045			0.0005214 0.0005214 0.0010428 0.00117315

Table 4. Air Pollution Capture System and Control Euipment Parameters

Capture systems	Emission Point F1	Emission Point F2	Emission Point F3	Emission Point F4	Emission Point F5
Unit ventilation system used: Control device fan Other (specify)	Control device fan	Furnace #1 has fan feeding into CS1 dust collection. Furnaces #2&3 tapping emissions are not abated @ E4	Pulse jet collector "Clean Air" suction	Pulse jet collector "Clean Air" suction	Pulse jet collector "Clean Air" suction
	3 furnaces, 3 center line submerged arc, owner designed	N/A	N/A	N/A	N/A
Enclosure or Hood Design: Complete enclosure Closed hood Canopy or suspended hood Other (describe)	Front opening hood, "dog house" design	N/A	Suction vents at crusher hopper	Suction vents and hoods at various places	Suction vents and hoods at various places
Curtain or baffling material	Suction fan runs at capacity with control at D.P. across house or by max. motor current	N/A	None – fans run @ capacity	None - fans run @ capacity	None – fans run @ capacity
Distance between hood and emission source Description of hood (general shape and size relative to emission source)		N/A	N/A	N/A	N/A
Building ventilation: Open roof monitor — natural ventilation, other	To atmosphere by roof monitor	To atmosphere by roof monitor	To atmosphere by roof monitor	To atmosphere, no roof monitor	To atmosphere, no roof monitor

Table 4. Air Pollution Capture System and Control Euipment Parameters

Control Device:	Emission Point		Emission Point	Emission Point	Emission Point
Baghouse	E6	3	E10	E14	E15
_	Wheelbrator	ICA			
Gas inlet temperature, F	450° F	450° F	near am bient	near am bient	near ambient
Bag material, weight, and coating	Fiberglass	Fiberglass	Dacron	Dacron	Dacron
Cleaning method & frequency	Reverse air on fixed time	Reverse air on fixed time	Pulse air	Pulse air	Pulse air
Air to cloth ratio, acfm/ft	1.77 to 1	1.55 to 1	5.14 to 1	6.27 to 1	5.14 to 1
Pressure drop across baghouse	7" H <sub>2</sub> O	12 H₂O	4" H <sub>2</sub> O	4" H <sub>2</sub> O	4* H <sub>2</sub> O
Stack opacity, percent	N/A	N/A	N/A	N/A	N/A
Controlled particulate emissions, gr/dscf	N/A	N/A	N/A	N/A	N/A



#### PROCESS FLOW CALCULATIONS FOR SILICON METALTECH, INC.

## Nomenclature and legend for all Unit Operations

#### **Emission Point**

E1 Raw	materials	unloading
--------	-----------	-----------

E2 Raw materials storage

E3 Raw materials conveying/weighing

E4 Tapping (Furnaces 2, 3)

E5 On-site waste disposal

E6 Fume baghouses

E7 Fume storage and classify

E8 Fume storage

E9 Fume bagging

E10 Classifier cyclone

E11 Fume settling pond

E12 Onsite solid waste disposal

E13 Si product cooling

E14 Si/Dross product crushing

E15 Si product screening

E16 Si product storage and loading

E17 Dross loading

## Fugitive Emissions

F1	Furnace	fuma

F2 Tapping (Furnace 1)

F3 Si product crushing

F4 Si product screening

F5 Dross crushing

# Captured Stream

CS1 Furnace fume

CS2 Tapping (Furnace 1)

CS3 Classifier cyclone excess air

CS4 Si product crushing

CS5 Si product screening

CS6 Dross crushing

## **Baghouse Catch**

BC1 Fume baghouses

BC2 Excess air baghouse

BC3 Si/Dross product crushing

BC4 Si product screening

#### Fume Collection and Handling

$$E6 + F1 + E4 + F2 = 1100$$
 tons per year (t/y)

Assumptions:

E11 = 0

CS1 includes CS2

#### Silicon Product Handling

$$CS4 = 0.1\%(13,076) = 13.1 \text{ t/y}$$

$$E14 = 0.5\%(CS4) + 0.5\%(CS6) = 0.07 t/y$$

$$BC3 = CS4 + CS6 - E14 = 13.1 + 1.4 - 0.07 = 14.43 t/y$$

$$CS5 = 0.2\%(13076) = 26.2 t/y$$

$$E15 = 0.5\%(CS5) = 0.5\% (26.2) = 0.13 t/y$$

Therefore BC4 = CS5 - E15 = 
$$26.2 - 0.13 = 26.07 \text{ t/y}$$

Assumption:

$$E13 = 0$$

#### Product hoods

Assume 90% effiency as a standard for product hoods

$$F3/1 = CS4/9$$

$$F3 = CS4(1/9) = 13.1(1/9) = 1.46 t/y$$

$$F4/1 = CS5/9$$

$$F4 = CS5(1/9) = 26.2(1/9) = 2.91 t/y$$

#### <u>Crushing - Dross = by product</u>

$$CS6 = 0.1\%(Dross) = 0.1\%(1364) = 1.4 \text{ t/y}$$

$$F5/1 = CS6/9$$

$$F5 = 1.4(1/9) = 0.16 t/y$$

## Storage Pile Emissions

Use Washington DOE formula for calculation of storage pile emissions

Inventory of quartz = 17,957 tons average quartz usage = 38,408 t/y average

Inventory of coal = 1252 tons average coal usage = 14934 t/y average

Assume ratio of usage/inventory = 3/1

Mass =  $M(0.044n + 0.11)(S/1.5)(100/PE)^2$ 

M = storage mass (tons)

n = storage activity, days per week)

S = assumed to be 1

PE = precip/evap index (96 for SMI site)

Average inventory Si product = 13076/3 = 4359 tons Average inventory dross = 1364/3 = 455 tons

E16 Si product = 
$$4359[0.044(7) + 0.11](1.0/1.5)(100/96)^2$$

$$E16 = 1324$$
 pounds per year or  $1324/2000 = 0.66$  t/y

E17 Dross = 
$$455[0.044(7) + 0.11](1.0/1.5)(100/96)2$$

$$E17 = 138$$
 pounds per year or  $138/2000 = 0.07$  t/y

#### Losses to ground

Fume to sell = 3357 t/y

Loss to ground, bagging operation = 100 t/y

Loss to ground, storage = 100 t/y

Loss to ground, storage and classify = 150 t/y

Total loss to ground = 350 t/y

## Baghouse & classifier cyclones

CS3 is assumed to be 150 t/y Efficiency on that baghouse = 99%

Therefore E10 = 
$$1.0\%(150)$$
 =  $1.5 \text{ t/y}$ 

$$BC2 = CS3 - E10 = 150 - 1.5 = 148.5 \text{ t/y}$$

## Storage, Storage & Classify, Bagging

Bagging:

$$E9 = (0.5)(3457) = 17.29 \text{ t/y}$$

#### Storage:

Into storage = 
$$3357 + 100 + 17.29 + 100 = 3574.29 \text{ t/y}$$

$$E8 = 0.2\%(3574.29) = 7.15 \text{ t/y}$$

#### Storage and Classify

Into storage and classify = 
$$3574.29 + 148.5 + 150$$
  
=  $3872.79 \text{ t/y}$ 

$$E7 = 0.2\%(3872.7) = 7.75 \text{ t/y}$$

$$BC1 = 3872.79 + 4364 = 8236.79 \text{ t/y}$$

$$E6/1.5 = 8236.79/98.5$$

$$E6 = 8236.79(1.5)/98.5 = 125.4 t/y$$

Assume cyclone/drop out = 200 t/y

$$CS1 = 8362.22 + 200 = 8562.22 t/y$$

## Tapping Emissions

Assume 0.4% loss; they relate to total production

Total exiting tapping = 
$$0 + 1.46 + 13.1 + 26.2 + 2.91 +$$

$$0.66 + 13076 + 1364 + 0.07 + 1.4 + 0.16 = 14,486 \text{ t/y}$$

$$CS2 + F2 + E4 = 1\%(14486) = 144.86 \text{ t/y}$$

#### Furnace operation

Furnace operating hours:

#1 = 305 days  
#2 + #3 = 456 days  

$$E4/(CS2 + F2) = 456/305 = 1.495$$
  
 $E4 + (CS2 + F2) = 144.86 t/y$   
 $CS2/9 = F2/1 = 90\%$  efficiency

Solving these three equations gives:

$$E4 = 86.8 \text{ t/y}$$
  
 $CS2 = 52.3 \text{ t/y}$   
 $F2 = 5.76 \text{ t/y}$ 

#### Furnace Hood

## Raw materials handling

Assume E1 = E2 = E3
$$E1 + E2 + E3 = 4.7 + 1.1 \text{ t/y}$$

$$E1 = E2 = E3 = 1.9 \text{ t/y}$$

## On-Site Solid Waste Disposal

At each of two on-site solid waste disposal areas, calculated emissions form one of these areas for the 1991 emission inventory was 2.6 t/y, based on WDOE formula. This emission rate was used for this questionnaire, i.e., E5 = E12 = 2.6 t/y.

1991 Emission Inventory C:\reports\229r.wr1

July 29, 1992



Ms. Rose Longoria Air Quality Technician Washington Department of Ecology Central Regional Office 106 South 6th Avenue Yakima, Washington 98902-3387

Dear Ms. Longoria:

At the request of your Office, attached is the Metaltech, Inc. (SMI) air emission inventory for 1991.

The inventory is summarized in Table 1, SMI EMISSION INVENTORY, Summary of calculated emissions, tons/year. On the pages following Table 1 are emission estimates for various individual sources and types of emissions, as follows:

Silicon Material Balance for Emission Calculations

SOx Emission Calculations

Fugitive Emissions from Plant Roads

Calculation of NOx, VOC & Lead Emissions from EPA SCC reference

Fugitive Emissions from Raw Material Piles

Fugitive Emissions from Raw Material & Product Handling

Emissions from Product Crushing/Handling Dust Collector Bag Houses

Fugitive Emissions from Onsite Solid Waste Disposal

Calculations were based on either material balances, on EPA calculation methods (e.g., CONTROL OF OPEN FUGITIVE DUST SOURCES, EPA-450/3-88-008, EPA AP-42 and the EPA SCC reference listing of emission factors) or upon WDOE fugitive emission Each type of emission calculation used is data sheets. documented in the attachments.



Ms. Rose Longoria July 29, 1992 Page 2

Note that we found some difficulty in using some of the EPA methods and emission factors. For example, The emission factor for SOx from the EPA SCC listing of emission factors was incorrect (too low), since SOx emissions would depend primarily on sulfur in raw materials. We had plant data from which a material balance could be calculated to estimate SOx emissions. The factor for VOC listed in EPA SCC listing of emission factors appeared too high, so the factor for ferrosilicon was used.

Also note that the emission estimate submitted last year was not reflected in the blank form you sent to Mr. Trunzo at SMI, so that form was not filled out.

It is understood you will send a copy of the regulations dealing with toxic air pollutants so that emissions estimates of those pollutants, if any, can be made.

Please do not hesitate to call me if there are any questions.

Sincerely,

Patrick H. Wicks, PE, CHMM

( the thene

President

Enclosures

cc: Mr. Jim Trunzo, SMI, via Fax to 509-884-3263



# Silicon Metaltech, Inc. AIR EMISSION INVENTORY for 1991

July 29, 1992

Patrick H. Wicks, PE, CHMM President

Prepared for: Silicon Metaltech, Inc. **ROCK ISLAND, WASHINGTON** 

Prepared by:

Environmental Engineering & Consulting, Inc. 19125 Northcreek Parkway, Suite 111 Bothell, Washington 98011-8002 (206) 485-3437 Fax (206) 483-1058

## Table 1. SMI EMISSION INVENTORY

# Summary of Calculated Emissions, tons in 1991

Sources:	PARTI- CULATES	PM10		SOx	NOx	VOC	CO	Lead	Estimation Method
PROCESS/NONFUGITIVE EMISSIO									
a. Furnace 1/wb bag house; Furnace 2/ca bag house;	1,100.3	946.3	(1)	597.6	0.7	29.4	none	0.02	0, 2, 3
Furnace 3/ca bag house;									
Furnace building roof monitor; other Furnace building									
uncaptured emissions									
b. Product crusher dust collector	0.1	0.1	(2)						4
c. Product screening dust collector	0.1	0.1	(2)						4
d. Fume bagging dust collector	0.3	0.3	(2)						4
FUGITIVE EMISSIONS									
a. Vehicles on Plant Roads		16.2							3
b. Raw Material Piles	4.7	2.4	(3)						3
c. Raw Material & Product Handling	1.1	0.4	(4)						3
(Loading onto & from piles)									
d. Solid Waste Disposal	2.6								5
e. Fume Ponds	0.0								0
( O' amarida a disa a disa da da da da da da da da da da da da da		0.5	<b>/</b> E\						
f. Si crushing/handling losses	9.0	0.5	(5)						4
Subtotal Emissions	1,118.3	965.6		597.6	0.7	29.4	0.0	0.0	
TOTAL EMISSIONS, not including	1,745.9								
tand									

<sup>(1)</sup> PM10 is 86% of total particulates according to AP42, Table 7.4-4

Lead

<sup>(2)</sup> PM10 is 87% of total particulates according to AP42, Table 7.4-4

<sup>(3)</sup> PM10 is 50% of total particulates according to AP42, Table in section 11.2.7.3

<sup>(4)</sup> PM10 is 35% of total particulates according to AP42, Table in section 11.2.3-2

<sup>(5)</sup> PM10 is 6% of total particulates according to AP42, Table 8.19.2-1

#### Silicon Material Balance

for

## **Emission Calculations**

1991

	Concen- tration	Concen- tration		Total as 100%	Total as 100%	
Total,	of Si,	of SiO2,	Ratio,	SiO2,	Si,	
tons	%	%	Si/SiO2	tons	tons	

# INCOMING

Quartz	38.407	00 004	0.4674	20 220	17 016	Totalla
IQUAIIZ	38,407	99.8%	0.4674	38,330	17.916	Total In
	,			,	,	

#### **OUTGOING**

					_
13,076	98.9%			12,932	
1,364	68%			928	
170	98.9%			168	
30	98.9%			30	
3,357		90%	0.4674	1,412	
350		90%	0.4674	147	
4,364		90%	0.4674	1,836	
	Total Out	Except U	ncaptured Emissions	17,453	
1,100		90%	0.4674	463	
				17,916	Total Out
	1,364 170 30 3,357 350 4,364	1,364 68%  170 98.9% 30 98.9% 3,357 350 4,364  Total Out	1,364 68%  170 98.9% 30 98.9% 3,357 90% 350 90% 4,364 90%  Total Out Except U	1,364 68%  170 98.9% 30 98.9% 3,357 90% 0.4674 350 90% 0.4674 4,364 90% 0.4674  Total Out Except Uncaptured Emissions	1,364 68% 928  170 98.9% 168 30 98.9% 30  3,357 90% 0.4674 1,412 350 90% 0.4674 147 4,364 90% 0.4674 1,836  Total Out Except Uncaptured Emissions 17,453 1,100 90% 0.4674 463

#### Notes:

<sup>(1)</sup> uncaptured emissions including roof monitor emissions, other furnace building uncaptured emissions and baghouse losses.

<sup>(2)</sup> Weight ratio of Si to SiO2 is 0.4674: @ROUND(28.086/(28.086+2\*15.9994),4)

<sup>(3)</sup> Estimated 9 tons/yr fugitive dust emitted of this amount.

# Calculation of NOx, VOC & Lead Emissions from EPA SCC reference 1991

#### NOx emissions:

NOx emission calculated from SCC factor of 0.1 lb NOx/ton Si:

0.6538 tons/y

#### **VOC emissions:**

VOC emission calculated from SCC factor of 4.5 lb VOC/ton for FeSi operations:

29.421 tons/y

VOC emission factor listed in EPA SCC reference of 72 lb VOC/ton Si is believed to be not applicable; use of that factor would have resulted in calculated emissions of 471 t/y.

#### Lead emissions:

Lead emissions calculated from SCC factor of 0.0031 lb lead/ton Si:

0.020267 tons/y

Note this emission is already included in particulate emissions above, so is not an additional emission.

#### **FUGITIVE EMISSIONS**

#### from

# Plant Roads

			1991	. –		Calculated	
						PM10 (1B)	alculated
			Number			Emission	PM10
	Average	Weight,	of	Travel,	Type	Factor,	missions,
	Speed, mph	lb	Wheels	miles	Surface	lb/VMT	t/y
Product Trucks-Outbound	8	80,000	18	305	unpaved	4.62	0.71
Product Trucks-Inbound	8	30,000	18	305	unpaved	2.33	0.36
Inbound Trucks	8	10,300	6	484	paved	3.41	0.83
Wood Chip Trucks-Inbound	8	60,000	10	377	unpaved	2.82	0.53
Wood Chip Trucks-Outbound	8	27,000	10	377	unpaved	1.61	0.30
920 Loader	5	22,000	4	1,391	paved	3.41	2.37
Fork Lifts	5	12,000	4	4,130	paved	3.41	7.05
980 Loader	5	89,000	4	4,704	unpaved	1.47	3.45
Dump Truck	8	10,400	6	78	paved	3.41	0.13
Dump Truck	8	10,400	6	78	unpaved	0.64	0.03
Boom Truck	5	17,860	6	157	paved	3.41	0.27
Maintenance utility vehicle	5	4,920	4	78	paved	3.41	0.13
						Total	16.15

<sup>(1</sup>B) Formula 2–3, with surface silt loading of 50 g/sq.m for paved roads or Formula 3–1, with silt loading of 10% for unpaved roads, from CONTROL OF OPEN FUGITIVE DUST SOURCES, EPA-450/3–88–008; miles traveled reduced to 86% of 1990 amount to reflect same reduction in production and throughput.

Weather data for 1990 indicate 87 days with at least .01" precipitation; 87 raindays mean wind speed about 12 mph; based on personal communication with National Weathers Service, Wenatchee.

# FUGITIVE EMISSIONS from Raw Material Piles 1991

		1331		
			Calculated	
			Fugitive	
	Tons Avg.	% Silt	missions,	
	Inventory	Content	lb/yr (1)	
Quartz	17,957	1.5	8,145	
Coke	397	1.5	180	
Coal	1,252	1.5	568	
Wood Chips	1,295	1.5	587	
To	tal Fugitive E	missions:	9,480	lb/yr
			4.7	tons/yr

<sup>(1)</sup> Storage pile fugitive emissions per WDOE calculation method

# SOx Emission Calculations 1991

	R	aw Mat'l &	Contained	
	Sulfur	Products,	Sulfur,	as SO2
	Content, %	tons	tons	(1), tons
SULFUR IN				
Coke	5.5%	3,989	219.4	
Coal	0.6%	14,934	89.6	
		Total	309.0	
SULFUR OU	Т			
Fume	0.10%	9,171	9.2	
Si & Dross	0.005%	14,440	0.7	
SOx emission	ns		299.1	597.6
		Total	309.0	

EPA SCC reference listed SOx emission factor of 0.07 lb SOx/ton Si. Using this value, 0.6 tons/y SOx emission would be calculated. This is highly questionable when compared to mass balance above; accordingly the value resulting from the mass balance is used.

(1) Sulfur converted to SO2 via ratio of molecular weights: (32.064+2\*15.9994)/32.064

# EMISSIONS from Product Crushing/Handling Dust Collector Bag Houses 1991

	Entrained Into Bag House Inlet	Capture Efficiency	Emission, t/y
13,076	0.1%	99.5%	0.07
13,076	0.2%	99.5%	0.13
3,357	1.0%	99.0%	0.34
		Total	0.53
	13,076 13,076	Product Entrained ughput Into Bag t/y House Inlet 13,076 0.1% 13,076 0.2%	ughput Into Bag Capture t/y House Inlet Efficiency 13,076 0.1% 99.5% 13,076 0.2% 99.5%

# FUGITIVE EMISSIONS from Onsite Solid Waste Disposal 1991

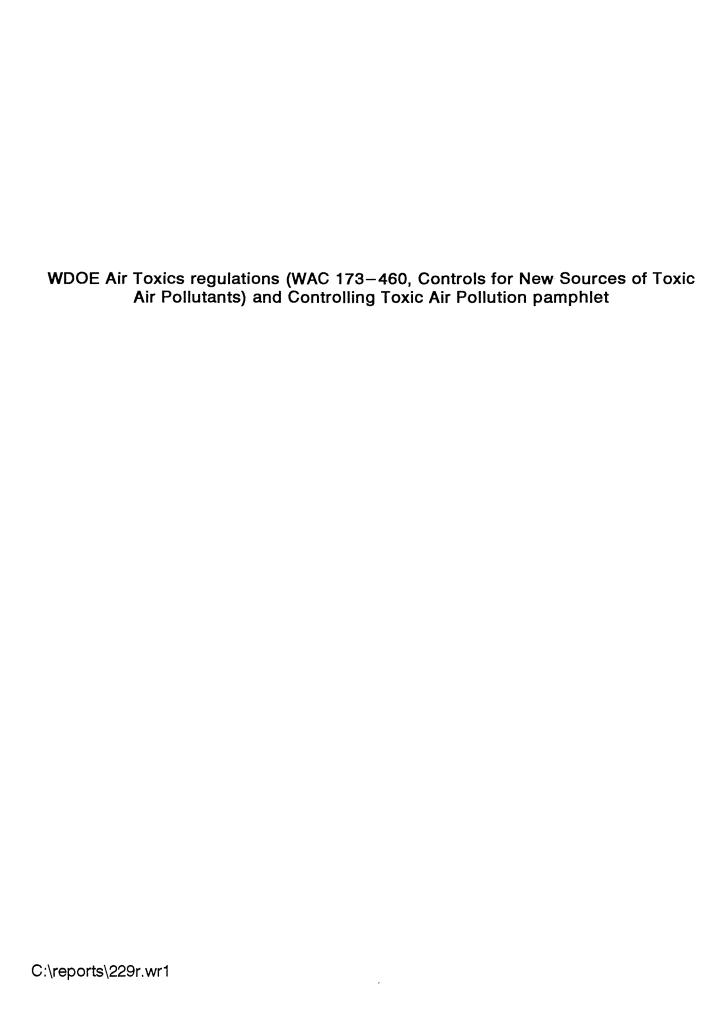
Fugitive emissions from onsite solid waste disposal, based on WDOE calculation method, for one acre:

2.6 tons/y

# FUGITIVE EMISSIONS from Raw Material & Product Handling 1991

					<b>Emission</b>	Calculated		
	Throughput,	<u>Emissio</u>	Emission formula variables Factor (1A)					
	tons	k	U, mph	M, %	lb/ton	t/y		
Raw Materials Handling								
Quartz	38,407	0.35	12	1	0.00922	0.177		
Coke	3,989	0.35	12	6	0.00075	0.001		
Coal	14,934	0.35	12	7	0.00061	0.005		
Wood Chips	21,948	0.35	12	10	0.00037	0.004		
Product Handling								
Si Metal	13,076	0.35	12	0.5	0.02434	0.159		
dross	1,364	0.35	12	0.5	0.02434	0.017		
					Subtotal	0.36		
Number of times each handled						3		
					Total	1.1		

(1A) Formula 4–1 from CONTROL OF OPEN FUGITIVE DUST SOURCES, EPA–450/3–88–008 Note: Estimated emissions from conveyors using WDOE transfer, conveying formula was less than 0.1 ton/yr; accordingly the above estimates should adequately include fugitive emissions from conveyors.





## Overview

The Department of Ecology adopted a statewide rule (Controls for New Sources of Toxic Air Pollution, WAC 173-460) in June 1991, to control air pollution from more than 500 toxic or cancer-causing chemicals. The regulation represents more than two years of research, planning and consultation with scientific, business and environmental experts. The purpose of the rule is to protect the public from exposure to unhealthful levels of toxic and cancer-causing emissions from new or modified commercial and industrial operations.

Traditionally, the regulation of air pollution has focused on six pollutants—carbon monoxide, particulate, sulfur dioxide, lead, ozone and nitrogen oxides—for which the federal government has set outdoor exposure standards designed to protect public health and the environment. These are known as criteria pollutants. But there are hundreds of additional substances emitted into the atmosphere that threaten public health, too. They have been virtually unregulated in the United States.

The regulation of these non-criteria pollutants signals a new era in air pollution control, much like when the regulation of the criteria pollutants began 20 years ago. The 1990 federal clean air act amendments set a timeline for the creation—during the next several years—of national regulations governing 189 toxic air pollutants. With Ecology's air toxics rule effective September 18, 1991, Washington benefits from controls on these pollutants years ahead of most other states.

## Assessing the problem

Washington's Environment 2010 study examined 24 of these non-criteria pollutants. The report estimated at least 2.5 million Washington residents are at risk of suffering the effects of these substances. Pound for pound, many of these toxic and cancer-causing emissions pose a more potent public health threat than the criteria pollutants.

Air pollution from these 24 substances was estimated at 121,000 tons per year, which the report noted represents only a fraction of Washington's total toxic air emissions. While much attention has focused on toxic ground and water pollution, the U.S. Environmental Protection Agency's annual Toxic Release Inventory—based on reports from 343 Washington companies—shows emissions into the air account for 57% of all reported toxic releases, discharges and transfers. The Inventory covers only a small percentage of the state's sources of toxic air pollution.

Pollutants regulated by Ecology's rule include:

■ Toluene, a common component of printing inks, paints and solvents. It can affect the nervous system. It is emitted into the outside air from such businesses as greeting card and plastic bread wrapper printing facilities, painting and coating operations. Switching to water based inks and paints avoids the use and air emission of this solvent.

■ Trichloroethylene, a solvent commonly used to clean and degrease metal aircraft and other mechanical parts. It is a probable carcinogen. Alternative non-toxic cleaning products and methods are available. **Perchloroethylene**, widely used as a cleaning agent by dry cleaners. It is a probable carcinogen. Emissions can be controlled by using a process to capture and reuse the solvent. Hexavalent chromium, emitted into the air by the chrome plating process used in making such goods as car bumpers, faucets and airplane parts. It is a known carcinogen. Control technology is available to limit pollution by this and other chromium compounds. These toxic air pollutants come from the production of goods and services people buy and use every day. Washington's new toxic air pollution rule Ecology has set up a permitting process for new or modified sources of toxic air pollutants. The requirements affect such enterprises as dry cleaners, auto paint shops, pulp mills, aluminum plants, chemical manufacturers, solid waste incinerators, fiberglassing operations, printers and aircraft parts manufacturers. This rule utilizes the same initial permitting process as is required for new sources of criteria pollutants. The toxic regulation applies to all new sources and to modifications of existing sources that increase emissions of any regulated pollutant. The Best Available Control Technology (BACT) is required for any new or increased toxic emission. BACT can include substitutions or process changes that prevent or reduce toxic emissions. When there is no new or increased toxic emission, the less stringent Reasonably Available Control Technology (RACT) is required for any toxic pollutant in a modified source that remains the same or is reduced. Acceptable Source Impact Levels For each of more than 500 chemicals, the rules define an acceptable increase in outdoor exposure to a toxic pollutant called an *Acceptable Source Impact Level*, or ASIL for short. These have been set conservatively to protect human health because there can be multiple sources of a particular toxic pollutant in a community. Here are the basic limits: The ASIL for each carcinogenic pollutant corresponds to a cancer risk of one additional case per million people, over a lifetime of constant exposure to the substance. The ASIL for the other toxic pollutants is set by dividing worker exposure limits by a safety factor of 300. Ecology used the American Council of Governmental Industrial Hygienists' Threshold Limit Values as the basis for its ASIL calculations.

#### **Permitting Steps**

The permitting process is divided into two basic tiers. It is expected that most applicants will need to pass only the first tier in order to receive a permit. Key first tier requirements include:

- Use of the Best Available Control Technology (BACT).
- Calculation of how much of the pollutant(s) would be emitted using that technology.
- Based on these projected emissions, and using approved air pollution dispersion models, an estimate of the resulting level of outdoor pollutants.

If these calculations show that the ambient level of a toxic air pollutant is less than or equal to the ASIL, the permit can be granted. If the calculations show that the ASIL would be exceeded, the applicant can proceed in either of two ways:

- Revise the proposed project so that it meets the first tier requirements.
- Pursue the second permitting tier, in which the applicant must demonstrate how public health will be protected. A health impact assessment is required. It must describe:
  - the pathway by which the pollutant would reach the public,
  - the number of people exposed, and
  - the pollutant's toxic effects.

In the case of carcinogenic emissions, the risk of an additional cancer case in the population must be no greater than one in 100,000 over 70 years.

A further permitting tier is available to sources of carcinogenic emissions that do not meet the one per 100,000 cancer risk requirement. Ecology will evaluate applicants at this level case by case. The process must produce an outcome that benefits the environment as a whole. Also, the applicant must discuss the project and risks with the affected community.

This permitting tier allows for innovation, with improvement to the overall environment as the general guiding principle. Many scenarios are possible. Some possibilities:

- A new source that would exceed the ASIL could address other toxic or carcinogenic sources in the community. This could include programs such as helping residents replace uncertified wood stoves, buying up older more polluting cars, and paying for programs such as car and van pools, which reduce vehicle emissions.
- An industry could change a manufacturing process to one that reduces the health risk to a community. For example, an old process exposes the community to a 70 year cancer risk of one in 100. The new process would reduce that risk to one in 10,000, a 100-fold improvement.
- A source could provide an overall environmental benefit in and of itself; a sewage treatment plant, for example.

# Gasoline vapors

Ecology's rule to recover and contain gasoline vapors addresses this source of benzene, toluene and other toxics, making it unnecessary to include a redundant provision in the toxic air pollution rule for gasoline refueling. (For more information on the gasoline vapor control regulation, see the Ecology fact sheet *Focus: Gasoline Vapor Control*.)

#### Fluoride

Ecology adopted the nation's first regulation for a non-criteria pollutant—hydrogen fluoride, an aluminum mill by-product—in 1972. The rule requires the use of scrubbers or filters to remove the fluoride before it can be released into the air. The regulation sets standards for ambient air quality, and fluoride levels in nearby forage plants. The air toxics rule will not affect the fluoride regulation for aluminum mills.

# King, Snohomish, Kitsap and Pierce counties

Regulation of toxic air pollutants is already in effect in these four counties, under a program adopted by the Puget Sound Air Pollution Control Agency (PSAPCA) in 1990. PSAPCA's rule is largely based on the ASIL exposure limits developed by Ecology. Ecology and PSAPCA have worked closely to coordinate their toxics regulations. PSAPCA's program is not altered by the state regulation.

#### **Future actions**

Ecology's air toxics rule presently addresses new sources and modified sources that increase toxic emissions. Ecology is preparing to begin development of rules for existing sources. (The PSAPCA air toxics regulation already addresses existing sources in King, Snohomish, Kitsap and Pierce counties.) Ecology will involve the business, environmental and scientific community in developing this rule. No timetable has been set, but the process could begin in late 1991. In addition, Ecology expects to update the air toxics regulation as needed to conform with federal air toxics rules scheduled for adoption over the next several years.

#### Permitting information

Most air quality permits, including air toxics permits, are granted by local air pollution control agencies. Projects permitted on or after September 18, 1991, come under the air toxics rule. The local agencies may adopt additional requirements. In counties with no local air pollution control agency Ecology issues all air quality permits.

#### For more information

For a copy of the air toxics rule (WAC 173-460) contact Judy Beitel, Department of Ecology, PV-11, Olympia, WA 98504-8711, (206) 459-6304, or your local air pollution control agency.

Questions concerning the rule may be addressed to your local air pollution control agency or to Ecology Program Development and Planning Supervisor, Dan Johnson, at the same address and phone listed above.

#### Air pollution control agencies

- Puget Sound Air Pollution Control Agency (King, Kitsap, Pierce, Snohomish counties): 200 W. Mercer St., Rm 205, Seattle, WA 98119-3958, (206) 296-7431.
- Northwest Air Pollution Authority (Island, Skagit, Whatcom counties): 302 Pine St., #207, Mt. Vernon, WA 98273, (206) 428-1617 (Mt. Vernon), 1-800-622-4627 (Island and Whatcom counties.)
- Olympic Air Pollution Control Authority (Clallam, Grays Harbor, Jefferson, Mason, Pacific, Thurston counties): 120 E State Ave., Olympia, WA 98501, (206) 586-0593, 1-800-422-5623.
- Southwest Air Pollution Control Authority (Clark, Cowlitz, Lewis, Skamania, Wahkiakum counties): 1308 NE 134th St, Ste. D, Vancouver, WA 98685-2747. (206) 574-3058, 1-800-633-0709.
- Spokane County Air Pollution Control Authority: W 1101 College Ave., Rm 230, Spokane, WA 99201, (509) 456-4727.
- Yakima County Clean Air Authority: County Courthouse, Yakima, WA 98901, (509) 575-4116.
- Douglas County Air Pollution Control Commission: 110 Third St., NE, East Wenatchee, WA 98802 (509) 884-1511.
- Grant County Clean Air Authority: PO Box 37, Ephrata, WA 98823-0337, (509) 754-2011.
- Benton Franklin Walla Walla Counties Air Pollution Control Authority: 650 George Washington Way, Richland, WA 99352, (509) 545-2354, (509) 946-4489.

#### Chapter 173-460 WAC

#### CONTROLS FOR NEW SOURCES OF TOXIC AIR POLLUTANTS

#### NEW SECTION

WAC 173-460-010 PURPOSE. (1) Pursuant to chapter 70.94 RCW, Washington Clean Air Act, the purpose of this chapter is to establish the systematic control of new sources emitting toxic air pollutants (TAPs) in order to prevent air pollution, reduce emissions to the extent reasonably possible, and maintain such levels of air quality as will protect human health and safety. Toxic air pollutants include carcinogens and noncarcinogens listed in WAC 173-460-150 and 173-460-160.

- (2) This chapter establishes three major requirements:
- (a) Best available control technology for toxics;
- (b) Toxic air pollutant emission quantification;
- (c) Human health and safety protection demonstration.
- (3) Policy. It is the policy of ecology to reduce, avoid, of eliminate toxic air pollutants prior to their generation whenever economically and technically practicable.

#### NEW SECTION

WAC 173-460-020 DEPINITIONS. The definitions of terms contained in chapter 173-400 WAC are incorporated into this chapter by reference. In the event of a conflict between the definitions provided in chapter 173-400 WAC and the definitions provided in this section, the definitions in this section shall govern. Unless a different meaning is clearly required by context, the following words and phrases as used in this chapter shall have the following meanings. Note: For copies of the above mentioned rule and any other rule cited in this chapter, contact the Department of Ecology, Records Section, Mailstop PV-11, Olympia, WA 98504-8711.

(1) "Acceptable source impact analysis" means a procedure for demonstrating compliance with WAC 173-460-070 and 173-460-080, that compares maximum incremental ambient air impacts with applicable

acceptable source impact levels (ASIL).

- (2) "Acceptable source impact level (ASIL)" means a concentration of a toxic air pollutant in the outdoor atmosphere in any area which does not have restricted or controlled public access that is used to evaluate the air quality impacts of a single source. There are three types of acceptable source impact levels: Risk-based, threshold-based, and special. Concentrations for these three types of ASILs are determined as provided in WAC 173-460-110. ASILs are listed in WAC 173-460-150 and 173-460-160.
- (3) "Authority" means an air pollution control authority activated pursuant to chapter 70.94 RCW that has jurisdiction over the subject source. Ecology is the authority if an air pollution control authority has not been activated or if ecology has jurisdiction over the source pursuant to RCW 70.94.395.
- (4) "Best available control technology for toxics (T-BACT)" applies to each toxic air pollutant (TAP) discharged or mixture of TAPs, taking in account the potency quantity and toxicity of each toxic air pollutant or mixture of TAPs discharged in addition to the meaning given in WAC 173-400-030(10).
- (5) "Carcinogenic potency factor" means the upper 95th percentile confidence limit of the slope of the dose-response curve and is expressed in units of (mg/kg-day)-1.

- (6) "Class A toxic air pollutant (Class A TAP)" means a substance or group of substances listed in WAC 173-460-150.
- (7) \*\*Class B toxic air pollutant (Class B TAP) \*\* means any substance that is not a simple asphyxiant or nuisance particulate and that is listed in WAC 173-460-160.
- (8) "EPA's Dispersion Modeling Guidelines" means the United States Environmental Protection Agency Guideline on Air Quality Models, ZPA 450/2-78-0277R and is hereby incorporated by reference.
- (9) "EPA's Risk Assessment Guidelines" means the United States Environmental Protection Agency's Guidelines for Carcinogenic Risk Assessment, 51 FR 33992 (September 24, 1986) and is hereby incorporated by reference.
- (10) "Increased cancer risk of one in one hundred thousand" means the 95th percent upper bound on the estimated risk of one additional cancer above the background cancer rate per one hundred thousand individuals continuously exposed to a Class A toxic air pollutant at a given average dose for a specified time.
- (11) "Increased cancer risk of one in one million" means the 95th percent upper bound on the estimated risk of one additional cancer above the background cancer rate per one million individuals continually exposed to a Class A toxic air pollutant at a given average dose for a specified time.
- (12) "Inhalation Reference Dose (Inhalation RfD)" means a reference dose published in the United States Environmental Protection Agency Integrated Risk Information System (IRIS).
- (13) "Mixture" means a combination of two or more substances mixed in arbitrary proportions.
- (14) "New toxic air pollutant source" means a source or emissions unit which may emit toxic air pollutants and which commenced construction after the effective date of this chapter. Addition to, enlargement, modification, replacement, or any alteration of any process or air pollutant source which may increase emissions or ambient air concentrations of any regulated air pollutant, including toxic air pollutants, shall be construed as construction or installation or establishment of a new toxic source.
- (15) "Reasonably available control technology for toxics (T-RACT)" applies to each toxic air pollutant (TAP) discharged or mixture of TAPs, taking into account the potency, quantity, and toxicity of each toxic air pollutant or mixture of TAPs discharged in addition to the meaning given in WAC 173-400-030(59).

  (16) "Second Tier Analysis" means an optional procedure used
- (16) "Second Tier Analysis" means an optional procedure used after T-BACT and acceptable source impact analysis for demonstrating compliance with WAC 173-460-070. The second tier analysis uses a health impact assessment as provided in WAC 173-460-090, instead of an acceptable source impact level.
- (17) "Simple asphyxiant" means a physiologically inert gas or vapor that acts primarily by diluting atmospheric oxygen below the level required to maintain proper levels of oxygen in the blood. Examples of simple asphyxiants are given in Appendix X of the TLV Booklet referred to in subsection (19) of this section and incorporated by reference.
- (18) "Threshold limit value-time weighted average (TLV-TWA)" means a concentration limit recommended by the American Conference of Governmental Industrial Hygienists (ACGIH) for a normal eight-hour workday and forty-hour workweek.
- (19) "TLV Booklet" means "TLVs, Threshold Limit Values and Biological Exposure Indices for 1987-88," published by the American Conference of Governmental Industrial Hygienists and is hereby incorporated by reference.
- (20) "Toxic air pollutant (TAP)" means any Class A or Class 5 toxic air pollutant listed in WAC 173-460-150 and 173-460-160. The term toxic air pollutant may include particulate matter and volatile organic compounds if an individual substance or a group of substances within either of these classes is listed in WAC 173-460-150 and 173-460-160. The term toxic air pollutant does not include particulate matter and volatile organic compounds as generic classes of compounds.

(21) "Upper bound unit risk factor" means the 95 percent upper confidence limit of an estimate of the extra risk of cancer associated with a continuous 70 year exposure to 1 uq/m3 of a Class A toxic air pollutant.

#### NEW SECTION

WAC 173-460-030 REQUIREMENTS, APPLICABILITY AND EXEMPTIONS. Applicability.

- (a) The provisions of this chapter shall apply state-wide. authority shall enforce WAC 173-460-010, 173-460-020, 173-460-030, 173-460-040, 173-460-050, 173-460-060, 173-460-070, 173-460-080, 173-460-130, 173-460-140, 173-460-150, and 173-460-160.

  (b) Except as provided in this chapter, any new toxic air pollu-
- tant source listed in (b)(i), (ii), or (iii) of this subsection that may emit a Class A or Class 8 TAP into the ambient air is subject to these regulations:
  - (i) Standard industrial classifications:
  - (A) Major group 10-Metal mining.
  - (B) Major group 12-Bituminous coal and lignite mining. (C) Major group 13-Oil and gas extraction.

  - (D) Manufacturing industries major groups 20-39.
- (E) Major group 49-Electric, gas, and sanitary services except 4971 irrigation systems.
  - (F) Dry cleaning plants, 7216.
  - (G) General medical surgical hospitals, 8062.

  - (H) Specialty hospitals, 8069.(I) National security, 9711.
- (ii) Any source or source category listed in WAC 173-400-100, 173-400-115(2), or 173-490-030(1) except WAC 173-490-030 (1) (e) gasoline dispensing facilities.
  - (iii) Any of the following sources:
  - (A) Landfills.
- (B) Sites subject to chapter 173-340 WAC Model Toxics Control Act--Cleanup regulation.
  - (2) Exempt sources.
- (a) Containers such as tanks, barrels, drums, cans, and buckets are exempt from the requirements of this chapter unless equipped with a vent other than those required solely as safety pressure release
- (b) Monprocess fugitive emissions of toxic air pollutants from stationary sources, such as construction sites, unpaved roads, coal piles, waste piles, and fuel and ash handling operations are exempt from WAC 173-460-060.
- The following sources are generally exempt from the require-(c) ments of WAC 173-460-050, 173-460-070, 173-460-080, and 173-460-090. However, the authority may on a case-by-case basis, require compliance with these sections if the authority determines that the amount of emissions, nature of pollutant, or source location indicate that the ambient impact should be evaluated.
  - (i) Perchloroethylene dry cleaners
  - (ii) Petroleum solvent dry cleaning systems
  - (iii) Solvent metal cleaners
  - (iv) Spray coating operations
  - (v) Abrasive blasting
- (d) Demolition and renovation projects involving asbestos removal and disposal are exempt from the requirements of this chapter.
- (e) Process vents subject to 40 C.F.R. Parts 264 and 265, Subpart AA are exempt from the requirements of this chapter.

#### NEW SECTION

WAC 173-460-040 NEW SOURCE REVIEW. (1) Applicability. This chapter supplements the new source review requirements of WAC 173-400-110 by adding additional new source review requirements for toxic air pollutant sources. If a notice of construction is required under both chapter 173-400 WAC and this chapter, the written applications shall be combined. A notice of construction is a written application to permit construction of a new source.

- (a) The owner or operator of a new toxic air pollutant source listed in WAC 173-460-030(2) shall notify the authority prior to the construction, installation, or establishment of a new toxic air pollutant source and shall file a notice of construction application with the authority for the proposed emission unit(s). Notification and notice of construction are not required if the source is an exempt source listed in WAC 173-460-030(3) or subsection (2) of this section.
- (b) The notice of construction and new source review applies only to the affected emission unit(s) and the contaminants emitted from the emission unit(s).
- (c) New source review of a modification is limited to the emission unit or units proposed to be modified and the emission unit or units whose emissions of TAPs may increase as a result of the modification.
- (2) The owner or operator of a new toxic air pollutant source listed in WAC 173-460-030(2) is not required to notify or file a notice of construction with the authority if any of the following conditions are met:
- (a) Routine maintenance or repair requires equivalent replacement of air pollution control equipment; or
- (b) The new source is a minor process change(s) that does not increase capacity and total toxic air pollutant emissions do not exceed the emission rates specified in small quantity emission rate tables in #AC 173-460-080; or
- (c) The new source is the result of minor changes in raw material composition and the total toxic air pollutant emissions do not exceed the emission rates specified in the small quantity emission rate tables in WAC 173-460-080.
- (3) Additional information. Within thirty days of receipt of a notice of construction, the authority may require the submission of additional plans, specifications, and other information necessary for the review of the proposed new or modified source.
- (4) Requirements for new toxic air pollutant sources. The authority shall review notice(s) of construction, plans, specifications, and other associated information to determine that:
- (a) The source will be in accord with applicable federal, state, and authority air pollution control rules and regulations;
- (b) The source will use T-BACT for emissions control for the toxic air pollutants which are likely to increase;
- (c) The source will use T-RACT for emissions control for the toxic air pollutants which are likely to remain the same or decrease; and
- (d) Sources required to use T-BACT for emission control demonstrate compliance with WAC 173-460-070 by using the procedures established in WAC 173-460-080 or, failing that, demonstrates compliance, by using the additional procedures in WAC 173-460-090 and/or 173-460-100.
- (4) Preliminary determination. Within thirty days after receipt of all information required, the authority shall:
- (a) Make preliminary determinations on the matters set forth in this section; and
- (b) Initiate compliance with the provisions of WAC 173-400-171 relating to public notice and public comment, as applicable.
- (5) Final determination. If, after review of all information received including public comment, the authority finds that all the conditions in this section are satisfied, the authority shall issue a regulatory order to approve the notice of construction for the proposed new source or modification. If the authority finds that the

conditions in this section are not satisfied, the authority shall issue an order for the prevention of construction, installation, or establishment of the toxic air pollution source(s). Where ecology has jurisdiction, it will endeavor to make final determinations as promptly as possible.

(6) Appeal of decision. A final notice of construction decision may be appealed to the pollution control hearings board pursuant to

chapter 43.21B RCW.

(7) Commencement of construction. The owner(s) or operator(s) of the new source shall not commence construction until the applicable notice of construction has been approved.

- (8) Operation and maintenance plan. As a condition of notice of construction approval; prior to start up, the authority may require a plan for the operation and maintenance of all equipment and procedures to assure continuous compliance with this chapter.
- (a) A copy of the plan shall be filed with the authority upon request.
- (b) The plan shall reflect good industrial practice and may include operating parameters and maintenance procedures, and shall be updated to reflect any changes in good industrial practice.

(c) Submittal of all plans should coincide with the authorities

reporting requirements where applicable.

(9) Jurisdiction. Emission of toxic air pollutants that exceed the acceptable source impact levels listed in WAC 173-460-150 and 173-460-160 requires ecology and, if applicable, authority approval as specified in WAC 173-460-090 and 173-460-100.

#### NEW SECTION

WAC 173-460-050 REQUIREMENT TO QUANTIFY EMISSIONS. (1) New sources.

- (a) When applying for a notice of construction, an owner or operator of a new toxic air pollution source shall quantify those emissions of each TAP or combination of TAPs that:
- (i) Will be used for the modeling procedures in WAC 173-460-080; and
- (ii) That may be discharged after applying required control technology. The information shall be submitted to the authority.
- (b) 2 missions shall be quantified in sufficient detail to determine whether the source complies with the requirements of this chapter.
  - (2) Small quantity sources.

Sources that choose to use small quantity emission rate tables instead of using dispersion modeling shall quantify emissions as required under WAC 173-460-080, in sufficient detail to demonstrate to the satisfaction of the authority that the emissions are less than the applicable emission rates listed in WAC 173-460-080.

(3) Level of detail.

An acceptable source impact level analysis under WAC 173-460-080, may be based on a conservative estimate of emissions that represents good engineering judgment. If compliance with WAC 173-460-070 and 173-460-080 cannot be demonstrated, more precise emission estimates shall be used prior to WAC 173-460-090.

(4) Mixtures of toxic air pollutants.

- (a) An owner or operator of a source that may discharge more than one toxic air pollutant may demonstrate compliance with WAC 173-460-070 and 173-460-080 by:
- (i) Quantifying emissions and performing modeling for each TAP individually; or
- (ii) Calculate the sum of all TAP emissions and perform modeling for the total TAP emissions and compare maximum ambient levels to the smallest ASIL; or
  - (iii) Equivalent procedures may be used if approved by ecology.

- (b) Dioxin and furan emissions shall be considered together as one TAP and expressed as an equivalent emission of 2,3,7,3 TCDD based on the relative potency of the isomers in accordance with United States Environmental Protection Agency (EPA) quidelines.
- Note: Copies of EPA "Interim procedures for estimating risks associated with exposures to mixtures of chlorinated dibenzo-pedioxins and dibenzofurans (CDDs and CDFs). 1989 Update" are available by requesting EPA /625/3-89/016, March 1989 from ORD Publications (513) 684-7562.
- (c) Polyaromatic hydrocarbon (PAH) emissions. The owner or operator of a source that may emit a mixture of polyaromatic hydrocarbon emissions shall quantify the following PAHs and shall consider them together as one TAP equivalent in botency to benzo(a) pyrene: benzo(a) anthracene, benzo(b) fluoranthene, benzo(k) fluoranthene, chrysene, libenzo(a,h) anthracene, indenol(1,2,3-cd) pyrene, benzo(a) pyrene. The acceptable source impact analysis shall be conducted using the polyaromatic hydrocarbon emission ASIL contained in VAC 173-460-150(3).
- (d) Uncontrolled roof vent emissions from primary aluminum smelters. The owner or operator of a primary aluminum smelter that may emit a mixture of polyaromatic hydrocarbons from uncontrolled roof vents shall quantify PAH emissions using either of the following methods:
- (i) Quantify PAH emissions using the procedures in (c) of this subsection; or
- (ii) Multiply the total particulate emission mass from the uncontrolled roof vents by the percent of the particulate that is extractable organic matter. The percent extractable organic matter shall be considered one percent of total particulate matter unless ecology determines that there is compelling scientific lata which demonstrates that the use of this value is inappropriate. The acceptable source impact analysis shall be conducted using the primary aluminum smelter uncontrolled roof vent PAH emission ASIL contained in WAC 173-460-150(3). Note: For example, 100 grams of particulate air emission mass times one percent yields one gram of PAH emissions.

#### NEW SECTION

WAC 173-460-060 CONTROL TECHNOLOGY REQUIREMENTS. Except as provided for in WAC 173-460-040, a person shall not establish, operate, or cause to be established or operated any new toxic air pollutant source which is likely to increase TAP emissions without installing and operating T-BACT. Satisfaction of the performance requirements listed below fulfill the T-BACT requirement for those particular sources. Authorities may develop and require performance requirements in lieu of T-BACT provided that ecology approves the performance requirements as equivalent to T-BACT.

- (1) Perchloroethylene dry cleaners. The entire iryer exhaust shall be vented through a control device which will reduce VOC emissions to 5 kg or less per 100 kg dry weight of cleaned articles.
- (a) The control device shall meet one of the following conditions:
- (i) The exhaust from a carbon adsorber shall contain less than 100 ppm perchloroethylene as measured over a period of one minute before dilution; or
- (ii) The air temperature at the outlet of a refrigerated contenser shall reach seven degrees centigrade or less during the cooldown period. A temperature gauge with a minimum range from negative thirty-two to seventy-five degrees centigrade shall be installed and maintained on the condenser outlet duct; or

- (iii) The demonstrated control efficiency for any other control device shall be ninety percent or greater by weight, prior to the discharge to the atmosphere measured over a complete control cycle.
- (b) The operation of any perchloroethylene dry cleaner shall meet all of the following conditions:
  - (i) All leaking components shall be repaired immediately; and
- (ii) All filtration cartridges shall be drained in the filter housing or other enclosed container before discarding the cartridges.
- (2) Petroleum solvent dry cleaning systems. A petroleum solvent dry cleaning system shall include the following:
- (a) All cleaned articles are dried in a solvent recovery dryer or the entire dryer exhaust is vented through a properly functioning control device which will reduce emissions to no more than 3.5 kg of VOC per 100 kg dry weight of cleaned articles; and
- (b) All cartridge filtration systems are drained in their sealed housing or other enclosed container before discarding the cartridges; and
  - (c) All leaking components shall be repaired immediately.
- (3) Chromic acid plating and anodizing. The facility-wide uncontrolled heravalent chromium emissions from plating or anodizing tanks shall be reduced by at least ninety-five percent using either of the following control techniques:
- (a) An antimist additive or other equally effective control method approved by ecology or authority; or
  - (b) The tank is equipped with:
- (i) A close capture system which shall be in place and in operation at all times electrical current is applied to the tank; and
- (ii) An emission control system which limits hexavalent chromium emissions to no more than 0.15 milligrams per ampere-hour of electrical charge applied to the tank or uncontrolled emissions shall be reduced by ninety-five percent.
- (4) Chromic acid and plating (greater than 1 kilogram). If the facility-wide hexavalent chromium emissions from chromic acid plating and anodizing are greater than 1 kilogram per year after the application of control techniques required by subsection (3) of this section, the facility-wide hexavalent chromium emissions shall be reduced by at least ninety-nine percent using either of the following control techniques:
- (a) An antimist additive or other equally effective control method approved by ecology or authority; or
  - (b) The tank is equipped with:
- (i) A close capture system which shall be in place and in operation at all times electrical current is applied to the tank; and
- (ii) An emissions control system which limits hexavalent chromium emissions to no more than 0.03 milligrams per ampere-hour of electrical charge applied to the tank or uncontrolled emissions shall be reduced by ninety-nine percent.
  - (5) Solvent metal cleaners.
- (a) Any solvent metal cleaner shall include all of the following equipment:
- (i) A cover for the solvent tank which shall be closed at all times except when processing work in the legreaser. However, the cover shall be closed to the maximum extent possible when parts are being degreased;
- (ii) A facility for draining cleaned parts such that the drained solvent is returned to the solvent tank;
- (iii) For cold solvent cleaners, a freeboard ratio greater than or equal to  $0.75\,$ ;
  - (iv) Vapor degreasers shall have:
  - (A) A high vapor cutoff thermostat with manual reset; and
- (B) for degreasers with spray devices, a vapor-up thermostat which will allow spray operation only after the vapor zone has risen to the design level; and
- (C) Either a freeboard ratio greater than or equal to 0.75 or a refrigerated freeboard chiller; and
  - (v) Conveyorized vapor degreasers shall have:

- (A) A drying tunnel or a rotating basket sufficient to prevent cleaned parts from carrying liquid solvent out of the degreaser; and
  - (B) A high vapor cutoff thermostat with manual reset; and
- (C) A vapor-up thermostat which will allow conveyor movement only after the vapor zone has risen to the design vapor level.
- (b) The operation of any solvent metal cleaner shall meet the following requirements:
- (i) Solvent shall not leak from any portion of the degreasing equipment;
- (11) Solvent, including waste solvent, shall be stored in closed containers and shall be disposed of in such a manner as to prevent its evaporation into the atmosphere;
- (iii) For cold cleaners, cleaned parts shall be drained until dripping ceases; and
- (iv) Degreasers shall be constructed to allow liquid solvent from cleaned parts to drain into a trough or equivalent device and return to the solvent tank.
- (c) For open-top vapor degreasers, solvent drag-out shall be minimized by the following measures:
  - (i) Racked parts shall be allowed to fully drain;
- (ii) The work load shall be degreased in the vapor zone until condensation ceases;
  - (iii) Spraying operations shall be done within the vapor layer;
- (iv) When using a powered hoist, the vertical speed of parts in and out of the vapor zone shall be less than three meters per minute (ten feet per minute);
- (v) When the cover is open, the lip of the degreaser shall not be exposed to steady drafts greater than 15.3 meters per minute (fifty feet per minute); and
- (vi) When equipped with a lip exhaust, the fan shall be turned off when the cover is closed.
- (d) For conveyorized vapor degreasers, solvent drag-out shall be minimized by the following measures:
  - (i) Racked parts shall be allowed to fully drain; and
- (ii) Vertical conveyor speed shall be maintained at less than three meters per minute (ten feet per minute).
  - (6) Abrasive blasting.
- (a) Abrasive blasting should be performed inside a booth or hangar designed to capture the blast grit or overspray.
- (b) Outdoor blasting of structures or items too large to be reasonably handled indoors should employ control measures such as curtailment during windy periods and enclosure of the area being blasted with tarps.
- (c) Outdoor blasting should be performed with either steel shot or an abrasive containing less than one percent (by mass) which would pass through a No. 200 sieve.
- (d) All abrasive blasting with sand shall be performed inside a blasting booth or cabinet.

#### NEW SECTION

WAC 173-460-070 AMBIENT IMPACT REQUIREMENT. When applying for a notice of construction under WAC 173-460-040, the owner or operator of a new toxic air pollutant source which is likely to increase TAP emissions shall demonstrate that emissions from the source are sufficiently low to protect human health and safety from potential carcinogenic and/or other toxic effects. Compliance shall be demonstrated in any area which does not have restricted or controlled public access. The source shall demonstrate compliance by using procedures established in this chapter after complying with the control technology requirements in WAC 173-460-060.

#### NEW SECTION

WAC 173-460-080 DEMONSTRATING AMBIENT IMPACT COMPLIANCE. (1) When applying for a notice of construction under WAC 173-460-040, the owner or operator of a new toxic air pollutant source which is likely to increase TAP emissions shall complete an acceptable source impact level analysis for Class A and Class B TAPs. The authority may complete this analysis.

(2) Acceptable source impact analysis.

(a) Carcinogenic effects. The owner or operator shall use dispersion modeling to estimate the maximum incremental ambient impact of each Class A TAP from the source and compare the estimated incremental ambient values to the Class A acceptable source impact levels in WAC 173-460-150. If applicable, the source may use the small quantity emission rate tables in (e) of this subsection.

(b) Other toxic effects. The owner or operator shall use dispersion modeling to estimate the maximum incremental ambient impact of each Class B TAP from the source and compare the estimated ambient values to the Class B acceptable source impact levels in WAC 173-460-160. If applicable, the source may use the small quantity emission rate tables in (e) of this subsection.

(c) Dispersion modeling. The owner or operator shall use dispersion modeling techniques in accordance with EPA guidelines. If concentrations predicted by dispersion screening models exceed applicable acceptable source impact levels, more refined modeling and/or emission estimation techniques shall be used. Refined modeling techniques shall be approved by ecology and the authority. (Note: EPA's guideline on Air Quality Models, EPA 450/2-78-0277R, can be obtained through NTIS (703) 487-4650).

(d) Averaging times. The owner or operator shall use the averaging times in (d)(i), (ii), (iii) of this subsection unless alternate averaging times are approved by ecology. Ecology may allow the use of an alternate averaging time if it determines that the operating procedures of the source may cause a high concentration of a TAP for a short period and that consideration of potential health effects due to peak exposures may be warranted for the TAP.

(i) An annual average shall be used for Class A TAPs listed in WAC 173-460-150(2).

(ii) The averaging times specified in WAC 173-460-150(3) shall be used for Class A TAPS listed in WAC 173-460-150(3).

(iii) A twenty-four-hour averaging time shall be used for Class B TAPs listed in WAC 173-460-160.

(e) Small quantity emission rates. Instead of using dispersion modeling to show compliance with ambient impact demonstration requirements in WAC 173-460-080 and 173-460-090, a source may use the small quantity emission rate tables for all toxic air pollutants with acceptable source impact levels equal to or greater than 0.001 ug/m3. A source must first meet control technology and emission quantification requirements of WAC 173-460-050 and 173-460-060, then demonstrate that the source emission rate does not exceed the rates specified in the appropriate table below.

### SMALL QUANTITY EMISSION RATES CLASS A TOXIC AIR POLLUTANTS

Acceptable Source Impact
Level (Annual ug/m3)
Pounds per Year
(10 meter stack
and downwash)

0.001 to 0.0099	0.5
0.01 to 0.06	10
0.07 to 0.12	20
0.13 to 0.99	50
1.0 to 10	500

[ 9 ]

### SMALL QUANTITY EMISSION RATES CLASS B TOXIC AIR POLLUTANTS

Acceptable Source Impact		7	AP Emis	sions		
Level (24 hour ud/x3)	Pounds	ber	iear	Pounds	D+6€	Hour

Less than 1	175	0.02
1 to 9.9	175	J.02
10 to 29.3	1,750	J.20
30 to 59.9	5,250	0.60
60 to 99.9	10,500	1.20
100 to 129.9	17,500	2.0
130 to 250	22,750	2.6
Greater than 250	43,748	5.0

(3) Criteria for compliance. Compliance with WAC 173-460-070 is lemonstrated if the authority determines that, on the basis of the acceptable source impact analysis, the source's maximum incremental ambient air impact levels do not exceed the Class A or Class B acceptable source impact levels in WAC 173-460-150 and 173-460-160; or, if applicable, the source TAP emission rates do not exceed the rates specified in subsection (2) (e) of this section.

#### HEW SECTION

WAC 173-460-090 SECOND TIPE ANALYSIS. (1) Applicability.

- (a) The owner or operator who cannot demonstrate class A or class 3 TAP source compliance with WAC 173-460-070 and 173-460-080 using an acceptable source impact level analysis as provided in WAC 173-460-080(2), may submit a petition requesting ecology perform a second tier analysis evaluation to determine a means of compliance with WAC 173-460-070 and 173-460-080 by establishing allowable emissions for the source. Petitions for second tier analysis evaluation shall be submitted to the local authority or ecology if ecology has jurisdiction over the source. Petitions received by local authorities shall be submitted to ecology within ten days of receipt. A second tier analysis evaluation may be requested when a source wishes to more accurately characterize risks, to justify risks greater than acceptable source impact levels, or to otherwise modify assumptions to more accurately represent risks. Risks may be more accurately characterized by utilizing updated EPA unit risk factors, inhalation reference doses, or other EPA recognized or approved methods. Ecology shall specify the maximum allowable emissions of any class A or class H. TAP, source based on ecology's second tier analysis evaluation.
- (b) Ecology shall evaluate a source's second tier analysis only if:
- (i) The authority has advised ecology that other conditions for processing the notice of construction have been met; and
- (ii) Emission controls contained in the conditional notice of construction represent at least T-BACT; and
- (iii) Ambient concentrations exceed acceptable source impact levels after using more refined emission quantification and air dispersion modeling techniques.
- (c) Ecology shall determine whether the conditions in (b) (i), (ii), and (iii) of this subsection for a second tier analysis have been satisfied within ten working days of receipt of all information needed to make the determination. The matter shall be returned to the authority if ecology finds the conditions for a second tier analysis evaluation have not been met.
  - (2) Jurisdiction.

- (a) Any second tier analysis application submitted by a source wishing to emit toxic air pollutants at levels greater than the acceptable source impact level contained in WAC 173-460-150 or 173-460-160 shall be approved or rejected by ecology.
- (b) Any new emission limits approved by ecology as a result of the second tier analysis evaluation shall be enforced by the authority provided the authority approves the new emission limits.
  - (3) Approval criteria.
- (a) Based on the second tier analysis, ecology may approve the emissions of TAPs from a source where ambient concentrations exceed acceptable source impact levels only if it determines that emission controls represent at least T-BACT and the source demonstrates that emissions of Class A TAPs are not likely to result in an increased cancer risk of more than one in one hundred thousand. The emission of Class A TAPs at levels likely to result in an increased cancer risk of more than one in one hundred thousand requires the approval of the director after complying with WAC 173-460-100.

  (b) Ecology shall consider the second tier analysis and other
- (b) Ecology shall consider the second tier analysis and other information submitted by the applicant as well as department of health comments.
- (i) Comments from other agencies and universities with appropriate expertise may also be considered in the decision to approve emissions that exceed acceptable source impact levels.
- (ii) Public comments shall be considered if the source applies for a risk management decision under WAC 173-460-100.
  - (4) Contents of the second tier analysis.
- (a) The second tier analysis consists of a health impact assessment. The applicant shall complete and submit a health impact assessment to ecology which includes the following information. Ecology may approve the submittal of less information if it determines that such information is sufficient to perform the second tier analysis evaluation. The health impact assessment shall be prepared in accordance with EPA's risk assessment guidelines as defined in WAC 173-460-020(6).
- Demographics such as population size, growth, and sensitive subgroups;
- (ii) Toxicological profiles of all toxic air pollutants that exceed the ASIL;
- (iii) Characterization of existing pathways and total daily intake for toxic air pollutants that exceed the ASIL;
- (iv) Contribution of the proposed source toward total daily intake for toxic air pollutants that exceed the ASIL;
- (v) Using existing data, characterization of risk from current exposure to the toxic air pollutants that exceed the ASIL. This includes existing TAP sources in the area, and anticipated risk from the new source;
- (vi) Additive cancer risk for all Class A toxic air pollutants which may be emitted by the source;
- (vii) Other information requested by ecology and pertinent to ecology's decision to approve the second tier application:
  - (viii) Uncertainty in the data; and
  - (ix) Length of exposure and persistence in the environment.
- (b) The health assessment shall utilize current scientific information. New scientific information on the toxicological characteristics of toxic air pollutants may be used to justify modifications of upper bound unit risk factors used to calculate ASILs in WAC 173-460-150 and/or absorption rates of individual toxic air pollutants if ecology determines there is compelling scientific data which demonstrates that the use of EPA recognized or approved methods are inappropriate.
  - (5) Additional information.
- (a) If approved by ecology, newly discovered scientific information which was unavailable at the time of the original submission of the health assessment may be used to justify modifications of the original health assessment. Ecology may approve the additional information if the source exercised due diligence at the time of original submission.

- (b) Within thirty days after receipt of the second tier analysis and all supporting data and documentation, ecology may require the submission of additional information needed to evaluate the second tier analysis.
  - (6) Determination.
- (a) If the second tier analysis is approved by ecology, ecology will return the petition to the authority and the authority may approve the notice of construction.
- (b) The authority shall specify allowable emissions consistent with ecology's second tier analysis evaluation determination expressed in weight of pollutant per unit time for each emissions unit involved in the application. The notice of construction shall also include all requirements necessary to assure that conditions of this chapter and chapter 173-400 WAC are satisfied.
  - (7) Public notification requirements.
- Ecology decisions regarding second tier analysis or decisions under WAC 173-460-100 shall comply with public notification requirements contained in WAC 173-400-171.

#### NEW SECTION

- WAC 173-460-100 REQUEST FOR RISK MANAGEMENT DECISION. (1) Applicability. The owner or operator of a source that emits Class A TAPs that are likely to result in an increased cancer risk of more than one in one hundred thousand may request that ecology establish allowable emissions for the source.
  - (2) Contents of the application.
- The applicant shall meet the submittal requirements of WAC 173-460-090(1) and submit all materials required under WAC 173-460-090 (4) and (5). The applicant may submit the request for a risk management decision concurrently with the second tier analysis application. Prior denial of the second tier analysis application under WAC 173-460-090(6) is not required.
- (3) Criteria for approval. Ecology may approve the emissions of TAPs from a source where ambient concentrations are likely to result in an increased cancer risk of more than one in one hundred thousand only if the source first demonstrates the following:
- (a) Proposed emission controls represent all known available and reasonable technology; and
- (b) Application of all known available toxic air pollution prevention methods to reduce, avoid, or eliminate toxic air pollutants prior to their generation including recycling, chemical substitution, and efforts to redesign processes; and
- (c) The proposed changes will result in a greater benefit to the environment as a whole.
- (4) Additional methods to reduce toxic air pollutants. In addition to the requirements in subsection (3) of this section, the owner or operator may propose and ecology may consider innovative or established measures that are likely to reduce community exposure to toxic air pollutants provided that such measures are not already required. Examples of innovative measures include but are not limited to:
- (a) Reducing vehicle miles traveled to the facility through vanpool programs and transportation management plans;
  - (b) Permanent removal of woodstoves; and
- (c) Purchasing used automobiles. Examples of established methods include, but are not limited to, emission bubbles and offsets.
- (5) Public involvement. Ecology will endeavor to initiate public notice and comment within thirty days of receipt of a completed risk management decision application. In addition to the public notice and comment requirements of WAC 173-400-171, the owner or operator shall:
- (a) Present the results of the second tier analysis, the proposed emission controls, pollution prevention methods, additional proposed measures, and remaining risks; and

(b) Participate in discussions with and answer questions from the iffected community.

(6) Time limitation. The owner or operator shall commence construction within eighteen months of the director's approval.

#### MEW SECTION

AAC 173-460-110 ACCEPTABLE SOURCE IMPACT LEVELS. There are three types of acceptable source impact levels: Risk-based, threshold-based, and special acceptable source impact levels. They are computed as follows:

(1) Risk-based acceptable source impact levels for Class A TaPs. Risk-based acceptable source impact levels means the annual average concentration, in micrograms per cubic meter, that may cause in increased cancer risk of one in one million. Ecology shall calculate the risk-based acceptable source impact levels for Class A TAPs in AAC173-460-150(2) using the following equation:

Risk based ASIL = (ug/m3)

1198

There:

RISK=Cancer risk level (1 in 1,000,009)

TRF=Upper bound unit risk factor as nublished in IRIS data base or other appropriate sources (ug/m3)-1.

- (2) Threshold-based acceptable source impact levels for Class 3 TAPS. Threshold-based acceptable source impact levels in \*AC 173-460-160 shall be determined as follows:
- (a) If a Class 8 TAP has an Environmental Protection Agency Inhalation Reference Jose, the innalation reference dose and specified averaging time shall be used.
- (b) Other Class 3 TAP acceptable source impact levels shall be determined by dividing the TLY-TWA by three numbered to calculate a twenty-four hour TWA acceptable source impact level.
  - (3) Special acceptable source impact levels.
- (a) Scology may establish special acceptable source impact levels for TAPs for which upper bound risk factors or TLVs have not been established, or for mixtures of compounds if it determines that the above acceptable source impact level methods are not appropriate, lo not adequately protect human health or are overly stringent.
- (b) The averaging times for special ASILs are listed in WAC 173-450-150(3).

#### JEH SECTION

JAC 173-460-120 SCIENTIFIC REVIEW AND AMENDMENT OF ACCEPTABLE

SOURCE IMPACT LEVELS AND LISTS. (1) Ongoing scientific review.

(a) To use the sest available scientific information, ecology shall conduct an ongoing review of information concerning whether to add or delete toxic air pollutants to #AC 173-460-150 or 173-460-160, what acceptable source impact levels should be used to review emissions of TAPs, source applicability and exemptions.

(b) A complete review shall be made at least once every three

years at which time ecology shall consider scientific information developed by the E.P.A., Washington department of health, other states or other scientific organizations, scientific information provided by any person, and results of second tier analyses evaluations.

[ 13 ]

- (2) Criteria for listing as Class A or Class B TAP.
- (a) Ecology shall list a substance or group of substances as Class A or Class B TAPs if the department has reason to believe that the compound or group of compounds are likely to be emitted to the air from an air pollution source and the air emission of such compound or compounds could impact public health. The compounds shall be removed from the list if ecology determines that these conditions no longer exist.
- (b) Ecology may list mixtures of compounds as Class A and/or Class B TAPs if ecology determines that the health impact of the emission mixture is likely to be different from the known individual chemical impacts.
  - (3) Acceptable source impact level (ASIL).

Ecology may adopt an ASIL only if ecology determines that concentrations at that level will not unreasonably endanger human health.

#### NEW SECTION

WAC 173-460-130 FEES. (1) Pursuant to RCW 70.94.152, ecology or the authority may charge a fee for the review of notices of construction.

(2) The fee imposed under this section may not exceed the cost of reviewing plans, specifications, and other information and administering such notice.

#### YEW SECTION

WAC 173-460-140 REMEDIES. Violations of this chapter are subject to the behalty provisions and/or other remedies provided in chapter 70.94 RCW.

#### NEW SECTION

CAS I

WAC 173-460-150 CLASS A TOXIC AIR POLLUTANTS: KNOWN, PROBABLE AND POTENTIAL HUMAN CARCINOGENS AND ACCEPTABLE SOURCE IMPACT LEVELS.

(1) TABLE I CLASS A TOXIC AIR POLLUTANTS Known and Probable Carcinogens

SUBSTANCE

75-07-0	Acetaldehyde
107-13-1	Acrylonitrile
309-00-2	Aldrin
	Aluminum smelter polyaromatic hydrocarbon emissions
117-79-3	2-Aminoanthraquinone
97-56-3	o-Aminoazotoluene
92-67-1	4-Aminobiphenyl
61-82-5	Amitrole
	Arsenic and inorganic arsenic compounds
1332-21-4	Asbestos

```
CAS #
                 SUBSTANCE
2465-27-2
                 Auramine (technical grade)
56-55-3
                  Benz (a) anthracene
71-43-2
                  Benzene
92-97-5
                  Benzidine and its salts
50-32-8
                  Benzo (a) pyrene
204-99-2
                  Benzo(b) fluoranthene
205-82-3
                  Benzo (j) fluoranthene
205-08-9
                  Benzo(k) fluoranthene
1694-09-3
                  Benzyl violet 4b
                  Beryllium and compounds
111-44-4
                  Bis (2-chloroethyl) ether
117-81-7
                  Bis (2-ethylhexyl) phthalate
542-88-1
                  Bis (chloromethyl) ether and technical-grade
                   chloromethyl methyl ether
106-99-0
                  1.3-Butadiene
3068-88-0
                  B-Butyrolactone
                  Cadmium and compounds
56-23-5
                  Carbon tetrachloride
57-74-9
                  Chlordane
74-87-3
                  Chlorodibromoethane
67-66-3
                  Chlorofor.
107-30-2
                  Chloromethyl methyl ether (technical-grade)
108-43-0
                  Chlorophenols
126-99-8
                  Chloroprene
                  Chromium, hexavalent metal and compounds
  ___
                  Coke oven emissions
8001-58-9
                  Creosote
135-20-6
                  Cupferron
94-75-7
                  2.4-D and esters
50-29-3
                  DDT (1,1,1 Trichloro-2,2-Bis(p-chlorophenyl)-ethane)
613-35-4
                  N, N-Diacetylbenzidine
101-80-4
                  4,4*-Diaminodiphenyl ether
226-36-8
                  Dibenz (a, h) acridine
53-70-3
                  Dibenz (a, h) anthracene
224-42-0
                  Dibenz (a, j) acridine
189-64-0
                  Dibenzo (a, h) pyrene
191-30-0
                  Dibenzo(a, 1) pyrene
189-55-9
                  1,2:7,8-Dibenzopyrene (dibenzo(a,i)pyrene)
192-65-4
                  Dibenzo(a,e) pyrene
                  1,4-Dichloro-2-butene
28434-86-8
                  3,3°-Dichloro-4,4°-diaminodiphenyl ether
106-46-7
                  1,4-Dichlorobenzene
91-94-1
                  3,3'-Dichlorobenzidine
107-06-2
                  1,2-Dichloroethane (ethylene chloride)
                  Dichloromethane (methylene chloride)
75-09-2
696-28-6
                  Dichlorophenylarsine (arsenic group)
78-87-5
                  1.2-Dichloropropane
60-57-1
                  Dieldrin
1615-80-1
                  1,2-Diethylhydrazine
101-90-6
                  Diglycidyl resorcinol ether
                  3,3'-Dimethoxybenzidine (ortol-dianisidine)
119-90-4
77-78-1
                  Dimethyl sulfate
540-73-8
                  1,2-Dimethylhydrazine
25321-14-6
                  Dinitrotoluenes (mixed)
123-91-9
                  1,4-Dioxane
                  Dioxins and furans
122-66-7
                  1,2-Diphenylhydrazine
106-93-4
                  Ethylene Dibromide
 75-21-8
                  Ethylene oxide
50-00-0
                  Pormaldehyde
                  Purium (nitrofuran group)
765-34-4
                  Glyciadaldehyde
76-44-8
                  Heptachlor
```

CAS #	SUBSTANCE
118-74-1	Hexachlorobenzene
319-84-6	Herachlorocycloherane (Lindane) Alpha BHC
319-85-7	Hexachlorocyclohexane (Lindane) Beta BHC
580-89-9	Hexachlorocyclohexane (Lindane) Gamma BHC
67-72-1	Hexachloroethane
193-39-5	Indeno (1, 2, 3-cd) pyrene
	Isopropyl oils
301-04-2	Lead acetate
7446-27-7	Lead phosphate
129-15-7	2-Methyl-1-mitroanthraquinone
592-62-1	Methylazoxymethanol & acetate
3697-24-3	5-Methylchrysene
101-14-4	4,4'-Methylenebis(2-chloroaniline) (MBOCA)
838-88-0	4,4°-Methylenebis(2-methylaniline)
101-77-9	4,4-Methylenedianiline
13552-44-8	4,4-Methylenedianiline dihydrochloride
64091-91-4	4- (Methylnitrosamino) -1- (3-pyridyl) -1-butanone
	Mirex
1 39-91-3	5- (Morpholinomethyl) - 3- ((5-nitrofurfurylidene) amino) -
	2-oxazolidin one
924-16-3	N-Mirtrosodi-a-butylamine
134-32-7	1-Napthylamine
7440-02-0	Nickel and compounds
531-82-8	N-(4-(5-Wirto-2-furyl)-2-thiazolyl) acetamide
759-73-9	N-Nirtoso-n-ethylurea (NEU)
621-64-7	N-Nirtosodi-n-propylamine
10595-95-6	N-Nirtosomethylethylamine
59-89-2	N-Nirtosomorpholine
86-30-6	N-Nirtrosdiphenylamine
55-18-5	N-Nirtrosodiethylamine (diethylnitrosoamine) (DEN)
62-75-9 602-97-9	N-Nirtrosodimethylamine 5-Nitroacenaphthene
1836-75-5	Nitrofen
1030-13-3	Nitrofurans Purazolidone
59-87-0	Nitrofurazone
555-84-9	1-(5-Nitrofurfurylidene) amino)-2-imidazolidinone
126-85-2	Nitrogen mustard N-oxide
302-70-5	Witrogen mustard n-oxide hydrochloride
79-46-9	2-Nitropropane
615-53-2	N-Nitroso-n-methylurethane
2646-17-5	Oil orange SS
794-93-4	Panfuran S (dihydroxymethylfuratrizine)
127-18-4	Perchloroethylene (tetrachloroethylene)
63-92-3	Phenoxybenzamine hydrochloride
	N-Phenyl-2-napthylamine
	Polyaromatic Hydrocarbons (PAH)
1336-36-3	Polychlorinated biphenyls (PCBs)
3761-53-3	Ponceau MX
	P(p)(alpha, alpha, alpha)-Tetra-chlorotoluene
1746-01-6	2,3,7,8-Tetrachlorodihenzo-p-dioxin (2,3,7,8-TCDD)
139-65-1	4,4'-Thiodianiline
1314-20-1	Thorium dioxide
584-84-9	2,4-Toluene diisocyanate
95-53-4	o-Toluidine ε its hydrochloride
8001-35-2	Toxaphene
55738-54-0	Trans-2((Dimethylamino) methylimino) -5-
	(5-nitro-2-furyl)
70-01-5	vinyl-1,3,4-oxadiazole
79-01-6	Trichlorophenol (mixed)
25167-82-2 75-01-0	Trichlorophenol (mixed) Vinyl Chloride
75-01-4	ATRAT CHIOFING

# (2) TABLE II CLASS A TOXIC AIR POLLUTANTS WITH ESTABLISHED ACCEPTABLE SOURCE IMPACT LEVELS

		10-6 RISK
		ASIL MICRO- GRAMS/M <sup>3</sup>
		ANNUAL
CAS #	SUBSTANCE	AVERAGE
75-07-0	Acetaldehyde	0.4500000
107-13-1	Acrylonitrile	0.0150000
309-00-2	Aldrin	0.0002000
	Arsenic and inorganic arsenic compounds	0.0002300
1332-21-4	Asbestos (Note: fibers/ml)	0.0000042
71-43-2	Benzene	0.1200000
92-87-5	Benzidine and its salts	0.0000150
50-32-8	Benzo (a) pyrene	0.0006000
111 00 0	Beryllium and compounds	0.0004200
111-44-4 542-88-1	Bis(2-chloroethyl)ether Bis(chloromethyl)ether and technical-	0.0030000
342-00-1		0.0000160
	grade chloromethyl methyl ether Cadmium and compounds	0.0005600
56-23-5	Carbon tetrachloride	0.0670000
57-74-9	Chlordane	0.007000
67-66-3	Chloroform	0.0430000
108-43-0	Cholorphenols	0.1800000
	Chromium, hexavalent metal and compounds	
	Coke oven emissions	0.0016000
50-29-3	DDT (1,1,1 Trichloro-2,2-Bis-	
	(p-chlorophenyl) -ethane)	0.0100000
	1,4-Dichloro-2-butene	0.0003800
107-06-2	1,2-Dichloroethane (ethylene chloride)	0.0400000
75-09-2	Dichloromethane (methylene chloride)	2.0000000
60-57-1	Dieldrin	0.0002000
122-66-7	1,2-Diphenylhydrazine	0.0045000
106-93-4	Ethylene Dibromide	0.0045000
75-21-8	Ethylene oxide	0.0100000
50-00-0	Pormaldehyde	0.0770000
76-44-8	Heptachlor	0.0007700
118-74-1	Hexachlorobenzene	0.0020000
67-72-1	Hexachloroethane	0.2500000
127-18-4	Perchloroethylene (tetrachloroethylene)	1.1000000
1746-01-6	2,3,7,8-Tetrachlorodibenzi-p-dioxin	
	(2,3,7,8-TCDD)	0.0000003
8001-35-2	Toraphene	0.0030000
79-01-6	Trichloroethylene	0.8000000
25167-82-2	Trichlorophenol (mixed)	0.1800000
75-01-4	Vinyl Chloride	0.0230000

## (3) TABLE III CLASS A TOXIC AIR POLLUTANTS WITH SPECIAL ACCEPTABLE SOURCE IMPACT LEVELS

CAS •	SUBSTANCE	ASIL HICRO- GRAHS/H <sup>3</sup>	AVERAGING TIME
	Primary aluminum smelter uncontrolled roof vent polyaromatic hydrocarbon (PAH) emissions (Note: Quantify according to WAC 173-460-050 (4)(d))	0.0013	Annual
61-82-5	Amitrole	0.6	24 hour
106-99-0	1.3-Butadiene	73.3	24 hour
126-99-8	B-Chloroprene	116.6	24 hour
94-75-7	F1377 - 1334 S13770 W W 100 - 1647 S1377 S137 S137 L L L L L L L L L L L L L L L L L L L	33.3	24 hour
106-46-7	1.4-Dichlorobenzene	1500	24 hour
78-87-5	1,2-Dichloropropane	1166.6	24 hour
77-78-1	Dimethyl sulfate	1.6	24 hour
540-73-8	1,2-Dimethylhydrazine	3.3	24 hour
123-91-9	1,4-Dioxane	300	24 hour
58-89-9	Lindane	1.6	24 hour
101-14-4	4,4'-Methylenebis	A-7-7-7	
	(2-Chloroaniline) (MBOCA)	0.7	24 hour
101-77-9	4,4-Methylenedianiline	2.6	24 hour
7440-02-0	Nickel and compounds	3.3	24 hour
79-46-9	2-Nitropropane	116.6	24 hour
	Polyaromatic hydrocarbon (PAH) emissions (Note: Quantify according to WAC 173-460-050 (4)(d))	0.0006	Annual
584-84-9	2.4-Toluene diisocyanate	0.1	24 hour
95-53-4	O-Toluidine	30	24 hour

#### HEW SECTION

WAC 173-460-160 CLASS B TOXIC AIR POLLUTANTS AND ACCEPTABLE SOURCE IMPACT LEVELS. The following table lists class B toxic air pollutants and acceptable source impact levels:

## CLASS B TOXIC AIR POLLUTANTS AND ACCEPTABLE SOURCE IMPACT LEVELS

		ASIL MICROGRAMS/MITMENTY-FOUR-HOUR	
CAS #	SUBSTANCE	AVERAGE	
5471			
86-88-4	ANTU	1.0	
75-07-0	Acetic acid	83.3	
108-24-7	Acetic anhydride	66.6	
67-64-1	Acetone	5927.4	
75-05-8	Acetonitrile	233.1	
79-27-6	Acetylene tetrabromide	50.0	
107-02-8	Acrolein	0.8	

		ASIL MICROGRAMS/N3
AS #	SUBSTANCE	TWENTY-FOUR-HOUR AVERAGE
9-06-1	Acrylamide	0.1
9-10-7	Acrylic acid	99.9
07-18-6	Allyl alcohol	16.7
06-92-3	Allyl glycidyl ether (AGE)	73.3
179-59-1		40.0
429-90-5	Aluminum, Al alkyls	6.7
429-90-5		33.3
429-90-5	• • • • • • • • • • • • • • • • • • • •	16.7
429-90-5	Aluminum, as Al soluble salts	6.7
429-90-5		16.7
04-29-0	2-Aminopyridine	6.7
664-41-7	Assonia	59.9
2125-02-9	Ammonium chloride fume	33.3
825-26-1	Ammonium perfluorooctanoate	0.3
773-06-0	Ammonium sulfamate	33.3
28-63-7		1764.9
26-38-0	sec-Amyl acetate	2214.5
2-53-3	Aniline & homologues	33.3
	Anisidine (o-,p- isomers)	1.7
440-36-0	Antimony 6 compounds as Sb	1.7
309-64-4		1.7
784-42-1		0.7
1052-42-4	market (factorial) rando	16.7
912-24-9		16.7
6-50-0	Azinphos-methyl	0.7
440-39-3		1.7
7804-35-2 14-36-0		33.3
		16.7
00-44-7 12-52-4	Benzyl chloride	16.7
304-82-1	•	5.0
304-82-1		33.3
304-02-1		16.7
303-96-4		3.3
303-96-4		16.7
303-86-2	Borates, pentahydrate Boron oxide	3.3
	Boron tribromide	33.3
726-95-6		33.3
14-40-9		10.0
726-95-6		33.3
789-30-2		2•3 2•3
5-25-2	Bromoform	16.7
06-97-8	Butane	6327.0
11-76-2	2-Butoxyethanol	399.6
23-86-4	n-Butyl acetate	2364.3
05-46-4	sec-Butyl acetate	3163.5
40-88-5	tert-Butyl acetate	3163.5
41-32-2	Butyl acrylate	183.2
1-36-3	n-Butyl alcohol	499.5
8-92-2	sec-Butyl alcohol	1015.7
5-65-0	tert-Butyl alcohol	999.0
189-85-1	tert-Butyl chromate, as Cr03	0.3
426-08-6	n-Butyl glycidyl ether (BGE)	449.6
38-22-7	n-Butyl lactate	83.3
09-79-5	Butyl mercaptan	5.0
09-73-9	Butylamine	50.0
9-72-5	o-sec-Butylphenol	99.9
8-51-1	p-tert-Butyltoluene	199.8
56-62-7	Calcium cyanamide	1.7
305-62-0	Calcium hydroxide	16.7
305-78-8	Calcium oxide	

ris 1	SUBSTANCE	ASIL MICROGRAMS/M <sup>3</sup> TWENTY-FOUR-HOUR AVERAGE
76-22-2	Camphor, synthetic Caprolactam, dust	40.0
105-60-2	Caprolactam, dust	3.3
	Caprolactam, vapor	66.6
2425-06-1	Captafol	0.3
133-06-2	Captan Carbaryl	16.7
63-25-2 1563-66-3	Carbafyl	16.7 0.3
	Carbofuran Carbon black	11.7
1333-00-4 75-15-0	Carbon disulfide	99.9
558-13-4	Carbon disulfide Carbon tetrabromide	4.7
353-50-4	Carbonyl fluoride	16.7
120-80-9	Catechol	66.6
21351-79-1	Cesium hydroxide	6.7
8001-35-2	Chlorinated camphene	1.7
		1.7
7782-50-5	Chlorine	10.0
	Chlorine dioxide	1.0
	Chlorine trifluoride	1.3
	1-Chloro-1-nitropropane	33.3
	Chloroacetaldehyde	10.0
532-27-4	a-Chloroacetophenone Chloroacetyl chloride	1.0
79-04-9	Chloroacetyl chloride	0.7
	o-Chlorobenylidene malonitrile	1.3
108-90-7		1165.5
74-97-5	Chlorobromomethane Chlorodifluoromethane	3496.5 11655.0
75-45-6		21045.6
76-06-2	Chloropentafluoroethane Chloropicrin	2.3
2039-87-4	o-Chlorostyrene	949.1
95-49-8	o-Chlorotoluene	832.5
	Chlorpyrifos	0.7
	Chromium (II) compounds, as Cr	1.7
	Chromium (III) compounds, Cr	1.7
	Chromium (metal)	1.7
	8 Chromyl chloride	0.5
2971-90-6	Clopidol	33.3
	Cobalt as Co metal Dust and fu	0.2
	1 Cobalt carbonyl as Co	0.3
	8 Cobalt hydrocarbonyl	0.3
	Copper, Dusts and mists, as Cu	3.3
	Copper, Pume	0.7
4340 77 3	Cotton dust, rav	0.7 73.3
4170-30-3	Cresol, all isomers Crotonaldehyde	20.0
299-86-5	Crufomate	16.7
98-82-2	Cumene	815.9
420-04-2	Cyanamide	6.7
151-50-8	Cyanides, as CN	16.7
460-19-5	Cyanogen	66.6
506-77-4	Cyanogen chloride	2.0
110-82-7	Cyclohexane	3496.5
108-93-0	Cyclohexanol	666.0
108-94-1	Cyclohexanone	333.0
110-83-8	Cycloherene	3380.0
108-91-8	Cyclohexylamine	133.2
121-82-4	Cyclomite	5.0
542-92-7	Cyclopentadiene	666.0
287-92-3	Cyclopentane	5727.6
	5 Cyhexatin	16.7
94-75-7	2,4-D	33.3 1.0
	9 Decaborane	

CAS #	SUBSTANCE	ASIL MICROGRAMS/M <sup>3</sup> TWENTY-FOUR-HOUR AVERAGE
8065-48-3	Demeton	
117-81-7	Di (2-ethylhexyl) phthalate	0.3 16.7
123-42-2	Diacetone alcohol	799.2
333-41-5	Diazinon	0.3
334-68-3	Diazomethane	1.3
19287-45-7	Diborane	0.3
107-66-4	Dibutyl phosphate	16.7
84-74-2	Dibutyl phthalate	16.7
102-81-8	2-N-Dibutylaminoethanol	46.6
594-72-9	1,1-Dichloro-1-nitroethane	33.3
118-52-5	1,3-Dichloro-5,5-dimethyl hydantoin	0.7
7572-29-4	Dichloroacetylene	1.3
95-50-1	o-Dichlorobenzene	999.0
106-46-7	p-Dichlorobenzene	1498.5
75-71-8	Dichlorodifluoromethane	16483.5
75-34-3	1,1-Dichloroethane	2697.3
111-44-4	Dichloroethyl ether	99.9
540-59-0	1,2-Dichloroethylene	2630.7
75-43-4	Dichlorofluoromethane	133.2
78-87-5 543-75-6	1,2-Dichloropropane	1165.5
542-75-6 75-00-0	Dichloropropene	16.7
75-99-0	2,2-Dichloropropionic acid	20.0
76-14-2 62-73-7	Dichlorotetrafluoroethane	23310.0
141-66-2	Dichlorvas Dicrotophos	3.3
77-73-6	Dicyclopentadiene	0.8
102-54-5	Dicyclopentadienyl iron	99.9
60-57-1	Dieldrin	33.3
111-42-2	Diethanolamine	0.8
96-22-0	Diethyl ketone	50.0
84-66-2	Diethyl phthalate	2347.7
109-89-7	Diethylamine	16.7 99.9
100-37-8	Diethylaminoethanol	166.5
111-40-0	Diethylene triamine	13.3
75-61-6	Difluorodibromomethane	2863.8
2238-07-5	Diglycidyl ether	1.7
108-83-8	Diisobutyl ketone	499.5
108-18-9	Diisopropylamine	66.6
127-19-5	Dimethyl acetamide	116.6
1 24-40-3	Dimethylamine	59.9
121-69-7	Dimethylaniline	83.3
68-12-2	Dimethylformamide	99.9
57-14-7	1,1-Dimethylhydrazine	3.3
131-11-3	Dimethylphthalate	16.7
148-01-6	Dinitolmide	16.7
534-52-1	Dinitro-o-cresol	0.7
528-29-0	Dinitrobenzene, all isomers	3.3
78-34-2	Dioxathion	0.7
122-39-4	Diphenylamine	33.3
123-19-3	Dipropyl ketone	782.6
3459U-94-8	Dipropylene glycol methyl ether	1998.0
85-00-7	Diquat	1.7
97-77-8 398-04-4	Disulfiram	6.7
298-04-4 128-27-0	Disulfuton	0.3
1 28-37-0 3 20-54-1	2,6-Ditert. butyl-p-cresol	33.3
330-54-1 1321-74-0	Diuron	33.3
2104-64-5	Divinyl benzene EPN	166.5
115-29-7	Endosulfan	1.7
72-20-8	Endosultan Endrin	0.3
		0.3
13838-16-9	Paflorana	1914.8

į		ASIL MICHOGRAMS/M3 TWENTY-FOUR-HOUR
CAS #	SUBSTANCE	AVERAGE
41-43-5	Ethanolamine	26.6
63-12-2	Ethion	1.3
10-80-5	2-Ethoxyethanol 2-Ethoxyethyl acetate	63.3
11-15-9	2-Ethoxyethyl acetate	89.9
0-29-7	Ethyl Ether	3996.0
41-78-6	Ethyl acetate	4662.0
40-88-5	Ethyl acrylate Ethyl alcohol	66.6
4-1/-5 41-06-6	Ethyl alcohol	6327.0
641-85-5 00-41-4	Ethyl amyl ketone Ethyl benzene	432.9
4-96-4	Ethyl bromide	1448.6
06-35-4	Ethyl butyl ketone	2963.7
5-00-3	Ethyl chloride	765.9
09-94-4		8658.0
5-08-1	Ethyl mercaptan	999.0 3.3
8-10-4	Ethyl silicate	
	Ethylamine	283.1 59.9
07-07-3		10.0
	Ethylene glycol	416.3
28-96-6		1.0
07-15-3	Ethylenediamine	83.3
151-56-4	Ethylenimine	3.3
	Ethylidene norbornene	83.3
100-74-3		76.6
22224-92-6	Fenamiphos	0.3
	Pensulfothion	0.3
55-38-9	Penthion	0.7
4484-64-1		33.3
	Ferrovanadium dust	3.3
	Pebrous glass dust	33.3
	Fluorides, as F	8.3
7782-41-4	<del></del> -	6.7
44-22-9		0.3
5-12-7		50.0
4-18-6	Formic acid	30.0
08-01-1	Purfural	26.6
8-00-1	Purfuryl alcohol	133.2
782-65-2		2.0
11-30-8		2.3
556-52-5		249.8
440-58-6		1.7
51-67-7 42-82-5		1332.0
142-62-5 17-68-3	fleptane (n-Heptane) Hexachlorobutadiene	5328.0
7-47-4	flexachlorocyclopentadiene	0.8
335-87-1	Hexachloronaphthalene	0.3
684-16-2	Herafluoroacetone	0.7
22-06-0	Hexamethylene diisocyanate	2.3 0.1
00-54-3	Hexane (n-Hexane)	599.4
	Herane, other isomers	5994.0
91-78-6	2-Hexanone (MBK)	66.6
08-84-9	sec-Heryl acetate	999.0
07-41-5	Hexylene glycol	416.3
	Hydrogen bromide	33.3
647-01-0	Hydrogen chloride	23.3
4-90-8	Hydrogen cyanide	33.3
664-39-3	Hydrogen fluoride, as F	8.3
722-84-1	Rydrogen peroxide	5.0
7783-07-5	Hydrogen selenide, as Se	0.7
783-06-4	Hydrogen sulfide	46.6
23-31-9	Rydroquinone	6.7

		ASIL HICROGRAMS/H3 TWENTY-FOUR-HOUR	
:AS #	SUBSTANCE	AVERAGE	
99-61-1	2-Hydroxypropyl acrylate	10.0	
5-1 3-6	Indene	149.9	
440-74-6	Indium, & compounds as In	0.3	
553-56-2		3.3	
75-47-8	Iodoform	33.3	
309-37-1	Iron oxide fume, Fe203 as Fe	16.7	
	Iron pentacarbonyl, as Pe	2.7	
22 02 2	Iron salts, soluble as Fe	3.3	
23-92-2	Isoamyl acetate	1748.3	
23-51-3 10-19-0	Isoamyl alcohol Isobutyl acetate	1198.8	
8-83-1	Isobutyl alcohol	2331.0	
	Isocytl alcohol	499.5	
8-59-1	Isophorone	899.1	
098-71-9	Isophorone diisocyanate	83.3	
109-59-1	Isopropoxyethanol	0.1	
108-21-4	Isopropyl acetate	349.7 3163.5	
57-63-0	Isopropyl alcohol	3263.4	
08-20-3	Isopropyl ether	3496.5	
016-14-2	Isopropyl glycidyl ether (IGE)	799.2	
75-31-0	Isopropylamine	40.0	
768-52-5	N-Isopropylaniline	33.3	
163-51-4	Ketene	3.0	
3687-31-8	Lead arsenate, as Pb3 (AsO4)2	0.5	
7758-97-6	Lead chromate, as Cr	0.2	
8476-85-7	Liquified petroleum gas	5994.0	
7580-67-8	Lithium hydride	0.1	
309-48-4	Magnesium oxide fume	33.3	
21-75-5	Malathion	33.3	
08-31-6	Maleic anhydride	3.3	
7439-96-5	Manganese Dust & compounds	16.7	
7439-96-5		3.3	
2079-65-1	Manganese cyclopentadienyl tricarbon	yl 0.3	
7439-97-6	manage capa	0.3	
7439-97-6	Mercury, as Hg Alkyl compounds	0.03	
7439-97-6	Mercury, vapors except alkyl	0.2	
141-79-7	Mesityl oxide	199.8	
79-41-4	methacrylic acid	233.1	
6752-77-5		8.3	
/2-43-5   09-86-4	Methoxychlor	33.3	
10-49-6	2-Methoxyethanol	53.3	
50-76-5	2-Methoxyethyl acetate	79.9	
37-05-3	4-Methoxyphenol	16.7	
9-20-9	Methyl 2-cyanoacrylate Methyl acetate	26.6	
4-99-7	Methyl acetylene	2031.3	
	Methyl acetylene-propadiene	5494.5	
	mixture (MAPP)	<b>59</b> 0# 0	
6-33-3	Methyl acrylate	5994.0	
7-56-1	Methyl alcohol	116.6 865.8	
00-61-8	N-Methyl aniline	6.7	
4-83-9	Methyl bromide		
4-87-3	Methyl chloride	66.6	
1-55-6	Methyl chloroform	349.7	
022-00-2	Methyl demeton	6327.0	
8-93-3	Methyl ethyl ketone (MEK)	1.7	
338-23-4	Methyl ethyl ketone peroxide	1964.7	
07-31-3	Methyl formate	5.0	
0-34-4	Methyl hydrazine	832.5	
74-88-4	Methyl iodide	1.2	
10-12-3	Methyl isoamyl ketone	33.3 799.2	
	******** *********************	744.7	

CAS #	SUBSTANCE	ASIL HICROGRAMS/M3 TWENTY-FOUR-HOUR	
		AVERAGE	
108-11-2	Methyl isobutyl carbinol	333.0	
108-10-1	Methyl isobutyl ketone (MIBK)	682.7	
624-83-9	Methyl isocyanate	0.2	
563-80-4	Methyl isopropyl ketone	2347.7	
74-93-1	Methyl mercaptan	3.3	
80-62-6	Methyl methacrylate	1365.3	
110-43-0	Methyl n-amyl ketone	782.6	
591-78-6	Methyl n-butyl ketone	66.6	
298-00-0	Methyl parathion	0.7	
107-87-9	Methyl propyl ketone	2331.0	
681-84-5	Methyl silicate	20.0	
98-83-9 136-00-7	a-Methyl styrene	799.2	
126-98-7 109-87-5	Methylacrylonitrile	10.0	
74-99-5	Methylal Methylamine	10323.0	
108-87-2		40.0	
	Methylcyclohexane Methylcyclohexanol	5328.0	
583-60-8	o-Methylcyclohexanone	782.6	
	Methylcyclopentadienyl	765.9	
	manganese tricarbonyl	0.7	
5124-30-1	Methylene bis (4-cyclo-herylisocyana	te) 0.2	
101-68-8	Methylene bisphenyl isocyanate	0.2	
101-77-9	4,4 -Methylene dianiline	2.7	
	Metribuzin	16.7	
7786-34-7	Mevinphos	0.3	
7439-98-7		16.7	
7439-98-7	Molybdenum, insoluble cpds	33.3	
6923-22-4	Honocrotophos	0.8	
110-91-8	Morpholine	233.1	
300-76-5	Naled	10.0	
91-20-3	Napthalene	166.5	
54-11-5	Nicotine	1.7	
7607-37-3	Nitrapyrin Nitric acid	33.3	
	Witric oxide	16.7	
100-01-6	p-Nitroaniline	99.9	
98-95-3	Mitrobenzene	10.0	
100-00-5	p-Witrochlorobenzene	16.7	
79-24-3	Nitroethane	2.0	
7783-54-2	Nitrogen trifluoride	1032.3 99.9	
55-63-0	Nitroglycerin	1.7	
75-52-5	Nitromethane	832.5	
108-03-2	1-Nitropropane	299.7	
88-72-2	Nitrotoluene	36.6	
111-84-2		3496.5	
2234-13-1	Octachloronaphthalene	0.3	
111-65-9	Octane	4828.5	
8012-95-1	Oil mist, mineral	16.7	
	Osmium tetroxide, as Os	0.007	
144-62-7	Oxalic acid	3.3	
7783-41-7		0.3	
8002-74-2		6.7	
4685-14-7	•	0.3	
56-38-2	Parathion Pentaborane	0.3	
1321-64-8		0.0	
87-86-5	Pentachlorophenol	1.7	
109-66-0	Pentane	1.7	
594-42-3	Perchloromethyl mercaptan	5994.0	
		2.7	
7616-94-6	Perchloryl fluoride	46.6	

CAS #	SUBSTANCE	ASIL MICROGRAMS/M <sup>3</sup> TWENTY-FOUR-HOUR AVERAGE
92-84-2		16.7
101-84-8	Phenyl ether	23.3
122-60-1 108-98-5	Phenyl glycidyl ether Phenyl mercaptan	20.0
106-50-3	p-Phenylene diamine	6.7
100-63-0	Phenylhydrazine	0.3 66.6
638-21-1	Phenylphosphine	0.8
298-02-2	Phorate	0.2
75-44-5	Phosgene	1.3
7803-51-2	Phosphine	1.3
7664-38-2	Phosphoric acid	3.3
7723-14-0	Phosphorus	0.3
10025-87-3	Phosphorus oxychloride	2.0
	Phosphorus pentachloride	3.3
1314-80-3		. 3.3
7719-12-2	Phosphorus trichloride	5.0
636-17 F	Phthalic anhydride m-Phthalodinitrile	20.0
		16.7
1918-02-1 88-89-1		33.3
83-26-1	Pictic acid	0.3
142-64-3	Pindone Piperazine dihydrochloride	0.3
7440-06-4	Platinum, Metal	16.7
7440-06-4	Platinum, Soluble salts as Pt	3.3 0.0
1310-58-3	Potassium hydroxide	6.7
107-19-7		6.7
57-57-8	B-Propiolactone	5.0
114-26-1		1.7
79-09-4 109-60-4	Proprionic acid	99.9
109-60-4	n-Propyl acetate	2797.2
71-23-8		1665.0
627-13-4		349.7
78-87-5		1165.5
6423-43-4	Propylene glycol dinitrate	1.0
107-98-2		
75-55-8 8003-34-7	Propylene imine	16.7
8003-34-7 110-86-1		16.7
106-51-4		50.0
108-46-3	Quinone Resorcinol	1.3
7440-16-6	· · · · · · · · · · · · · · · · · · ·	149.9
7440-16-6		3.3
7440-16-6		3.3 0.03
299-84-3	Ronnel	33.3
83-79-4	Rotenone	16.7
	Rubber solvent (Naphtha)	5328.0
7782-49-2	Selenium compounds, as Se	0.7
7783-79-1	Selenium hexafluoride, as Se	0.7
136-78-7	Sesone	33.3
7803-62-5	Silcon tetrahydride	23.3
7440-22-4		0.3
7440-22-4		0.03
	Sodium azide	1.0
7631-90-5		16.7
62-74-8	Sodium fluoroacetate	0.2
1310-73-2	Sodium hydroxide	6.7
7681-57-4		16.7
7803-52-3	Stibine	1.7
57-24-9 100-42-5	Strychnine	0.5
1395-21-7	Styrene	716.0
1373-21-1	Subtilisins	0.0

ASIL HICROGRAMS/M3

	_	WENTY-FOUR-HOUR	
CAS #	SUBSTANCE	AVERAGE	
3689-24-5	Sulfoten	0.7	
2551-62-4		19980.0	
	Sulfur monochloride	20.0	
5714-22-7	Sulfur pentafluoride	0.3	
7783-60-0		1.3	
7664-93-9	Sulfuric acid	3.3	
2699-79-8	Sulfuryl fluoride	66.6	
35400-43-2	Sulprofos	3.3	
93-76-5	2,4,5-T	33.3	
107-49-3	TEPP	0.2	
7440-25-7		16.7	
	Tellurium & compounds as Te	0.3	
7783-80-4	•	0.7	
3383-96-8		33.3	
	Terphenyls	16.7	
76-12-0	1,1,2,2-Tetrachloro-1,2-difluoroethan		
76-11-9	1,1,1,2-Tetrachloro-2,2-difluoroethan		
79-34-5	1,1,2,2-Tetrachloroethane	23.3	
1335-88-2 78-00-2	Tetrachloronaphthalene Tetraethyl lead, as Pb	6.7 0.3	
109-99-9	Tetrahydrofuran	1964.7	
175-74-1	Tetramethyl lead, as Pb	0.5	
3333-52-6		10.0	
509-14-8	Tetranitromethane	26.6	
7722-88-5		16.7	
479-45-8		5.0	
7440-28-0	•	0.3	
96-69-5	4,4-Thiobis(6-tert, butyl-m-cresol)	33.3	
68-11-1	Thioglycolic acid	13.3	
7719-09-7		16.7	
137-26-8	Thirum	16.7	
7440-31-5	fin, Metal	6.7	
7440-31-5	Tin, Organic compounds, as Sn	0.3	
7440-31-5	Tin, oxide & inorganic except SnH4	6.7	
108-88-3	Toluene	1248.8	
584-84-9	Toluene-2,4-diisocyanate, (TDI)	0.1	
108-44-1	n-Toluidine	30.0	
106-49-0	p-Toluidine	30.0	
126-73-8	Tributyl phosphate	8.3	
76-13-1	1,1,2-Trichloro-1,2,2-trifluorethane	25308.0	
76-03-9	Trichloroacetic acid	23.3	
120-82-1	1,2,4-Trichlorobenzene	133.2	
79-00-5	1,1,2-Trichloroethane	149.9	
71-55-6	1,1,1-Trichloroethane	6327.0	
75-69-4	Trichlorofluoromethane	18648.0	
1321-65-9	Trichloronaphthalene	16.7	
96-18-4	1,2,3-Trichloropropane Triethylamine	199.8	
121-44-8 75-63-8	Trifluorobromomethane	133.2 20313.0	
552-30-7	Trimellitic anhydride	0.1	
2551-13-7		416.3	
121-45-9	Trimethyl phosphite	33.3	
75-50-3	Trimethylamine	79.9	
118-96-7	2,4,6-Trinitrotoluene	1.7	
78-30-8	Triorthocresyl phosphate	0.3	
603-34-9	Triphenyl amine	16.7	
115-86-6	Triphenyl phosphate	10.0	
7440-33-7	• • •	16.7	
7440-33-7		3.3	
8006-64-2	Turpentine	1864.8	

		ASIL MICROGRAMS/M3 TWENTY-POUR-HOUR	
CAS #	SUBSTANCE	AVERAGE	
8032-32-4	VM & P Naphtha	4495.5	
	n-Valeraldehyde	582.8	
	Vanadium, as V205	0.2	
	Vinyl acetate	99.9	
593-60-2	Vinyl bromide	66.6	
106-87-6	Vinyl cycloherene dioxide	199.8	
75-35-4	Vinylidene chloride	66.6	
25013-15-4	Vinyl toluene	799.2	
81-81-2		0.3	
	Welding fumes	16.7	
1477-55-0	m-Xylene a,a'-diamine	0.3	
1330-20-7	Xylenes (m-,o-,p-isomers)	1448.6	
1300-73-8	Xylidine	33.3	
7440-65-5	Yttrium, metal and cpds as Y	3.3	
7646-85-7	Zinc chloride fume	3.3	
13530-65-9	Zinc chromates	0.03	
1314-13-2	Zinc oxide, fume	16.7	
7440-67-2		16.7	



December 3, 1992

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300		-9/3	

Camas, Washington 98607

Dear Mr. Peterson:

Enclosed is the final addendum to our report on SMI.

It has been a pleasure working with you on this project. Please do not hesitate to call me if there are any questions.

Sincerely,

⊭atrick H. Wicks, PE, CHMM

President

Enclosures

CC:

Mr. (b) (6)

Mr. Jim Trunzo